

Supporting information:

Ruthenium complexes with an N-heterocyclic carbene NNC-pincer ligand: preparation and catalytic properties

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1. NMR Spectra

1.1. ¹H and ¹³C NMR spectra of **1**

1.2. ¹H and ¹³C NMR spectra of **2**

1.2. ¹H and ¹³C NMR spectra of **3**

1.2. ¹H and ¹³C NMR spectra of **4**

2. X-Ray Crystallography

2.1 Crystal data for **2** and **4**

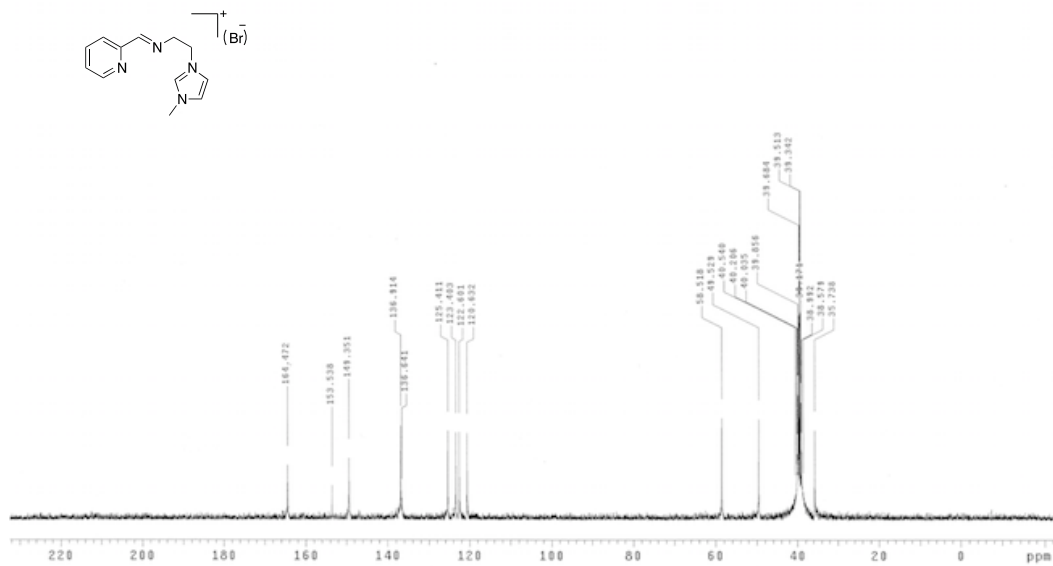
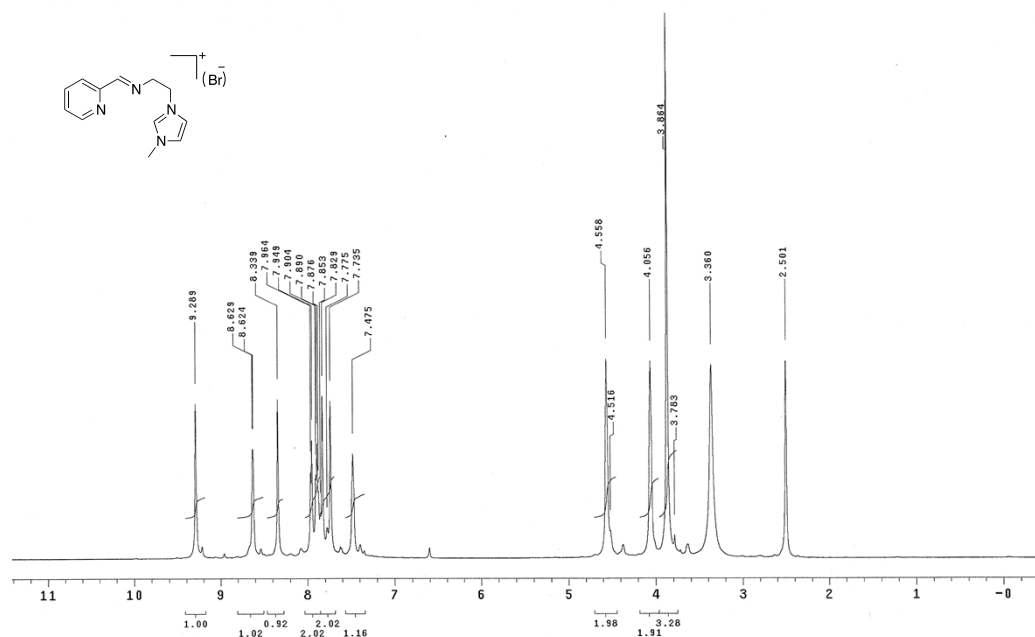
3. Catalytic studies:

3.1 General Procedure for the reduction of of ketones and aldehydes by transfer hydrogenation

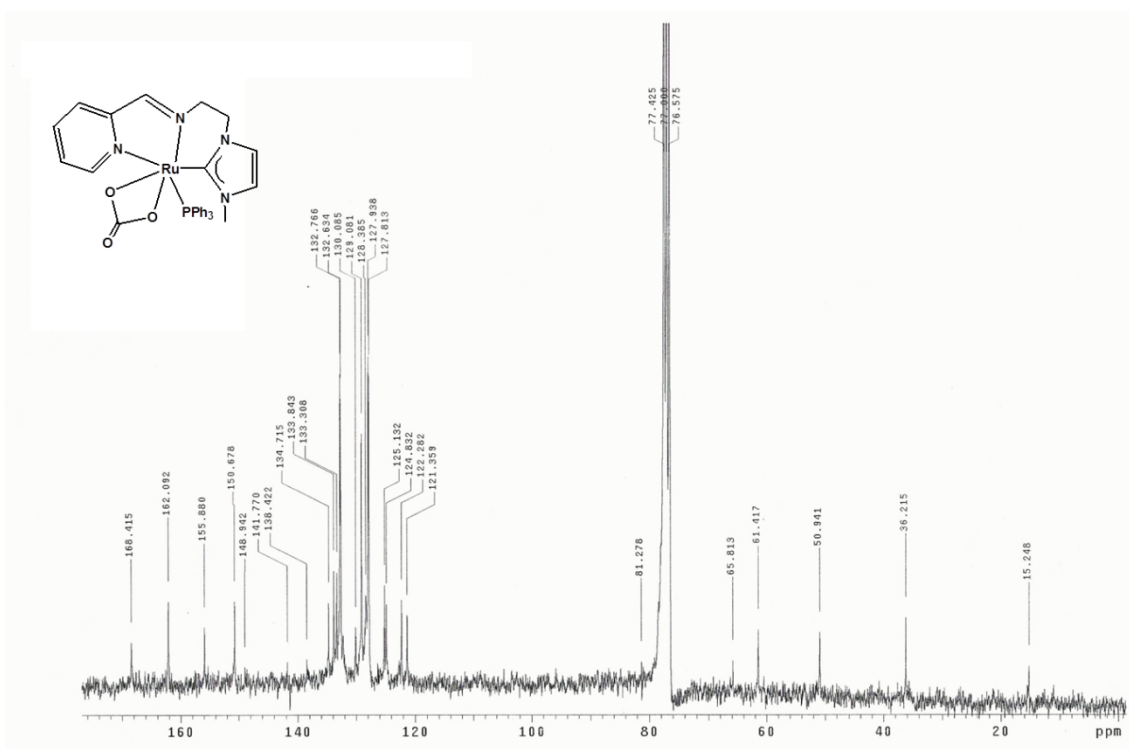
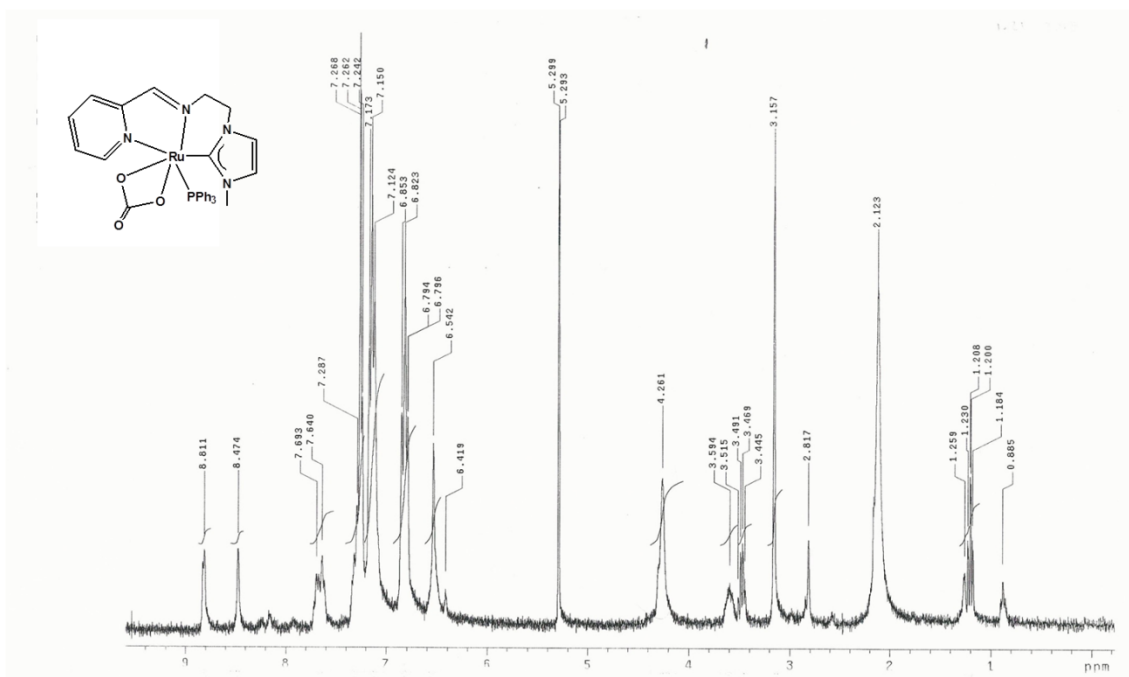
4. Bibliography

