

Supporting Information

Silica Microspheres Containing High Density Surface Hydroxyl Groups as Efficient Epoxidation Catalysts

Prakash Chandra^a, Dhananjay S. Doke^a, Shubhangi B. Umbarkar^{a,c}, Kumar Vanka^b and Ankush V. Biradar^{a,c}

S-1: Materials: Chemical: Tetraethyl orthosilicate (TEOS), cyclohexene, styrene, n-heptene, *cis*-cyclooctene, allylbenzene and 50 % H₂O₂ were purchased from Sigma-Aldrich chemical India, Absolute ethanol (99.9%), NH₃ (25%) solution, acetonitrile, *tert*-butanol, ethyl acetate were purchased from Thomas Bakar India Ltd. and KOH has purchased from Rankem India Ltd. Furthermore, all the chemicals were used as received unless stated otherwise.

S-2: Catalyst characterization

The synthesized catalysts were characterized by Scanning electron microscopy (SEM) measurements were performed on a FEI quanta 200 3D dual beam (ESEM) having thermionic emission tungsten filament in the 3 nm range at 30 kV and HRTEM experiment was performed with Tecnai FEI G2 microscope, using an accelerating voltage of 300 kV. For TEM analysis, a sample was deposited on a coated 200 mesh Cu grid. N₂ adsorption–desorption isotherms were recorded at 77 K by using an automated quantasorb instrument from quantachrome. Before each run, a known mass of sample (around 0.200 g) was heated at 200 °C under vacuum for 2.5 h. Specific surface areas were calculated from the linear part of the Brunauer–Emmett–Teller line. Pore-size distributions were obtained by applying the Barrett–Joyner–Halenda (BJH) equation to the desorption branch of the isotherm. The total pore volume was estimated from the N₂ uptake at a P/P_0 value of 0.991. FTIR spectra were obtained using a Thermo-Nicolet 670 spectrometer with samples loaded onto a KBr disk in the range of 4000–400 cm^{−1} region with 4 cm^{−1} resolution. Solid-state ²⁹Si NMR spectra was obtained using a 300 MHz Bruker Advance NMR spectrometer. For ²⁹Si MAS NMR experiments, 8.0 kHz spin rate, 5 s recycle delay, 10 ms contact time, $\pi/2$ pulse width of 5.6 μ s, and 10000–20000 scans using TPPM 1H decoupling was employed.

S-3: Catalyst surface area analysis by BET method

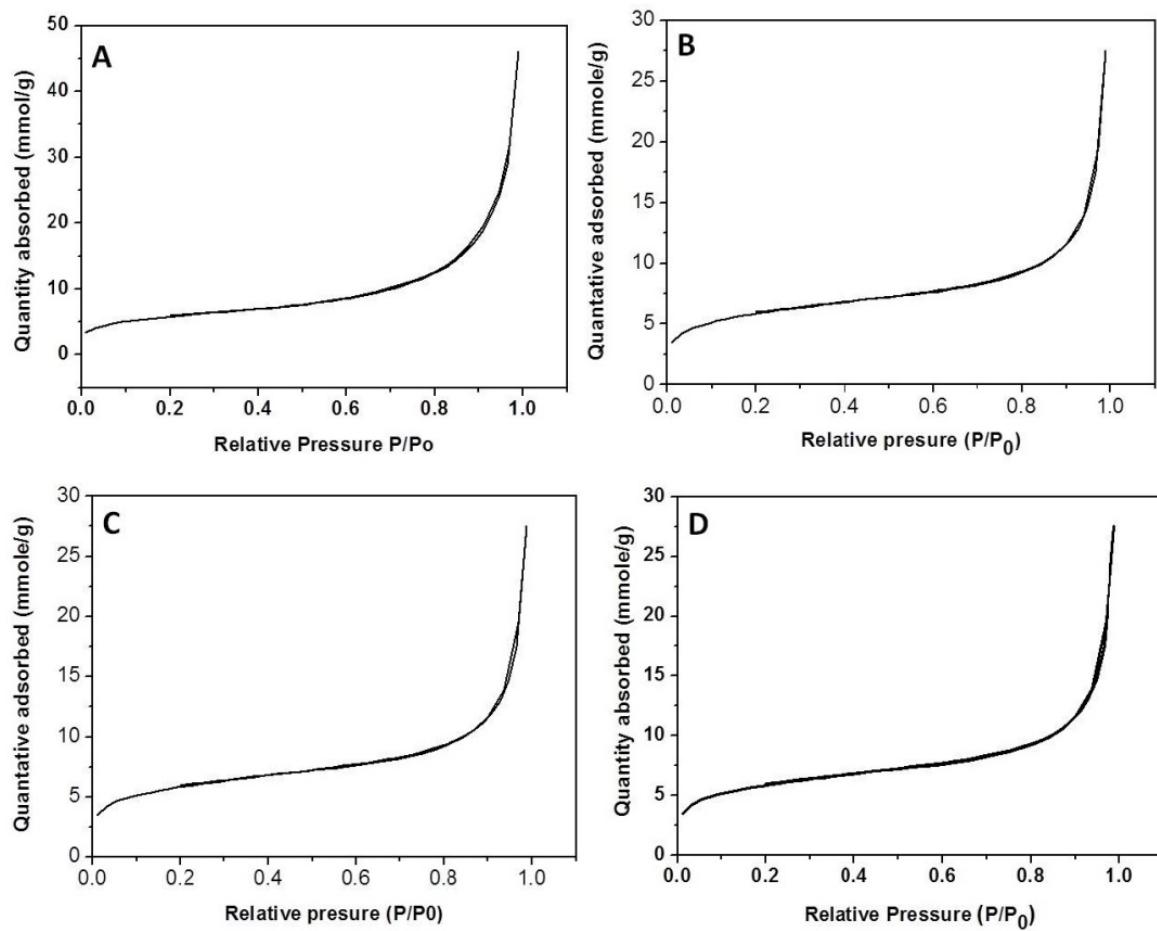


Figure-S1: Nitrogen adsorption-desorption isotherms for (A) silica microspheres (B) after etching for 2 h., (c) after etching for 3 h., and (d) after etching for 4 h.

