

Electronic Supplementary Information

Synthesis and characterization of TEMPO- and viologen-polymers for water-based redox-flow batteries

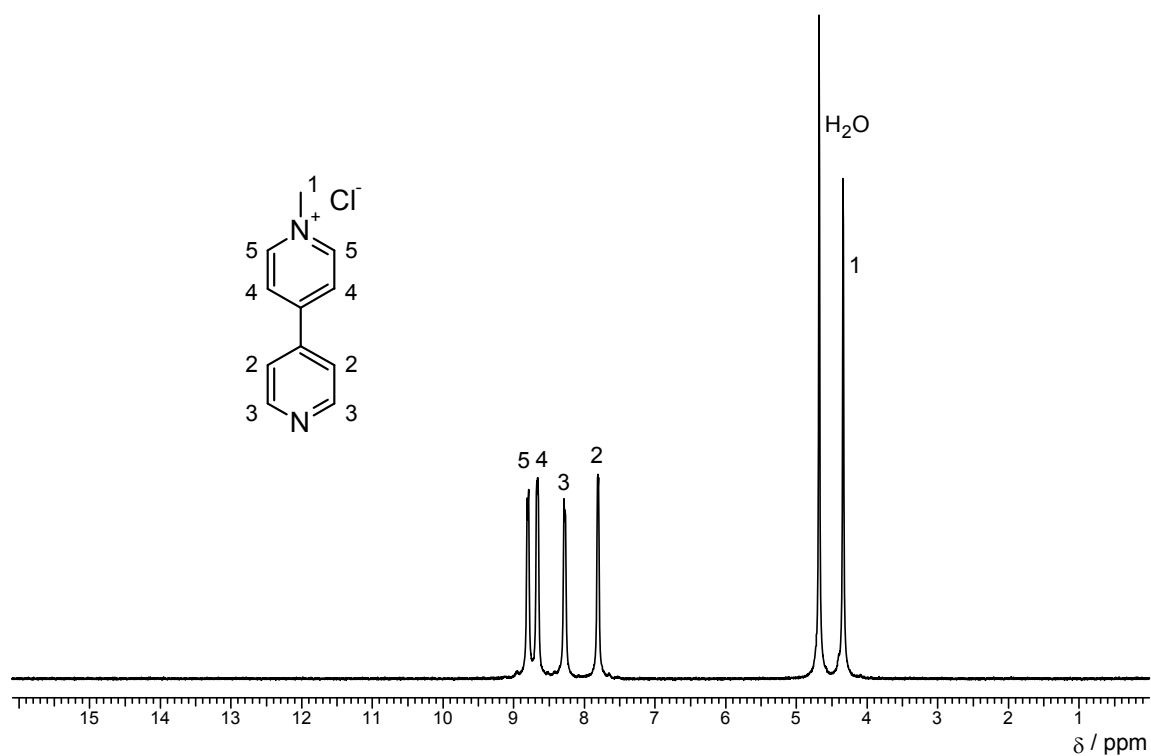
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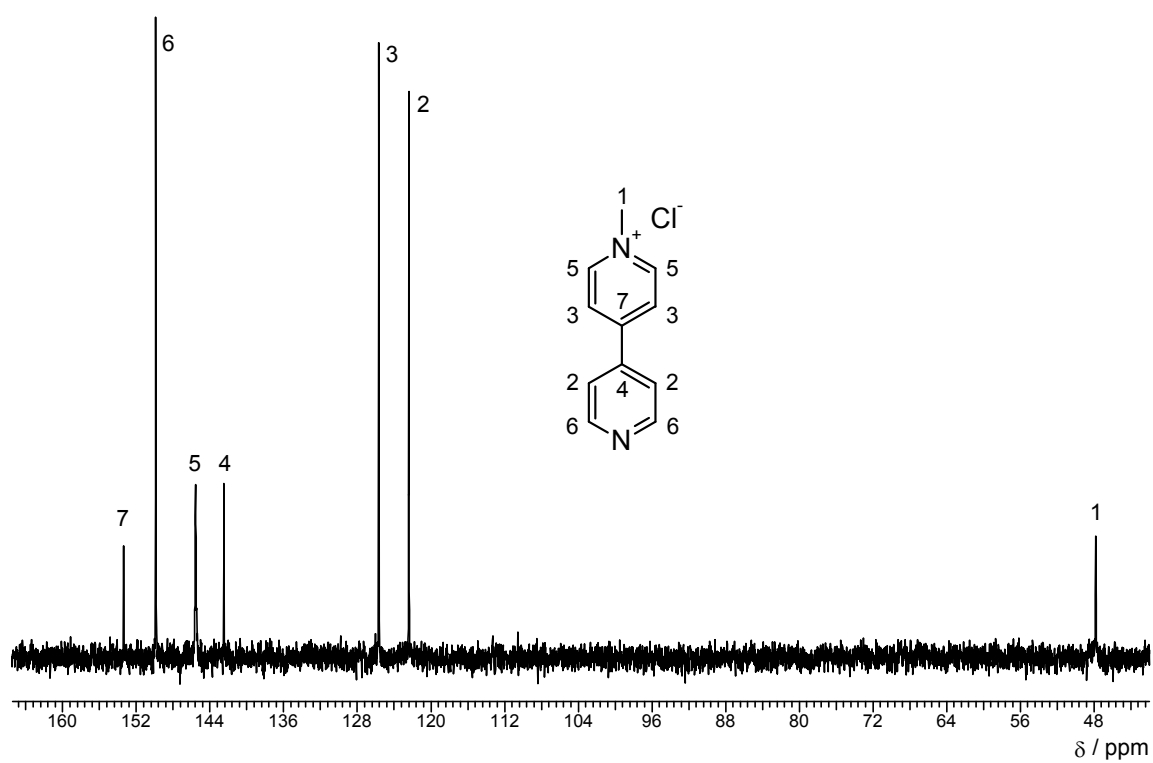
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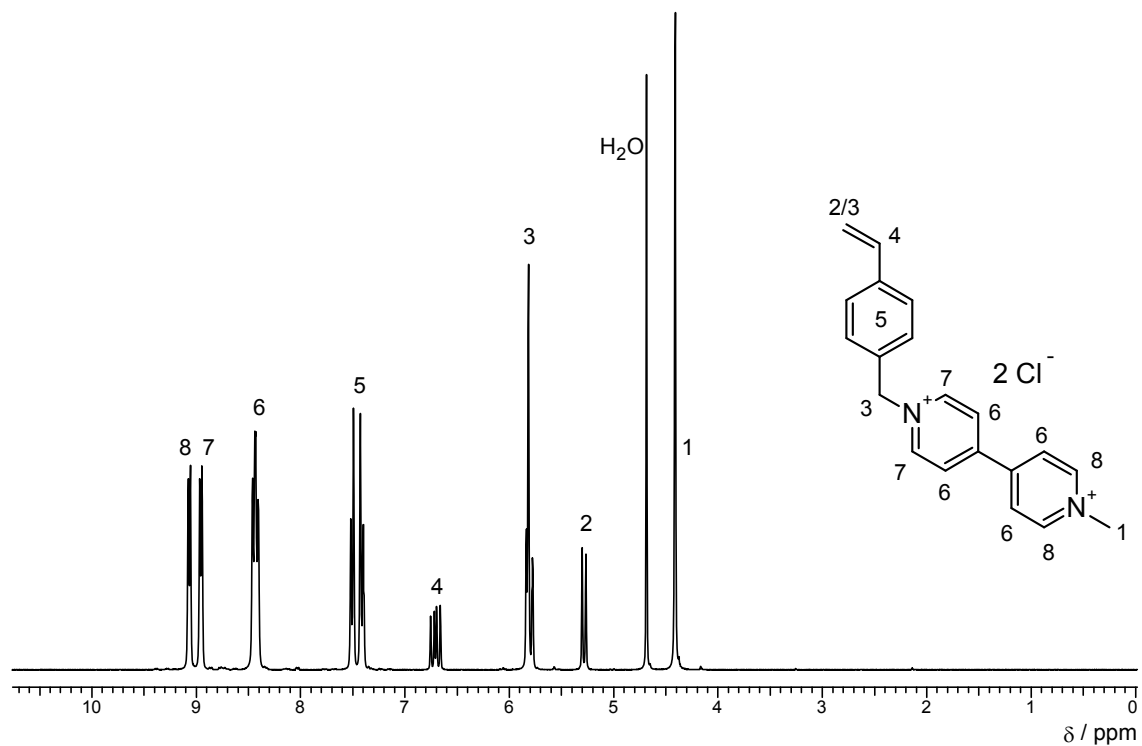
ESI Figure 1: ^1H NMR spectrum (300 MHz, D_2O): 1-methyl-[4,4'-bipyridine]-1-ium chloride (1a).



