

# Chelation-assisted CuAAC of star-shaped polymers enables fast self-healing at low temperatures.

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Here shown exemplary for Table 2, entry 5 (**2a + 3**, 20 °C, CuF(PPh<sub>3</sub>)<sub>3</sub>) and for Table 2, entry 7 (**2b + 3**, 10 °C, CuF(PPh<sub>3</sub>)<sub>3</sub>).
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1. NMR- and IR-investigations for the synthesis of 2-(6-azidomethyl)-pyridine-4-carboxylic acid (**1a**) and 2-(6-azidomethyl)-pyridine-5-carboxylic acid (**1b**).

Synthetic-route to obtain 2-(6-azidomethyl)-pyridine-4-carboxylic acid (**4 = 1a**):

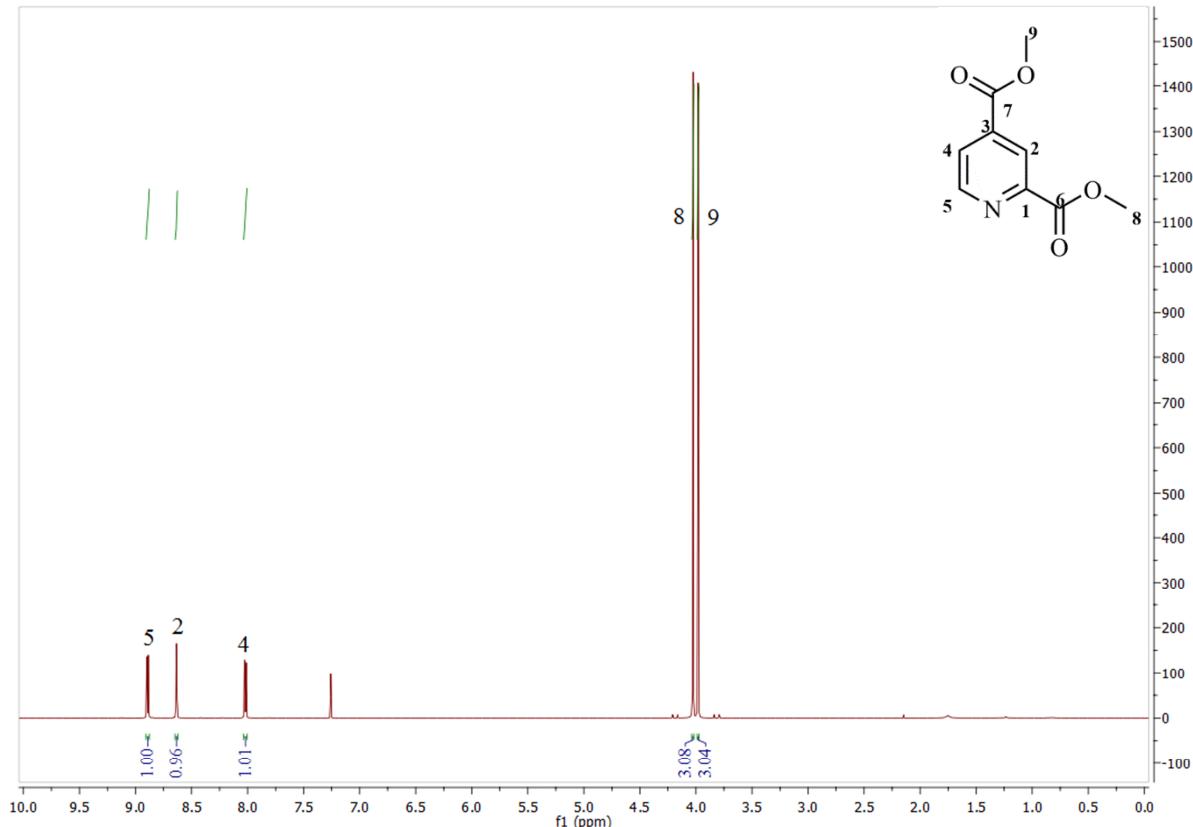
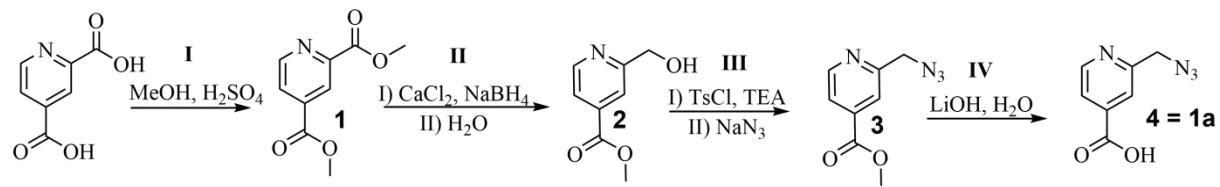


Figure S1. <sup>1</sup>H-NMR of pyridine-2,4-carboxylic acid dimethylester (**1**) in CDCl<sub>3</sub>.

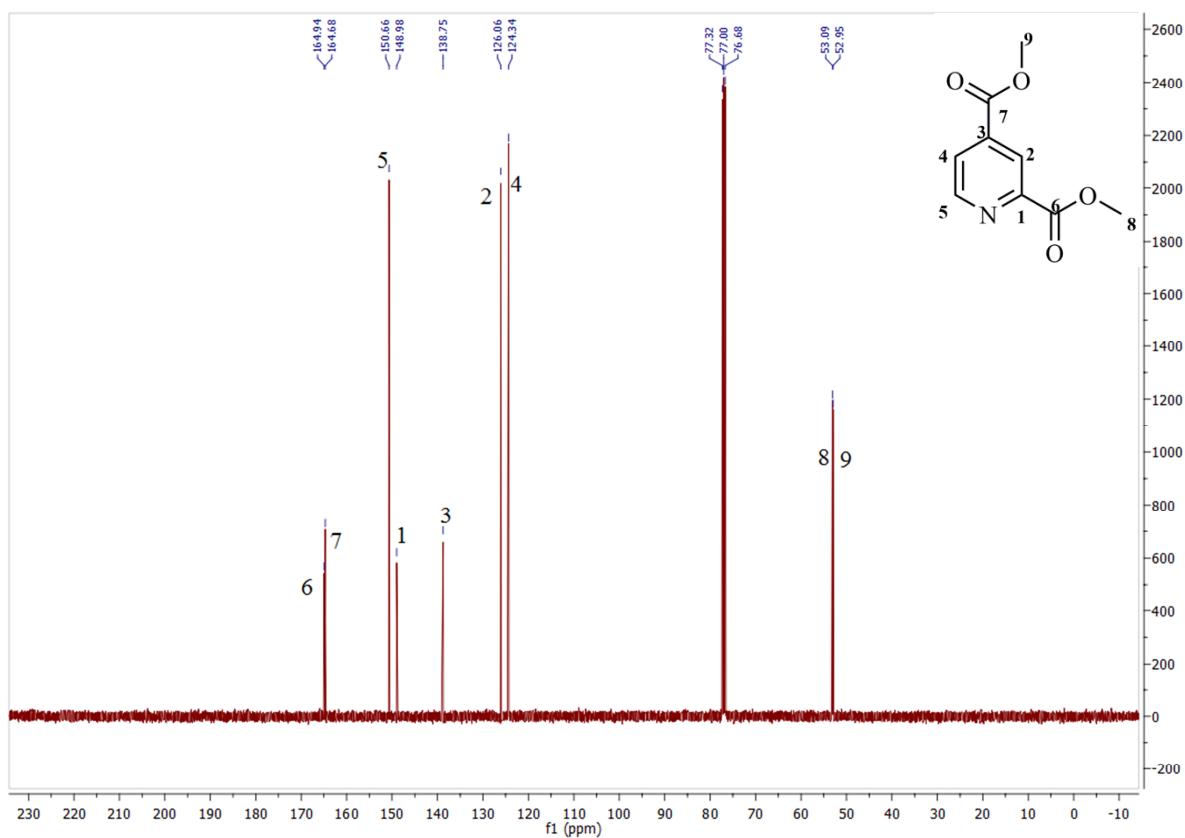


Figure S2.  $^{13}\text{C}$ -NMR of pyridine-2,4-carboxylic acid dimethylester (**1**) in  $\text{CDCl}_3$ .

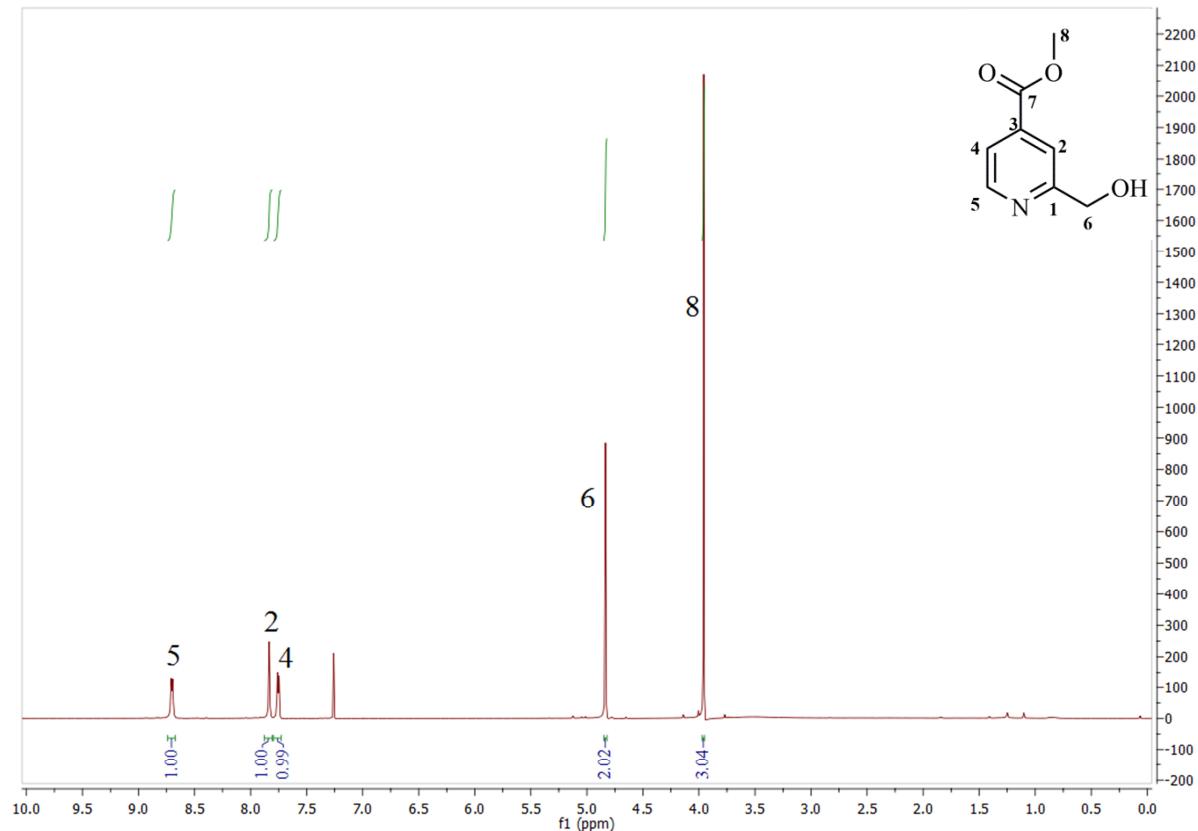


Figure S3.  $^1\text{H}$ -NMR of 2-(6-hydroxymethyl)-pyridine-4-carboxylic acid methylester (**2**) in  $\text{CDCl}_3$ .