S-4: FTIR analysis of silica microspheres

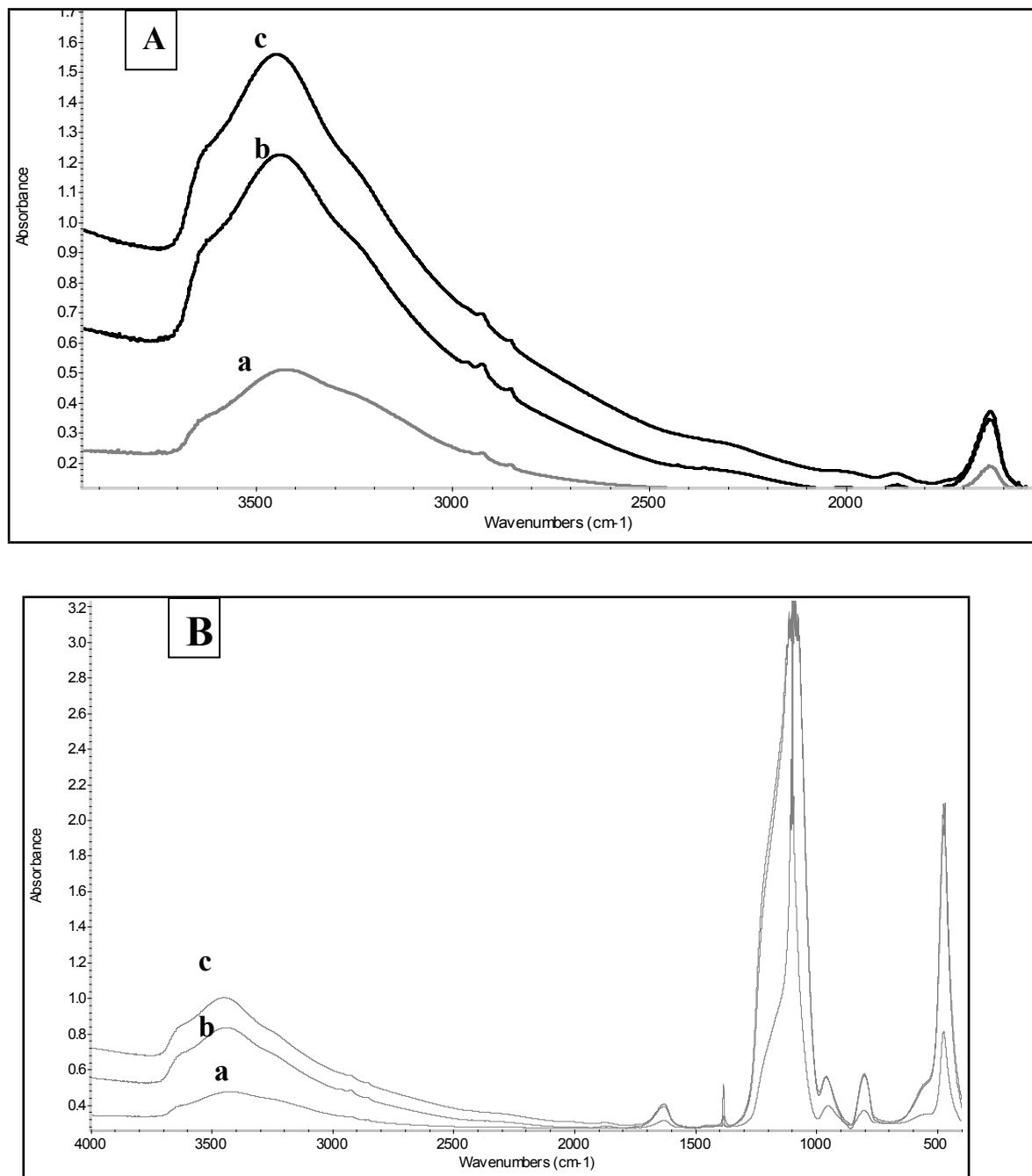


Figure-S2: FT-IR spectra of (a) Silica microspheres, (b) 3h and (4) 4h silica in (A) 1800-4000 cm⁻¹ and (B) 400 – 4000 cm⁻¹