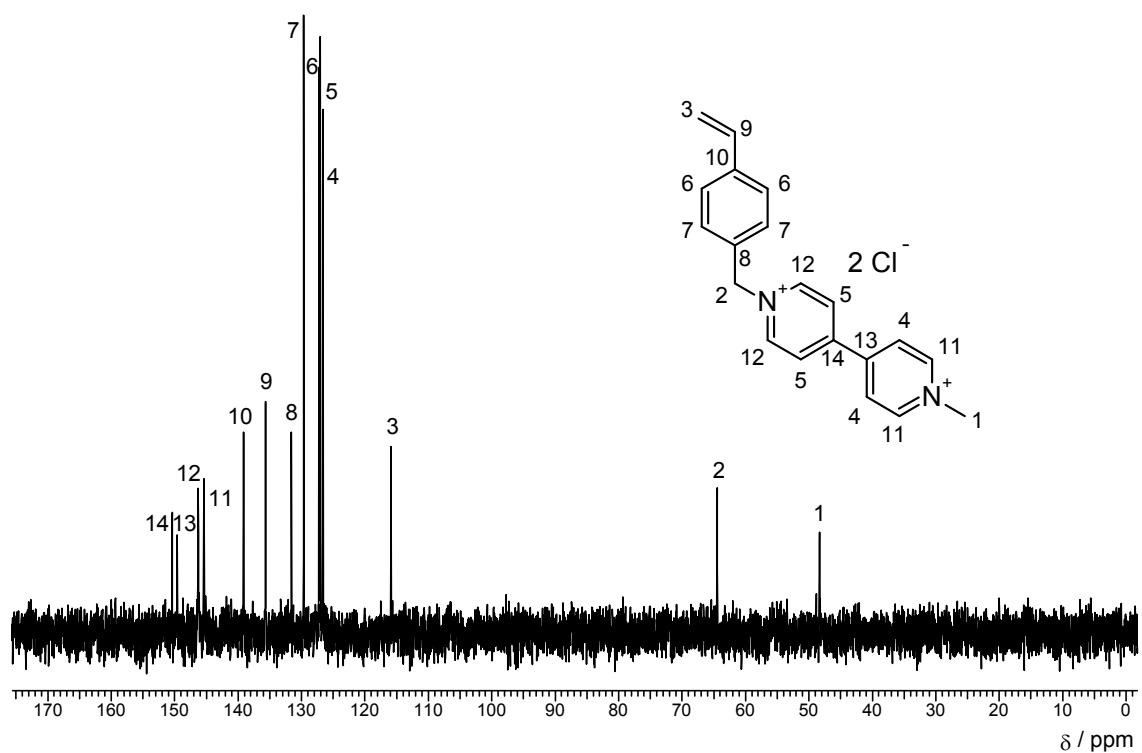
ESI Figure 2: ^{13}C NMR spectrum (75 MHz, D_2O): 1-methyl-[4,4'-bipyridine]-1-ium chloride (1a).