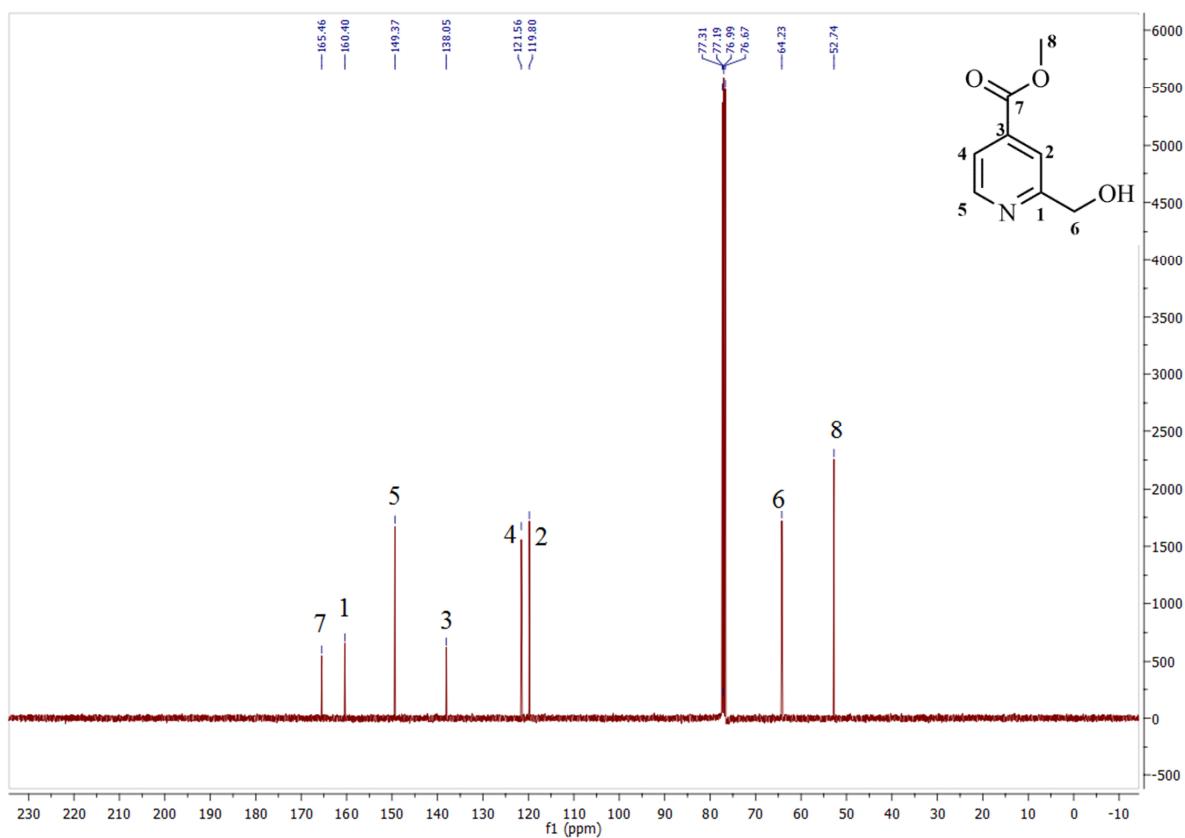


Figure S4.  $^{13}\text{C}$ -NMR of 2-(6-hydroxymethyl)-pyridine-4-carboxylic acid methylester (**2**) in  $\text{CDCl}_3$ .

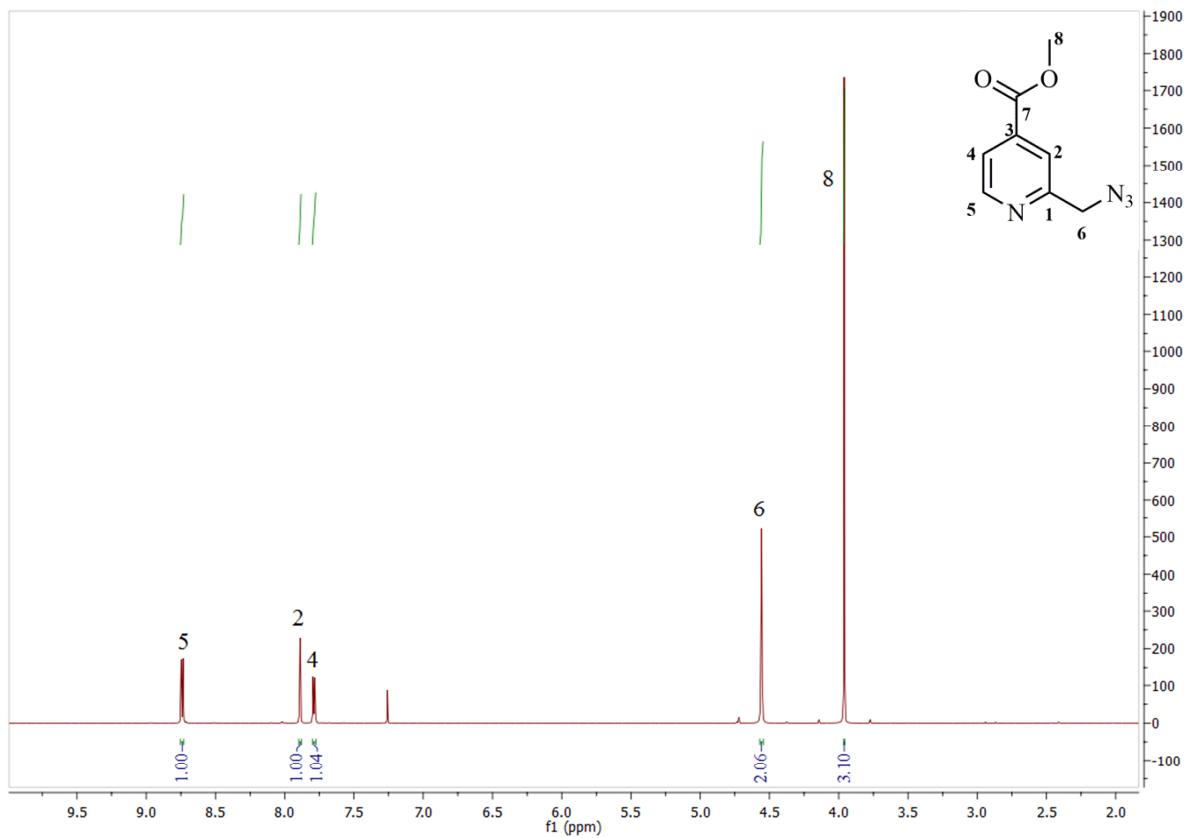


Figure S5.  $^1\text{H}$ -NMR of 2-(6-azidomethyl)-pyridine-4-carboxylic acid methylester (**3**) in  $\text{CDCl}_3$ .

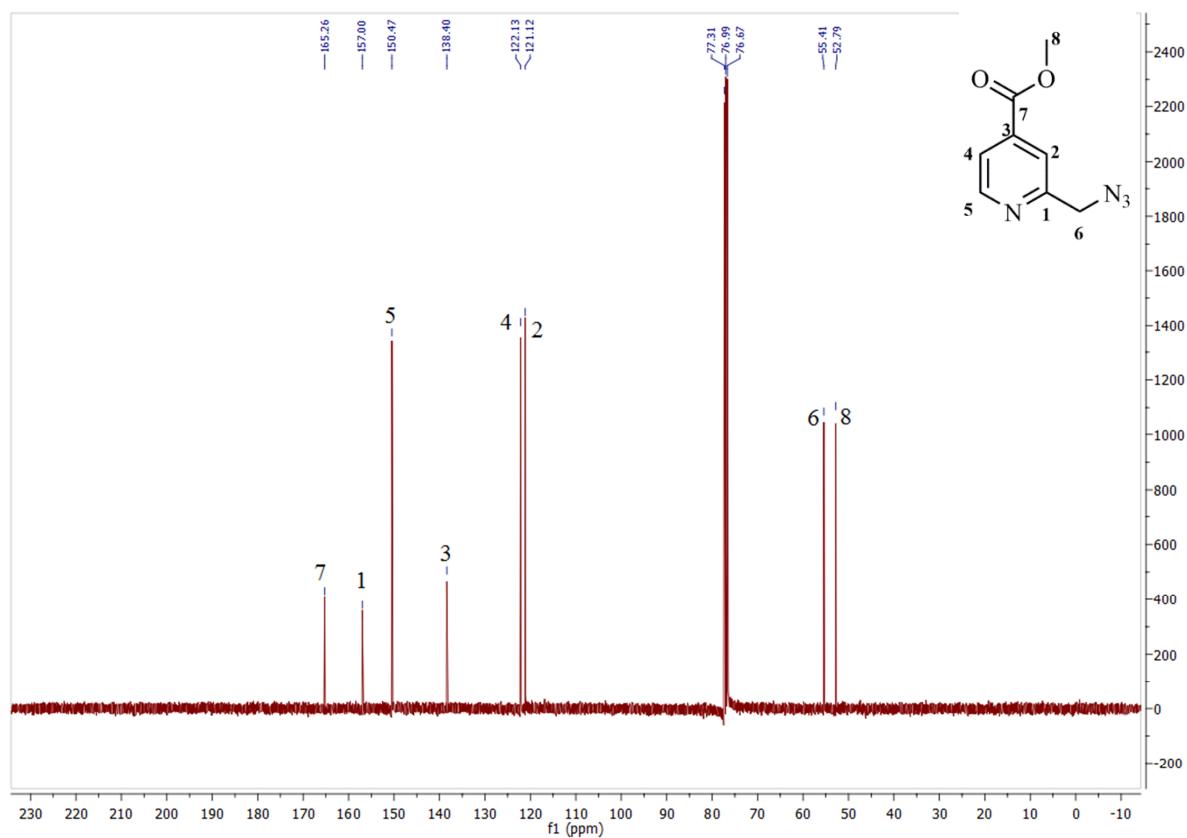


Figure S6.  $^{13}\text{C}$ -NMR of 2-(6-azidomethyl)-pyridine-4-carboxylic acid methylester (**3**) in  $\text{CDCl}_3$ .