1. Spectra

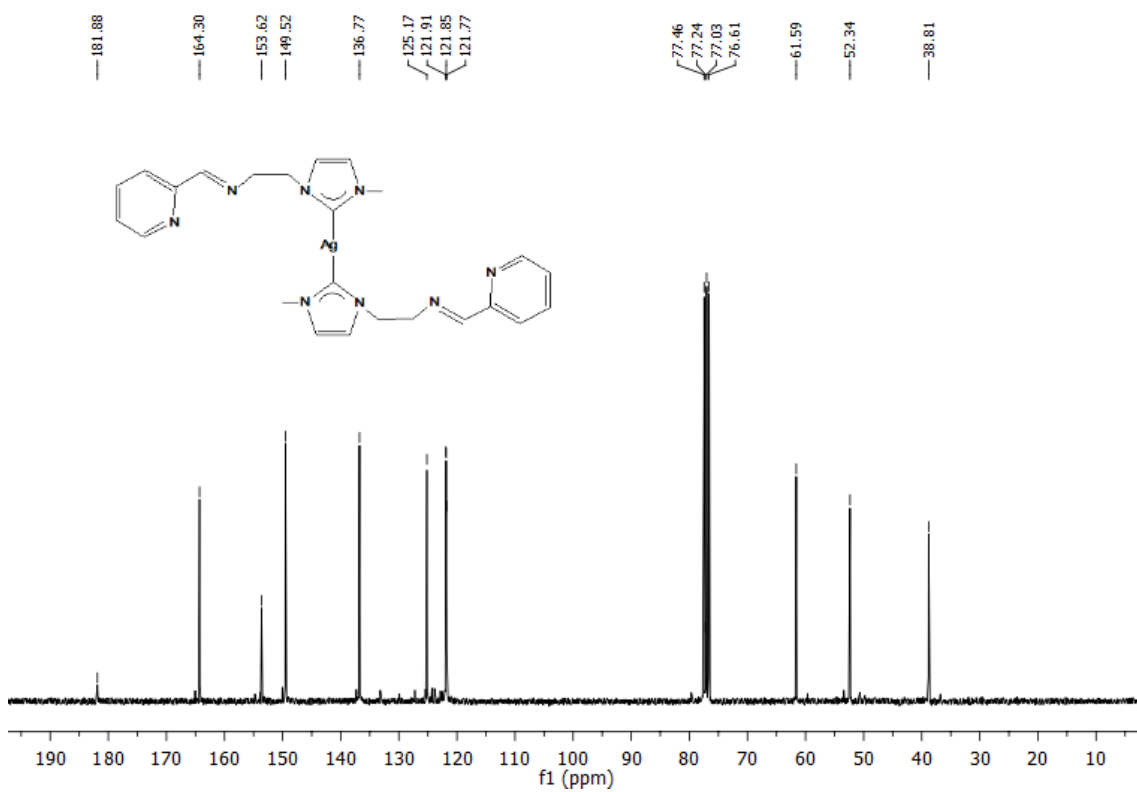