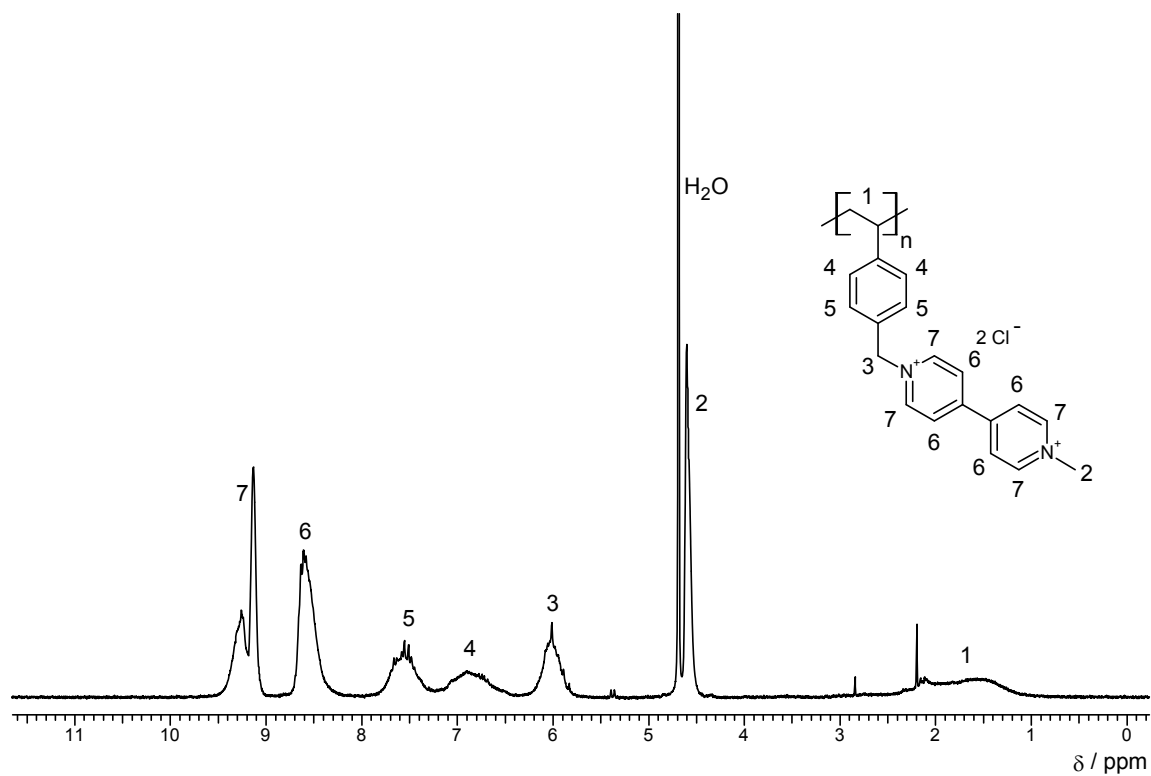
ESI Figure 3: ^1H NMR spectrum (300 MHz, D_2O): 1-Methyl-1'-(4-vinylbenzyl)-[4,4'-bipyridine]-1,1'-diium dichloride (2).



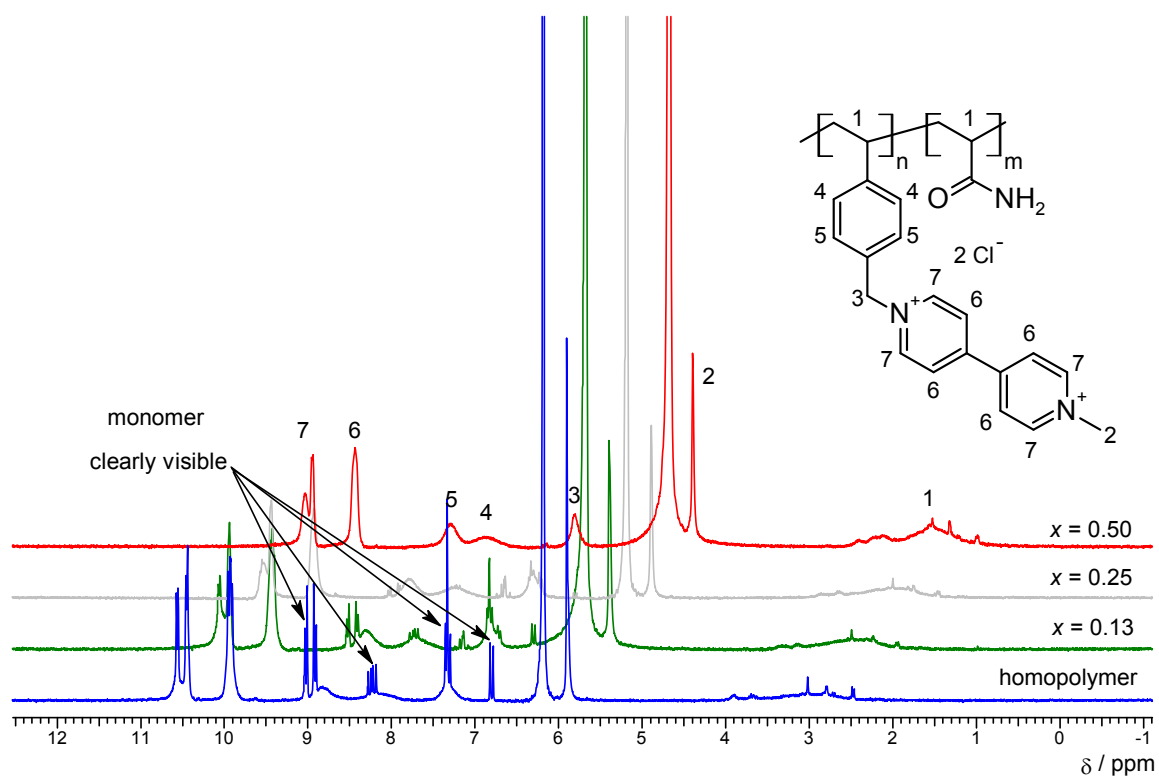
ESI Figure 4: ^{13}C NMR spectrum (75 MHz, D_2O): 1-Methyl-1'-(4-vinylbenzyl)-[4,4'-bipyridine]-1,1'-diium dichloride (2).