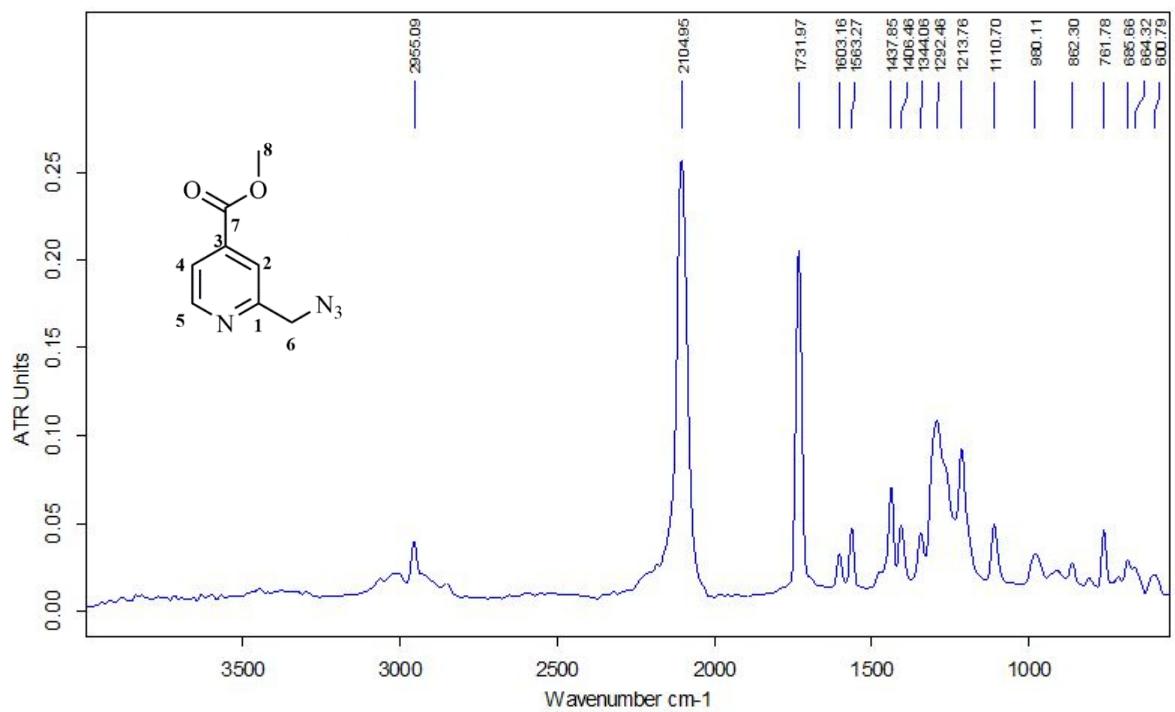


Figure S7. IR-spectrum of 2-(6-hydroxymethyl)-pyridine-4-carboxylic acid methylester (**3**).

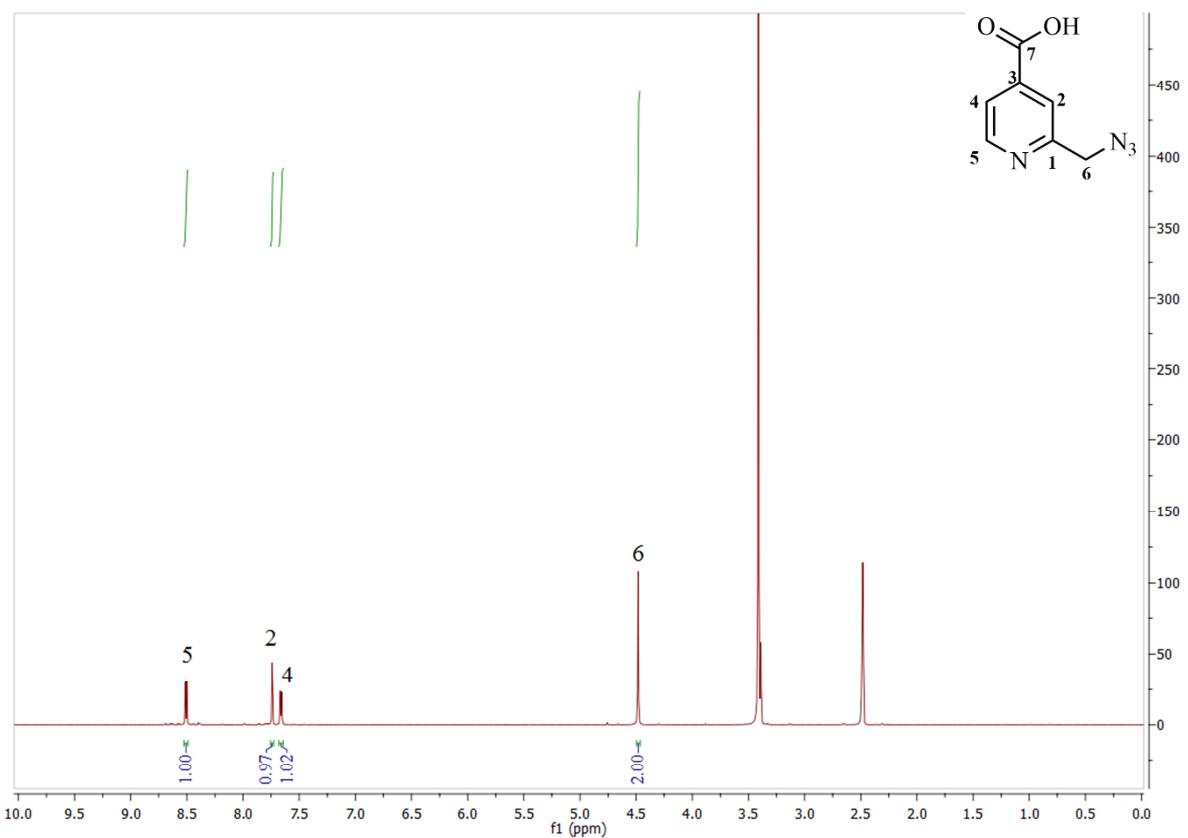


Figure S8. <sup>1</sup>H-NMR of 2-(6-azidomethyl)-pyridine-4-carboxylic acid (**4 = 1a**) in DMSO-d<sub>6</sub>.

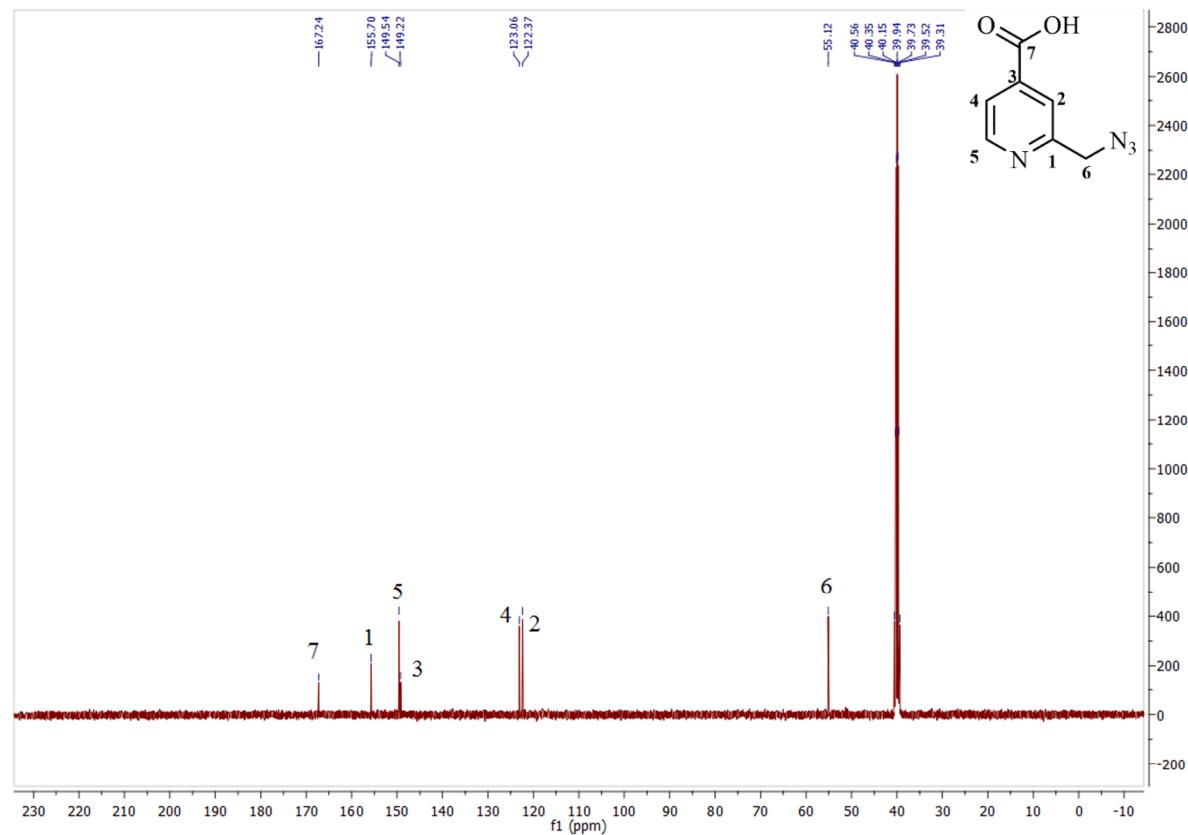


Figure S9. <sup>13</sup>C-NMR of 2-(6-azidomethyl)-pyridine-4-carboxylic acid (**4 = 1a**) in DMSO-d<sub>6</sub>.

