

# Ultrahighly Electron-Deficient Pyrrolo-acenaphtho-pyridazine-dione Based Donor-Acceptor Conjugated Polymers for Electrochromic Applications

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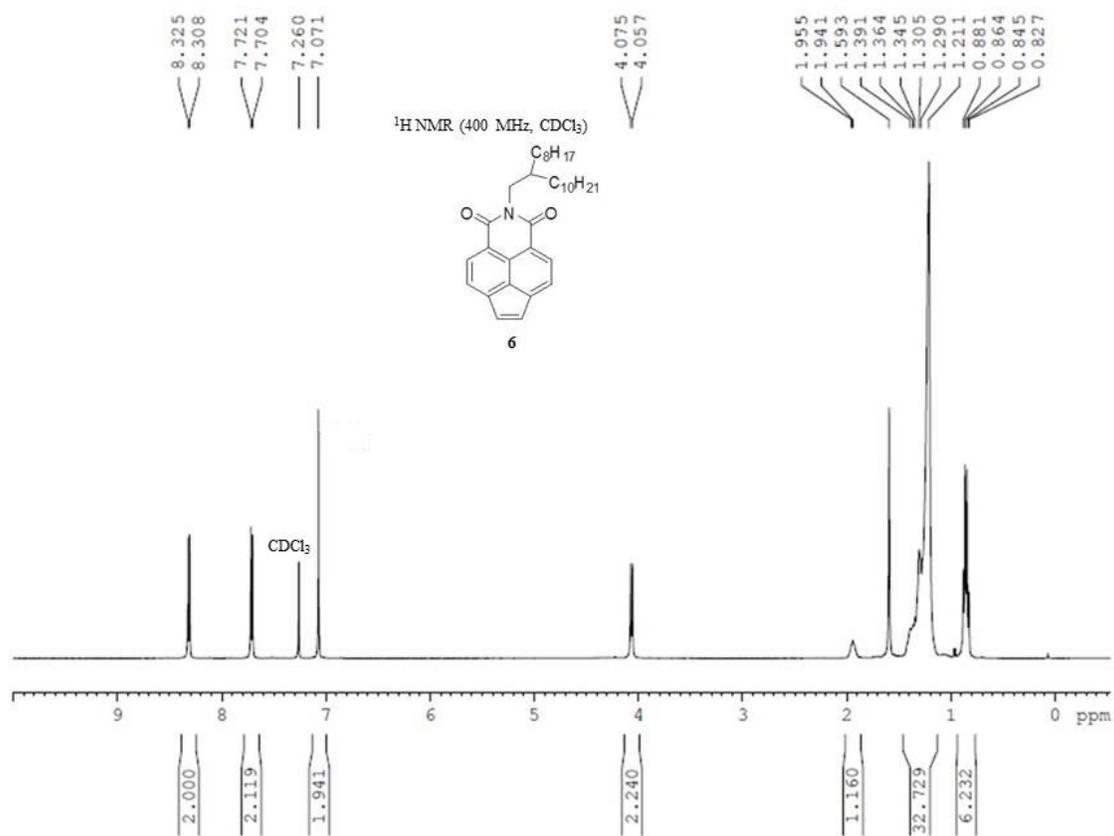
Email: [jw-xu@imre.a-star.edu.sg](mailto:jw-xu@imre.a-star.edu.sg); [ASXHLu@ntu.edu.sg](mailto:ASXHLu@ntu.edu.sg)

<sup>‡</sup>These authors contribute equally to this work.

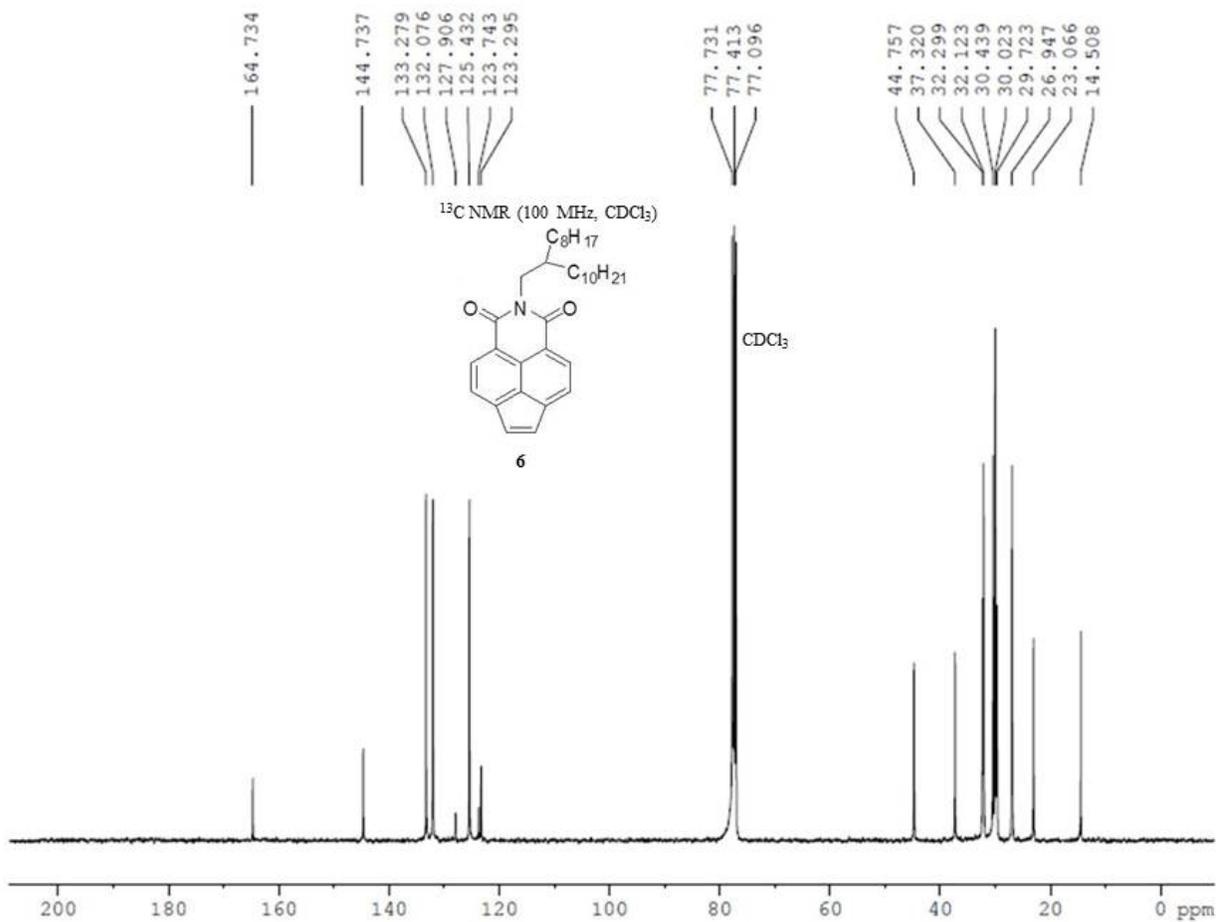
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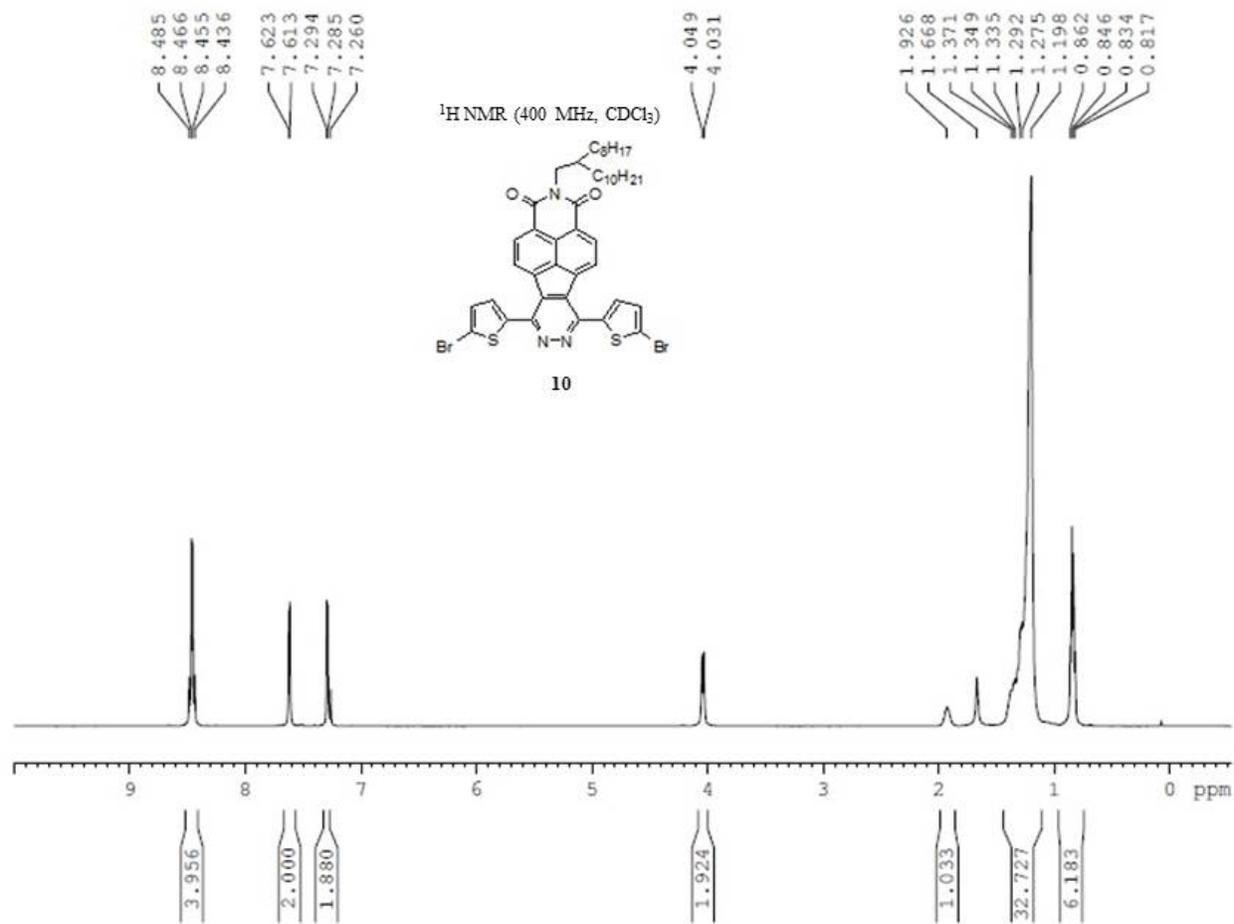
## 1. NMR Spectra of all new compounds