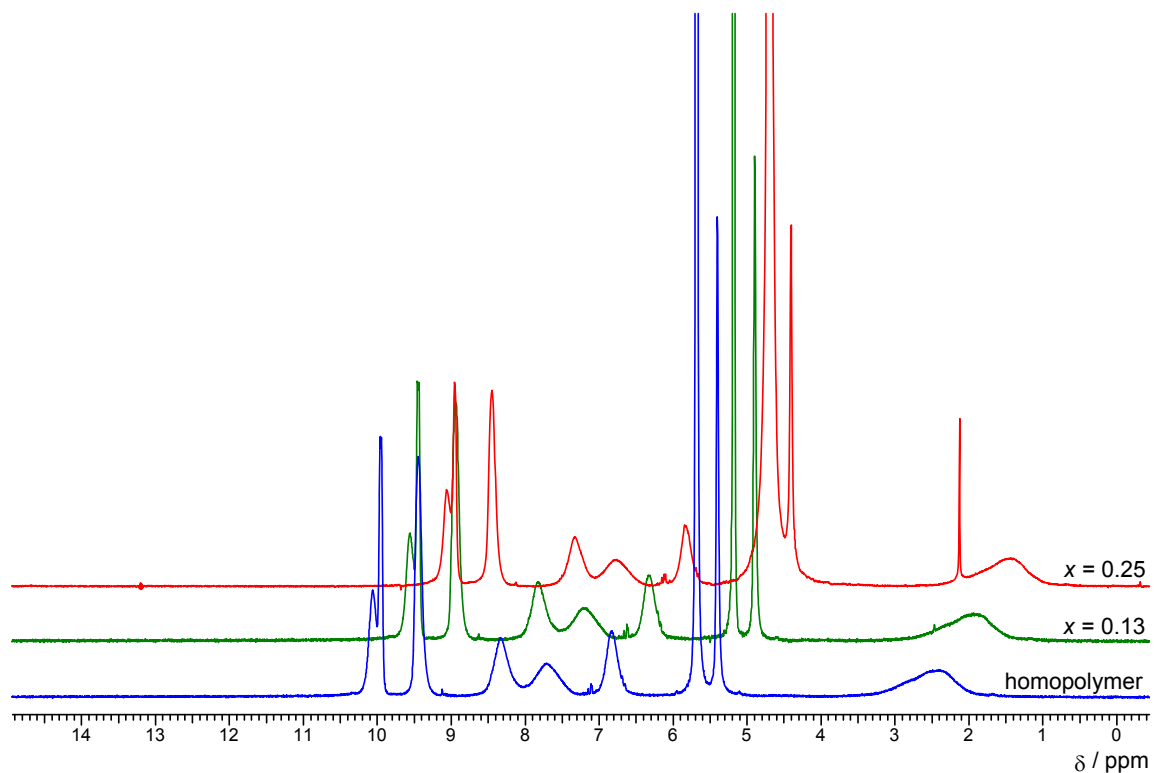
ESI Figure 5: Representative ^1H NMR spectrum (300 MHz, D_2O): poly(1-methyl-1'-(4-vinylbenzyl)-[4,4'-bipyridine]-1,1'-dium dichloride) (P2a), prepared with 7 mol-% thioacetic acid.



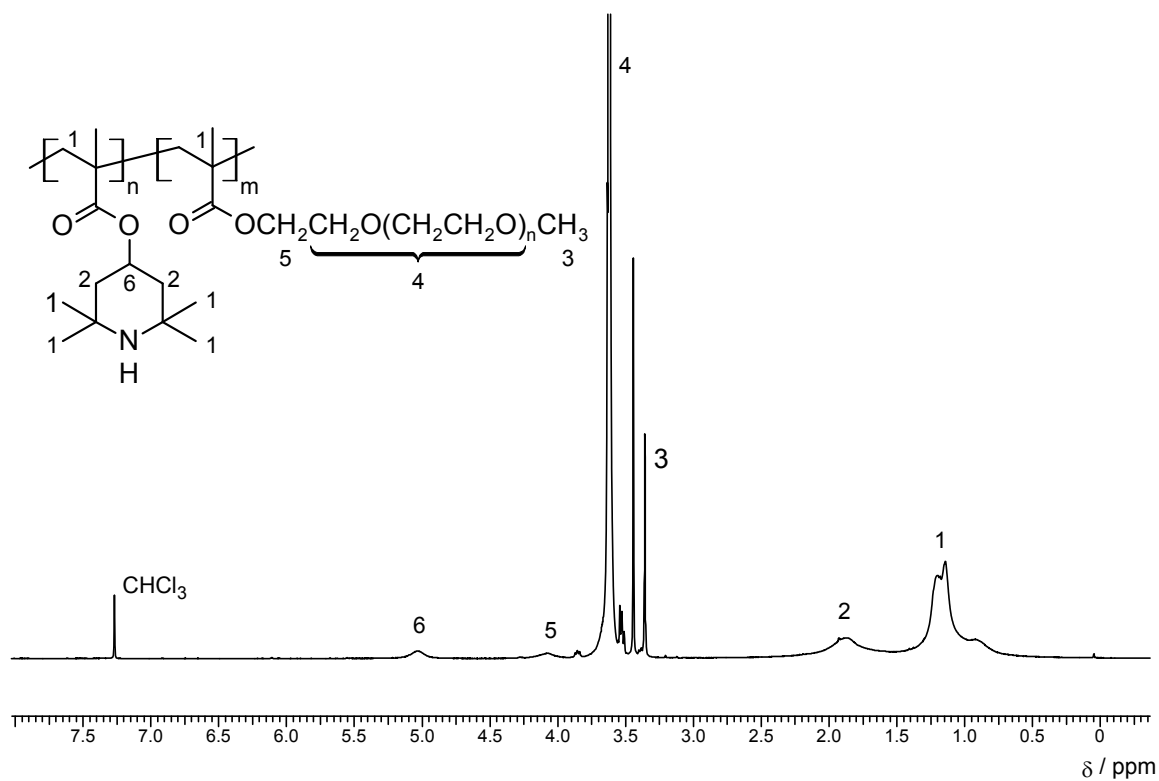
ESI Figure 6: ^1H NMR spectrum (300 MHz, D_2O) of poly(1-methyl-1'-(4-vinylbenzyl)-[4,4'-bipyridine]-1,1'-dium dichloride-*co*-acrylamide) (P2b) prepared using 4,4'-azobis(4-cyanovaleric acid) at different mole fractions of acrylamide comonomer, 16 h reaction time.