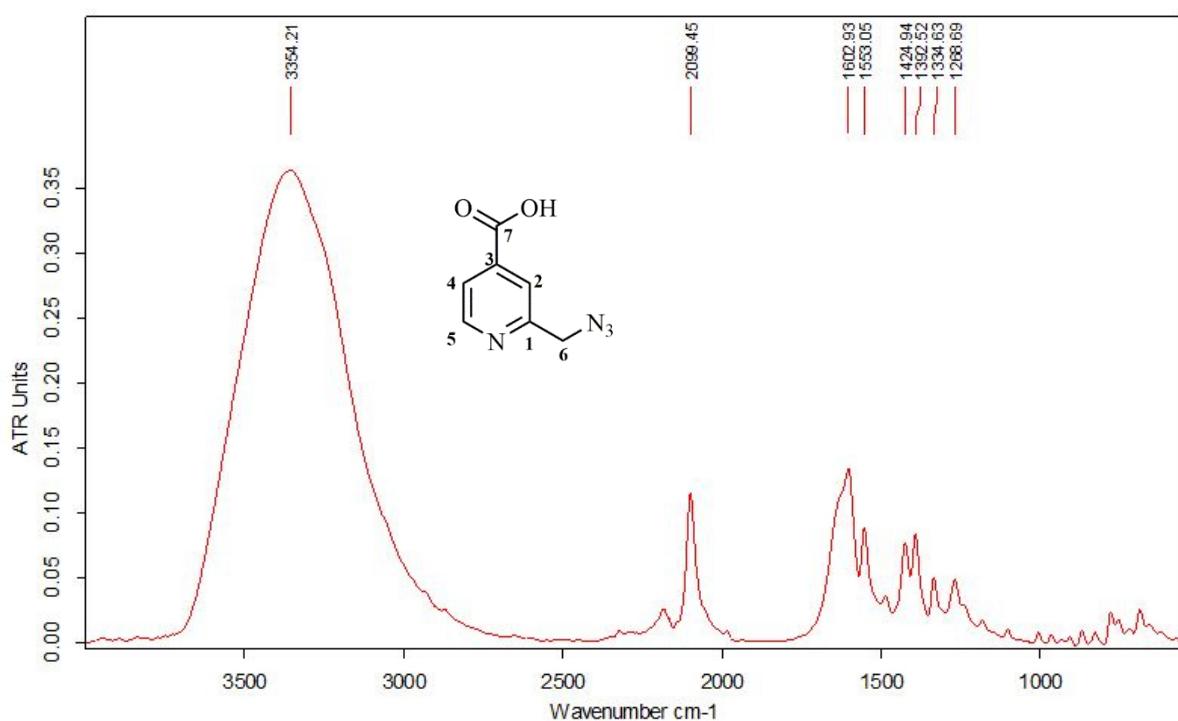


Figure S10. IR-spectrum of 2-(6-azidomethyl)-pyridine-4-carboxylic acid (**4 = 1a**).

Synthetic route to obtain 2-(6-azidomethyl)-pyridine-5-carboxylic acid (**7 = 1b**):

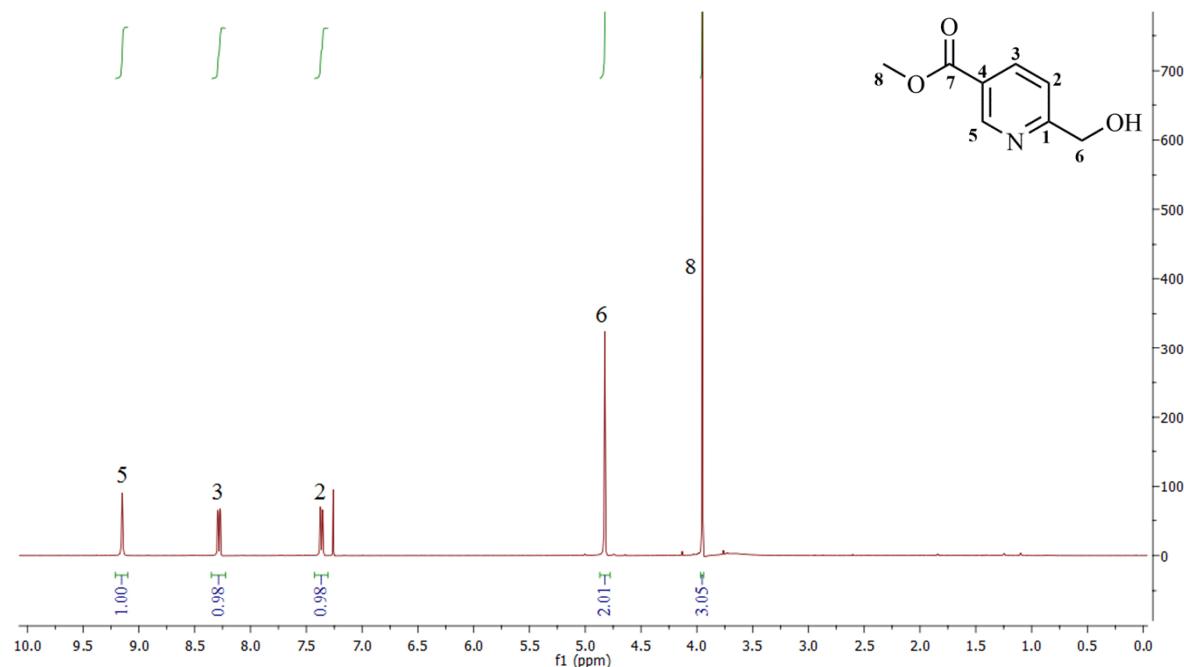
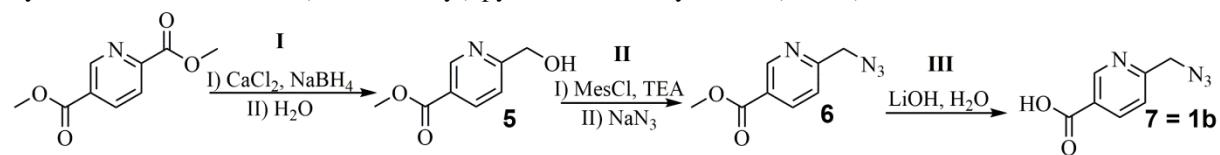


Figure S11.  $^1\text{H-NMR}$  of 2-(6-hydroxymethyl)-pyridine-5-carboxylic acid methylester (**5**) in  $\text{CDCl}_3$ .

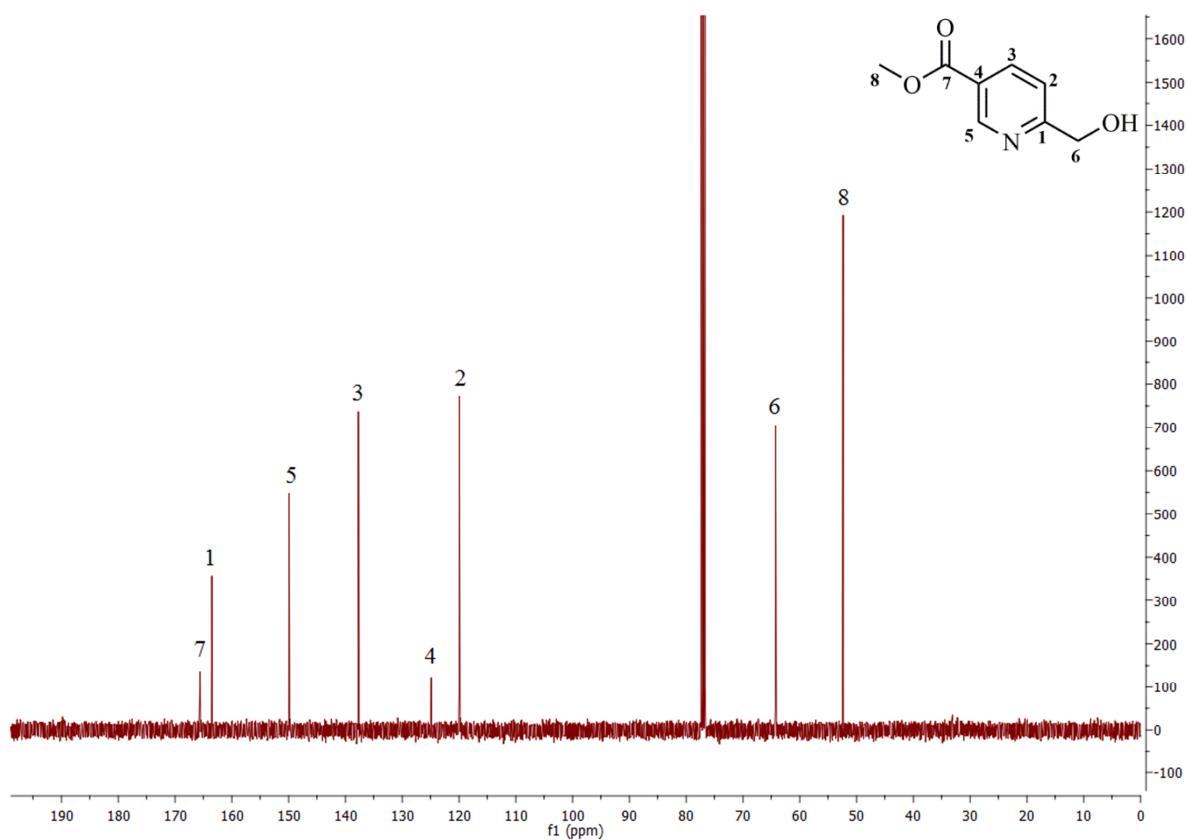


Figure S12.  $^{13}\text{C}$ -NMR of 2-(6-hydroxymethyl)-pyridine-5-carboxylic acid methylester (**5**) in  $\text{CDCl}_3$ .

