

Supporting Information

Fe(III)-Mediated ICAR ATRP in *p*-Xylene/PEG-200 Biphasic System: Facile and Highly Efficient Separation and Recycling of Iron Catalyst

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Table S1. Recycling experiments for Fe(III)-mediated TPSC-based ICAR ATRP of MMA.

Entry	Recycling times ^a	Monomer conversion (%)	$M_{n,th}^b$ (g/mol)	$M_{n,GPC}$ (g/mol)	M_w/M_n
1	1	23.5	5000	9000	1.27
2	2	73.8	15000	16500	1.21
3	3	82	16600	16000	1.28
4	4	80.2	16300	17300	1.19
5	5	84.1	17000	16800	1.28
6	6	81.6	16600	16800	1.30
7	7	77.8	15900	17700	1.34
8	8	83.9	17000	16500	1.45
9	9	83.6	17000	18000	1.39
10	10	80.2	16300	20100	1.45

15 Polymerization conditions: [monomer]₀: [EBPA]₀: [FeCl₃·6H₂O]₀: [TBABr]₀: [AIBN]₀ = 200:1:1:3:1, $V_{monomer} = 1.0$ mL, $V_{p-xylene} = 2.0$ mL and $V_{PEG-200} = 1.0$ mL, temperature = 70 °C. ^aPolymerization was conducted using the iron catalyst solution in PEG-200 after every recycling experiment without adding any fresh FeCl₃·6H₂O and TBABr. ^b $M_{n,th} = ([M]_0/[EBPA]_0) \times M_{w,MMA} \times conv.\%$.

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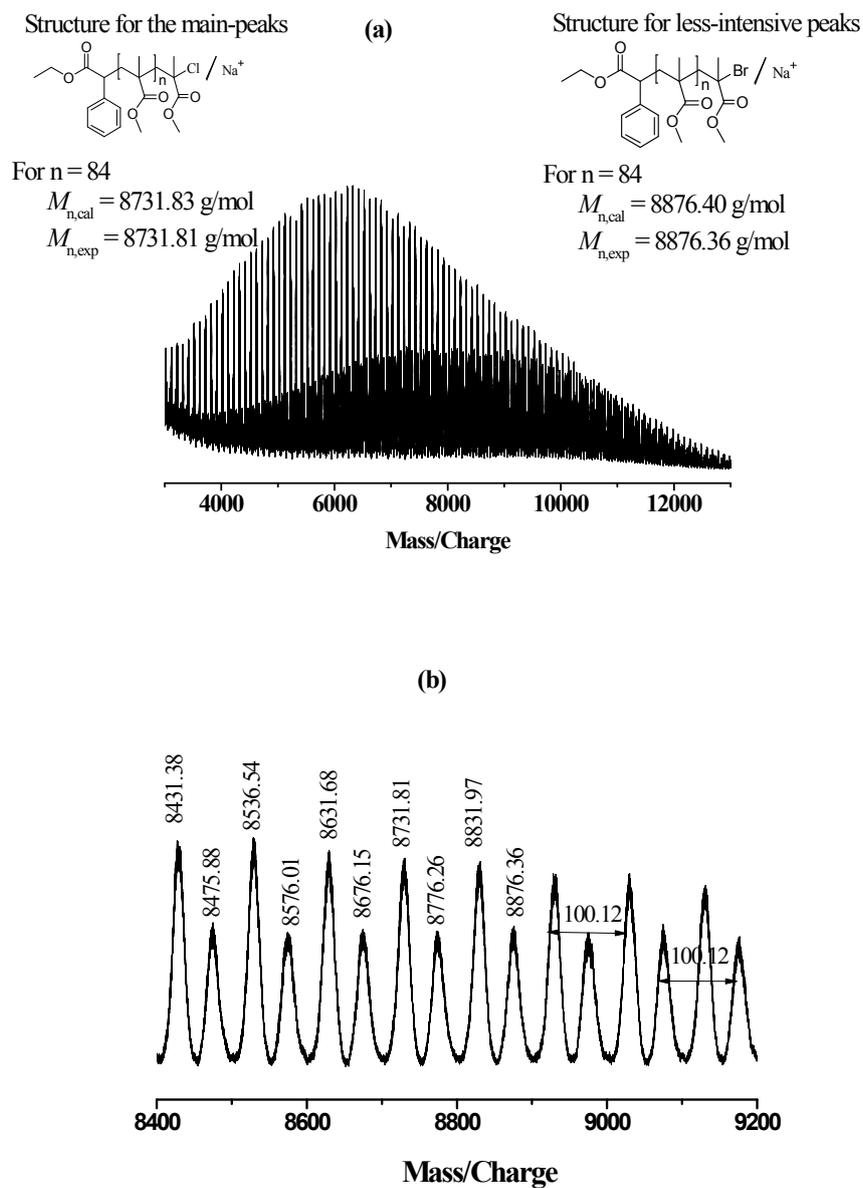


Figure S1. MALDL-TOF-MS in the linear mode of PMMA-Br and PMMA-Cl ($M_{n,GPC} = 8700 \text{ g/mol}$, $5 M_w/M_n = 1.25$) (a) and enlargement of the MALDL-TOF-MS from m/z 8400 to 9200 of PMMA-Br and PMMA-Cl (b).

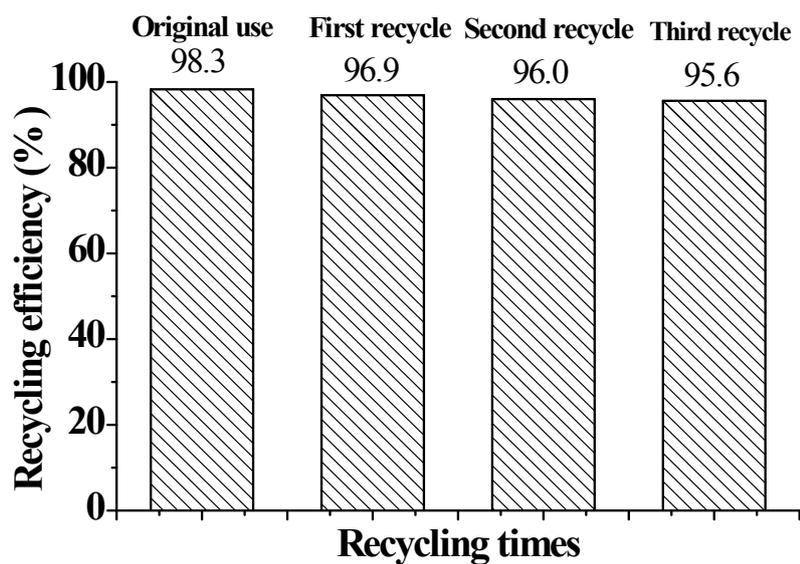


Figure S2. Recycling efficiency of iron catalyst at different recycle times. Polymerization conditions : $[MMA]_0:[EBPA]_0:[FeCl_3 \cdot 6H_2O]_0:[TBABr]_0:[AIBN]_0 = 200:1:1:3:1$, $V_{MMA} = 1.0$ mL, $V_{p\text{-xylene}} = 2.0$ mL and $V_{PEG-200} = 1.0$ mL, temperature = 70 °C. Recycling efficiency is the percentage of residual iron catalyst in PEG-200 phase to the original iron catalyst, which was determined by inductively coupled plasma (ICP) of Vista MPX.