

**Electronic Supplementary Information**

**Ba<sub>2</sub>GeS<sub>4</sub> and Mg<sub>2</sub>SnS<sub>4</sub>: synthesis, structures, optical properties and electronic structures**

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Figure S1. (a) EDX spectrum of Ba<sub>2</sub>GeS<sub>4</sub>.

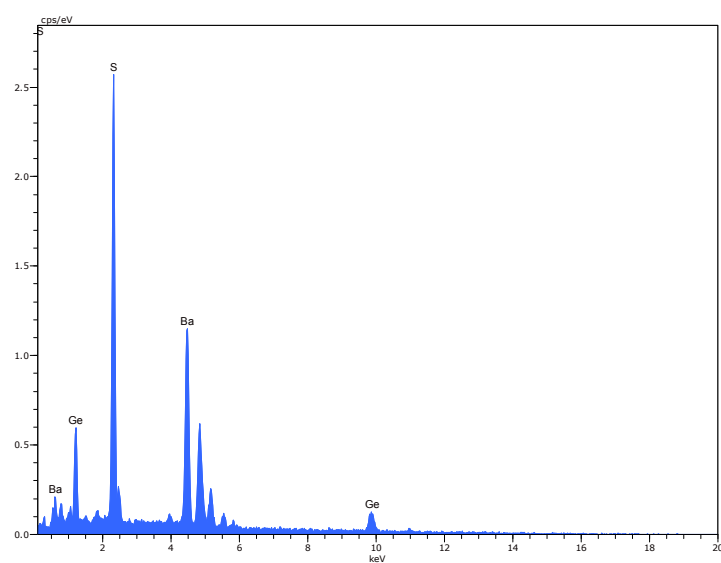


Table S1 (a) Average EDX data for Ba<sub>2</sub>GeS<sub>4</sub>.

| Element | Weight % | Atomic % | Formula |
|---------|----------|----------|---------|
| S       | 25.72    | 55.76    | 3.96    |
| Ge      | 14.70    | 14.08    | 1       |
| Ba      | 59.57    | 30.16    | 2.14    |

**Formula: Ba<sub>2.14</sub>GeS<sub>3.96</sub>**

Figure S1. (b) EDX spectrum of  $\text{Mg}_2\text{SnS}_4$ .

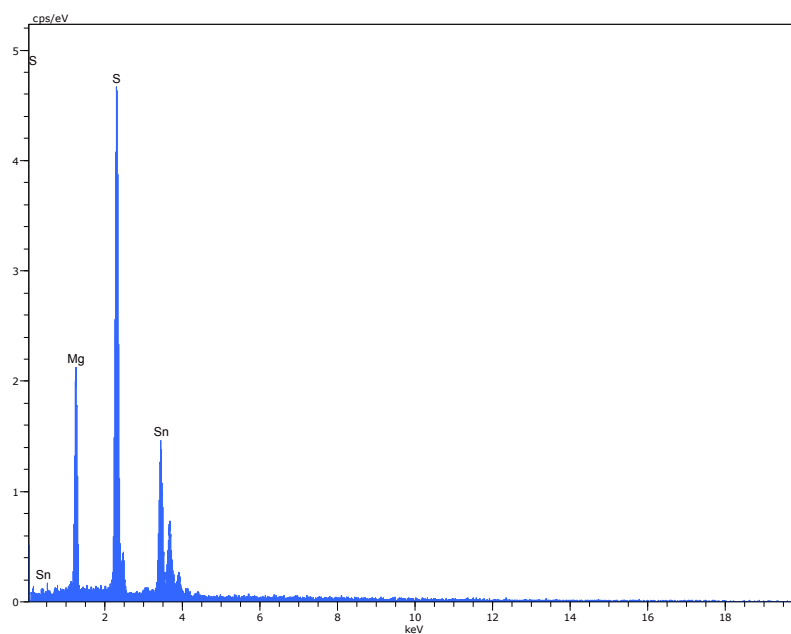


Table S1 (b) Average EDX data for  $\text{Mg}_2\text{SnS}_4$ .

| Element | Weight % | Atomic % | Formula |
|---------|----------|----------|---------|
| S       | 39.94    | 55.22    | 3.81    |
| Sn      | 44.64    | 16.67    | 1.15    |
| Mg      | 15.41    | 28.11    | 1.9     |