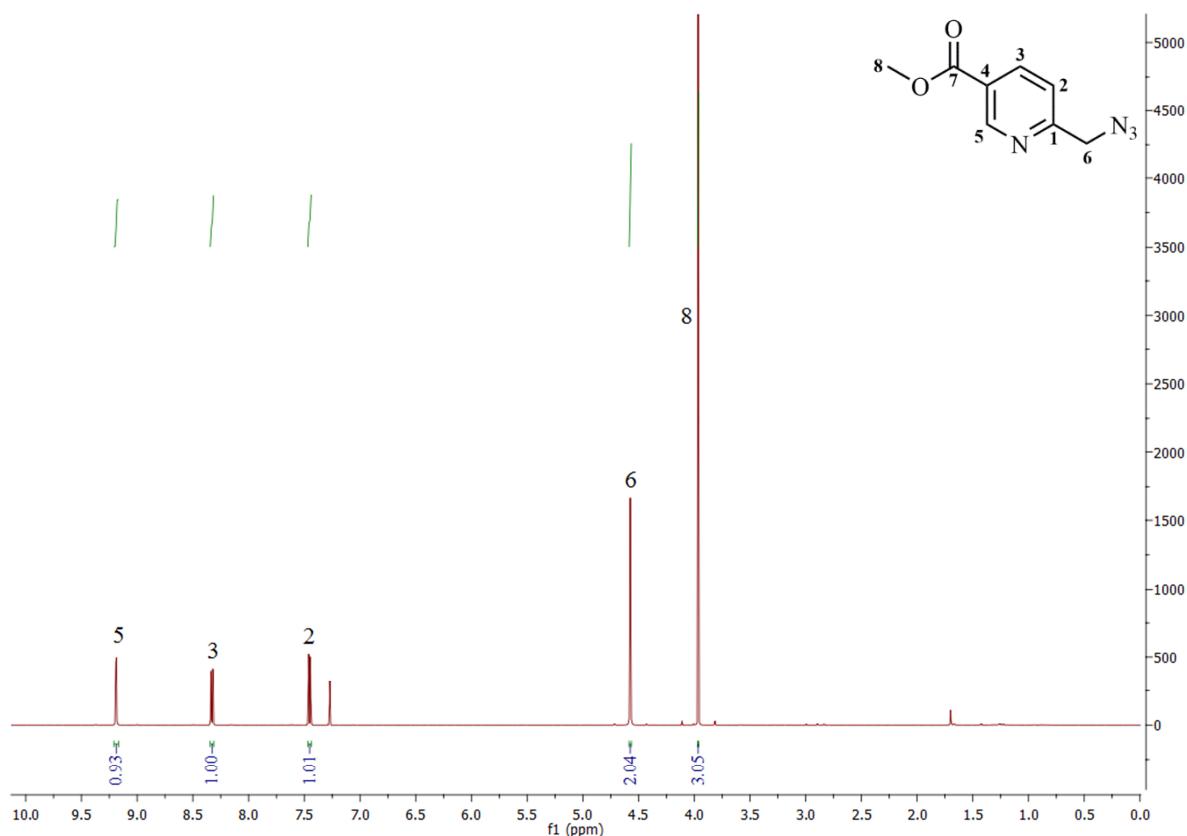


Figure S13.  $^1\text{H}$ -NMR of 2-(6-azidomethyl)-pyridine-5-carboxylic acid methylester (**6**) in  $\text{CDCl}_3$ .

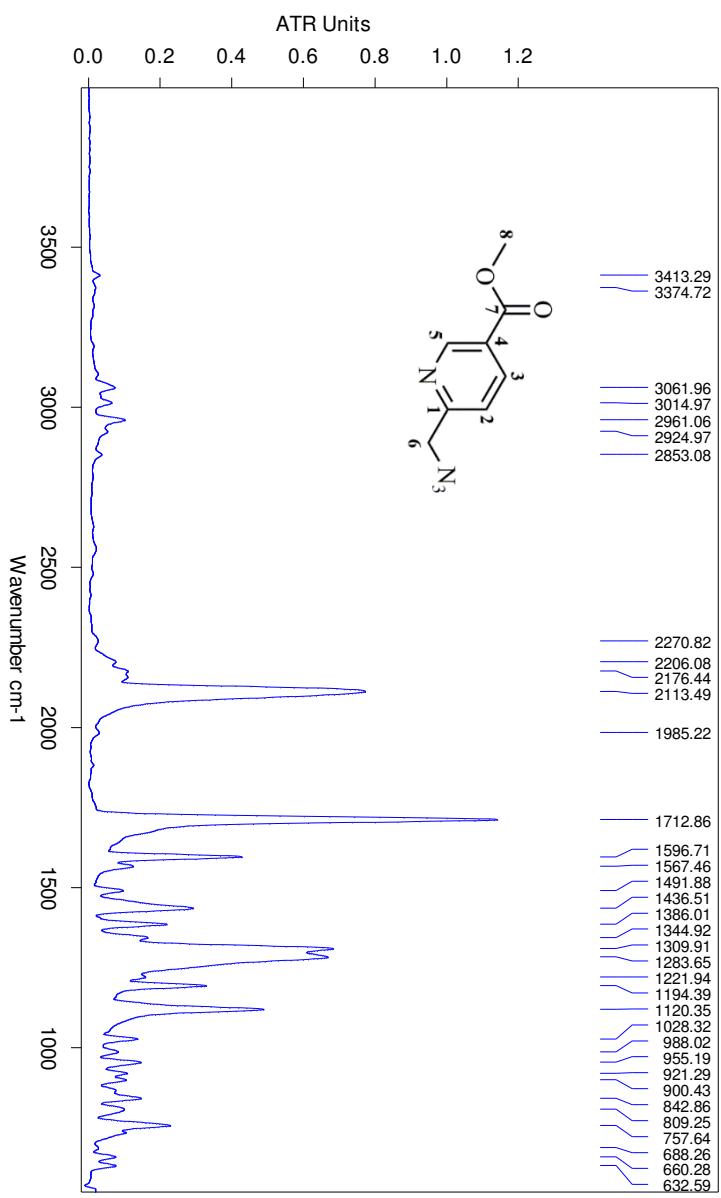


Figure S14.  $^{13}\text{C}$ -NMR of 2-(6-hydroxymethyl)-pyridine-5-carboxylic acid methylester (**6**) in  $\text{CDCl}_3$ .

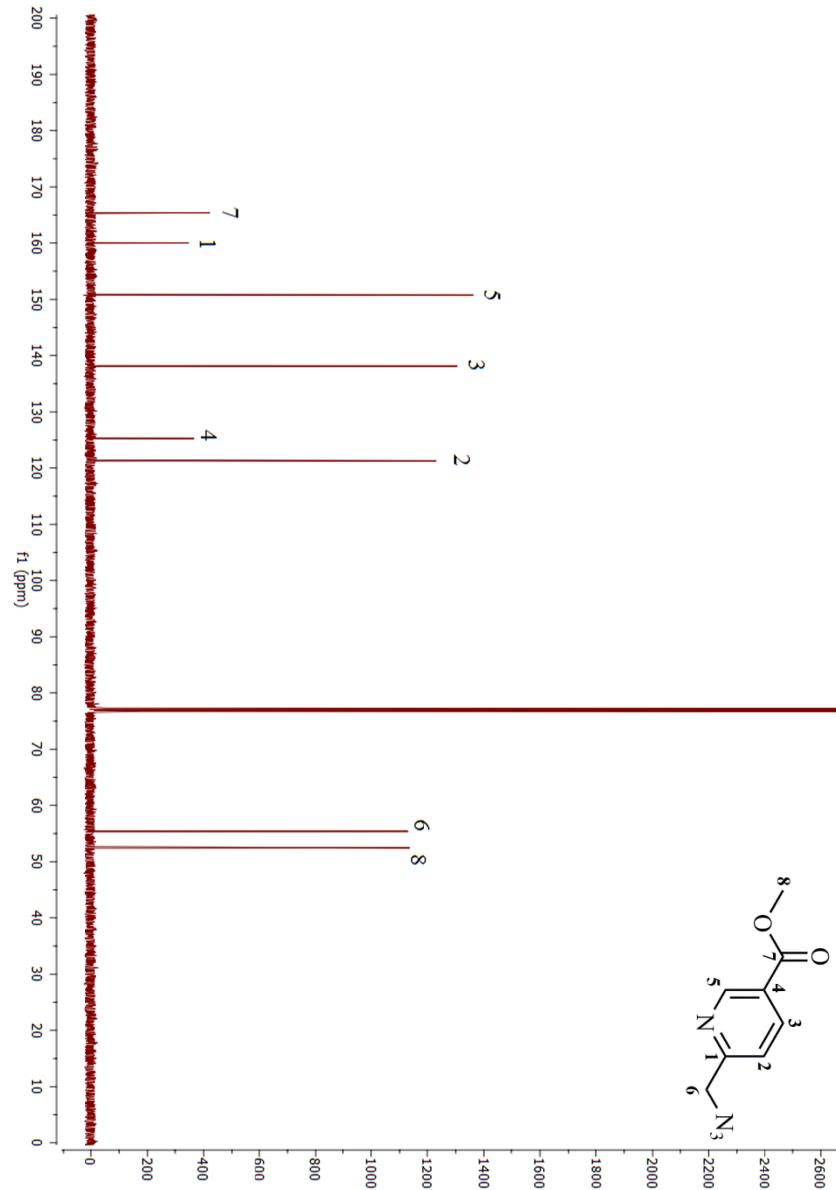


Figure S15. IR-spectrum of 2-(6-hydroxymethyl)-pyridine-4-carboxylic acid methylester (**6**).

