

# On the Properties of $\text{Au}_2\text{P}_3^z$ ( $z = -1, 0, +1$ ): Analysis on Geometry, Interaction, and Electron Density

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## Electronic Supplementary Information

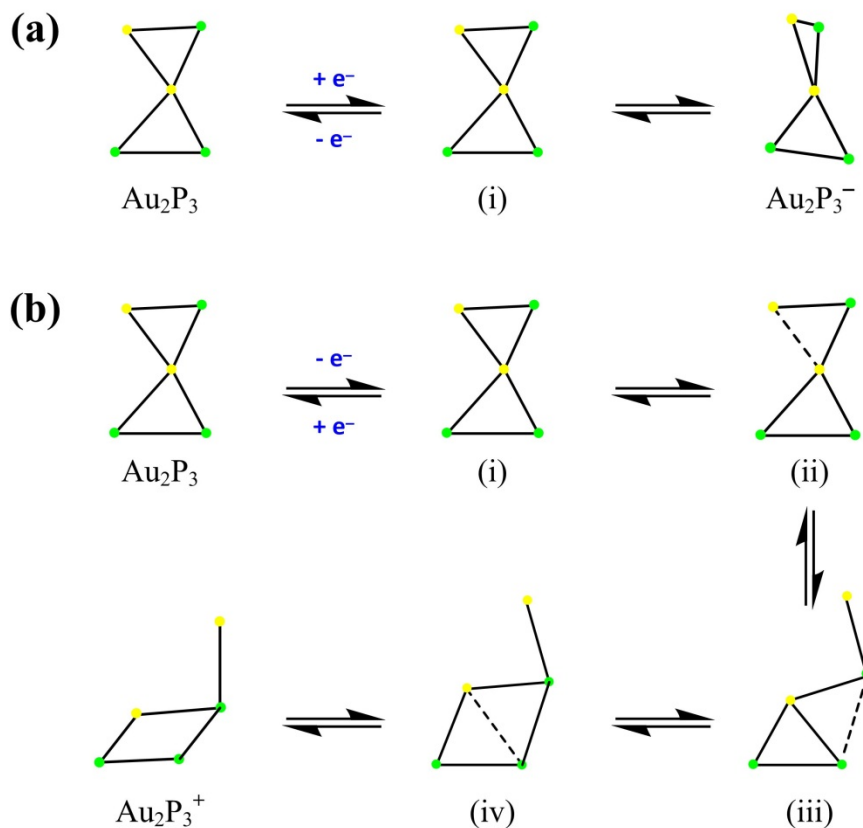
**Table S1** The Au-Au bond length at different value states of Au<sub>2</sub> (theo. & expt.) and Au<sub>2</sub>P<sub>3</sub> (theo.) clusters, respectively. All bond lengths are in Å.

|    | Au-Au (Au <sub>2</sub> P <sub>3</sub> ) | Au-Au (Au <sub>2</sub> ) |                    |       |
|----|-----------------------------------------|--------------------------|--------------------|-------|
|    | Theo.                                   | Theo.                    | Expt. <sup>b</sup> |       |
| +1 | 2.860                                   | 2.653                    | 2.639 <sup>a</sup> | --    |
| 0  | 2.742                                   | 2.536                    | 2.512 <sup>a</sup> | 2.470 |
| -1 | 2.922                                   | 2.660                    | 2.632 <sup>a</sup> | 2.582 |

<sup>a</sup> Results calculated in CCSD(T) level of theory, Ref. [1].

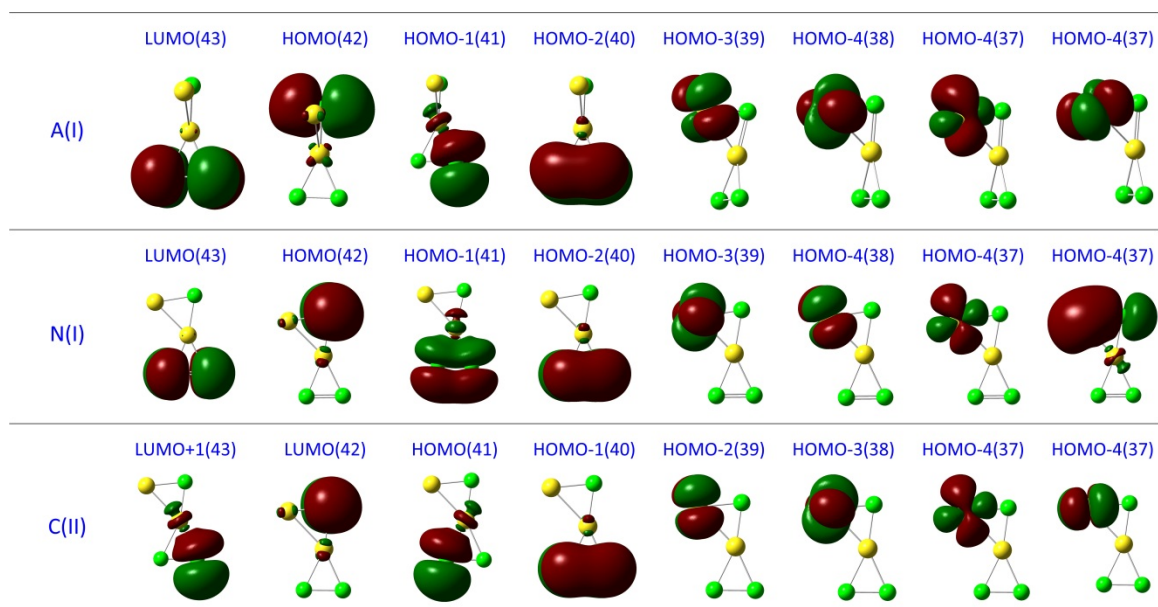
<sup>b</sup> Ref. [2].

**Fig. S1 Conjectural interchange between N(I) and A(I) (a), and between N(I) and C(I) (b), respectively. Illustrated to proceed via C(II) (2b.(i))<sup>a</sup>**



<sup>a</sup> Structures (i), (ii), (iii), and (iv) represent incrementally changed geometries along a transformation pathway.

**Fig. S2 The frontier molecular orbitals of  $\text{Au}_2\text{P}_3^z$  ( $z = -1, 0, 1$ ).**



[1] R. Wesendrup, T. Hunt, and P. Schwerdtfeger, J. Chem. Phys. 112, 9356 (2000).

[2] J. Ho, K. Ervin, and W. Lineberger, J. Chem. Phys. 93, 6987 (1990).