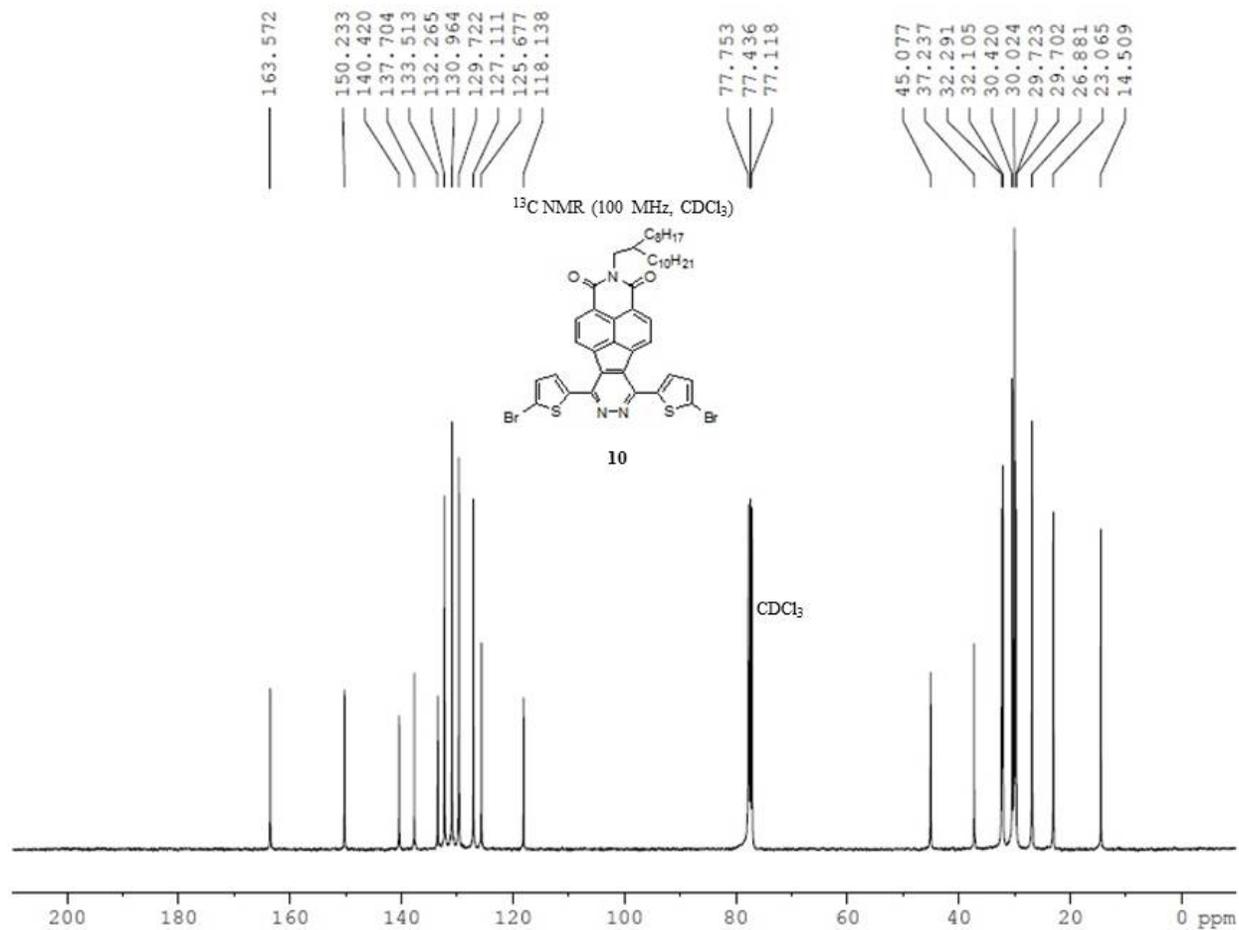
**Figure S1.** <sup>1</sup>H NMR spectrum of compound **6** (CDCl<sub>3</sub>, room temperature).



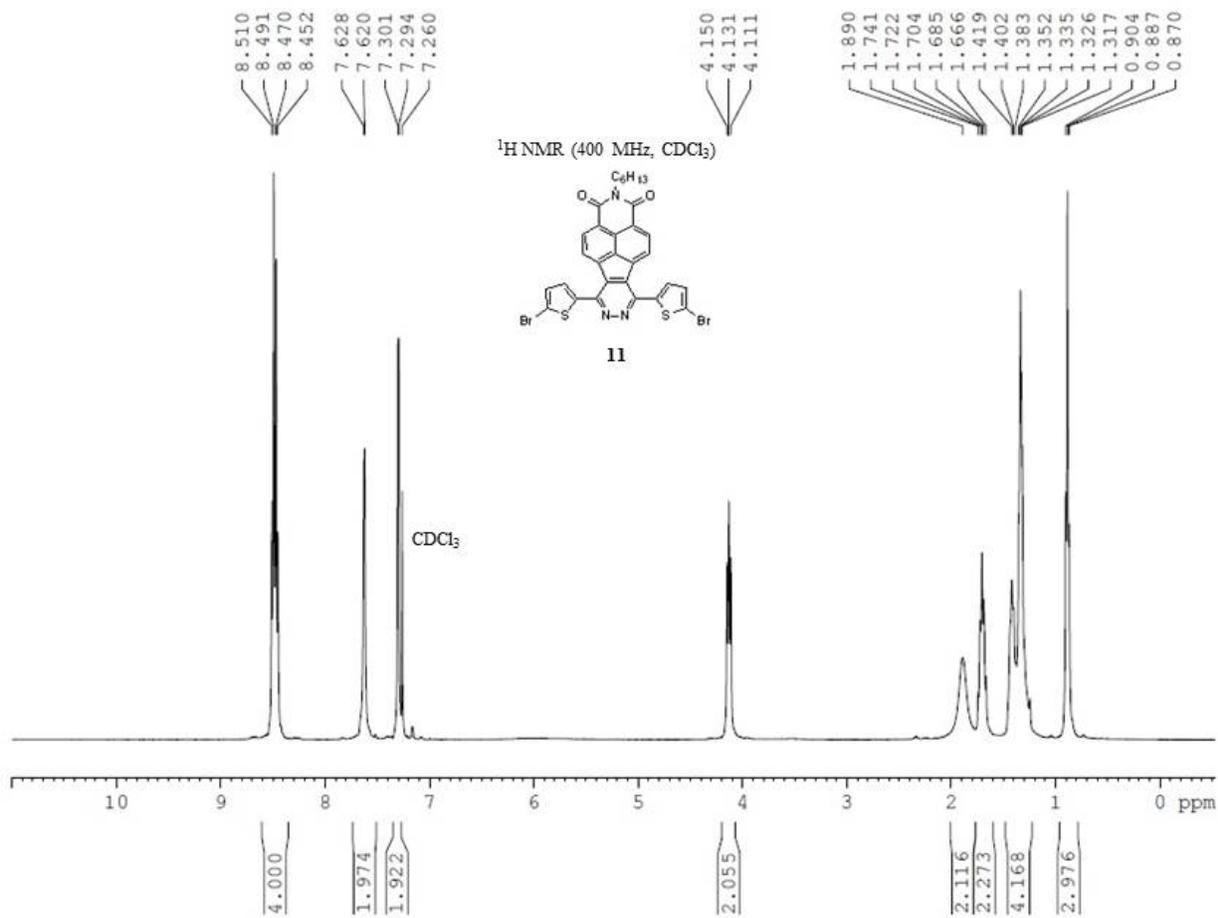
**Figure S2.** <sup>13</sup>C NMR spectrum of compound **6** (CDCl<sub>3</sub>, room temperature).



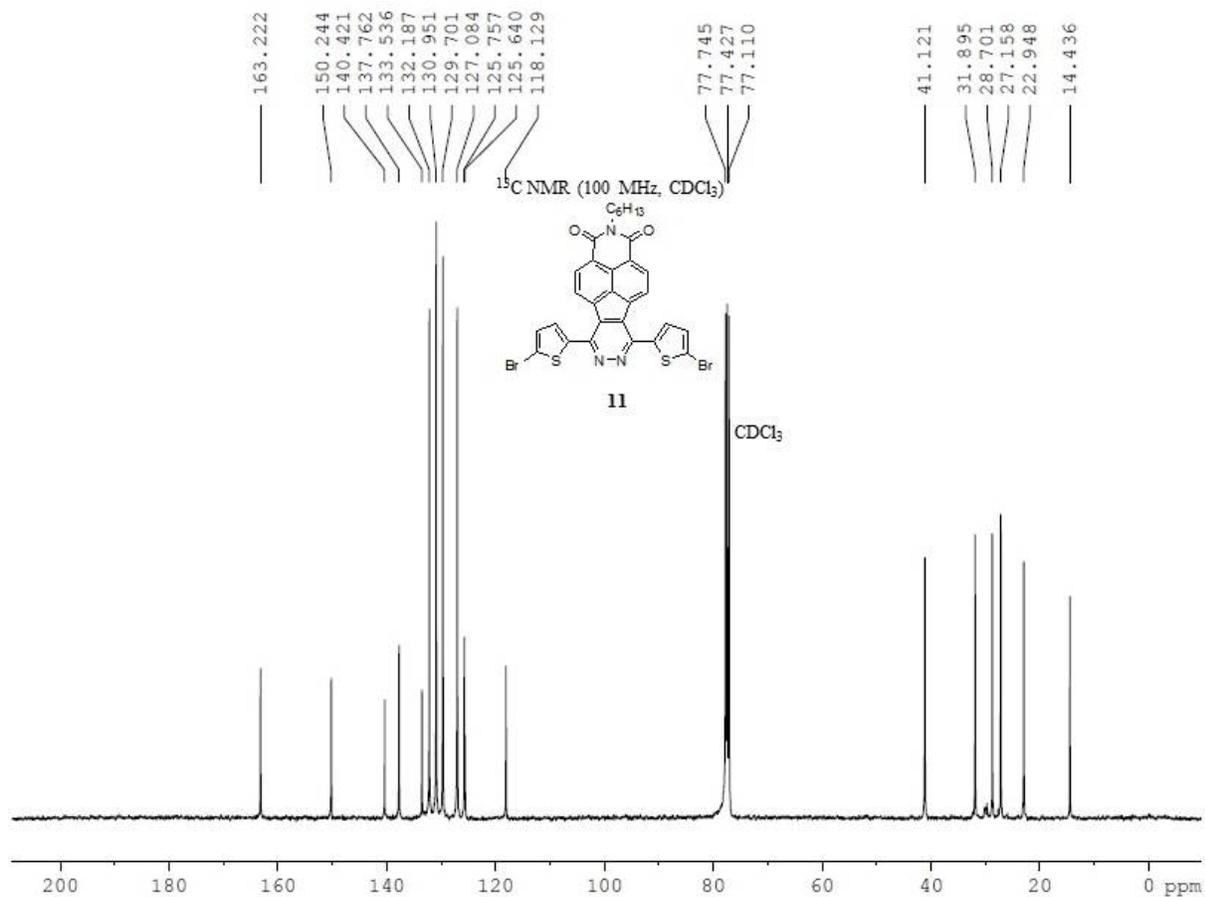
**Figure S3.** <sup>1</sup>H NMR spectrum of compound **10** (CDCl<sub>3</sub>, room temperature).