S-5: SEM analysis of the catalysts

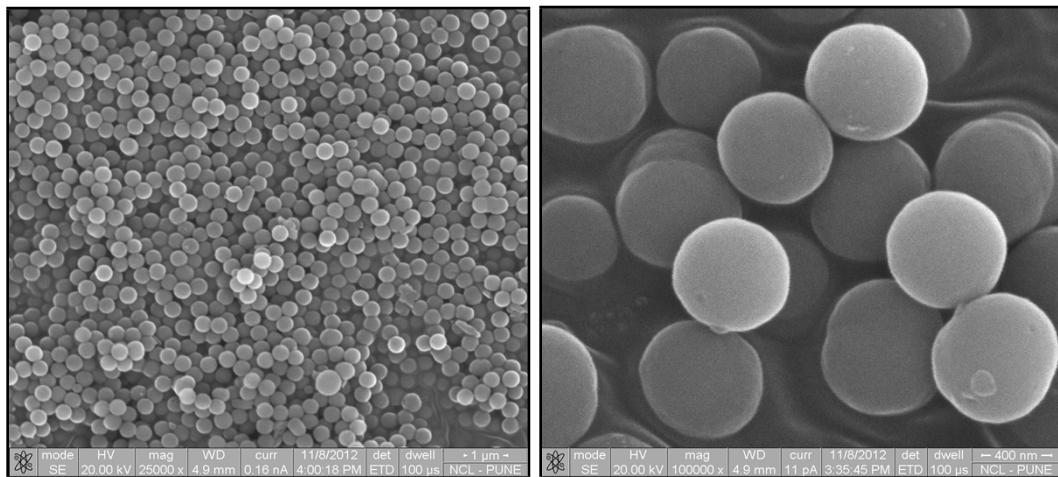


Figure-S3: SEM images of silica spheres A) 1 μm B) 400 nm

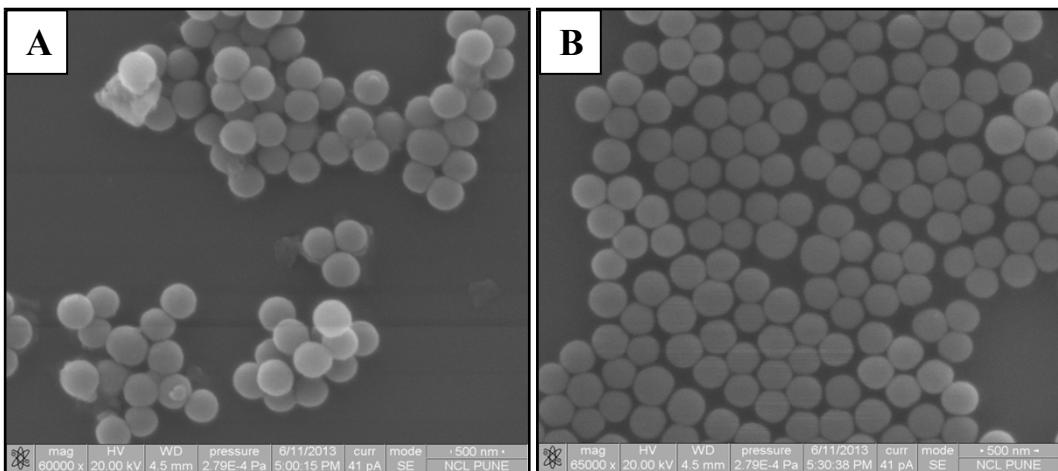


Figure-S4: SEM image of A) 3 h and B) 4 h silica

S-6: TEM analysis of the catalysts

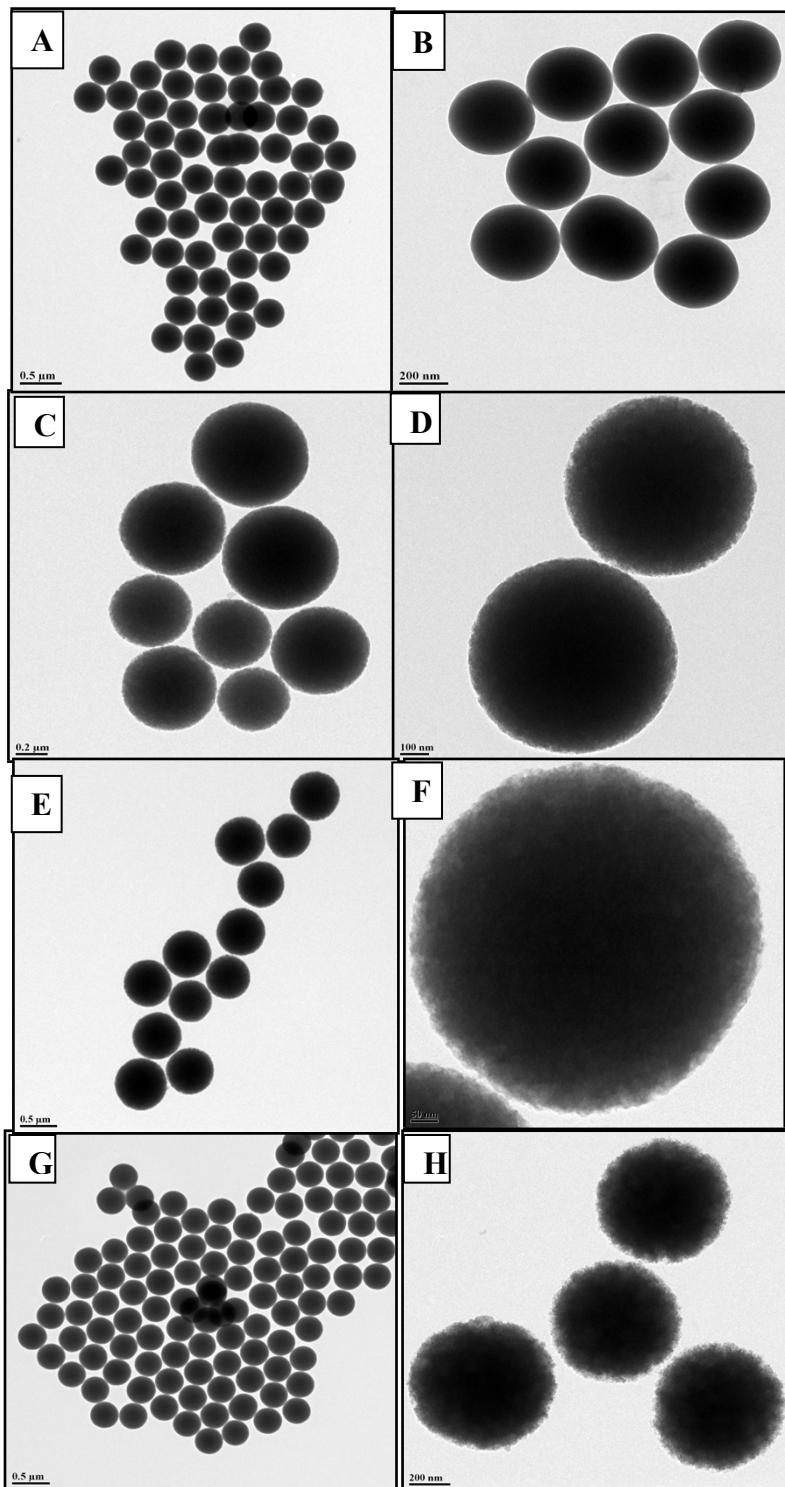


Figure-S5: HRTEM images of silica A) silica spheres; D) magnified image of silica spheres; C) 2h silica spheres; D) magnified image of 2h silica spheres; E) 3h silica spheres; F) magnified image of 3h silica spheres; G) 4h silica spheres and H) magnified image of 4h silica spheres.

S-7: Catalytic activity

Table ST1: Control experiments carried out using 4h etched SiO₂

Entry	Catalyst	% Conv.	% Sel.			
		1	2	3	4	5
1.	Etched SiO ₂ - 500 °C calcined	25.1	62	-	38	-
2.	Hot filtration ^[a]	30.0	93	-	7	-
3.	Hot filtration ^[b]	42.0	85	-	15	-

Reaction condition: Cyclohexene: 0.825 g (0.01 mol); 50% H₂O₂: 2.76 g (0.04 mol); Temperature: 80 °C; Solvent: Acetonitrile (10 g); Catalyst: 0.08 g; time-24h; ^b: catalyst 4h etched silica^[a] Reaction up to 10 h with catalyst; ^[b] Reaction without catalyst.

Table ST2: results of epoxidation of cyclohexene using different oxidant molar ratios^a

Entry	Oxidant molar ratio	% Conv. 1	% Selectivity		
			2	3	4
1.	1:2	25.2	92.8	--	7.2
2.	1:4	72.3	90.4	4.1	5.5
3.	1:6	95.0	89	8.4	2.1

Reaction condition: Cyclohexene: 0.82 g (0.01 mol); Solvent: Acetonitrile (10 g); Catalyst (4h etched silica sphere): 0.08 g; Temperature: 80 °C; Time-24 h.