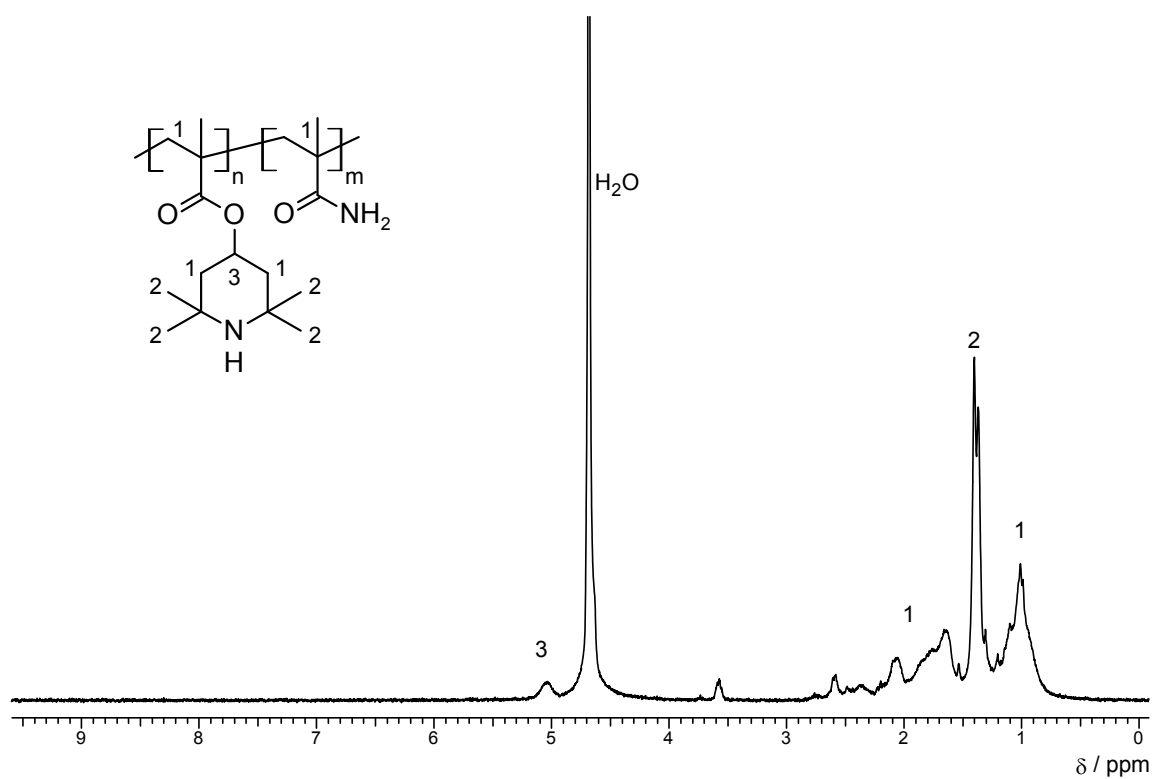
ESI Figure 7: ^1H NMR spectrum (300 MHz, D_2O) of poly(1-methyl-1'-(4-vinylbenzyl)-[4,4'-bipyridine]-1,1'-dium dichloride-*co*-acrylamide) (P2b) prepared using potassium peroxodisulfate at different mole fractions of acrylamide comonomer, 1 h reaction time.



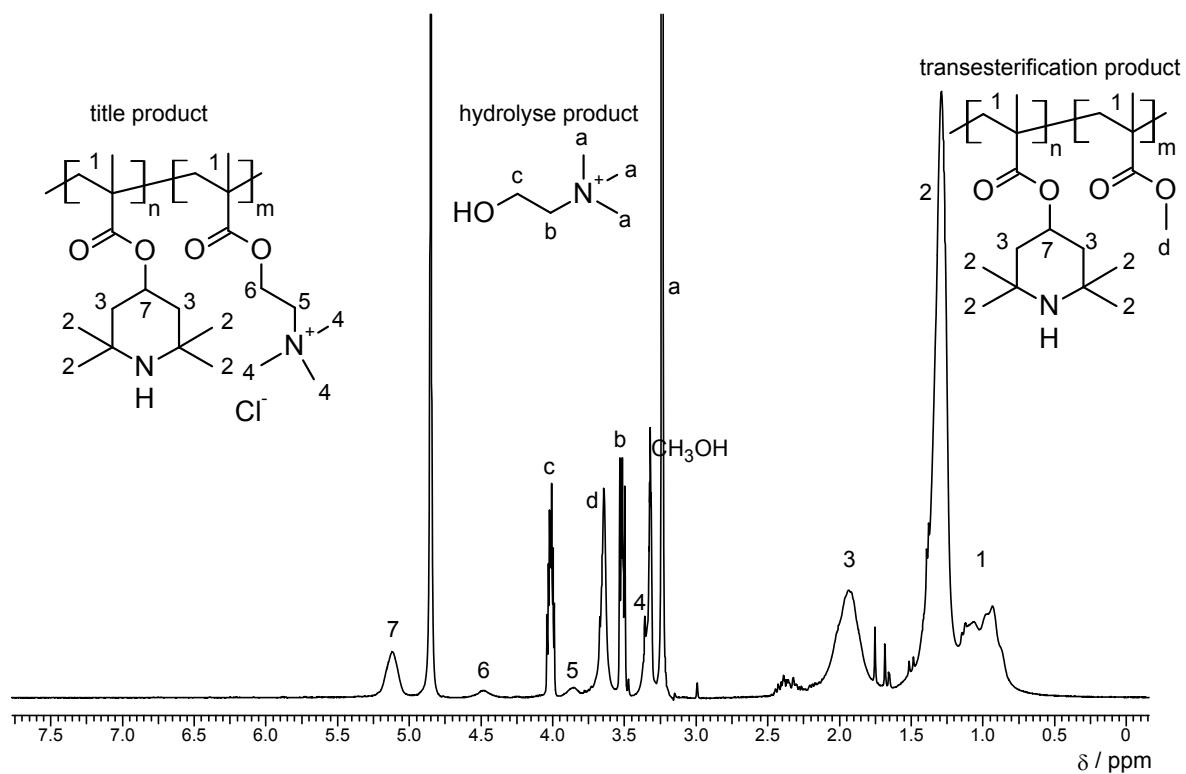
ESI Figure 8: Representative ^1H NMR spectrum (300 MHz, CDCl_3): Poly(2,2,6,6-tetramethylpiperidin-4-yl methacrylate-co-poly(ethylene glycol) methyl ether methacrylate) (P3b), $x_{4b} = 0.33$.