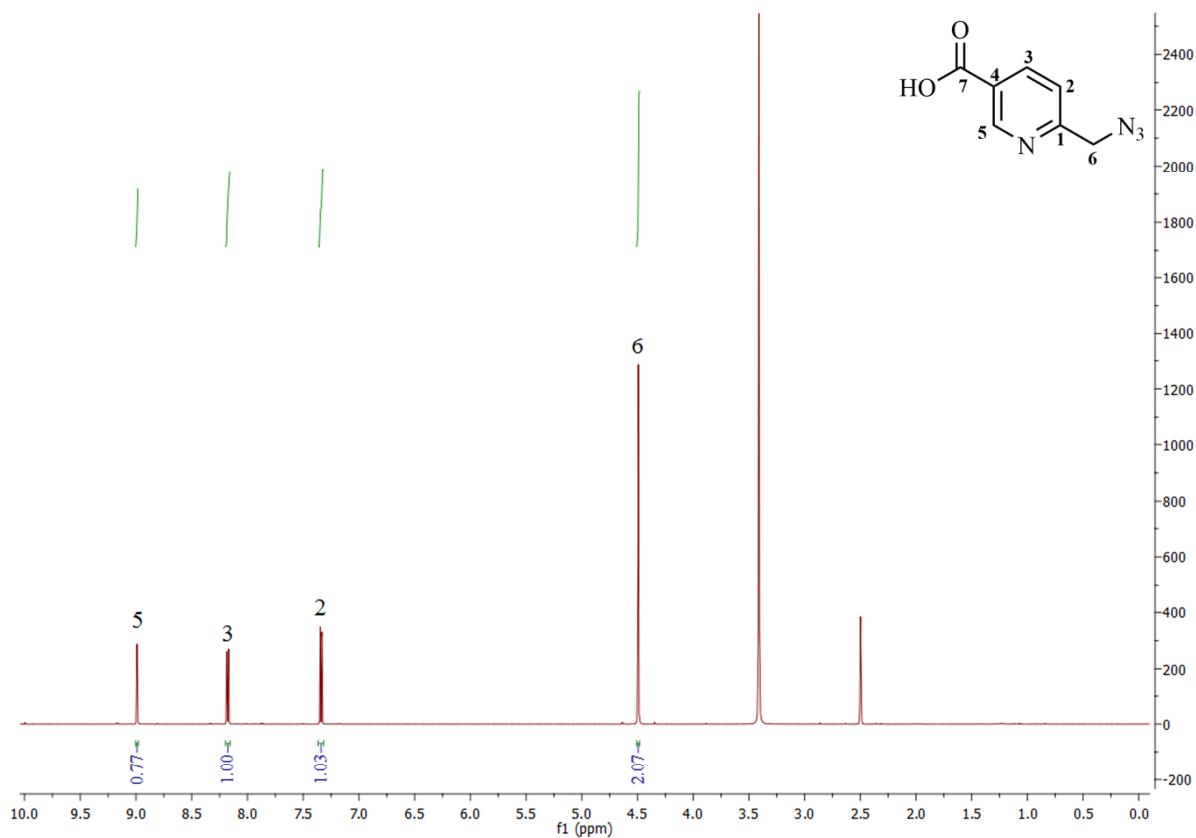


Figure S16.  $^1\text{H}$ -NMR of 2-(6-azidomethyl)-pyridine-5-carboxylic acid (**7 = 1b**) in  $\text{DMSO-d}_6$ .

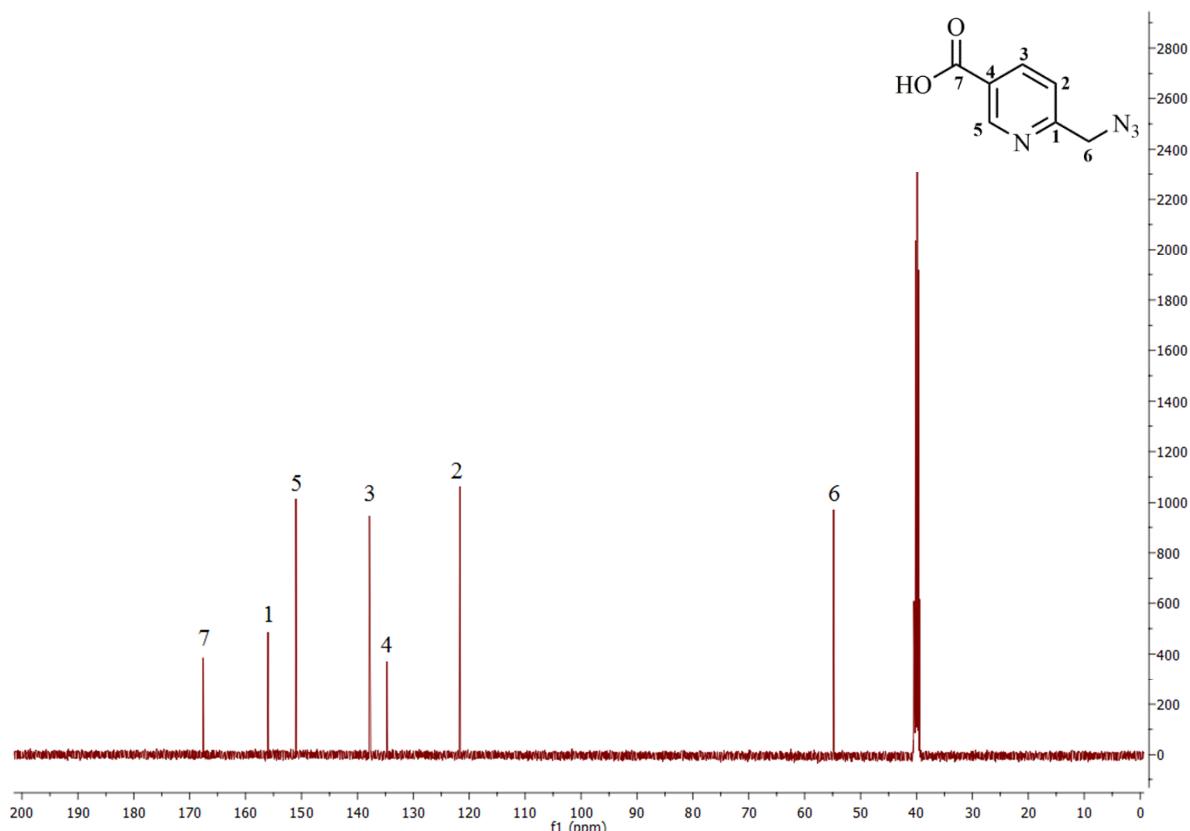


Figure S17.  $^{13}\text{C}$ -NMR of 2-(6-azidomethyl)-pyridine-5-carboxylic acid (**7 = 1b**) in  $\text{DMSO-d}_6$ .

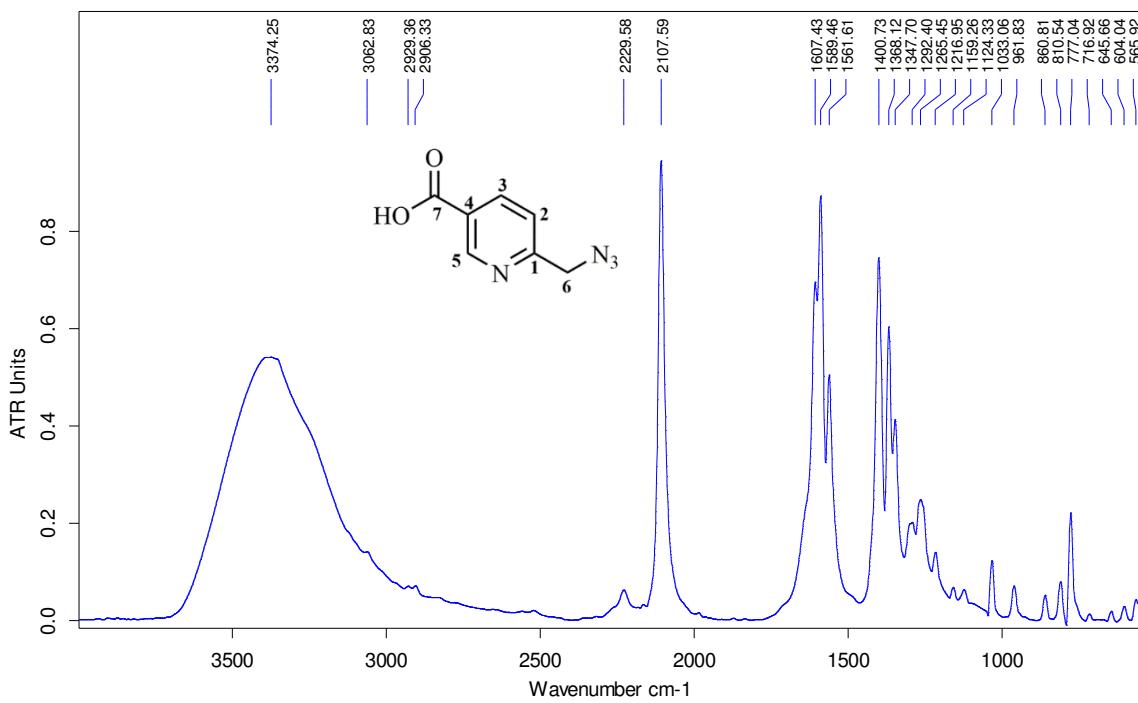


Figure S18. IR-spectrum of 2-(6-azidomethyl)-pyridine-5-carboxylic acid (**7** = **1b**).

