Side-Chain Engineering Approach to Solvent-Resistant

Semiconducting Polymer Thin Films

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Supplementary Information

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1. General information

6,6'-dibromoisoinigo¹ and bis(trimethylstannyl)thiophene² were synthesized according to procedure reported in the literature. Vinyl group end-functionalized polyisobutylene was purchased from BASF as Glissopal[®] 1000 (Mn = 1000g / mol). Other starting materials and reagents were purchased from Aldrich and Alfa-Aesar, and used as received without further purification. Toluene, CH₂Cl₂, DMF and THF were dried through dry Al₂O₃ columns and used without further treatment. Analytical thin-layer chromatography (TLC) was performed on glass that is precoated with silica gel 60-F₂₅₄ (Sorbtech). Flash column chromatography was carried out using Biotage[®] IsoleraTM Prime instrument with various size of SiO₂ Biotage ZIP[®] cartridge. UV/vis absorption spectra were recorded using Shimadzu UV-2600, while the fluorescent emission spectra were measured on Horiba Fluoromax-4. ¹H and ¹³C NMR spectra were obtained on a 500 MHz Varian Inova at room temperature and processed by MestReNova 6.1.0. Chemical shifts are reported in ppm relative to the signals corresponding to the residual non-deuterated solvents (CDCl₃: δ 7.26 for ¹H and 77.16 for ¹³C at room temperature). Size exclusion chromatography (SEC) was performed on TOSOH EcoSEC (HLC-8320GPC) in THF solution at 40°C temperature and the molecular weights calculated using a calibration curve based on polystyrene standards, equipped with TSKgel SuperHM-M and TSKgel SuperH-RC. Thermal gravimetric analysis (TGA) was recorded under nitrogen atmosphere with heating rate of 10 °C min⁻¹ from 40 to 500 °C using TA Q500. Cyclic voltammetry (CV) was carried out in nitrogenpurged dichloromethane (oxidation scan) at room temperature with a CHI voltammetric analyzer. n-Bu₄PF₆ (0.1 M) was used as the supporting electrolyte. High-resolution Matrix-assisted laser desorption/ionization (HR-MALDI) mass spectra were measured on Applied Biosystems 4800 MALDI-TOF. Atomic force microscopy (AFM) images were recorded with Bruker Dimension

Icon AFM in a tapping mode and processed using NanoScope Analysis. The evaluations of the FETs and inverters were carried out in atmosphere (humidity 50-60%) on a probe stage using a Keithley 4200 SCS as parameter analyzer.

2. TGA Trace of Monomer 1



Figure S1. TGA Trace of Monomer 1.

3. Solvent Resistant Property of Annealed Thin Film



Pristine Film

Annealed Film

Figure S2. Solvent resistant properties of pristine film and annealed PIIT-Boc65 thin film in CH₂Cl₂.

3. FT-IR Spectra



Figure S3. IR spectra comparison of Pristine and Annealed PIIT-Boc75.



Figure S4. IR spectra comparison of Pristine and Annealed PIIT-Boc65.



Figure S5. IR spectra comparison of Pristine and Annealed monomer 1 and 6,6'-dibromoisoindigo.

4. OFET Profile



Figure S6. (a) Transfer curve and (b) output curve of PIIT-Boc75 in immersed annealed film (CH₂Cl₂).



Figure S7. (a) Transfer curve and (b) output curve of PIIT-Boc75 in immersed annealed film (DMSO).



Figure S8. (a) Transfer curve and (b) output curve of PIIT-Boc75 in immersed annealed film (*p*-xylene).



Figure S9. (a) Transfer curve and (b) output curve of PIIT-Boc70 in pristine film.



Figure S10. (a) Transfer curve and (b) output curve of PIIT-Boc70 in annealed film.



Figure S11. (a) Transfer curve and (b) output curve of PIIT-Boc75 in immersed annealed film (CH₃Cl).



Figure S12. (a) Transfer curve and (b) output curve of PIIT-Boc75 in immersed annealed film (CH₂Cl₂).



Figure S13. (a) Transfer curve and (b) output curve of PIIT-Boc75 in immersed annealed film (DMSO).



Figure S14. (a) Transfer curve and (b) output curve of PIIT-Boc75 in immersed annealed film (*p*-xylene).



Figure S15. Device fabrication of TG/BC OFET structure. (i) spin casting active layer; (ii) spin casting CYTOP dielectric layer; (iii) vacuum depositing gate electrode; (iv) immersing into solvent for 5 min at room temperature; (v) annealing for 10 min at 180°C.

PIIT-Boc75		μ (cm ² V ⁻¹ s ⁻¹)	$V_{th}\left(V ight)$
Pristine ^a		1.61E-04	1.98
Annealed ^b		2.15E-04	6.15
CH ₂ Cl ₂	P.°	7.22E-05	7.40
	A. ^d	3.15E-04	25.96
CF	Р.	1.09E-05	13.22
	A.	1.38E-04	23.69
<i>p</i> -Xylene	Р.	6.26E-05	37.25
	A.	4.90E-05	43.77
DMSO	P.	N. A ^e	N. A
	A.	1.22E-04	42.73

Table S1. OFET property of PIID-Boc75 in pristine film, annealed film and immersed annealed film.

a. Pristine film (Device 1); b Annealed film (Device 3); c. Immersed pristine film (Device 2); d. Immersed annealed film (Device 4); e. Not available.

Table S2. OFET property of PIID-Boc70 in pristine film, annealed film and immersed annealed film.

PIIT-Boc70		μ (cm ² V ⁻¹ s ⁻¹)	V _{th} (V)
pristine ^a		9.67E-05	29.31
annealed ^b		2.02E-04	30.15
CH ₂ Cl ₂	P.c	N. A. ^e	N. A.
	A. ^d	1.62E-04	27.30
CF	P.	N. A.	N. A.

	А.	1.38E-04	20.09
p-Xylene	Р.	N. A.	N. A.
	А.	1.99E-04	22.21
DMSO	Р.	8.95E-05	29.66
	А.	1.20E-04	37.58

a. Pristine film (Device 1); b Annealed film (Device 3); c. Immersed pristine film (Device 2); d. Immersed annealed film (Device 4); e. Not available.

5. GIXRD pattern



Figure 16. GIXRD pattern of (a) pristine and (b) annealed **PIIT-Boc75** films; and (c) pristine and (d) annealed **PIIT-boc70** films.

6. ¹H and ¹³C NMR Spectra



Figure S17. ¹H NMR of Monomer1 (500 MHz, CDCl₃, RT).



Figure S18. ¹³C NMR of Monomer 1 (500 MHz, CDCl₃, RT).



Figure S19. ¹H NMR of Monomer 2 (500 MHz, CDCl₃, RT).



Figure S20. ¹³C NMR of Monomer 2 (500 MHz, CDCl₃, RT).



Figure S21. H NMR of PIIT-Boc75 (500 MHz, CDCl₃, RT).



Figure S22. H NMR of PIIT-Boc70 (500 MHz, CDCl₃, RT).



Figure S23. H NMR of PIIT-Boc65 (500 MHz, CDCl₃, RT).

7. Reference

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- 2. G. Zhang, Y. Fu, Z. Xie and Q. Zhang, *Macromolecules*, 2011, 44, 1414–1420.