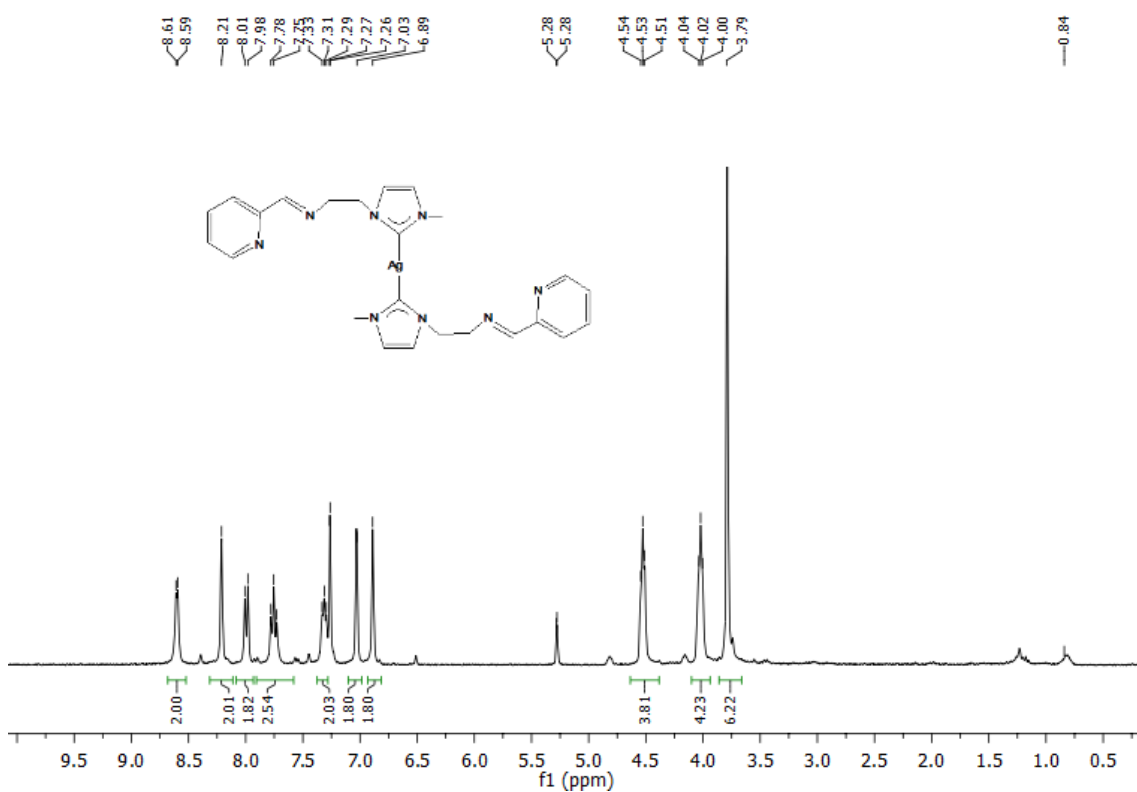
1.1. ¹H and ¹³C NMR spectra of 1