2. NMR- and IR-investigations for the synthesis of star-shaped picolinazido-telechelic PIBs (**2a**, **2b**).

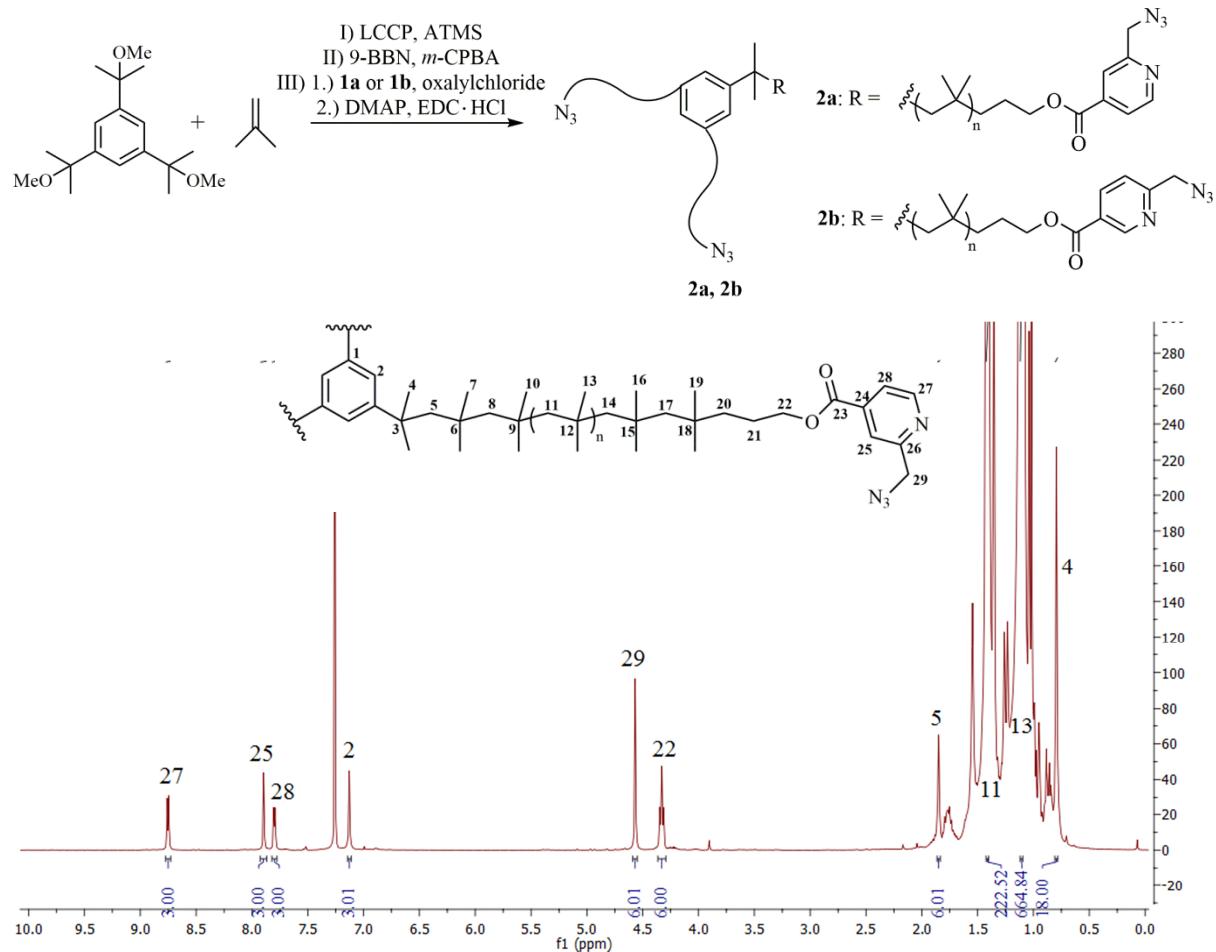


Figure S19. <sup>1</sup>H-NMR of star-shaped picolinazido-telechelic PIB (**2a**).

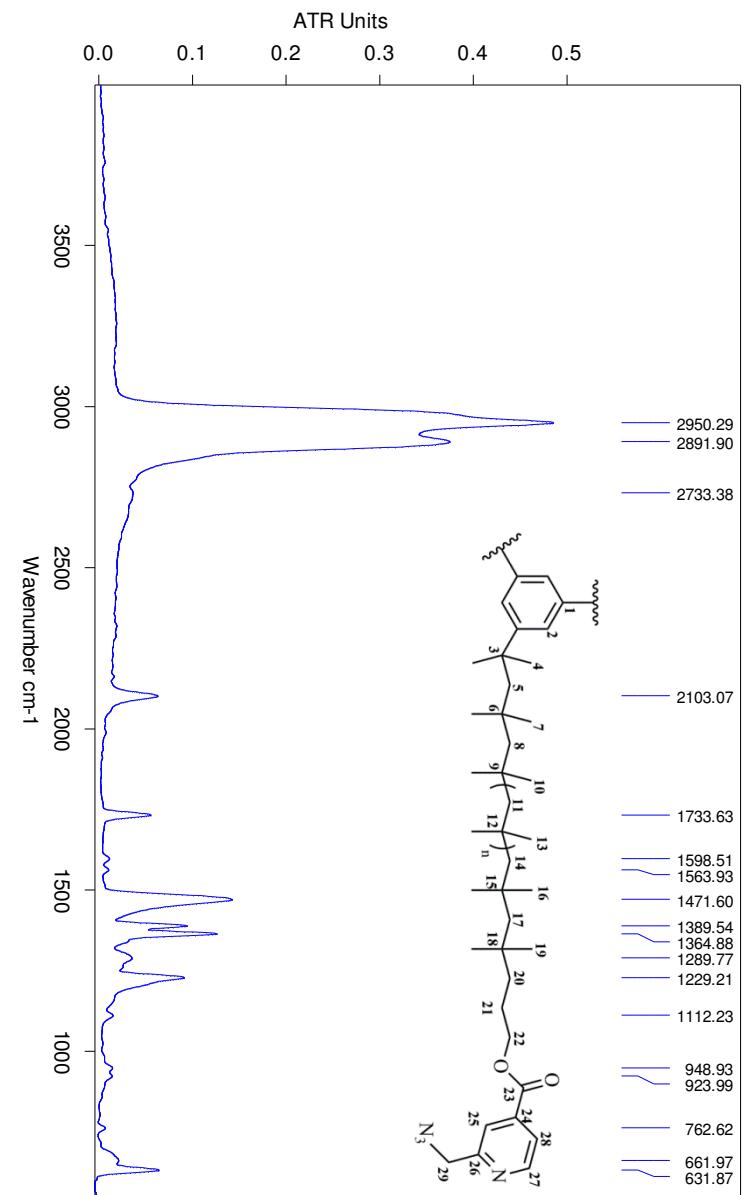


Figure S20. IR-spectrum of star-shaped picolinazido-telechelic PIB (**2a**).

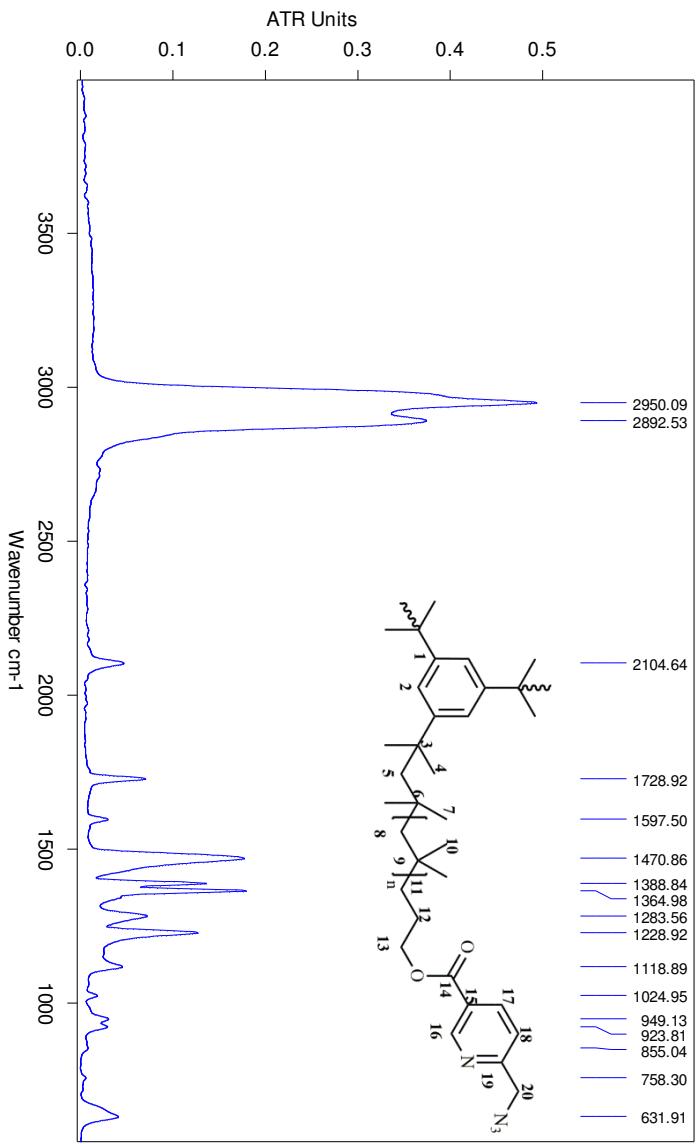


Figure S21. IR-spectrum of star-shaped picolinazido-telechelic PIB (**2b**).

3. *In Situ* NMR kinetic plot by conversion of 2-(6-azidomethyl)-pyridine-4-carboxylic acid methyl ester and phenylacetylene: - here shown exemplary for Table 1, entry 7: CuBr, DIPEA (0.01 eq.):

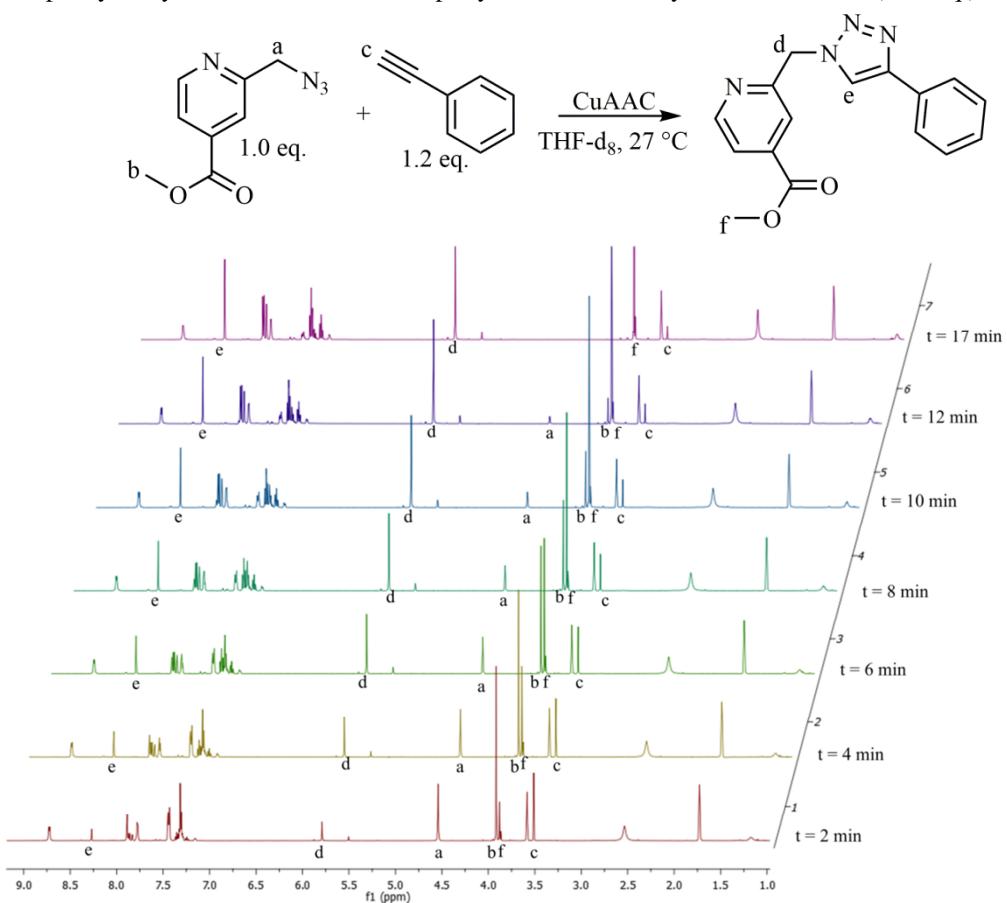


Figure S22. <sup>1</sup>H-NMR kinetic-plot (Table 1, entry 7) conducted in THF-d<sub>8</sub>.