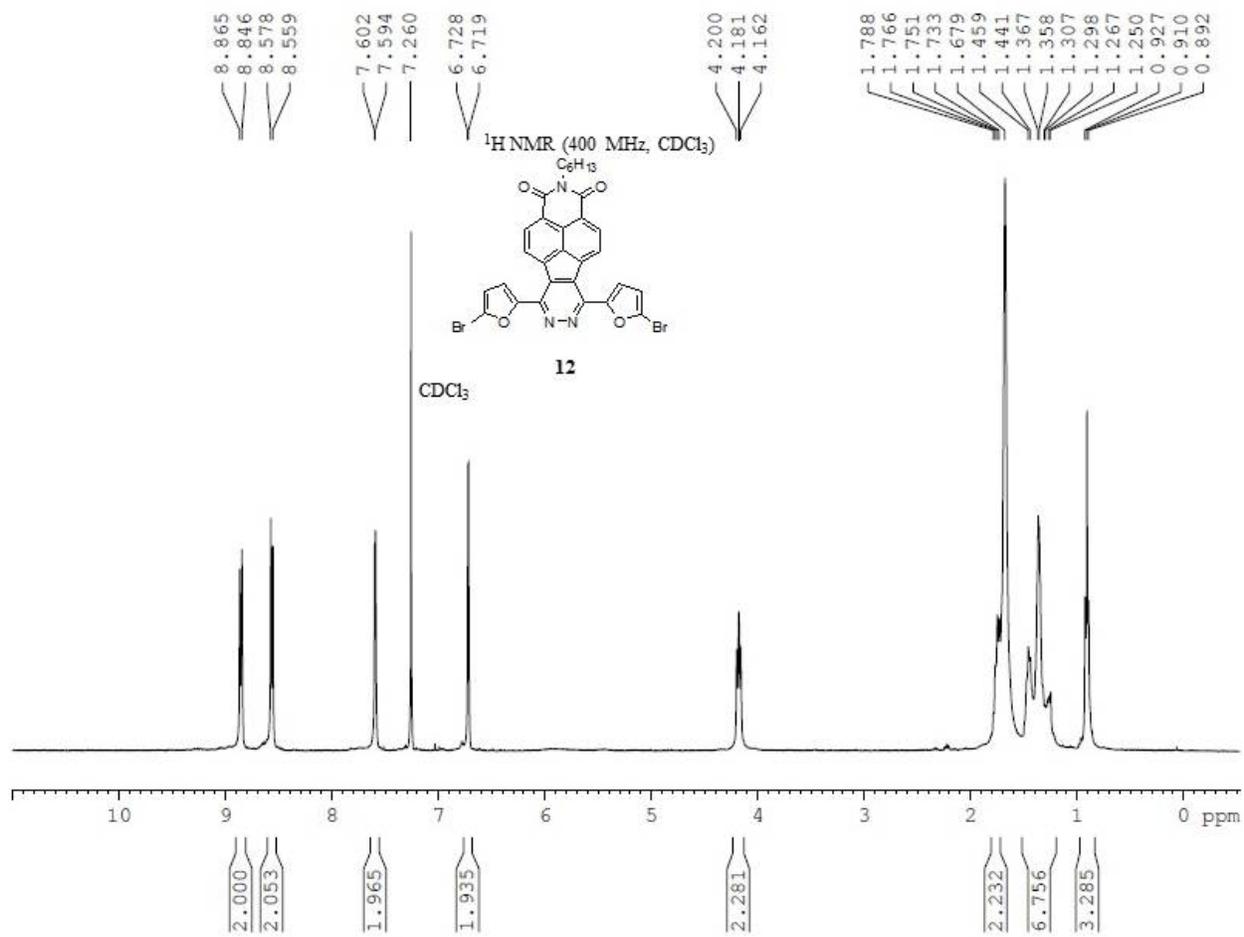
**Figure S4.** <sup>13</sup>C NMR spectrum of compound **10** (CDCl<sub>3</sub>, room temperature).



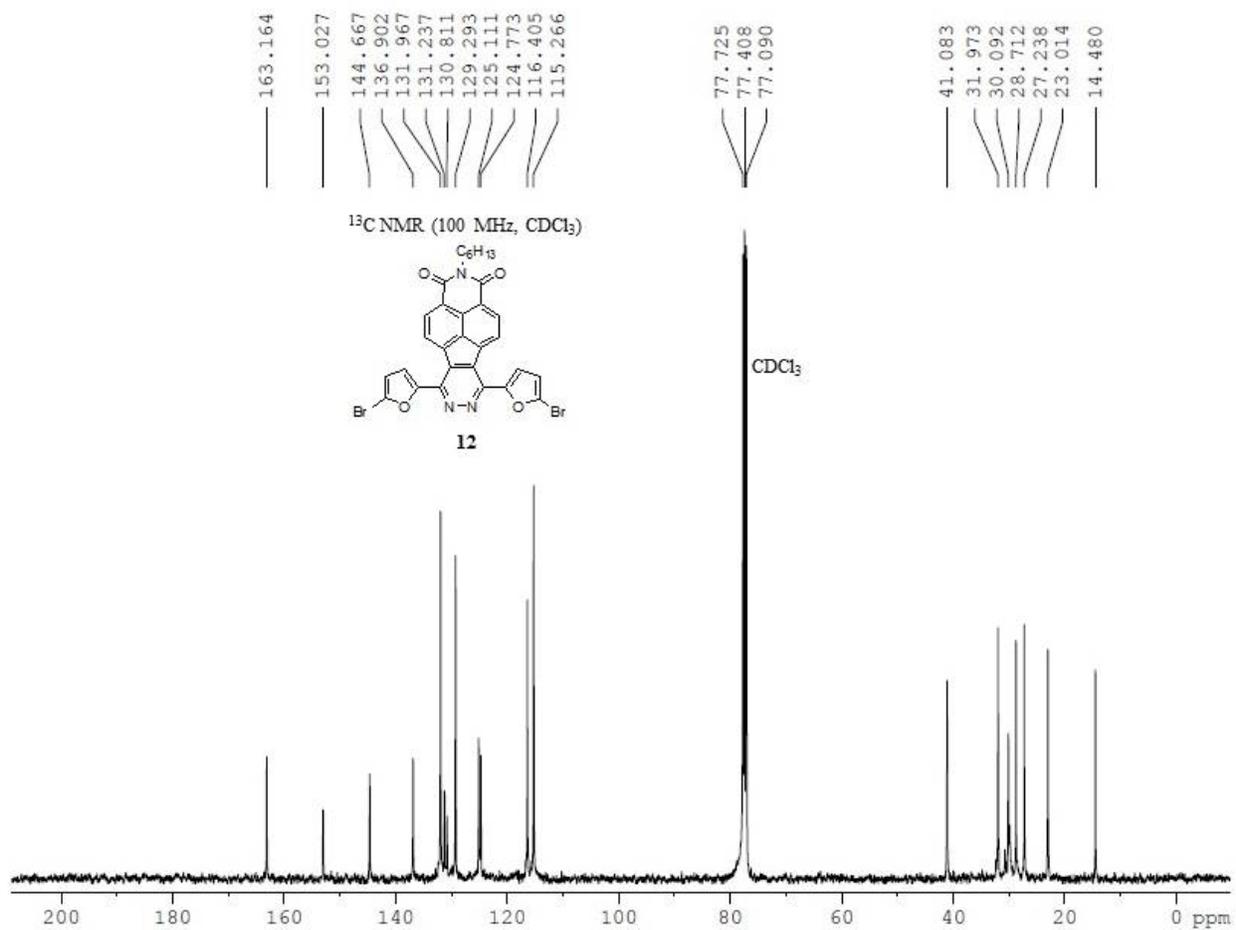
**Figure S5.** <sup>1</sup>H NMR spectrum of compound **11** (CDCl<sub>3</sub>, room temperature).



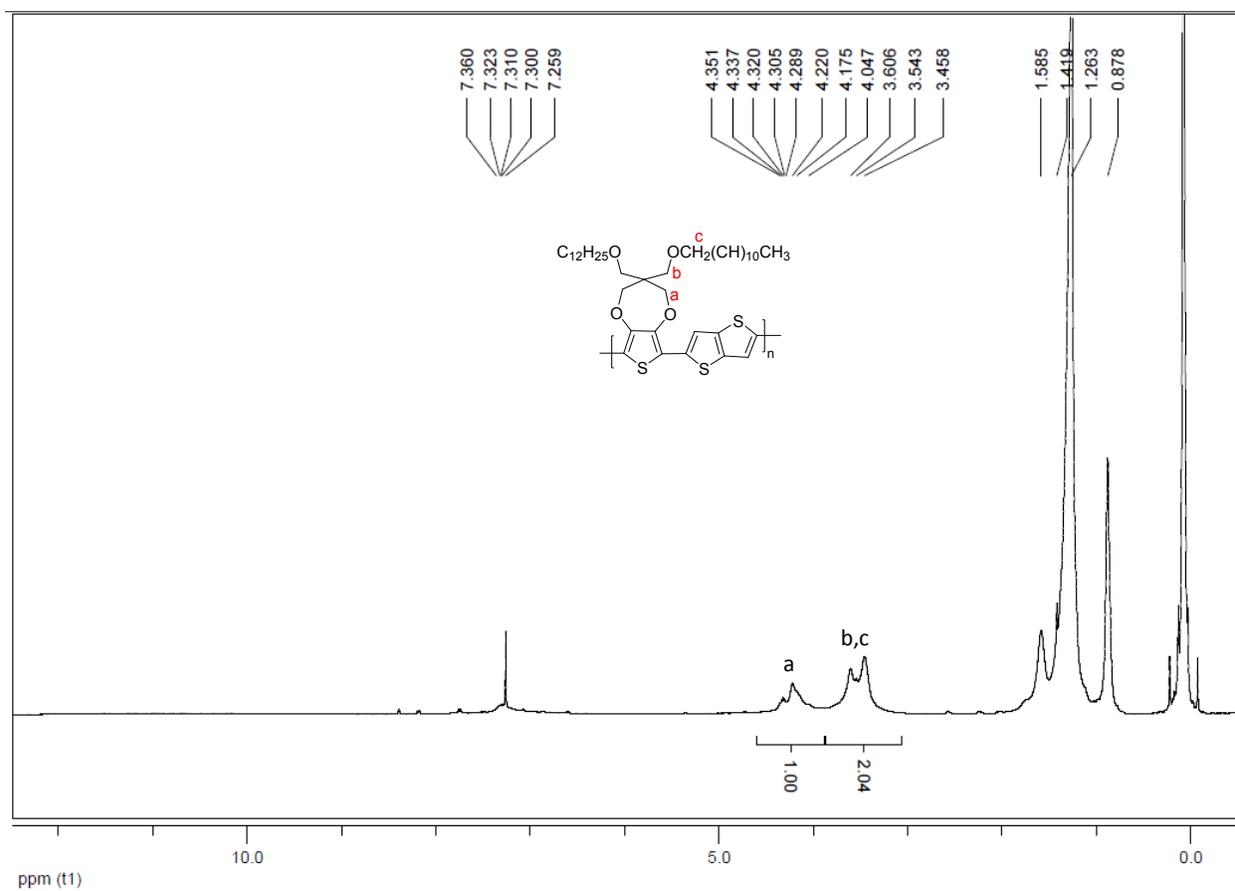
**Figure S6.** <sup>13</sup>C NMR spectrum of compound **11** (CDCl<sub>3</sub>, room temperature).



