## **Electronic Supplementary Information:**

## Sequence-Controlled Degradable Polymers by Controlled Cationic Copolymerization of Vinyl Ethers and Aldehydes: Precise Placement of Cleavable Units at Predetermined Positions

Marie Kawamura, Arihiro Kanazawa, Shokyoku Kanaoka, and Sadahito Aoshima\* Department of Macromolecular Science, Graduate School of Science, Osaka University, Toyonaka, Osaka 560-0043, Japan

## **Contents:**

**Figure S1.** Copolymer composition curves and Fineman-Ross plot for the polymerization of CEVE and *p*MeBzA

Figure S2. MWD curves of products obtained by the addition of a small amount of myrtenal or pMeBzA during the living cationic polymerization of IBVE or CEVE

**Figure S3.** Time-conversion curves, and  $M_n$  and  $M_w/M_n$  values of products obtained by the multiple addition of small amounts of *p*MeBzA during the living cationic polymerization of CEVE

**Figure S4.** Time-conversion curves,  $M_n$  and  $M_w/M_n$  values, and MWD curves of a product obtained by the addition of a large amount of *p*MeBzA during the living cationic polymerization of CEVE and its acid-hydrolysis product

Figure S5. GPC fractionation analyses and <sup>1</sup>H NMR spectra of acid-hydrolysis product of poly[IBVE-*b*-(IBVE-*alt*-myrtenal)]



**Figure S1.** (A) Copolymer composition curves of CEVE with *p*MeBzA and (B) the Fineman-Ross plot for the copolymerization of CEVE with *p*MeBzA. The broken curve shown in Figure S1A was drawn using the  $r_{\text{CEVE}}$  (0.03) and  $r_{\text{pMeBzA}}$  (0) values. The dashed-dotted line is an azeotropic line. Polymerization conditions: [monomer]<sub>total, 0</sub> = 1.2 M, [EtSO<sub>3</sub>H]<sub>0</sub> = 8.0 mM, [GaCl<sub>3</sub>]<sub>0</sub> = 4.0 mM, [1,4-dioxane] = 0.50 M, in toluene at -78 °C.



**Figure S2.** MWD curves of products obtained by the addition of a small amount of aldehyde during the living cationic polymerization of VE (middle) and its acid-hydrolysis product (lower). (A) A combination of IBVE and *p*MeBzA, (B) a combination of IBVE and myrtenal, and (C) a combination of CEVE and myrtenal. Polymerization:  $[VE]_0 = 0.40$  (for B) or 1.2 (for A and C) M,  $[aldehyde]_{added} = 40$  mM (five equivalent toward propagating chain),  $[EtSO_3H]_0 = 8.0$  mM,  $[GaCl_3]_0 = 4.0$  mM, [1,4-dioxane] = 0.50 (for A and C) or 1.0 (for B) M, in toluene at -78 °C; hydrolysis conditions: 0.50 M aqueous HCl-1,2-dimethoxyethane at room temperature for 2 h, 0.33 wt%.



**Figure S3.** (A) Time-conversion curves and (B)  $M_n$  and  $M_w/M_n$  values of products obtained by the multiple addition of small amounts of *p*MeBzA during the living cationic polymerization of CEVE. Polymerization conditions: [CEVE]<sub>0</sub> = 1.2 M, [*p*MeBzA]<sub>added</sub> = 40 mM (five equivalent toward propagating chain), [EtSO<sub>3</sub>H]<sub>0</sub> = 8.0 mM, [GaCl<sub>3</sub>]<sub>0</sub> = 4.0 mM, [1,4-dioxane] = 0.50 M, in toluene at -78 °C.



**Figure S4.** (A) Time-conversion curves, (B)  $M_n$  and  $M_w/M_n$  values, and (C) MWD curves of a product obtained by the addition of a large amount of *p*MeBzA during the living cationic polymerization of CEVE and its acid-hydrolysis product. Polymerization conditions: [CEVE]<sub>0</sub> = 1.2 M, [*p*MeBzA]<sub>added</sub> = 0.78 M, [EtSO<sub>3</sub>H]<sub>0</sub> = 8.0 mM, [GaCl<sub>3</sub>]<sub>0</sub> = 4.0 mM, [1,4-dioxane] = 1.0 M, in toluene at -78 °C; hydrolysis conditions: 0.50 M aqueous HCl-1,2-dimethoxyethane at room temperature for 2 h, 0.33 wt%.



**Figure S5.** GPC fractionation analyses of poly[IBVE-*b*-(IBVE-*alt*-myrtenal)]: MWD curves before and after fractionation and <sup>1</sup>H NMR spectra of each fraction. See also Figure 4.