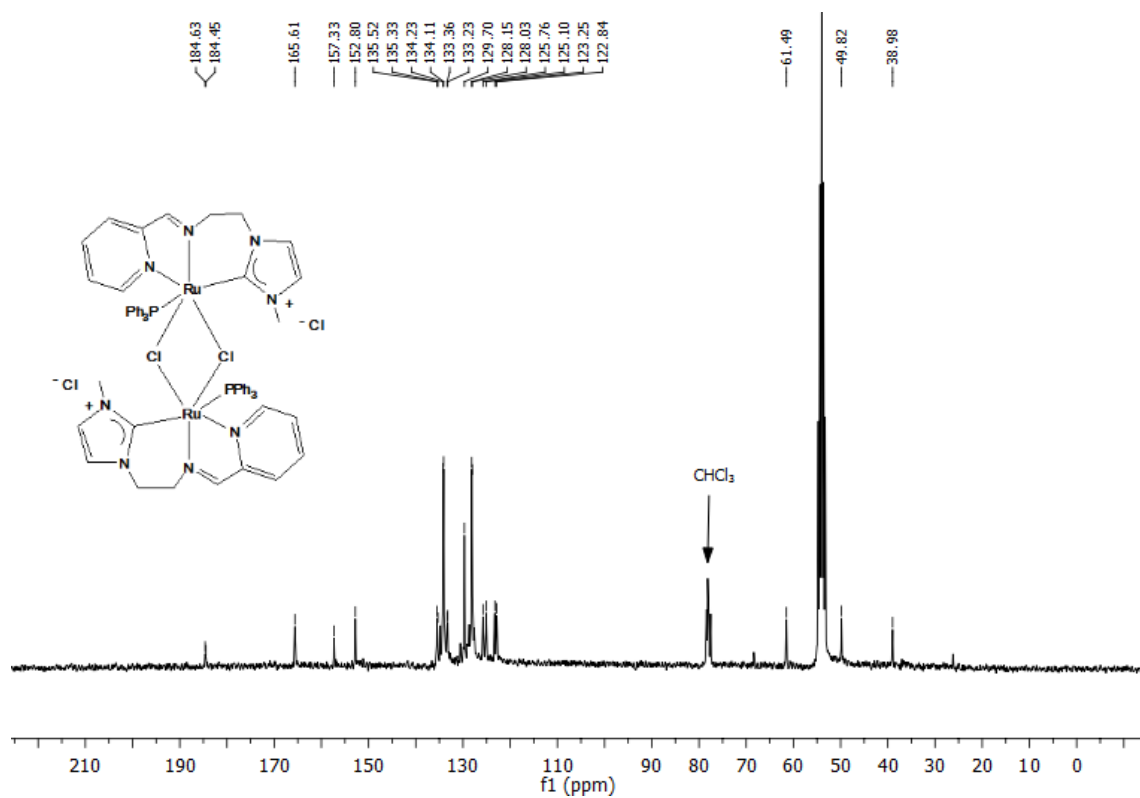
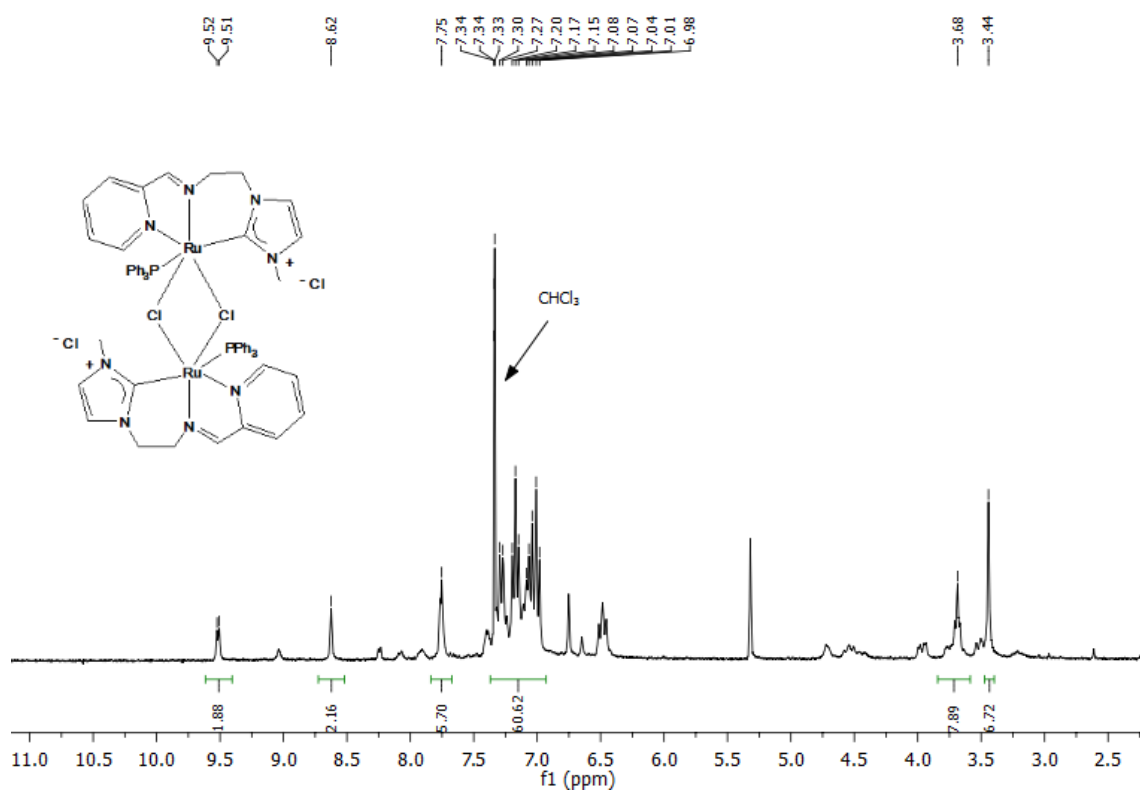
1.2. ^1H and ^{13}C NMR spectra of 2