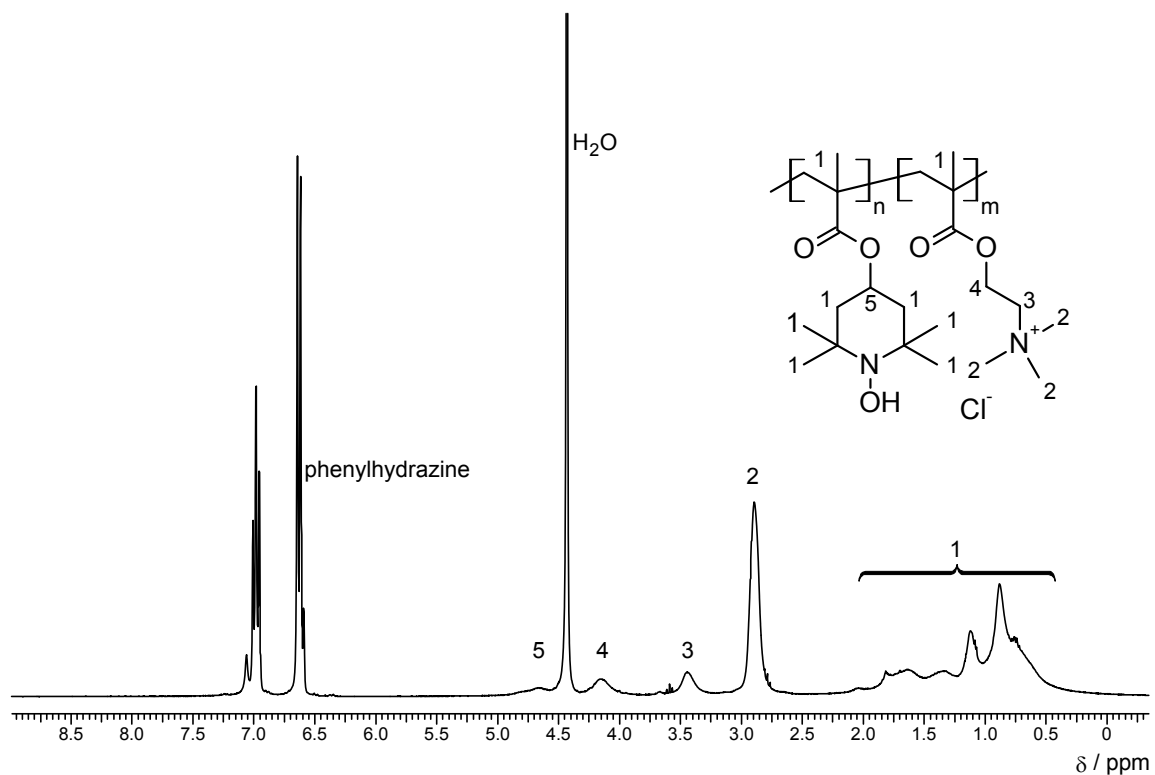
ESI Figure 9: Representative ^1H NMR spectrum (300 MHz, D_2O): Poly(2,2,6,6-tetramethylpiperidin-4-yl methacrylate-co-methacrylamide) (P3d), $x_{4d} = 0.67$ prepared with 8% 2-mercaptoethanol.