**Formula:  $\text{Mg}_{1.94}\text{Sn}_{1.15}\text{S}_{3.81}$**

Table S2. (a) Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for  $\text{Ba}_2\text{GeS}_4$ .  $U_{\text{eq}}$  is defined as one-third of the trace of the orthogonalized  $U_{ij}$  tensor.

| Atom | x       | y      | z       | $U_{\text{eq}}$ |
|------|---------|--------|---------|-----------------|
| Ba1  | 4811(1) | 2500   | 6715(1) | 20(1)           |
| Ba2  | 1291(1) | 2500   | 4113(1) | 32(1)           |
| Ge1  | 7285(1) | 2500   | 4247(1) | 14(1)           |
| S1   | 4862(2) | 2500   | 4051(2) | 28(1)           |
| S2   | 1756(1) | 106(2) | 6482(1) | 33(1)           |
| S3   | 8119(2) | 2500   | 5937(1) | 32(1)           |

Table S2. (b) Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for  $\text{Mg}_2\text{SnS}_4$ .  $U_{\text{eq}}$  is defined as one-third of the trace of the orthogonalized  $U_{ij}$  tensor.

| Atom | x       | y     | z       | $U_{\text{eq}}$ |
|------|---------|-------|---------|-----------------|
| Sn1  | 901(1)  | 2500  | 4245(1) | 10(1)           |
| Mg1  | 2698(1) | 2500  | 9988(2) | 13(1)           |
| Mg2  | 0       | 0     | 0       | 14(1)           |
| S1   | 1682(1) | 42(1) | 2428(1) | 12(1)           |
| S2   | -750(1) | 2500  | 2492(2) | 11(1)           |
| S3   | 923(1)  | 2500  | 8020(2) | 10(1)           |

Table S3. (a) Selected bond distances (Å) and angles (deg) for Ba<sub>2</sub>GeS<sub>4</sub>.