S-8: Computational analysis

S8-S15: Mechanistic studies for olefin epoxidation silica microspheres synthesized by hydrolysis-condensation (HC)

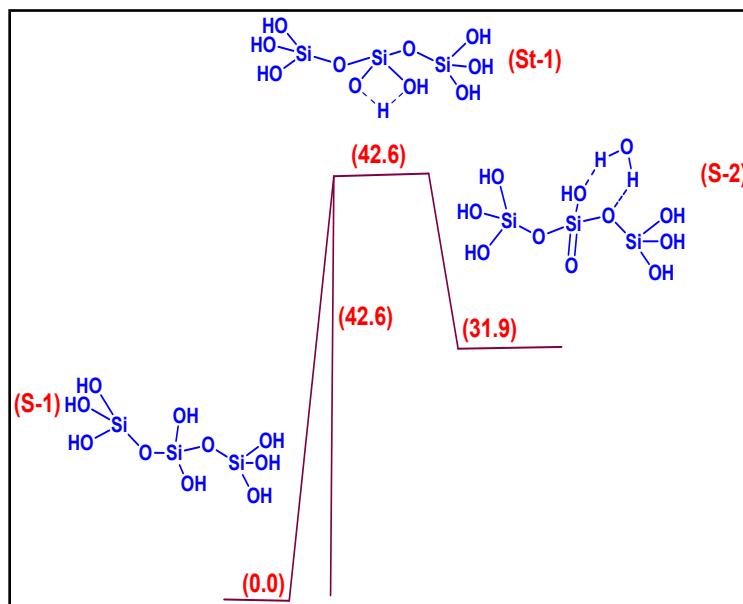


Figure-S8: The potential energy surface for the conversion of the germinal diols to $\text{Si}=\text{O}$ species; energy values (ΔG) are in kcalmol^{-1} .

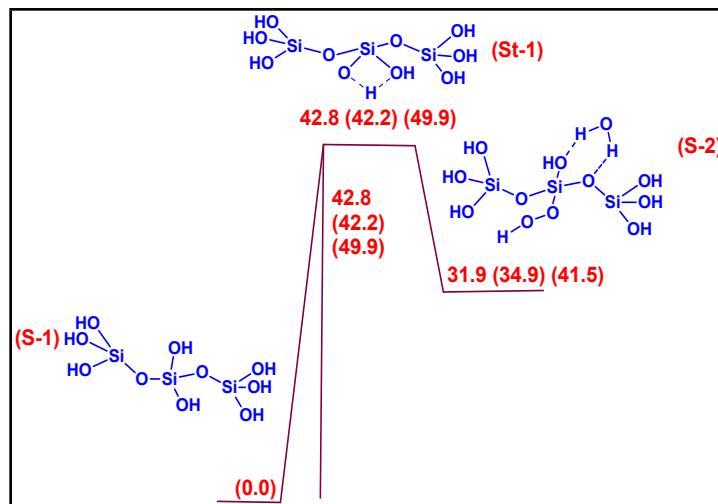


Figure-S9: The potential energy surface for the conversion of the germinal diols to $\text{Si}=\text{O}$ species; the values shown in the figure are the electronic energies calculated using the BP-86, PBE and B3LYP functionals respectively – the BP-86 values are shown outside the parenthesis, while the PBE and the B3LYP energies are shown (in that order), inside the parenthesis.

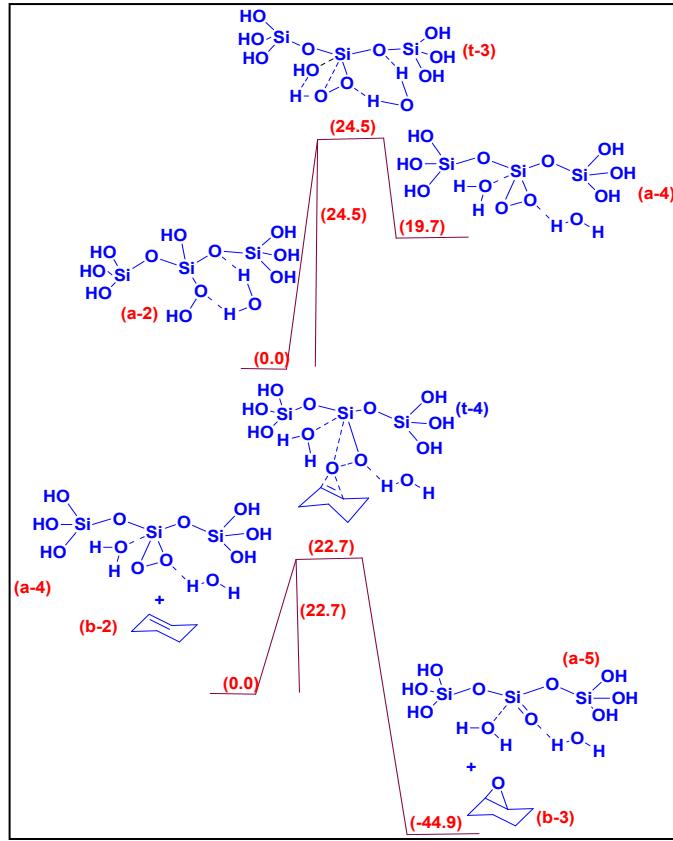


Figure-S10: The potential energy surface for olefin epoxidation catalyzed by secondary salinols present on the synthesized silica microspheres; energy values (ΔG) are in kcalmol^{-1} .

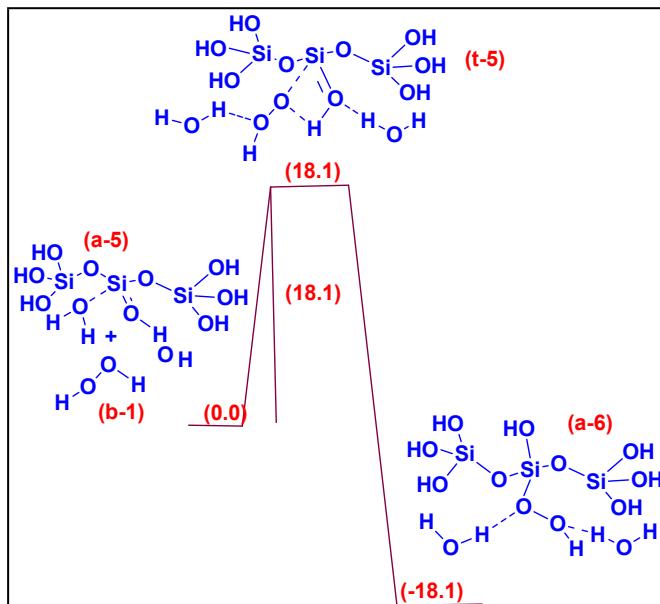


Figure-S11: The potential energy surface for attack at (a-4) on the silica surface by H_2O_2 to form (a-5); energy values (ΔG) are in kcalmol^{-1} .