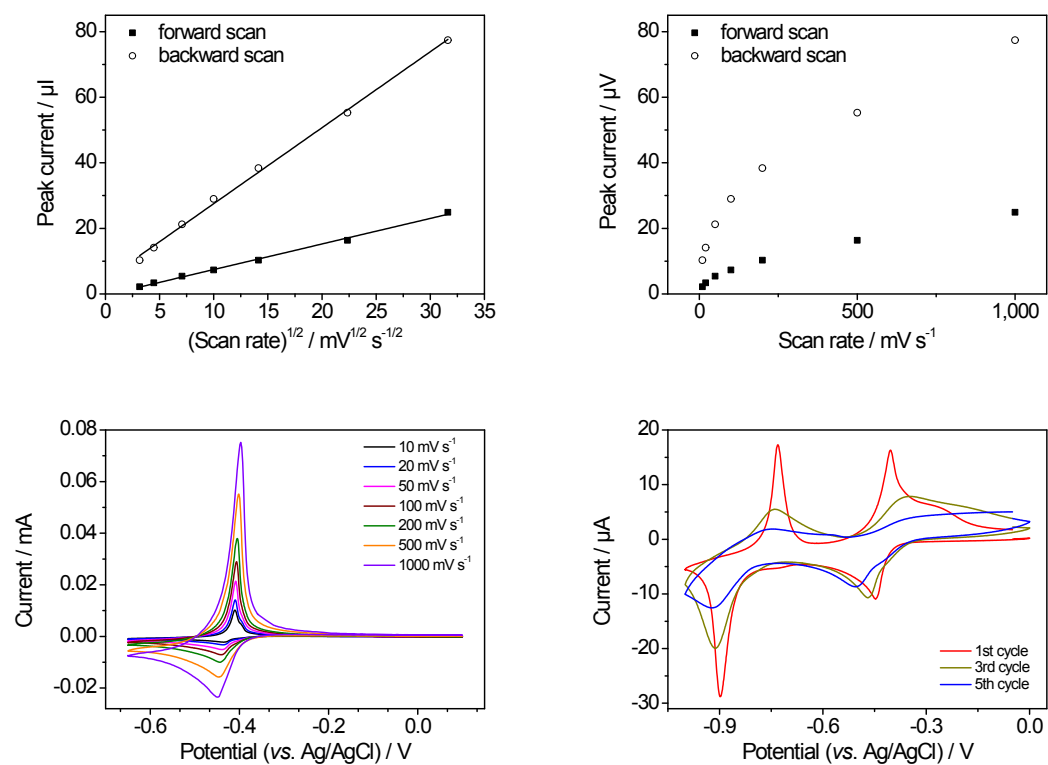
ESI Figure 10: ^1H NMR spectrum (300 MHz, methanol- d_4) of poly(2,2,6,6-tetramethylpiperidin-4-yl methacrylate-co-2-(methacryloyloxy)- N,N,N -trimethylethane ammonium chloride) (P3e) showing the partial hydrolysis of the ester in monomer 4e ($x_{4e} = 0.5$), occurring if the reaction mixture is not neutralized with HCl prior to polymerization.



ESI Figure 11: Representative ^1H NMR spectrum (300 MHz, D_2O): Poly(4-methacryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl-co-2-(methacryloyloxy)-*N,N,N*-trimethylethane ammonium chloride) (P4e), prepared with 8 mol-% 2-mercaptoethanol, quenched with phenyl hydrazine.

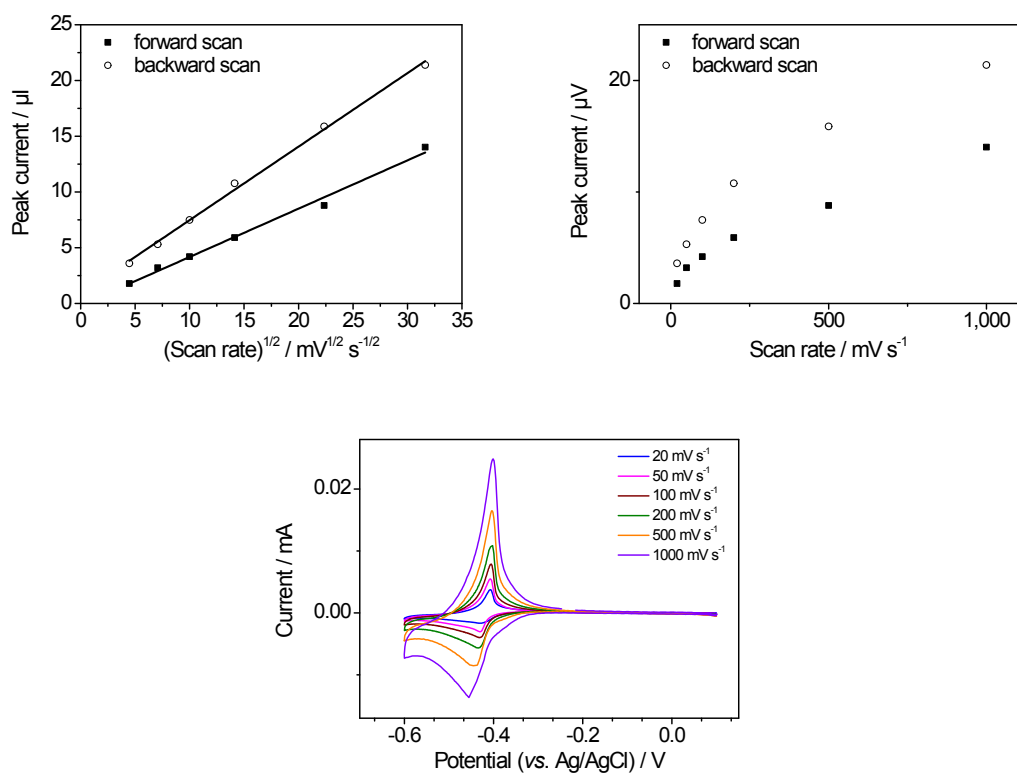


ESI Figure 12: Poly(1-methyl-1'-(4-vinylbenzyl)-[4,4'-bipyridine]-1,1'-diium dichloride) (P2a), prepared with 7 mol-% thioacetic acid.