1.3. ^1H and ^{13}C NMR spectra of **3**



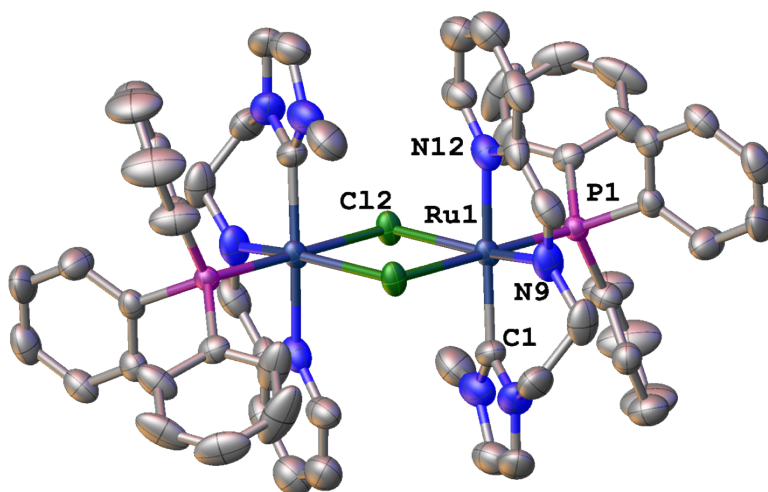
1.4. ^1H and ^{13}C NMR spectra of **4**



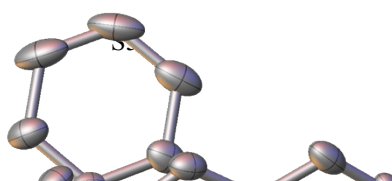
2. X-Ray Crystallography.

