Electronic Supplementary Information (ESI) for RSC Advances

## Nanostructuring of GNS-V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> Core/Shell Photocatalyst for Water Remediation Applications under Sun-light Irradiation

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## 1. Photocatalytic degradation of Methylene blue (MB) dye



Fig. S1. Photocatalytic degradation of MB using (a) V<sub>2</sub>O<sub>5</sub>, (b) V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub>, (c) GNS-V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub>, (d) Initial/residual concentration plot vs time and (e) degradation % of MB vs time.

The UV-Vis-absorption spectra (in Fig. S1 (a-c)) showed the MB degradation plot versus time with  $V_2O_5$ ,  $V_2O_5/TiO_2$ , GNS- $V_2O_5/TiO_2$  nanoarchitecture photocatalysts, respectively. C/C<sub>0</sub> spectra indicate that the  $2.5 \times 10^{-5}$  M concentration of MB dye was decomposed of about 37% for pure  $V_2O_5$  nanorods within 60 minutes, 83% for  $V_2O_5/TiO_2$  core/shell nanorods within 40

minutes and 93% for GNS-V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> nanoarchitectures within 20 minutes (in Fig. S1 (d and e)) under direct sunlight irradiation. GNS-V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> nanoarchitectures exhibited higher photocatalytic efficiency than that of the pure and core/shell nanomaterials. This result demonstrates that GNS enhances the photocatalytic activity of V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> core/shell nanorods under direct sunlight irradiation.

## 2. Nitrogen adsorption-desorption isotherm



Fig. S2. Nitrogen adsorption – desorption isotherm of  $V_2O_5$ ,  $V_2O_5/TiO_2$  and  $GNS-V_2O_5/TiO_2$ nanoarchitecture photocatalysts

To explore the specific surface area of V<sub>2</sub>O<sub>5</sub>, V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> and GNS-V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> nanoarchitecture photocatalysts, nitrogen adsorption-desorption investigations have been carried out. The BET surface area characteristic of these nanoarchitecture are shown in Fig. S2. Both adsorption and desorption curves demonstrate type IV curve characteristics. The surface area of GNS-V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> is determined to be 103.5 m<sup>2</sup>/g by fitting the isotherms to the BET model. This value is significantly higher than that of V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> (74.8 m<sup>2</sup>/g) indicating that graphene nanosheets have a better structure suitability due to their two dimensional nanostructures. Therefore, there are higher contact area between GNS and V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> core/shell nanostructure. These results

evidently highlight that the  $V_2O_5/TiO_2$  core/shell nanostructures are homogeneously anchored on GNS layers.



## 3. HRSTEM fringes of V<sub>2</sub>O<sub>5</sub>/TiO<sub>2</sub> core/shell nanostructures

Fig. S3. HRSTEM lattice fringes of V2O5/TiO2 core/shell nanostructures

The HRSTEM fringes pattern of  $V_2O_5/TiO_2$  core/shell nanorods for better understanding. The  $V_2O_5/TiO_2$  core-shell nanorod interface exhibits a strong alignment of the two different crystal lattices, resulting in the electrostatic interaction of TiO<sub>2</sub> on  $V_2O_5$  nanorods to form a core/shell like structure. The measured lattice distance of 0.43 nm corresponds to the (001) plane of orthorhombic  $V_2O_5$  nanorods. We have also observed the lattice fringes of 0.35 nm, which correspond to the interplanar distance of (101) plane of the anatase TiO<sub>2</sub> (in fig. S3 (a)). The SAED pattern reveals the diffraction rings are polycrystalline nature (in fig. S3 (b)).