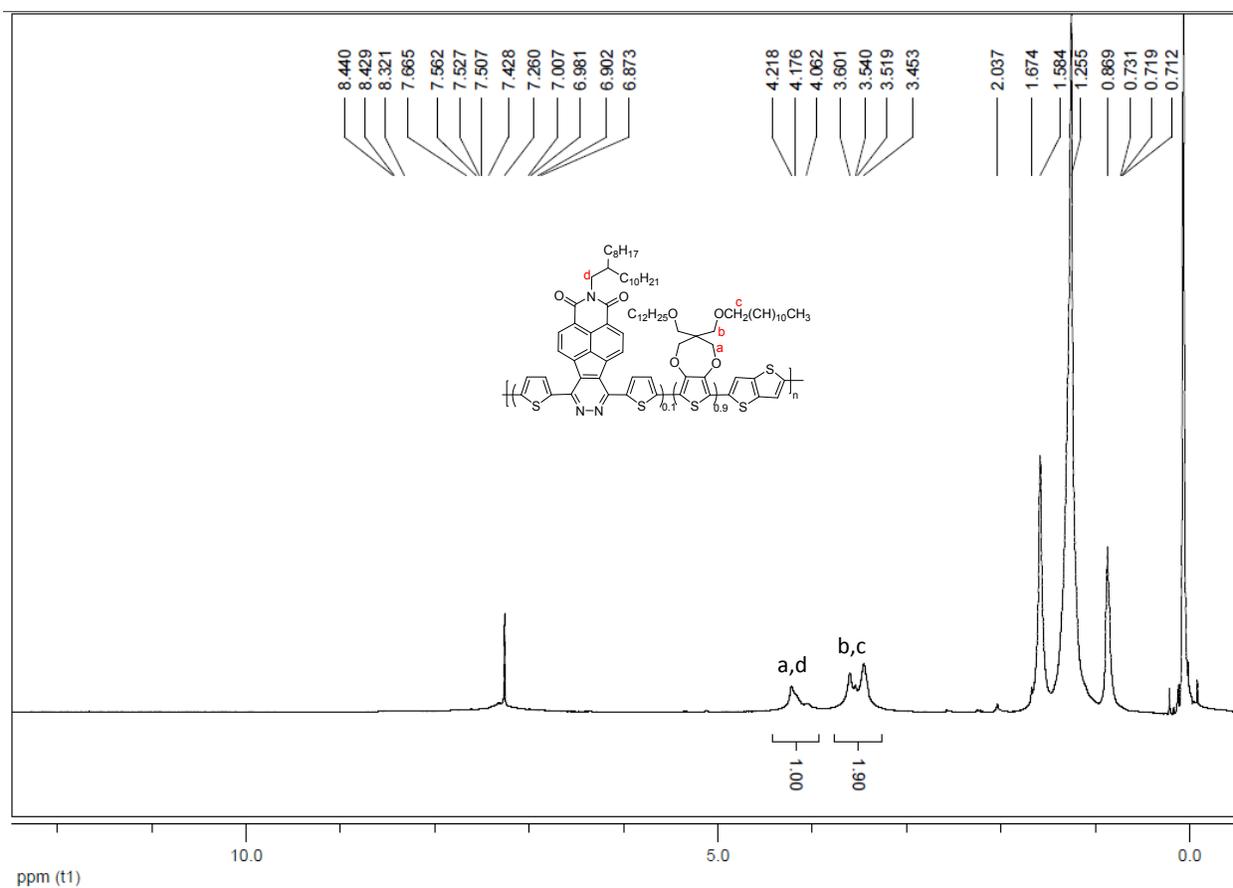
**Figure S7.** <sup>1</sup>H NMR spectrum of compound **12** (CDCl<sub>3</sub>, room temperature).



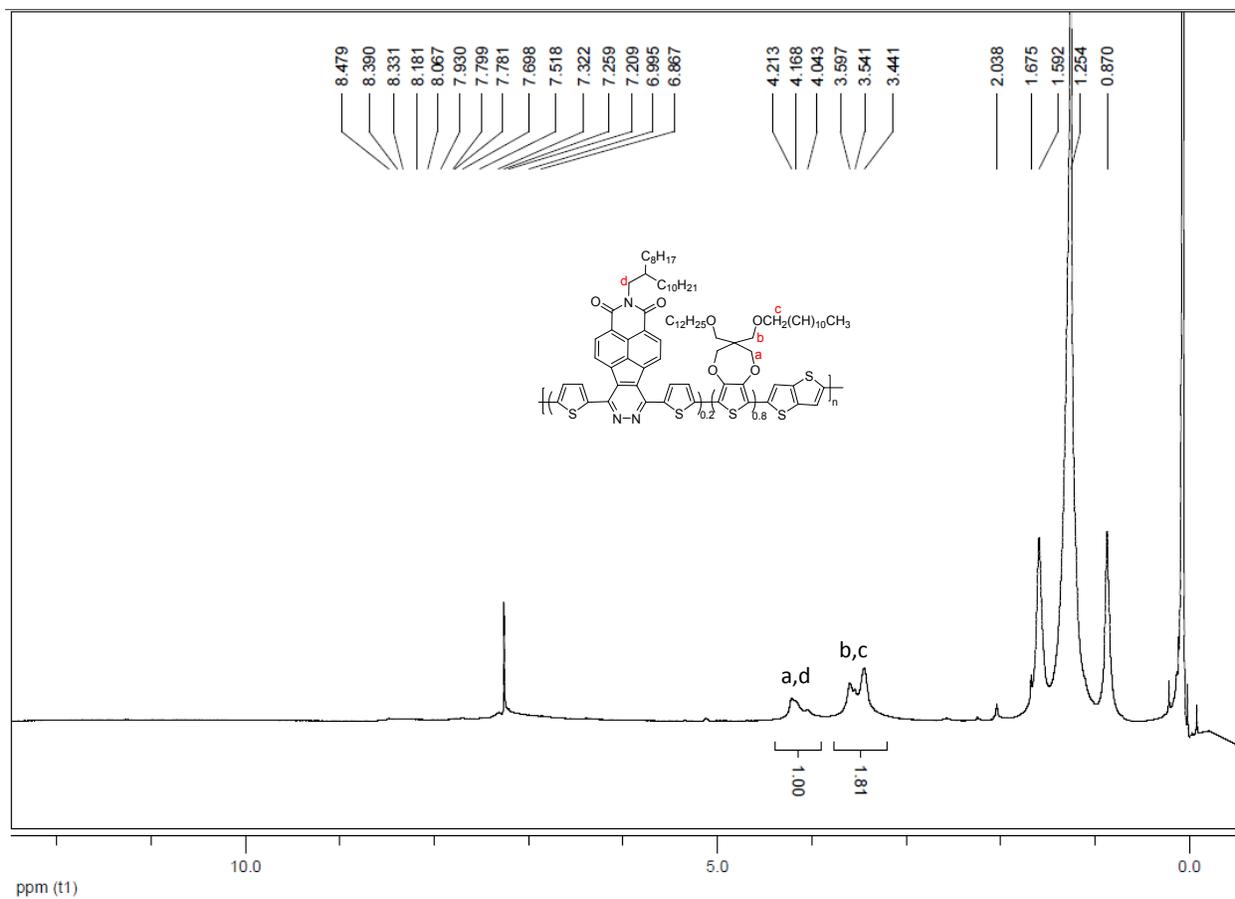
**Figure S8.**  $^{13}\text{C}$  NMR spectrum of compound **12** ( $\text{CDCl}_3$ , room temperature).



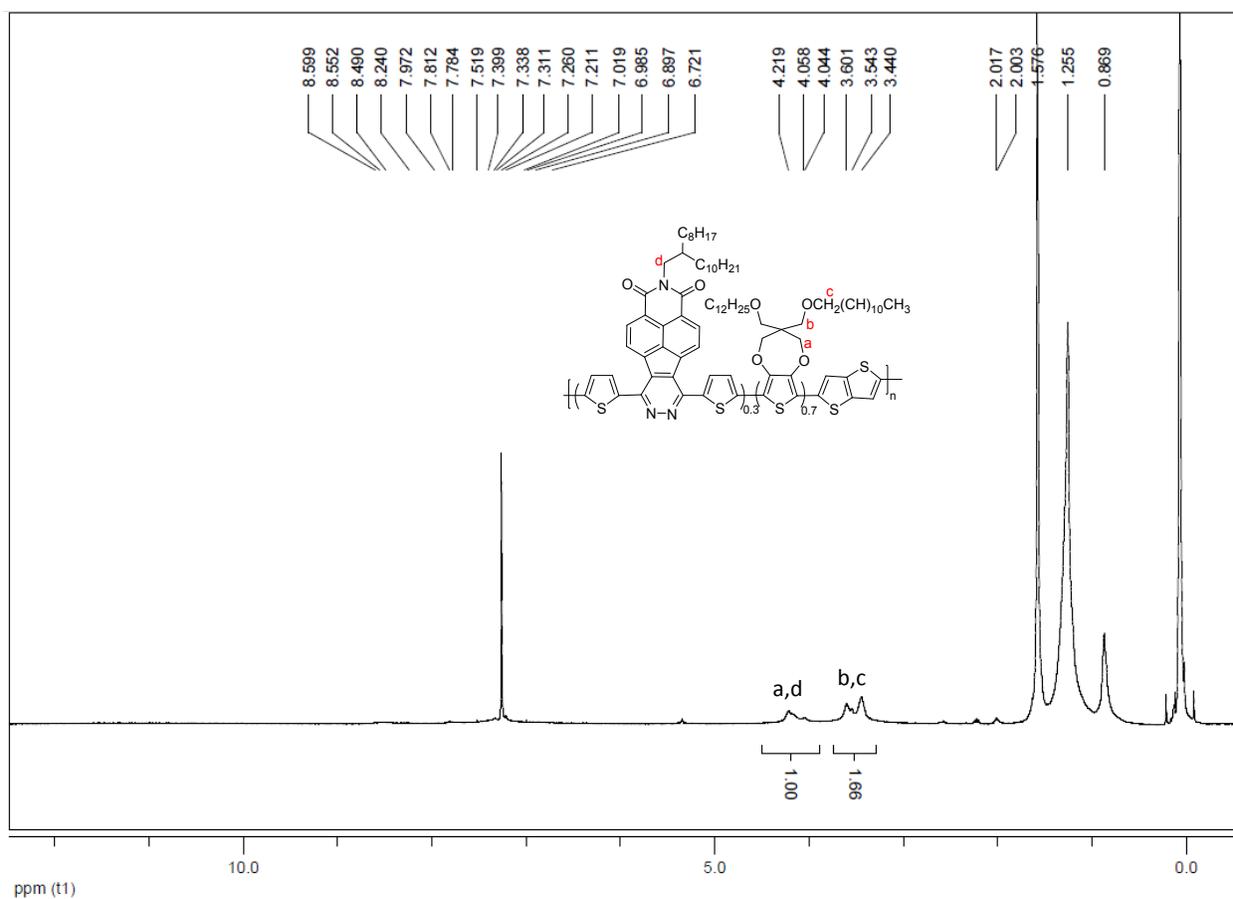
**Figure S9.**  $^1H$  NMR spectrum of Polymer **P1** ( $CDCl_3$ , room temperature).