|                |           |                 |           |
|----------------|-----------|-----------------|-----------|
| Ba1-S3         | 3.120(4)  | S3-Ba1-S2       | 141.87(6) |
| Ba1-S2         | 3.213(3)  | S3-Ba1-S2#1     | 141.87(6) |
| Ba1-S2#1       | 3.213(3)  | S2-Ba1-S2#1     | 61.63(10) |
| Ba1-S3#2       | 3.247(4)  | S3-Ba1-S3#2     | 135.65(4) |
| Ba1-S1         | 3.255(5)  | S2-Ba1-S3#2     | 71.25(6)  |
| Ba1-S2#3       | 3.258(3)  | S2#1-Ba1-S3#2   | 71.25(6)  |
| Ba1-S2#4       | 3.258(3)  | S3-Ba1-S1       | 71.46(5)  |
| Ba1-S1#5       | 3.575(4)  | S2-Ba1-S1       | 85.60(4)  |
| Ba1-S1#6       | 3.575(4)  | S2#1-Ba1-S1     | 85.60(4)  |
| Ba2-S1         | 3.209(4)  | S3#2-Ba1-S1     | 152.89(6) |
| Ba2-S2#9       | 3.350(3)  | S3-Ba1-S2#3     | 72.27(8)  |
| Ba2-S2#10      | 3.350(3)  | S2-Ba1-S2#3     | 140.96(4) |
| Ba2-S2#1       | 3.356(4)  | S2#1-Ba1-S2#3   | 105.02(8) |
| Ba2-S2         | 3.356(4)  | S3#2-Ba1-S2#3   | 69.71(9)  |
| Ba2-S3#6       | 3.479(4)  | S1-Ba1-S2#3     | 131.97(6) |
| Ba2-S3#5       | 3.479(4)  | S3-Ba1-S2#4     | 72.27(8)  |
| Ba2-S3#11      | 3.617(4)  | S2-Ba1-S2#4     | 105.02(8) |
| Ge1-S2#5       | 2.178(3)  | S2#1-Ba1-S2#4   | 140.96(4) |
| Ge1-S2#14      | 2.178(2)  | S3#2-Ba1-S2#4   | 69.71(9)  |
| Ge1-S1         | 2.190(3)  | S1-Ba1-S2#4     | 131.97(6) |
| Ge1-S3         | 2.198(3)  | S2#3-Ba1-S2#4   | 60.68(9)  |
| S1-Ba1#5       | 3.575(4)  | S3-Ba1-S1#5     | 80.90(3)  |
| S1-Ba1#6       | 3.575(4)  | S2-Ba1-S1#5     | 63.54(6)  |
| S2-Ge1#5       | 2.178(2)  | S2#1-Ba1-S1#5   | 122.66(5) |
| S2-Ba1#2       | 3.258(3)  | S3#2-Ba1-S1#5   | 105.67(4) |
| S2-Ba2#9       | 3.350(3)  | S1-Ba1-S1#5     | 74.75(4)  |
| S3-Ba1#4       | 3.247(4)  | S2#3-Ba1-S1#5   | 128.23(5) |
| S3-Ba2#6       | 3.479(4)  | S2#4-Ba1-S1#5   | 69.35(7)  |
| S3-Ba2#5       | 3.479(4)  | S3-Ba1-S1#6     | 80.90(3)  |
| S3-Ba2#15      | 3.617(4)  | S2-Ba1-S1#6     | 122.66(5) |
| S1-Ba2-S2#9    | 144.23(5) | S2#1-Ba1-S1#6   | 63.54(6)  |
| S1-Ba2-S2#10   | 144.23(5) | S3#2-Ba1-S1#6   | 105.67(4) |
| S2#9-Ba2-S2#10 | 64.65(10) | S1-Ba1-S1#6     | 74.75(4)  |
| S1-Ba2-S2#1    | 84.02(4)  | S2#3-Ba1-S1#6   | 69.35(7)  |
| S2#9-Ba2-S2#1  | 123.44(4) | S2#4-Ba1-S1#6   | 128.23(5) |
| S2#10-Ba2-S2#1 | 91.52(6)  | S1#5-Ba1-S1#6   | 148.12(8) |
| S1-Ba2-S2      | 84.02(4)  | S1-Ba2-S3#11    | 143.31(6) |
| S2#9-Ba2-S2    | 91.52(6)  | S2#9-Ba2-S3#11  | 59.37(7)  |
| S2#10-Ba2-S2   | 123.44(4) | S2#10-Ba2-S3#11 | 59.37(7)  |
| S2#1-Ba2-S2    | 58.74(10) | S2#1-Ba2-S3#11  | 64.32(7)  |
| S1-Ba2-S3#6    | 81.20(3)  | S2-Ba2-S3#11    | 64.32(7)  |
| S2#9-Ba2-S3#6  | 130.48(5) | S3#6-Ba2-S3#11  | 97.53(3)  |
| S2#10-Ba2-S3#6 | 65.94(7)  | S3#5-Ba2-S3#11  | 97.53(3)  |

|                |           |                |            |
|----------------|-----------|----------------|------------|
| S2#1-Ba2-S3#6  | 60.76(6)  | S3#6-Ba2-S3#5  | 162.33(7)  |
| S2-Ba2-S3#6    | 118.74(6) | S2#5-Ge1-S2#14 | 110.67(11) |
| S1-Ba2-S3#5    | 81.20(3)  | S2#5-Ge1-S1    | 110.42(6)  |
| S2#9-Ba2-S3#5  | 65.94(7)  | S2#14-Ge1-S1   | 110.42(6)  |
| S2#10-Ba2-S3#5 | 130.48(5) | S2#5-Ge1-S3    | 104.43(7)  |
| S2#1-Ba2-S3#5  | 118.74(6) | S2#14-Ge1-S3   | 104.43(7)  |
| S2-Ba2-S3#5    | 60.76(6)  | S1-Ge1-S3      | 116.18(8)  |

Note. Symmetry transformations used to generate equivalent atoms:

#1  $x, -y+1/2, z$     #2  $x-1/2, y, -z+3/2$     #3  $x+1/2, -y+1/2, -z+3/2$   
#4  $x+1/2, y, -z+3/2$     #5  $-x+1, -y, -z+1$   
#6  $-x+1, -y+1, -z+1$     #7  $-x+1/2, -y+1, z+1/2$   
#8  $-x+1/2, -y, z+1/2$     #9  $-x, -y, -z+1$   
#10  $-x, y+1/2, -z+1$     #11  $x-1, y, z$     #12  $-x+1/2, -y, z-1/2$   
#13  $-x+1/2, -y+1, z-1/2$     #14  $-x+1, y+1/2, -z+1$   
#15  $x+1, y, z$