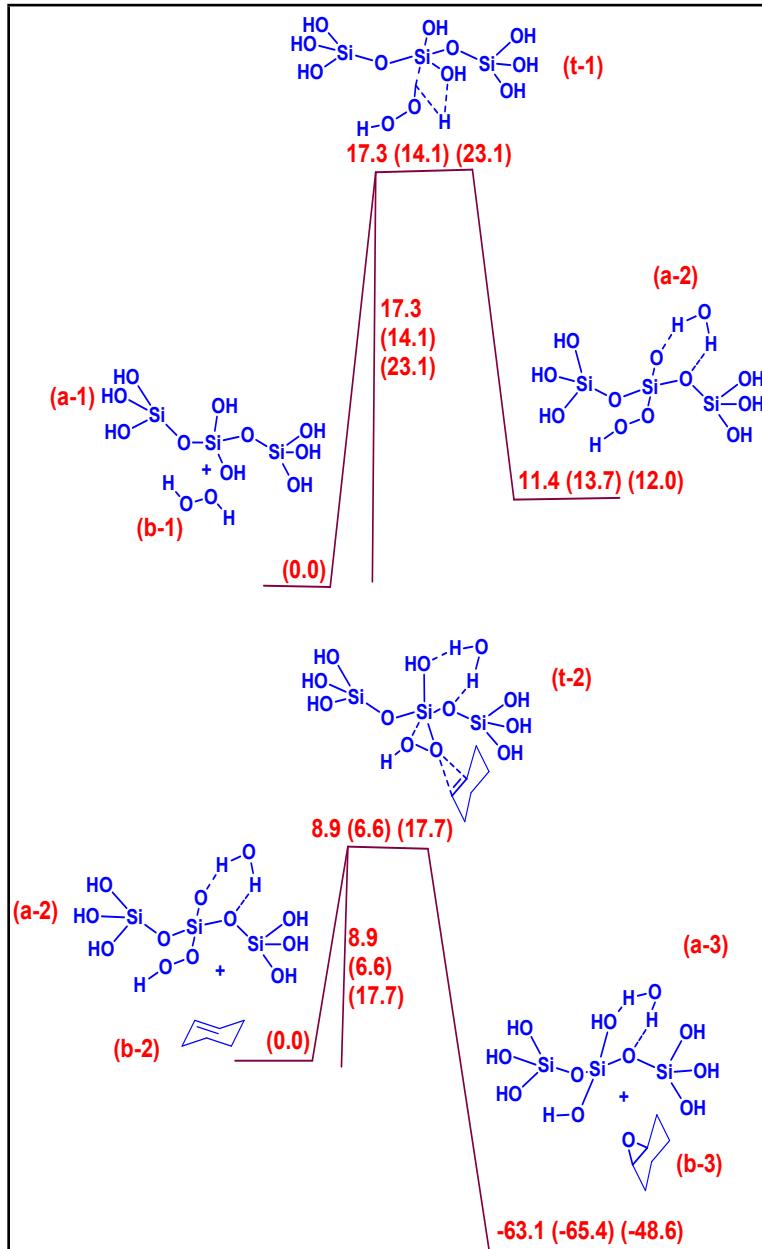


Figure-S12: The potential energy surface for olefin epoxidation catalyzed by secondary salinols present on the synthesized silica microspheres; the values shown in the figure are the electronic energies calculated using the BP-86, PBE and B3LYP functionals respectively – the BP-86 values are shown outside the parenthesis, while the PBE and the B3LYP energies are shown (in that order), inside the parenthesis.