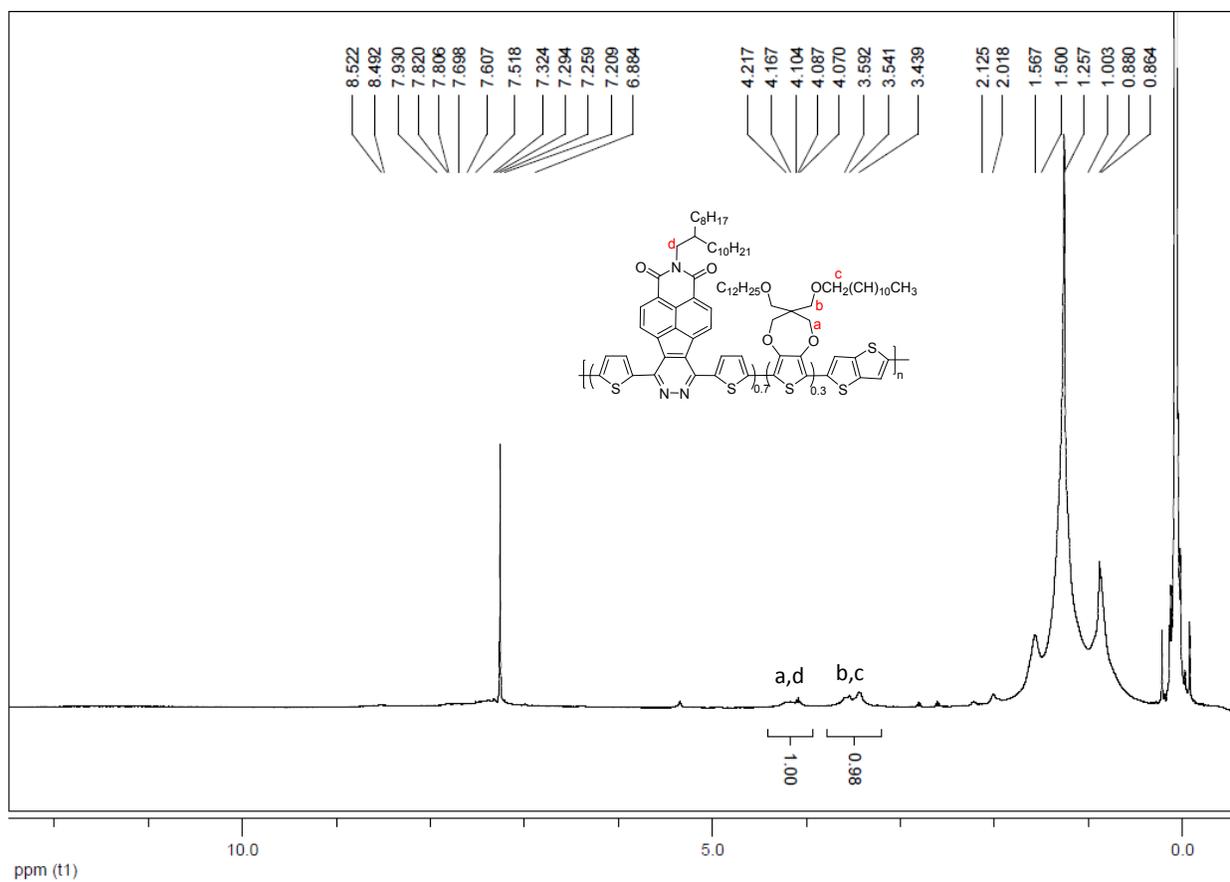
**Figure S10.** <sup>1</sup>H NMR spectrum of Polymer **P2** (CDCl<sub>3</sub>, room temperature).



**Figure S11.**  $^1\text{H}$  NMR spectrum of Polymer P3 ( $\text{CDCl}_3$ , room temperature).



**Figure S12.**  $^1\text{H}$  NMR spectrum of Polymer P4 ( $\text{CDCl}_3$ , room temperature).



**Figure S13.**  $^1\text{H}$  NMR spectrum of Polymer **P5** ( $\text{CDCl}_3$ , room temperature).

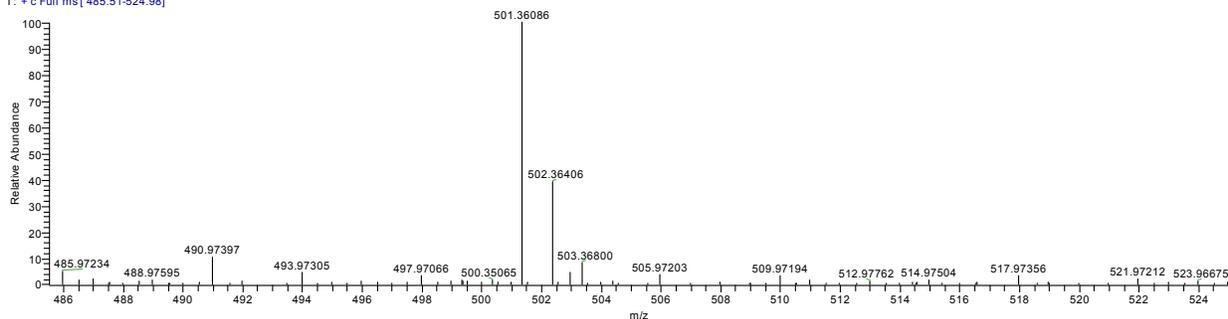
## 2. High resolution MS data of all new compounds

D:\MAT95\converted\NapC8C10-1-c1-av  
study.]

03/02/2015 03:01:00 PM

NapC8C10

NapC8C10-1-c1-av#1 RT: 4.09 AV: 1 NL: 2.48E6  
T: + c Full ms [ 485.51-524.98]



NapC8C10-1-c1-av#1 RT: 4.09

T: + c Full ms [ 485.51-524.98]

m/z	Intensity	Relative	Theo. Mass	Delta (ppm)	RDB equiv.	Composition
501.36086	2483799.0	100.00	501.36013	1.44	12.0	C <sub>34</sub> H <sub>47</sub> O <sub>2</sub> N <sub>1</sub>

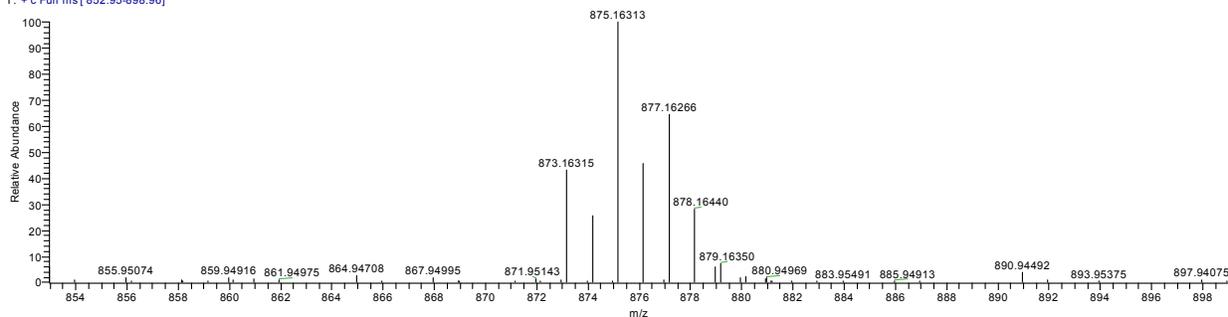
Figure S14. HR-EI-MS spectrum of compound 6.

D:\MAT95\...\20150113\Naph\_C8C10-c1-av  
study.]

13/07/2015 04:45:55 PM

Naph C8C10

Naph\_C8C10-c1-av#1 RT: 6.72 AV: 1 NL: 2.55E6  
T: + c Full ms [ 852.95-898.96]



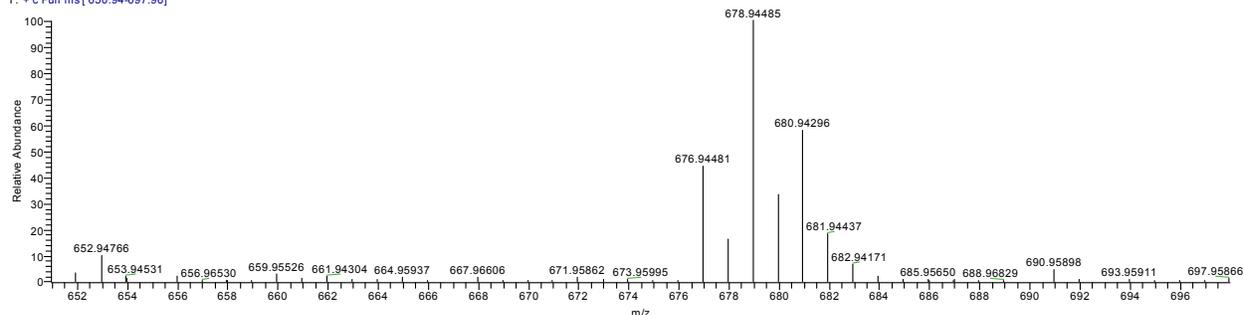
Naph\_C8C10-c1-av#1 RT: 6.72

T: + c Full ms [ 852.95-898.96]

m/z = 872.96694-873.39376

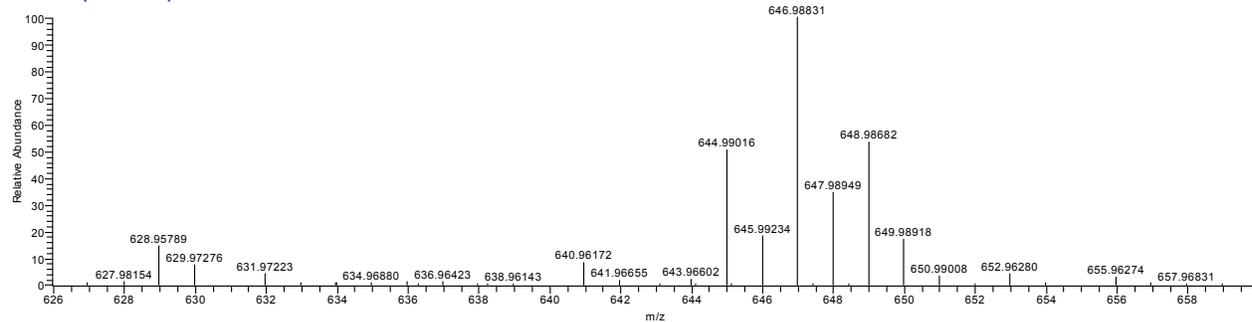
m/z	Intensity	Relative	Theo. Mass	Delta (ppm)	RDB equiv.	Composition
873.16315	1091169.0	100.00	873.16275	0.46	21.0	C <sub>44</sub> H <sub>49</sub> O <sub>2</sub> N <sub>3</sub> <sup>79</sup> Br <sub>2</sub> <sup>32</sup> S <sub>2</sub>

Figure S15. HR-EI-MS spectrum of compound 10.