To determine the conversion at different defined periods of time the integrals of the corresponding resonances (a) and (d) were chosen. Calculation was done according to the following equation (S1):

$$\text{conversion} = \frac{\text{integral of resonance d}}{(\text{integral of resonance d} + \text{integral of resonance a})} * 100 \% \quad (\text{S1})$$

When resonance (a) disappears the conversion to the final click-product was seen as complete (> 99 %).

4. IR investigation of cross-linked networks after rheological investigations:  
Here shown exemplary for Table 2, entry 5 (**2a + 3**, 20 °C, CuF(PPh<sub>3</sub>)<sub>3</sub>).

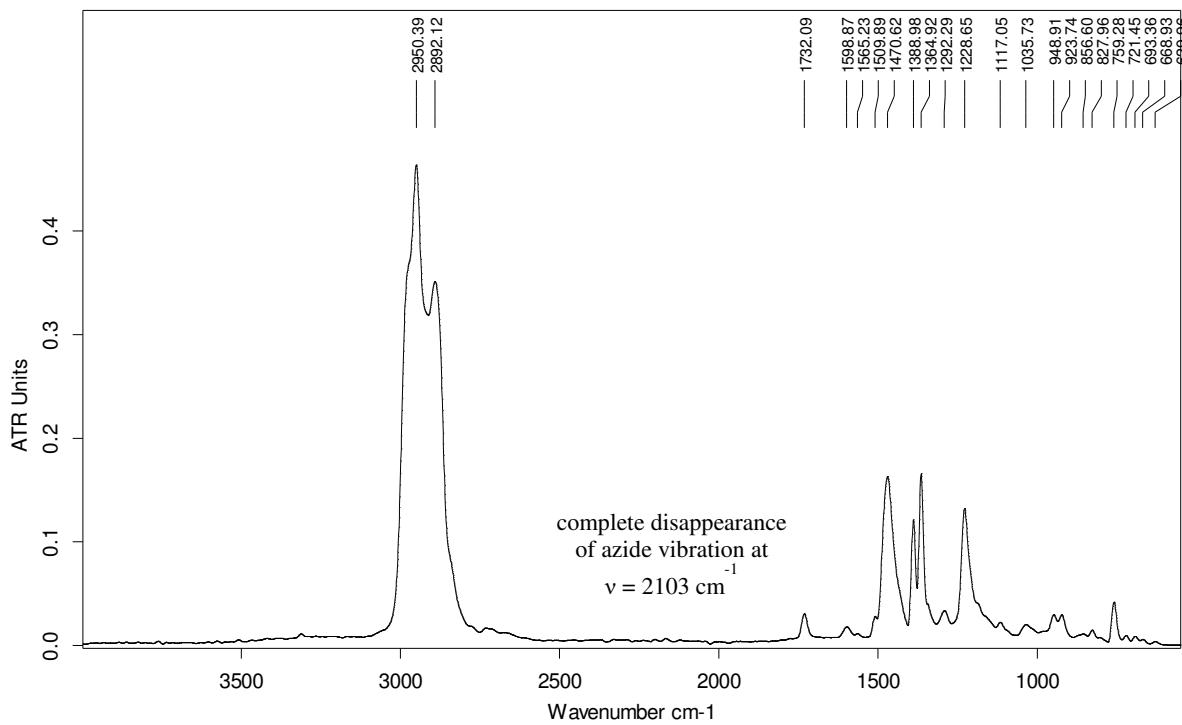


Figure S23. IR-spectrum of polymer network obtained by cross-linking of **2a + 3**.

Here shown exemplary for Table 2, entry 7 (**2b + 3**, 10 °C, CuF(PPh<sub>3</sub>)<sub>3</sub>).

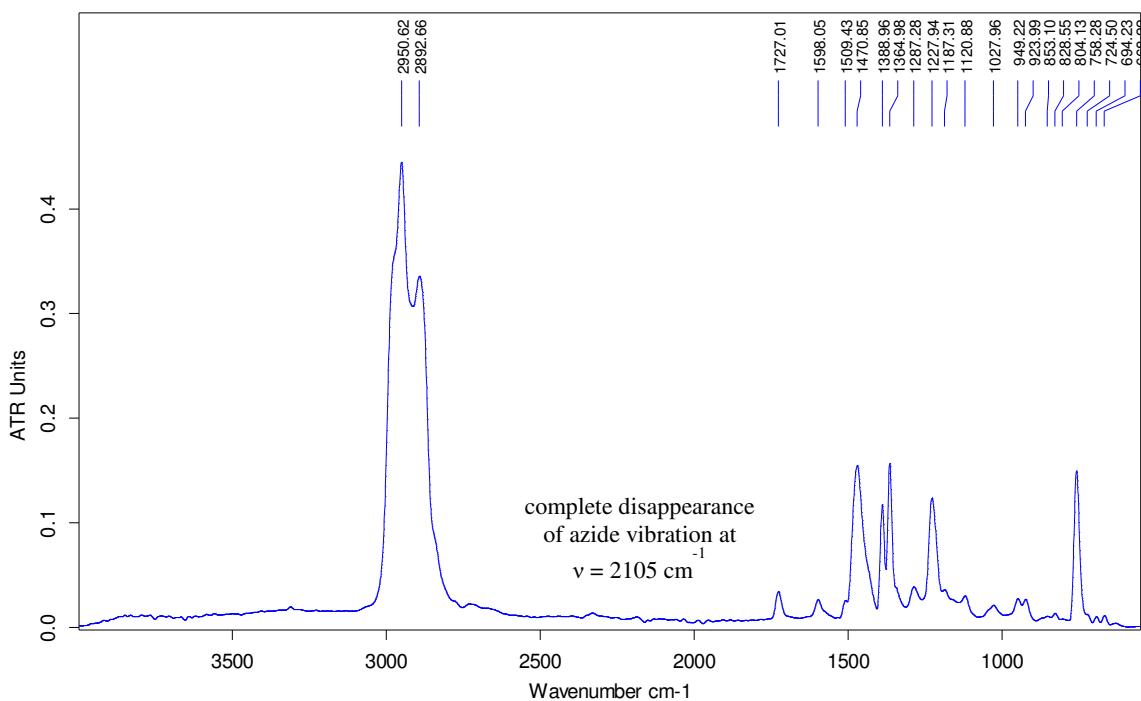


Figure S24. IR-spectrum of polymer network obtained by cross-linking of **2b + 3**.

5. Fluorescence measurements of reference samples.

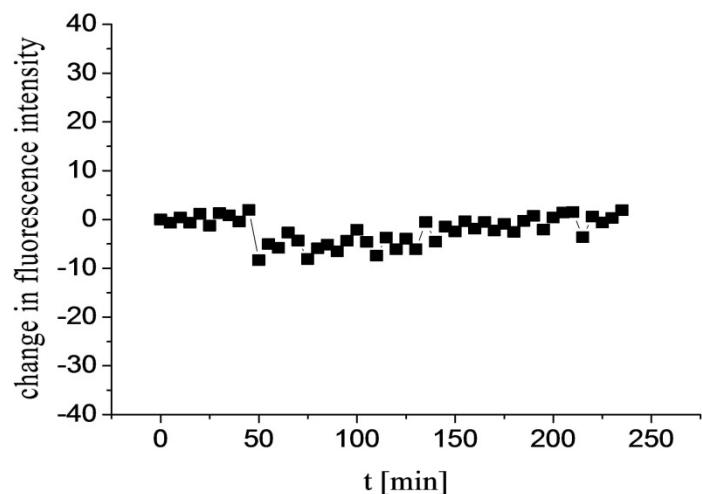


Figure S25. Fluorescence measurement of unscratched specimen.

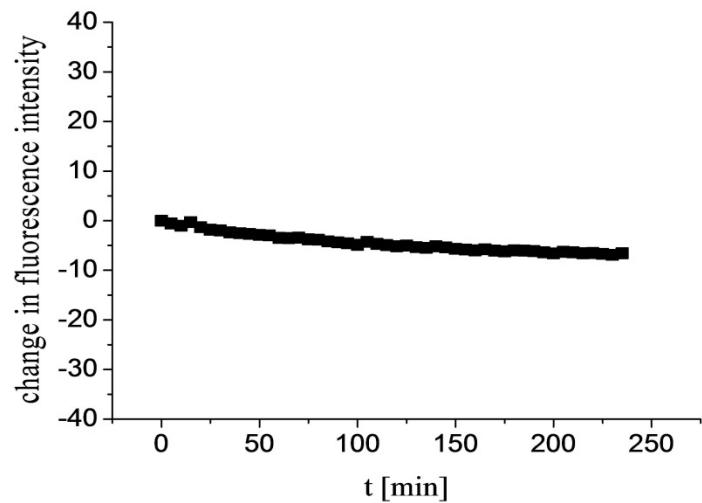


Figure S26. Fluorescence measurement of scratched specimen without  $\text{CuBr}(\text{PPh}_3)_3$ .