Table S3. (b) Selected bond distances (Å) and angles (deg) for Mg<sub>2</sub>SnS<sub>4</sub>.

|              |            |                 |           |
|--------------|------------|-----------------|-----------|
| Sn1-S3       | 2.3506(19) | S2#8-Mg1-S3     | 170.08(7) |
| Sn1-S1       | 2.3905(13) | S2#8-Mg1-S1#9   | 97.61(5)  |
| Sn1-S1#1     | 2.3905(13) | S3-Mg1-S1#9     | 89.14(5)  |
| Sn1-S2       | 2.4023(16) | S2#8-Mg1-S1#10  | 97.61(5)  |
| Sn1-Mg2#2    | 3.4459(16) | S3-Mg1-S1#10    | 89.14(5)  |
| S1-Mg1#3     | 2.6135(17) | S1#9-Mg1-S1#10  | 93.87(7)  |
| S1-Mg2       | 2.6542(15) | S2#8-Mg1-S1#11  | 92.19(6)  |
| S1-Mg1#4     | 2.7299(17) | S3-Mg1-S1#11    | 80.53(5)  |
| S2-Mg1#5     | 2.552(2)   | S1#9-Mg1-S1#11  | 169.05(6) |
| S2-Mg2       | 2.6228(13) | S1#10-Mg1-S1#11 | 89.64(5)  |
| S2-Mg2#2     | 2.6228(13) | S2#8-Mg1-S1#7   | 92.19(6)  |
| S3-Mg2#6     | 2.5450(13) | S3-Mg1-S1#7     | 80.53(5)  |
| S3-Mg2#7     | 2.5450(13) | S1#9-Mg1-S1#7   | 89.64(5)  |
| S3-Mg1       | 2.608(2)   | S1#10-Mg1-S1#7  | 169.05(6) |
| Mg1-S2#8     | 2.552(2)   | S1#11-Mg1-S1#7  | 85.10(7)  |
| Mg1-S1#9     | 2.6135(17) | S3#4-Mg2-S3#12  | 180.00(5) |
| Mg1-S1#10    | 2.6135(17) | S3#4-Mg2-S2     | 86.14(5)  |
| Mg1-S1#11    | 2.7299(17) | S3#12-Mg2-S2    | 93.86(5)  |
| Mg1-S1#7     | 2.7299(17) | S3#4-Mg2-S2#13  | 93.86(5)  |
| Mg2-S3#4     | 2.5450(13) | S3#12-Mg2-S2#13 | 86.14(5)  |
| Mg2-S3#12    | 2.5450(13) | S2-Mg2-S2#13    | 180.003   |
| Mg2-S2#13    | 2.6228(13) | S3#4-Mg2-S1     | 83.16(4)  |
| Mg2-S1#13    | 2.6542(15) | S3#12-Mg2-S1    | 96.84(4)  |
| S3-Sn1-S1    | 117.90(3)  | S2-Mg2-S1       | 87.66(4)  |
| S3-Sn1-S1#1  | 117.90(3)  | S2#13-Mg2-S1    | 92.34(4)  |
| S1-Sn1-S1#1  | 101.11(6)  | S3#4-Mg2-S1#13  | 96.84(4)  |
| S3-Sn1-S2    | 117.71(4)  | S3#12-Mg2-S1#13 | 83.16(4)  |
| S1-Sn1-S2    | 99.37(4)   | S2-Mg2-S1#13    | 92.34(4)  |
| S1#1-Sn1-S2  | 99.37(4)   | S2#13-Mg2-S1#13 | 87.66(4)  |
| S1-Mg2-S1#13 | 180.00(3)  |                 |           |

Note. Symmetry transformations used to generate equivalent atoms:

#1 x,-y+1/2,z #2 -x,y+1/2,-z #3 -x+1/2,-y,z-1/2 #4 x,y,z-1 #5 x-1/2,y,-z+3/2  
 #6 -x,y+1/2,-z+1 #7 x,y,z+1 #8 x+1/2,y,-z+3/2 #9 -x+1/2,-y,z+1/2  
 #10 -x+1/2,y+1/2,z+1/2 #11 x,-y+1/2,z+1 #12 -x,-y,-z+1 #13 -x,-y,-z

Figure S2. IR spectra of  $\text{Ba}_2\text{GeS}_4$  and  $\text{Mg}_2\text{SnS}_4$ .