Naph\_C6-c1-av #1 RT: 0.95 AV: 1 NL: 6.51E6  
T: + c Full ms [ 650.94-697.96]Naph\_C6-c1-av#1 RT: 0.95  
T: + c Full ms [ 650.94-697.96]  
m/z = 676.67285-677.23989

m/z	Intensity	Relative	Theo. Mass	Delta (ppm)	RDB equiv.	Composition
676.94481	2884069.0	100.00	676.94364	1.72	21.0	C <sub>30</sub> H <sub>21</sub> O <sub>2</sub> N <sub>3</sub> <sup>79</sup> Br <sub>2</sub> <sup>32</sup> S <sub>2</sub>

Figure S16. HR-EI-MS spectrum of compound 11.

Furan\_C6-c1-av #1 RT: 9.03 AV: 1 NL: 6.56E6  
T: + c Full ms [ 625.95-659.96]Furan\_C6-c1-av#1 RT: 9.03  
T: + c Full ms [ 625.95-659.96]  
m/z = 644.84867-645.06951

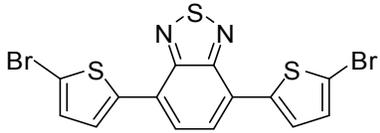
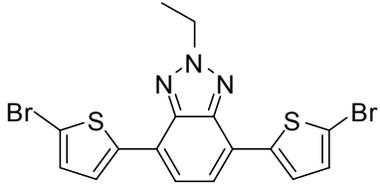
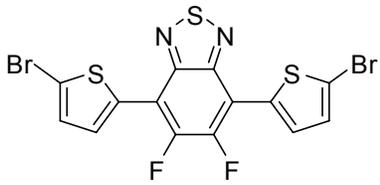
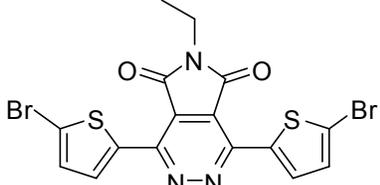
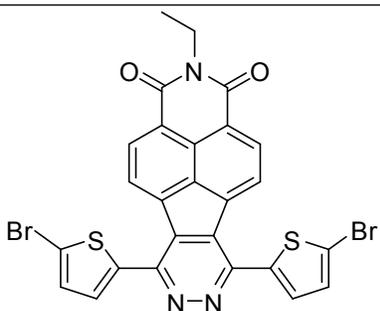
m/z	Intensity	Relative	Theo. Mass	Delta (ppm)	RDB equiv.	Composition
644.99016	3303838.0	100.00	644.98933	1.28	21.0	C <sub>30</sub> H <sub>21</sub> O <sub>4</sub> N <sub>3</sub> <sup>79</sup> Br <sub>2</sub>

Figure S17. HR-EI-MS spectrum of compound 12.

### 3. TD-DFT Calculation Details

The model compounds were built using GaussView and then optimized using DFT at the level of theory: B3LYP/6-31G with Gaussian 09.<sup>1</sup> The HOMO, LUMO and other properties were obtained at the DFT optimized geometries and summarized in Table S1.

**Table S1.** Summary of the frontier orbital energy levels of the model compounds listed in Figure 1.

Molecule Model	HOMO (eV)	LUMO (eV)	Dipole Moment (D)	Stoichiometry
	-5.434	-2.871	3.1025	C <sub>14</sub> H <sub>6</sub> Br <sub>2</sub> N <sub>2</sub> S <sub>3</sub>
	-5.237	-2.185	0.6424	C <sub>16</sub> H <sub>11</sub> Br <sub>2</sub> N <sub>3</sub> S <sub>2</sub>
	-5.622	-2.982	1.1988	C <sub>14</sub> H <sub>4</sub> Br <sub>2</sub> F <sub>2</sub> N <sub>2</sub> S <sub>3</sub>
	-5.906	-3.084	0.4392	C <sub>16</sub> H <sub>9</sub> Br <sub>2</sub> N <sub>3</sub> O <sub>2</sub> S <sub>2</sub>
	-6.086	-3.418	0.7746	C <sub>26</sub> H <sub>13</sub> Br <sub>2</sub> N <sub>3</sub> O <sub>2</sub> S <sub>2</sub>

<sup>1</sup>Gaussian 09, Revision A.02, Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, M. J.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, Ö.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian, Inc., Wallingford CT, 2009.

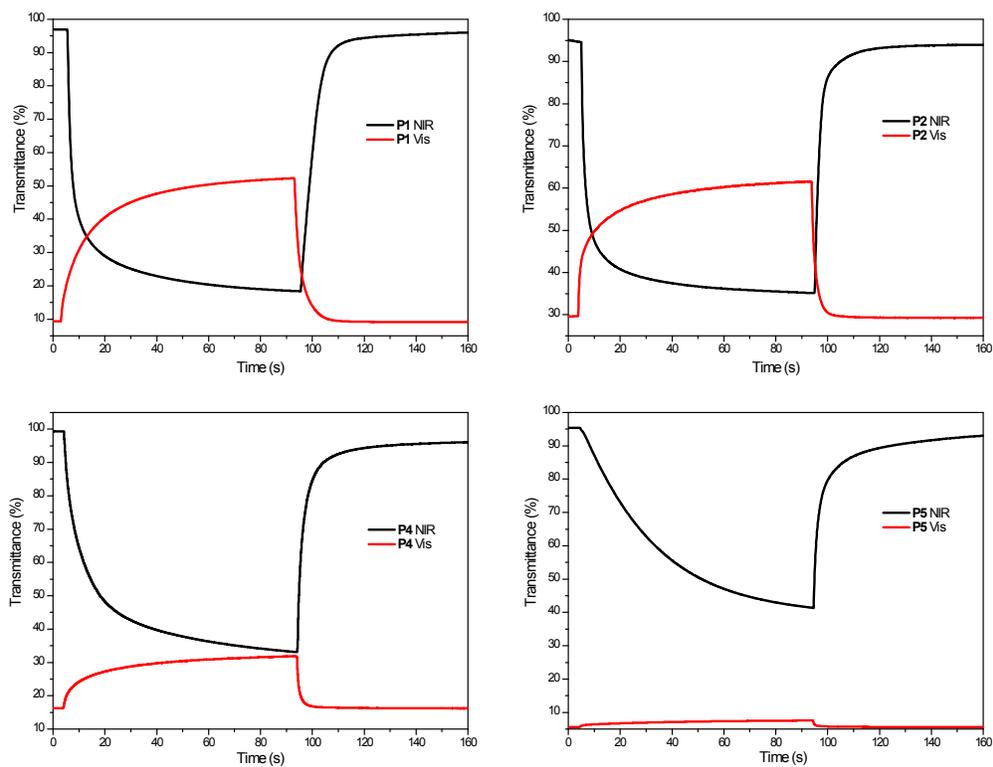
**Table S2.** Summary of the calculated free energies of molecules and reactions using (DMol<sup>3</sup>: GGA-PW91-VPSR/dnp (fine)).

Molecule	Et electronic (ha)	ZPVE (ha)	G, T=298.15K (ha)	Et (ha) T=0K	Et (ha) T=298.15K	Imaginary frequency (cm-1)
<b>Reactants</b>						
C10H4N4S2Br2 c	-6538.6029492	0.1217351	0.0729996	-6538.4812141	-6538.5299496	none
C10H4N4S2Br2 t	-6538.6029970	0.1220251	0.0731367	-6538.4809719	-6538.5298603	none
C10H4N4O2Br2 c	-5890.9393366	0.1272394	0.0823843	-5890.8120972	-5890.8569523	One, -9.7?
C10H4N4O2Br2 t	-5890.9396775	0.1279023	0.0809868	-5890.8117752	-5890.8586907	none
C16H11NO2	-822.4178467	0.2279119	0.1869038	-822.1899348	-822.2309429	none
<b>Intermediates</b>						
C26H15N5O2S2Br2	-7360.9828332	0.3503084	0.2873086	-7360.6325248	-7360.6955246	One, -3.6
C26H15N5O2S2Br2_2	-7360.9653869	0.3520948	0.2870169	-7360.6132921	-7360.6783700	none
C26H15N5O4Br2	-6713.3249086	0.3574429	0.2939618	-6712.9674657	-6713.0309468	none
C26H15N5O4Br2_2	-6713.3080570	0.3569776	0.2956479	-6712.9510794	-6713.0124091	One, -7.0

#### 4. Cyclic voltammetry of Compound 10

**Figure S18.** Cyclic voltammetry of compound **10** and its LUMO energy level.

## 5. Switching behaviour of devices



**Figure S19.** Switching behaviour of **P1**, **P2**, **P4** and **P5** device between +1.6 and -1.6 V.

## 6. Stability studies

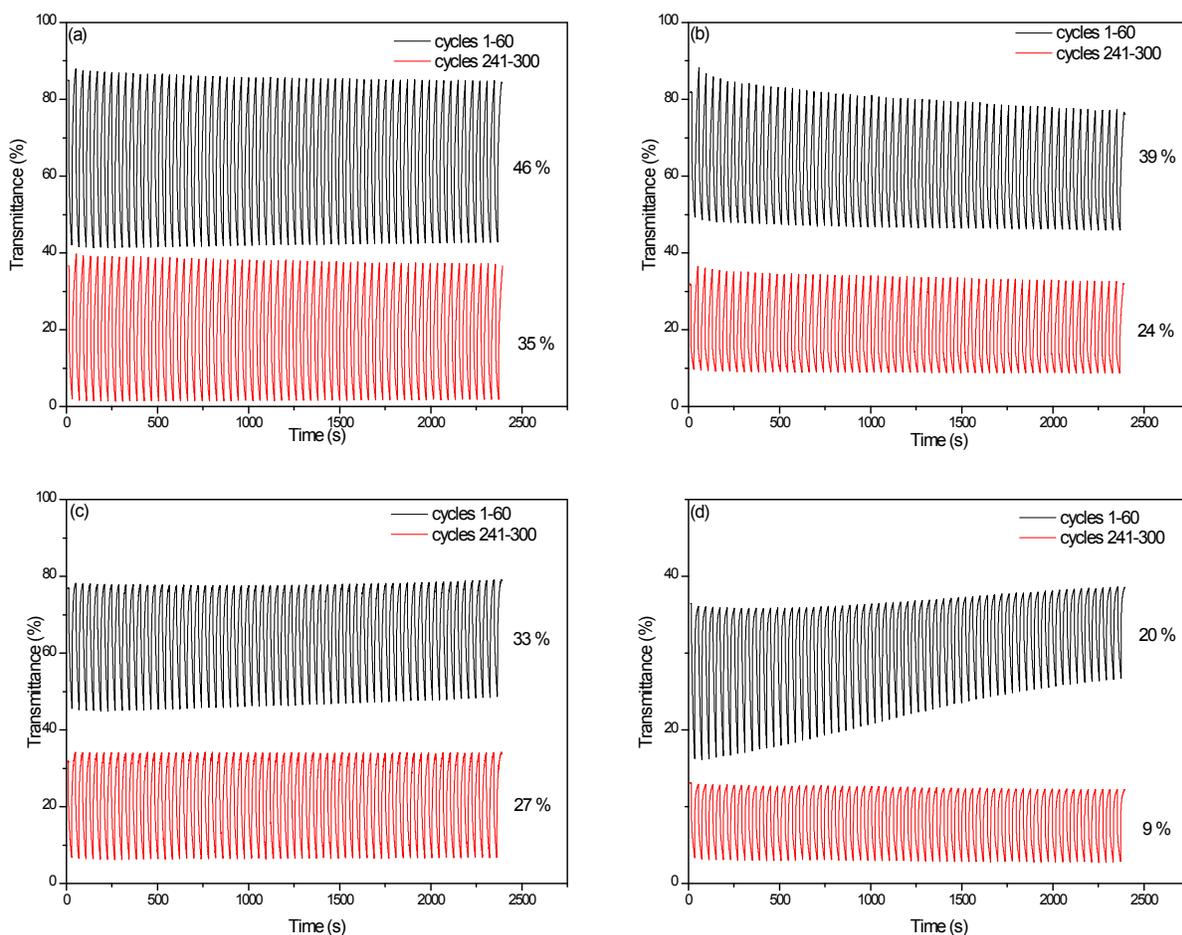


Figure S20. Stability studies of (a) P1, (b) P2, (c) P4 and (d) P5 devices in the NIR region.