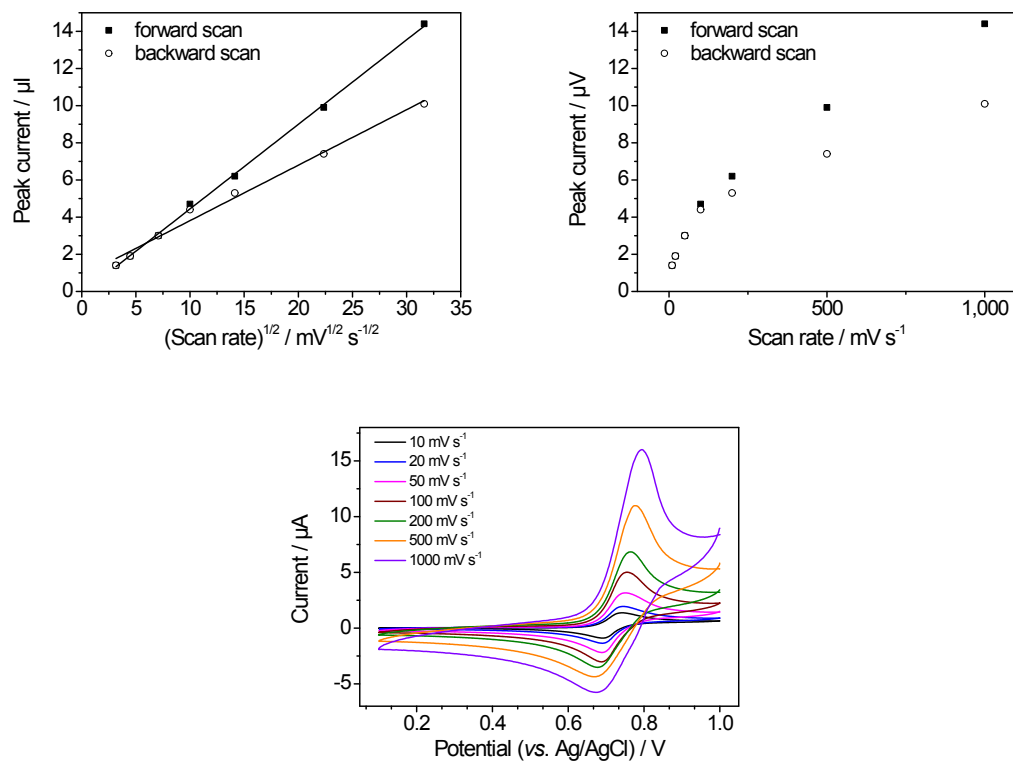
Scan rate (mV s^{-1})	$I_{\text{forward scan}}$ (μA)	$I_{\text{backward scan}}$ (μA)
1000	24.9	77.4
500	16.4	55.3
200	10.3	38.4
100	7.3	28.9
50	5.4	21.2
20	3.4	14.1
10	2.2	10.3

ESI Figure 13: Poly(1-methyl-1'-(4-vinylbenzyl)-[4,4'-bipyridine]-1,1'-diium dichloride) (P2a), prepared without thioacetic acid.



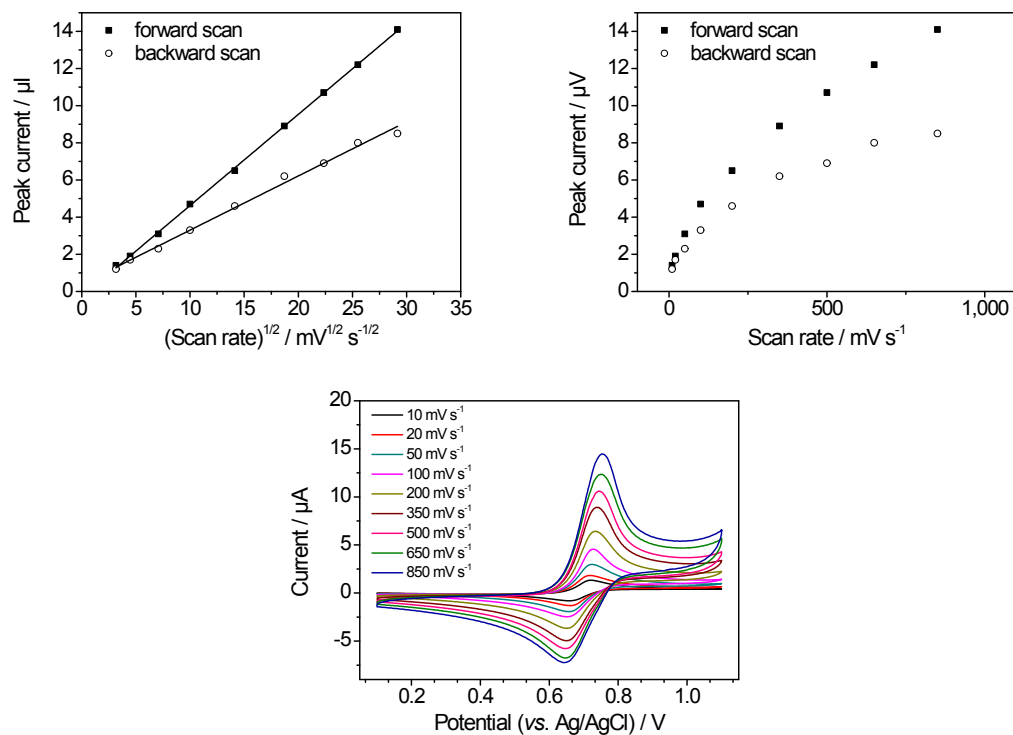
Scan rate (mV s^{-1})	$I_{\text{forward scan}}$ (μA)	$I_{\text{backward scan}}$ (μA)
1000	14.0	21.4
500	8.8	15.9
200	5.9	10.8
100	4.2	7.5
50	3.2	5.3
20	1.78	3.6

ESI Figure 14: Poly(4-methacryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl-co-2-(methacryloyloxy)-*N,N,N*-trimethylethane ammonium chloride) (P4e), prepared with 1 mol-% 2-mercaptoethanol.



Scan rate (mV s^{-1})	$I_{\text{forward scan}}$ (μA)	$I_{\text{backward scan}}$ (μA)
1000	14.4	10.1
500	9.9	7.4
200	6.2	5.3
100	4.7	4.4
50	3	3
20	1.9	1.9
10	1.4	1.4

ESI Figure 15: Poly(4-methacryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl-co-2-(methacryloyloxy)-*N,N,N*-trimethylethane ammonium chloride) (P4e), prepared with 8 mol-% 2-mercaptoethanol.



Scan rate (mV s^{-1})	$I_{\text{forward scan}}$ (μA)	$I_{\text{backward scan}}$ (μA)
850	14.1	8.5
650	12.2	8.0
500	10.7	6.9
350	8.9	6.2
200	6.5	4.6
100	4.7	3.3
50	3.1	2.3
20	1.9	1.7
10	1.4	1.2