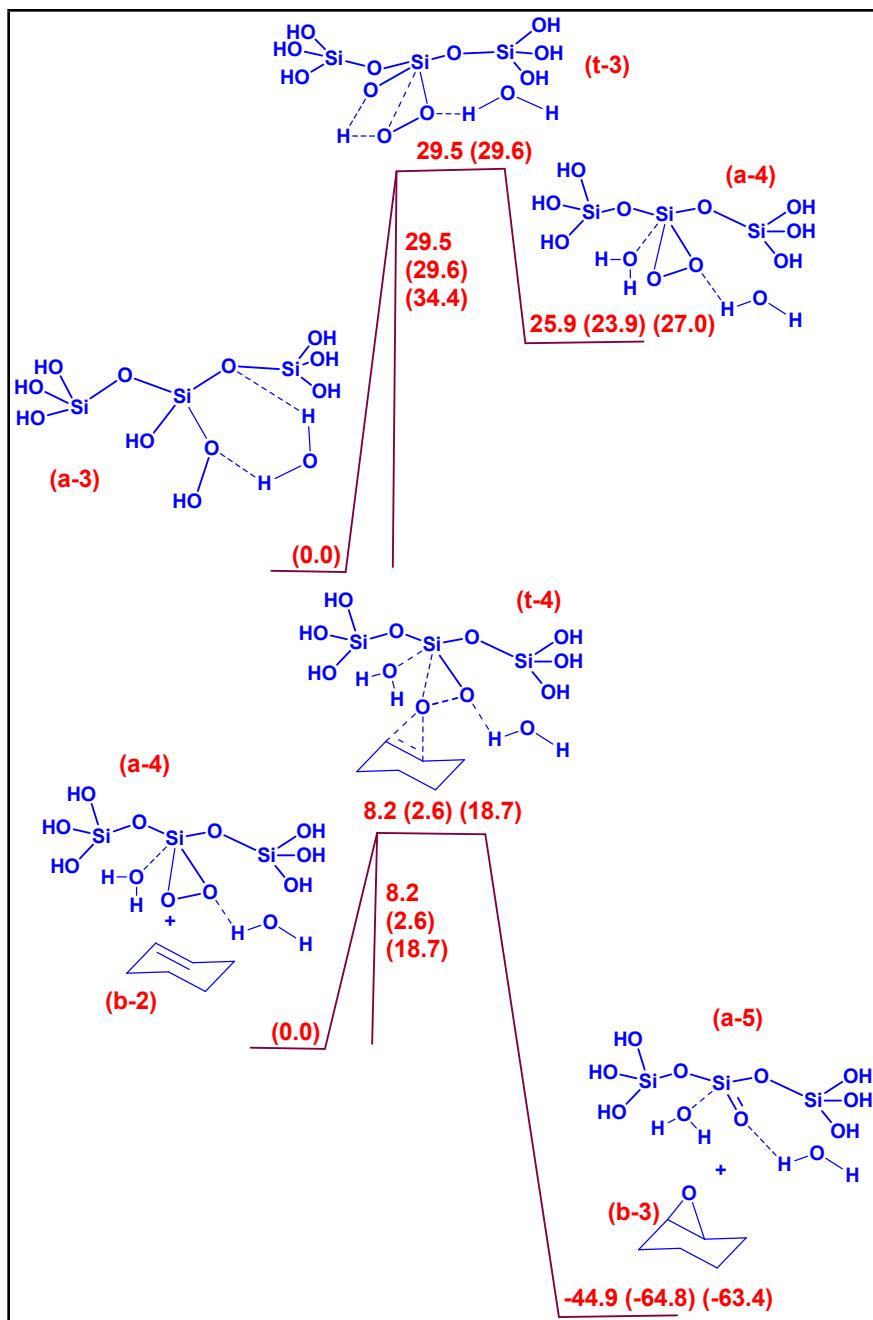


Figure-S13: The potential energy surface alternative pathway for olefin epoxidation catalyzed by secondary salinols present on the synthesized silica microspheres; the values shown in the figure are the electronic energies calculated using the BP-86, PBE and B3LYP functionals respectively – the BP-86 values are shown outside the parenthesis, while the PBE and the B3LYP energies are shown (in that order), inside the parenthesis.

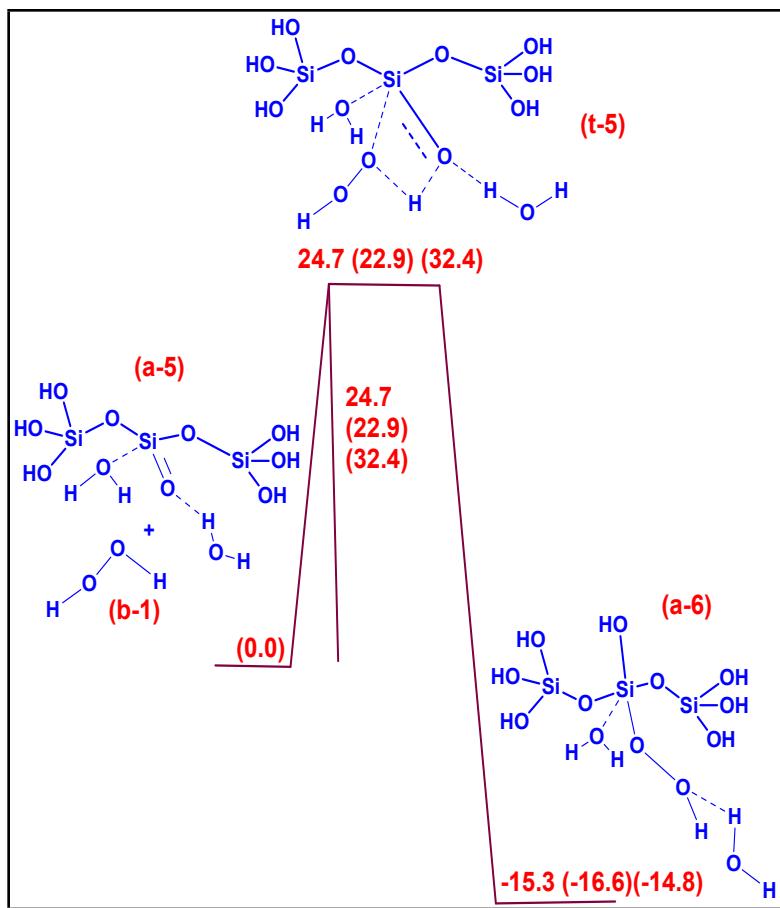


Figure-S14: The potential energy surface for double bond activation of secondary salinols by hydrogen peroxide present on the synthesized silica microspheres; the values shown in the figure are the electronic energies calculated using the BP-86, PBE and B3LYP functional respectively – the BP-86 values are shown outside the parenthesis, while the PBE and the B3LYP energies are shown (in that order), inside the parenthesis.