## 7. X-ray crystallographic data

Table S3. Crystal data and structure refinement for **11**.

<b>Empirical formula</b>	$C_{30}H_{21}Br_2N_3O_2S_2$
<b>Formula weight</b>	679.44
<b>Temperature</b>	100(2) K
<b>Wavelength</b>	1.54178 Å
<b>Crystal system</b>	Monoclinic
<b>Space group</b>	P 21/n
<b>Unit cell dimensions</b>	$a = 8.2952(4)$ Å, $\alpha = 90^\circ$ $b = 20.7323(9)$ Å, $\beta = 100.587(2)^\circ$ $c = 22.0577(10)$ Å, $\gamma = 90^\circ$
<b>Volume</b>	$3728.9(3)$ Å <sup>3</sup>

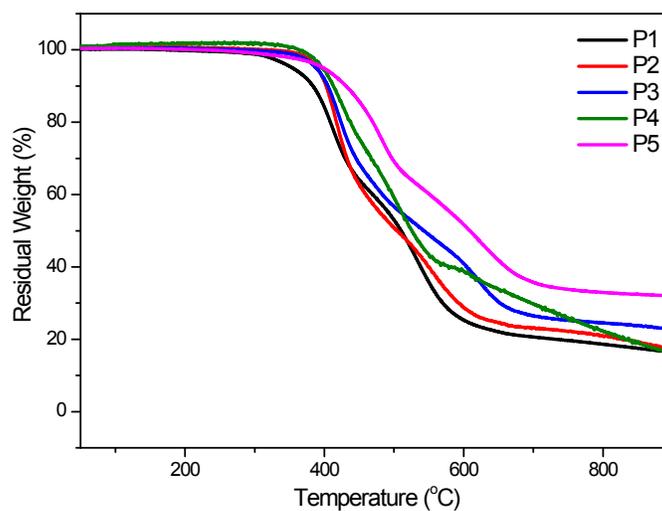
<b>Z</b>	4
<b>Density (calculated)</b>	1.210 Mg/m <sup>3</sup>
<b>Absorption coefficient</b>	4.010 mm <sup>-1</sup>
<b>F(000)</b>	1360
<b>Crystal size</b>	0.178 x 0.101 x 0.098 mm <sup>3</sup>
<b>Theta range for data collection</b>	2.949 to 54.230°
<b>Index ranges</b>	-8<=h<=8, -21<=k<=15, -23<=l<=18
<b>Reflections collected</b>	9028
<b>Independent reflections</b>	4225 [R(int) = 0.0332]
<b>Refinement method</b>	Full-matrix least-squares on F <sup>2</sup>
<b>Data / restraints / parameters</b>	4225 / 78 / 432
<b>Goodness-of-fit on F<sup>2</sup></b>	1.091
<b>Final R indices [I&gt;2sigma(I)]</b>	R1 = 0.0776, wR2 = 0.2322
<b>R indices (all data)</b>	R1 = 0.0891, wR2 = 0.2454
<b>Extinction coefficient</b>	n/a
<b>Largest diff. peak and hole</b>	0.651 and -0.594 e.Å <sup>-3</sup>

**Table S4.** Crystal data and structure refinement for **12**.

<b>Empirical formula</b>	C <sub>30</sub> H <sub>21</sub> Br <sub>2</sub> N <sub>3</sub> O <sub>4</sub>
<b>Formula weight</b>	647.315
<b>Temperature</b>	100(2) K
<b>Wavelength</b>	0.71073 Å
<b>Crystal system</b>	Triclinic
<b>Space group</b>	P -1
<b>Unit cell dimensions</b>	a = 13.4552(6) Å, α = 78.712(2)° b = 13.6974(7) Å, β = 73.880(2)° c = 14.8065(6) Å, γ = 76.746(2)°
<b>Volume</b>	2526.2(2) Å <sup>3</sup>
<b>Z</b>	4
<b>Density (calculated)</b>	1.702 Mg/m <sup>3</sup>
<b>Absorption coefficient</b>	3.253 mm <sup>-1</sup>
<b>F(000)</b>	1296
<b>Crystal size</b>	0.321 x 0.150 x 0.100 mm <sup>3</sup>
<b>Theta range for data collection</b>	2.259 to 28.078°
<b>Index ranges</b>	-16<=h<=17, -17<=k<=18, -19<=l<=19

<b>Reflections collected</b>	57642
<b>Independent reflections</b>	12264 [R(int) = 0.0751]
<b>Completeness to theta = 25.242°</b>	99.9 %
<b>Refinement method</b>	Full-matrix least-squares on F <sup>2</sup>
<b>Data / restraints / parameters</b>	12264 / 0 / 705
<b>Goodness-of-fit on F<sup>2</sup></b>	1.011
<b>Final R indices [I&gt;2sigma(I)]</b>	R1 = 0.0417, wR2 = 0.0760
<b>R indices (all data)</b>	R1 = 0.0842, wR2 = 0.0876
<b>Extinction coefficient</b>	n/a

## 8. TGA thermograms



**Figure S21.** TGA thermograms of **P1 – P5**.

## 9. DSC thermograms

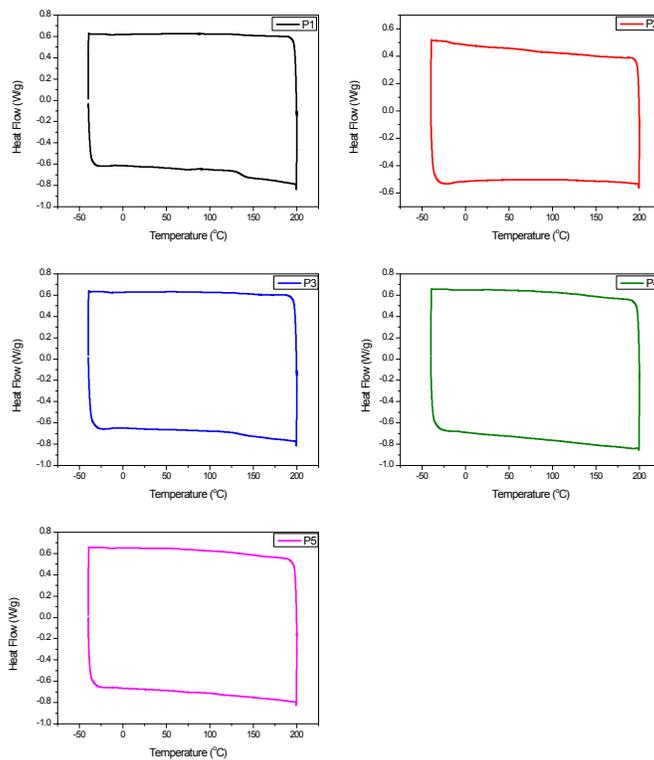


Figure S22. DSC thermograms of P1 – P5.