Single crystals of **2** and **4** suitable for X-ray crystallographic analysis were obtained as described above. Diffraction data were collected on a Agilent SuperNova diffractometer equipped with an Altas CCD detector using Cu K α radiation ($\lambda = 1.54184 \text{ \AA}$). Single mounted on a MicroMount polymer tip (MiteGen) in a random orientation. The crystals were kept at 293 K during data collection for **2** and at 290 K for **4**. The structures were solved by direct methods in SHELXS-97⁴ and refined by the full-matrix method based on F^2 with the program SHELXL-97 using the OLEX software package.¹⁻² Further crystallographic data may be found in the corresponding CIF files which were deposited at the Cambridge Crystallographic Data Centre CCDC, Cambridge, UK. The reference number for **2** and **4** were assigned as 1060997, 1060998 respectively.

2.1 Crystal data, data collection and structure refinement details for **2** and **4**:



Compound **2**: $C_{70}H_{66}Cl_{34}N_8P_2Ru_2$ ($M = 2488.68$): monoclinic, space group $P2_1/a$, $a = 13.2500(2) \text{ \AA}$, $b = 16.9695(3) \text{ \AA}$, $c = 23.3436(4) \text{ \AA}$, $\alpha = 90^\circ$, $\beta = 103.9482(18)^\circ$, $\gamma = 90^\circ$, $V = 5093.95(17) \text{ \AA}^3$, $Z = 2$, $T = 293(2) \text{ K}$, $\mu(\text{CuK}\alpha) = 11.254 \text{ mm}^{-1}$, $D_{\text{calc}} = 1.623 \text{ g/cm}^3$, 44887 reflections measured ($7.38 \leq 2\theta \leq 131.968$), 8883 unique ($R_{\text{int}} = 0.0512$, $R_{\text{sigma}} = 0.0321$), which were used in all calculations. The final R_1 was 0.0721 ($I \geq 2\sigma(I)$) and wR_2 was 0.2154 (all data).



Compound 4: C₃₂H₃₅Cl₂N₄O₅PRu (*M*=758.58): monoclinic, space group P2₁/c, *a* = 18.2792(3)Å, *b* = 11.74106(11) Å, *c* = 17.1049(2)Å, *V* = 3259.49(9) Å³, *Z* = 4, *T* = 199.95(10) K, μ(CuKα) = 1.54184 mm⁻¹, *D*_{calc} = 1.546 g/mm³, 27673 reflections measured (9.296 ≤ 2Θ ≤ 139.32), 6075 unique (*R*_{int} = 0.0499, *R*_{sigma} = 0.0291) which were used in all calculations. The final *R*₁ was 0.2354 (*I* > 2σ(*I*)) and *wR*₂ was 0.5147 (all data).

3. Catalytic studies: transfer hydrogenation.

3.1. General Procedure for the reduction of ketones and aldehydes by transfer hydrogenation.

In a 50 mL high-pressure Schlenk tube, ketone (1 mmol), catalyst (0.5 mol %), Base (0.1 mmol) and *i*PrOH (3 mL) were placed. The mixture was stirred and heated to 85 °C for 1h. The reaction yields were calculated by GC or ¹H NMR using Anisole as internal standard.

4. Bibliography

1. Dolomanov, O.V., Bourhis, L.J., Gildea, R.J, Howard, J.A.K. & Puschmann, H. (2009), *J. Appl. Cryst.* 42, 339-341.
2. Sheldrick, G.M. (2008). *Acta Cryst.* A64, 112-122.