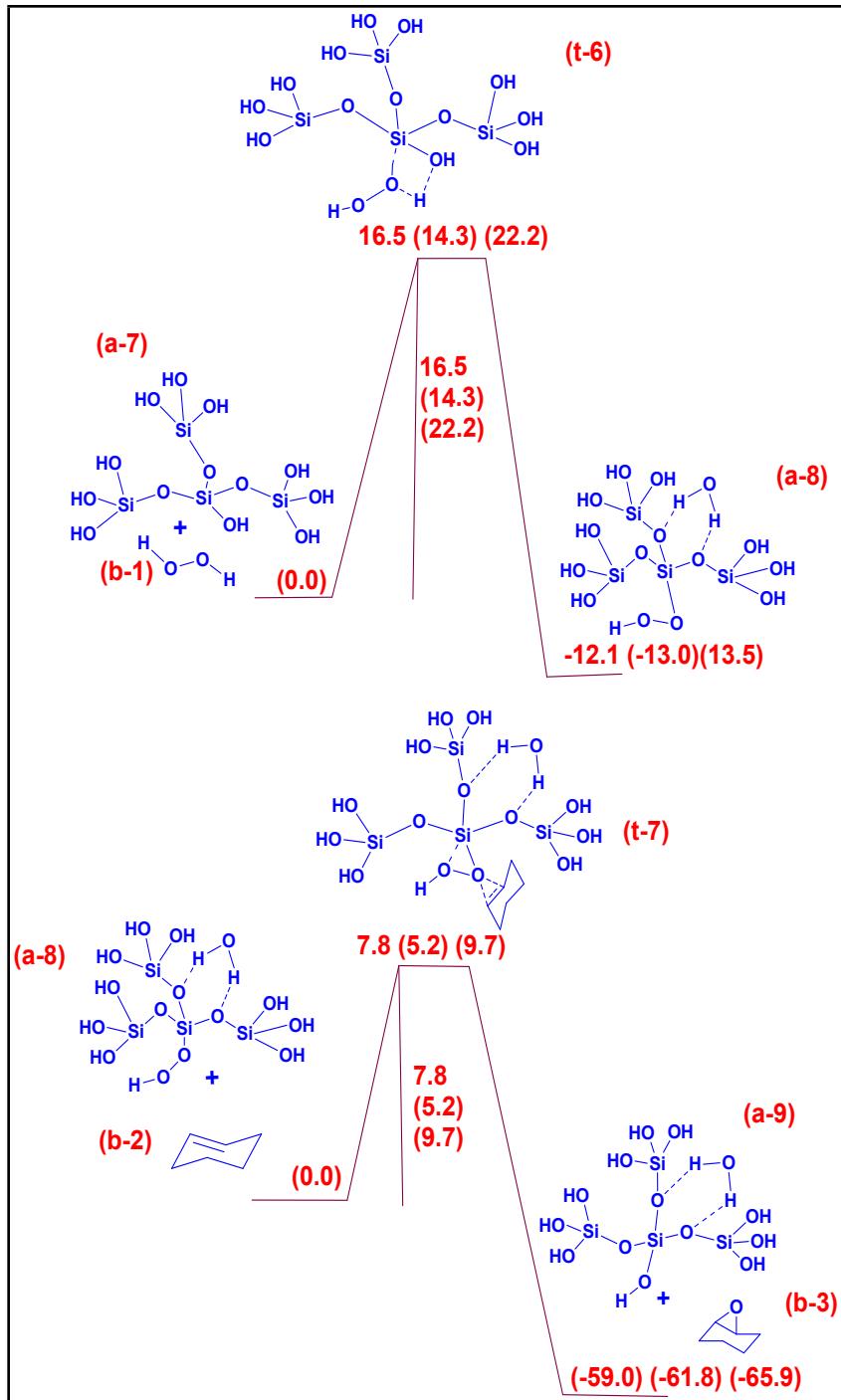
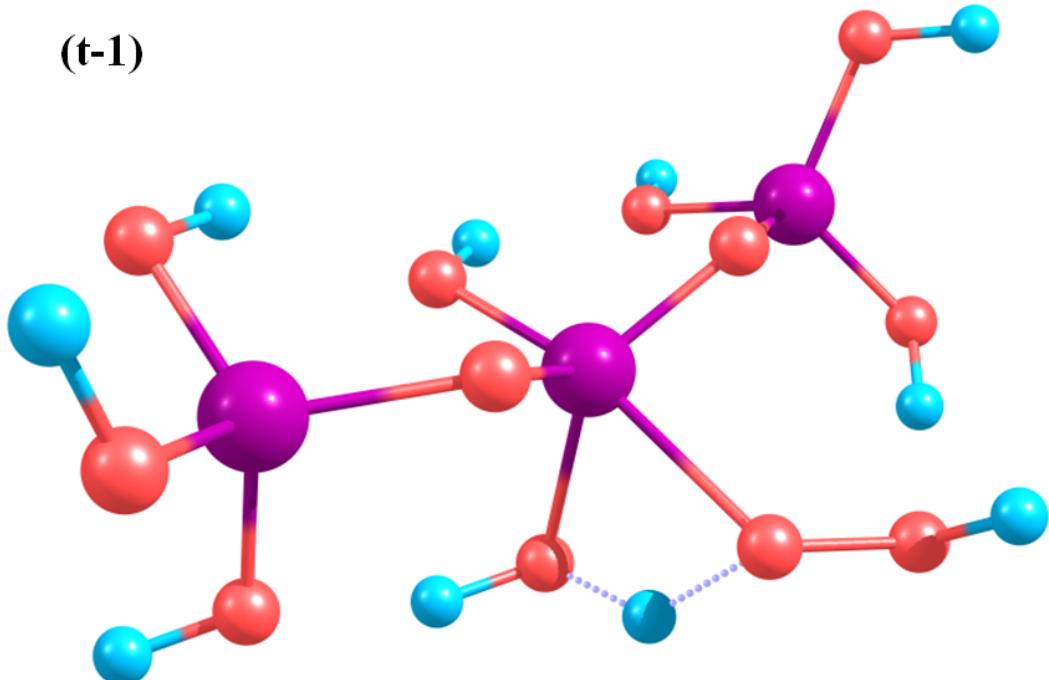


Figure S15: The potential energy surface for olefin epoxidation catalyzed by secondary salinols present on silica microspheres; the values shown in the figure are the electronic energies calculated using the BP-86, PBE and B3LYP functionals respectively – the BP-86 values are shown outside the parenthesis, while the PBE and the B3LYP energies are shown (in that order), inside the parenthesis.