## 10. XRD diffractograms

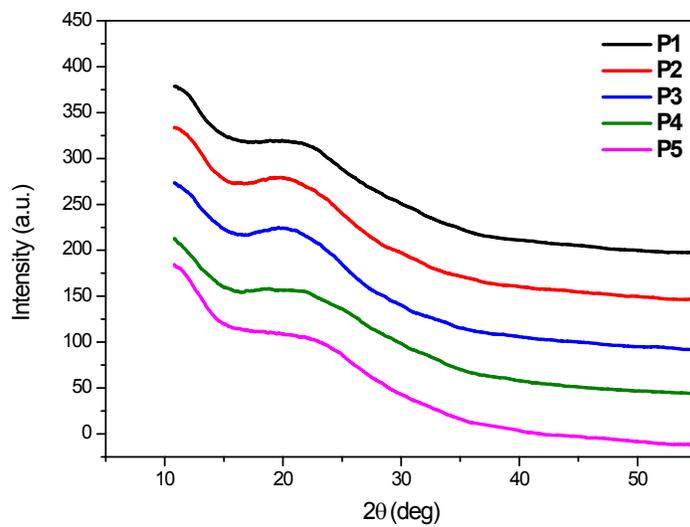
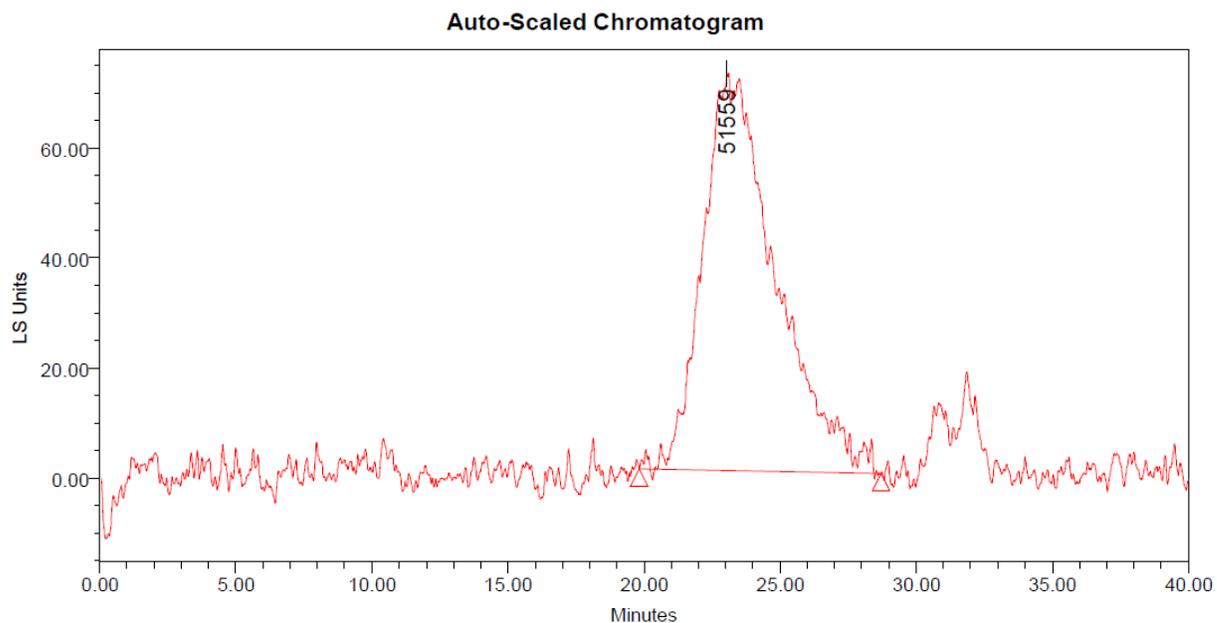
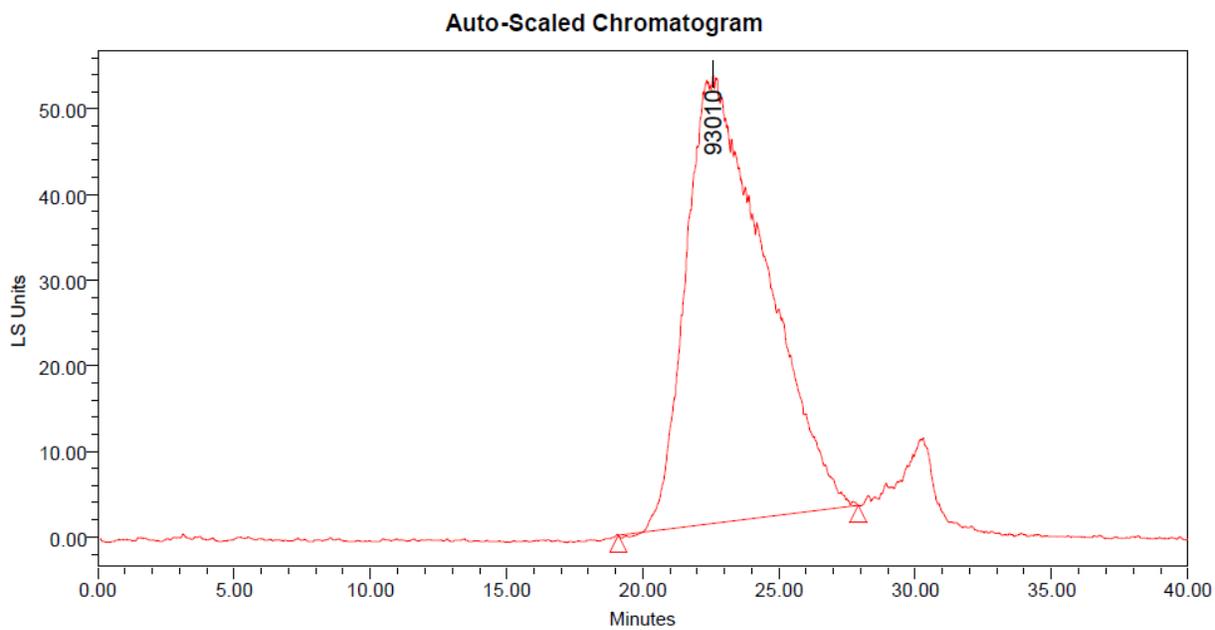


Figure S23. XRD diffractograms of P1 – P5

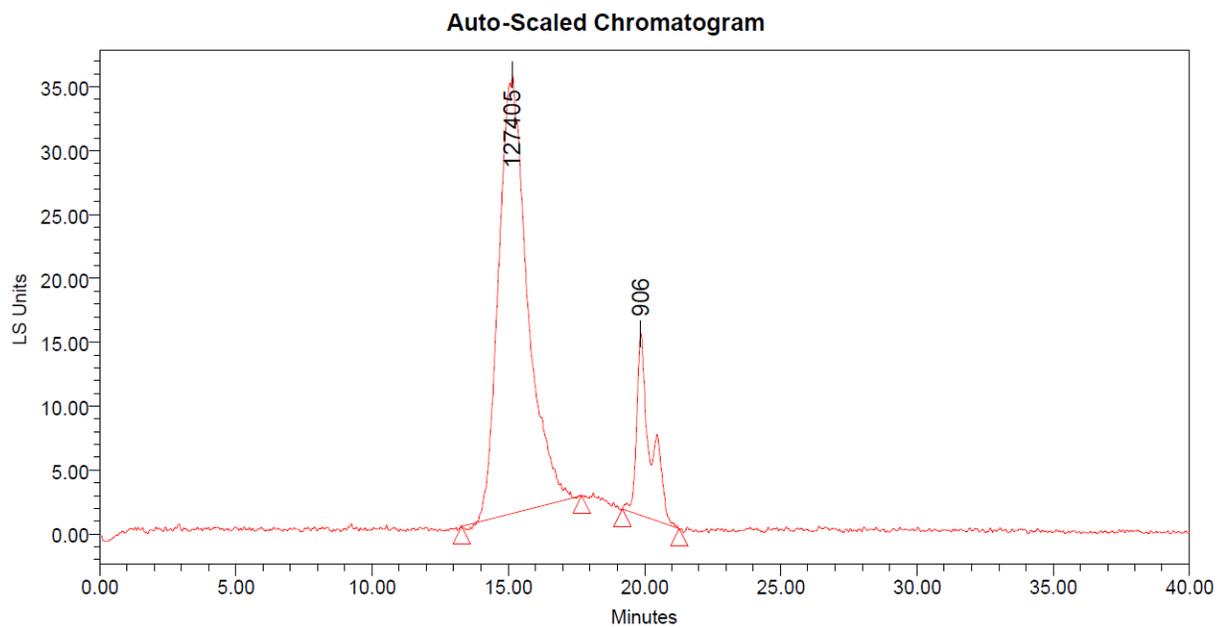
## 11. GPC chromatograms



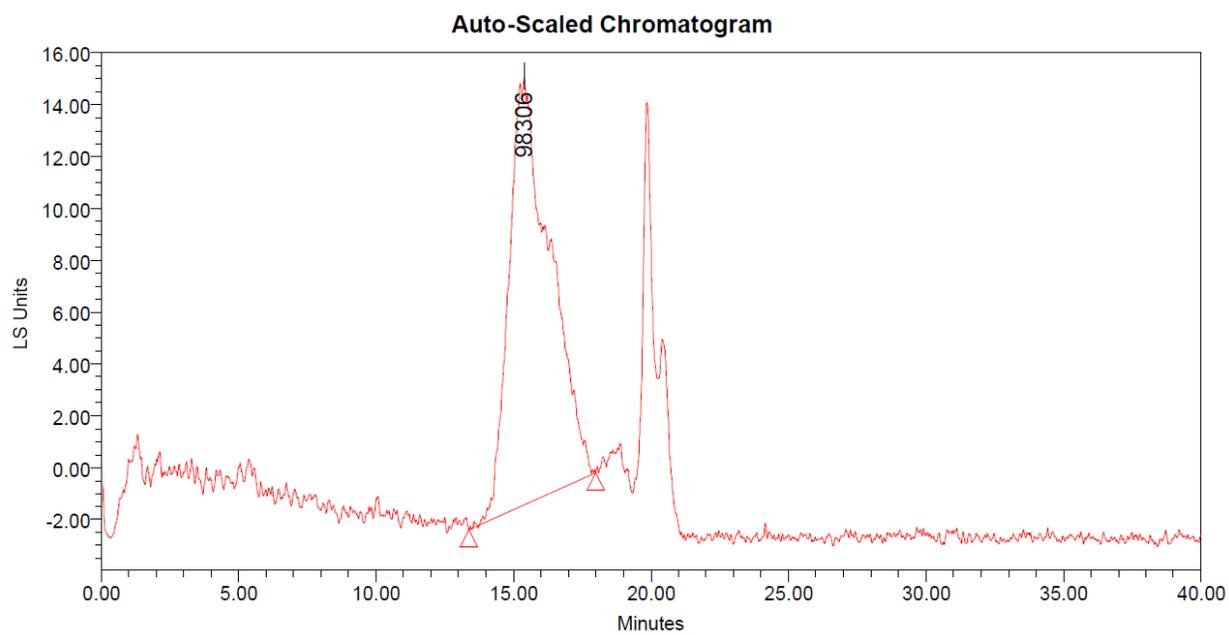
**Figure S24.** GPC chromatogram of **P1**.



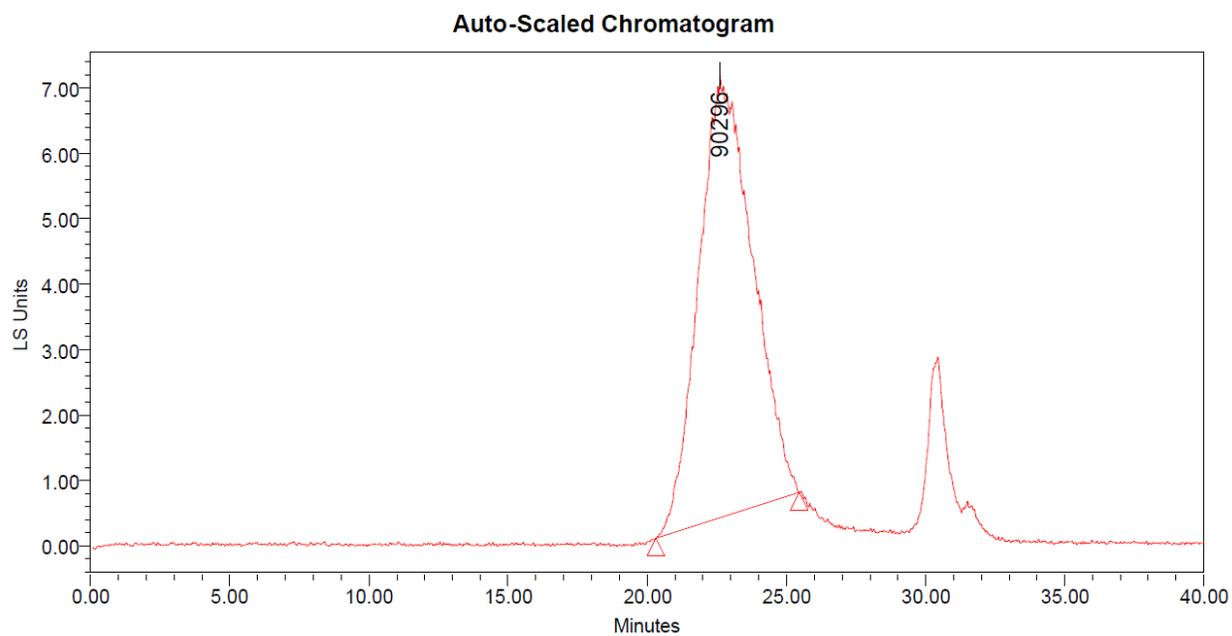
**Figure S25.** GPC chromatogram of **P2**.



**Figure S26.** GPC chromatogram of **P3**.



**Figure S27.** GPC chromatogram of **P4**.



**Figure S28.** GPC chromatogram of **P5**.