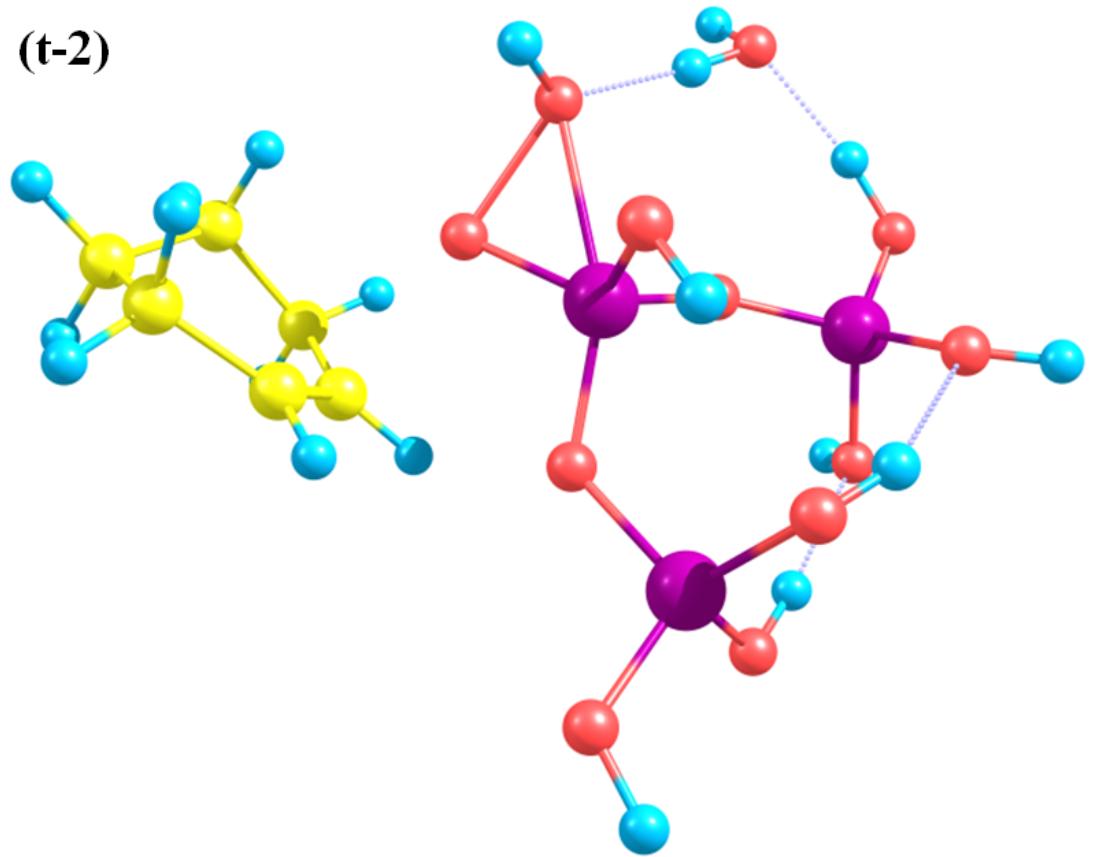
Table ST3: ΔH , ΔG and single imaginary vibrational frequencies for various transition states for silica catalyzed olefin epoxidation

Entry	Transition state	ΔH (298)	ΔG (298)	Imaginary Vibrational frequency (cm^{-1})
1	(t-1)	32.0	39.0	-554.55
2	(t-2)	20.5	23.7	-228.94
3	(t-3)	27.3	24.5	-405.95
4	(t-4)	19.6	22.7	-220.02
5	(t-5)	10.7	18.1	-874.78
6	(t-6)	28.4	32.6	-613.32
7	(t-7)	17.0	23.3	-223.02
8	(St-1)	42.6	42.6	-879.71

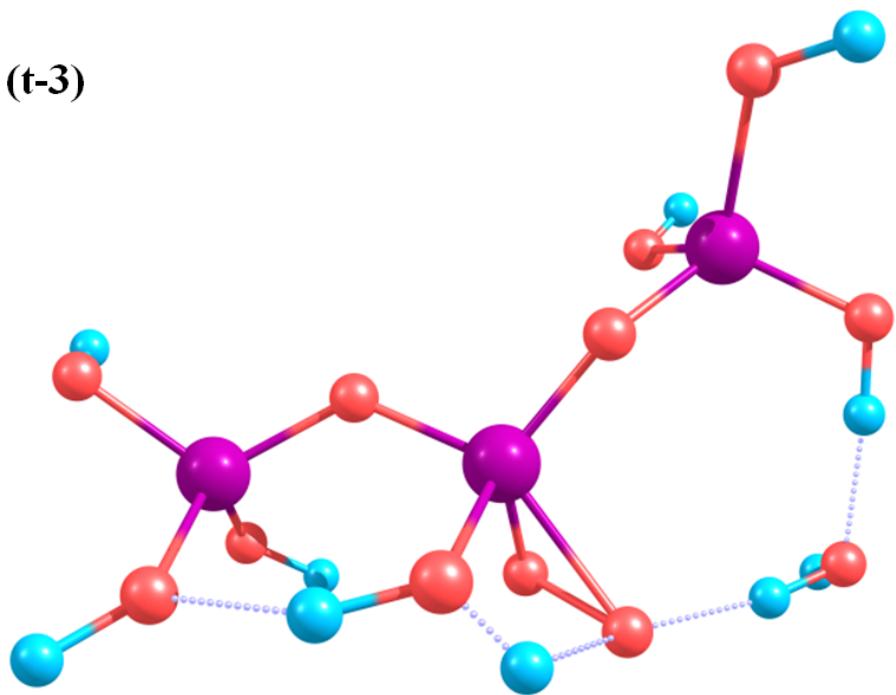
S-16- Optimized structures for various transition states involved during the silica catalyzed olefin epoxidation. (Nomenclature of each transition state t-1 to t-7 is according to as mentiond in the text)



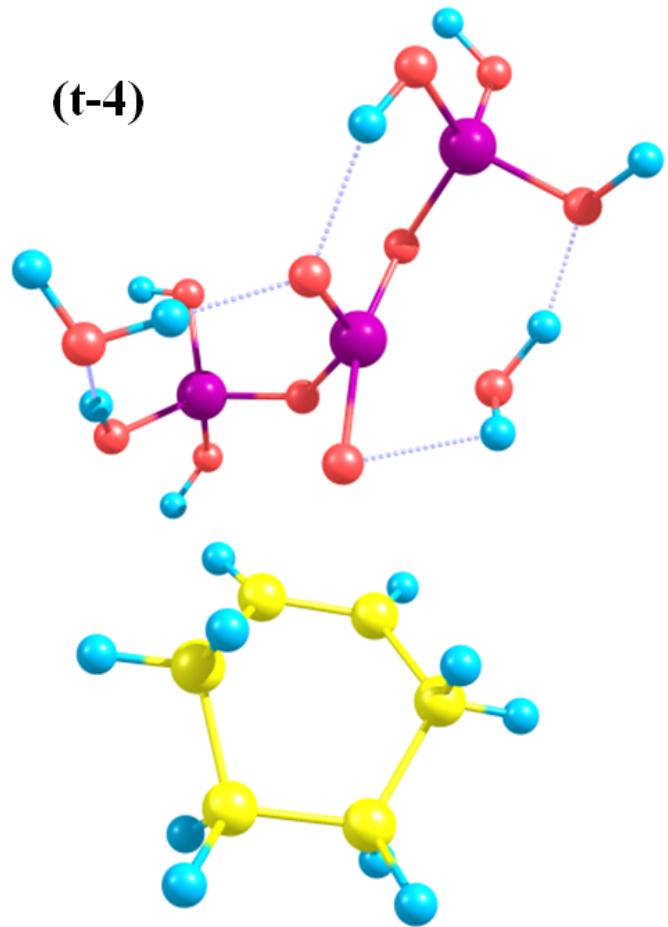
(t-2)



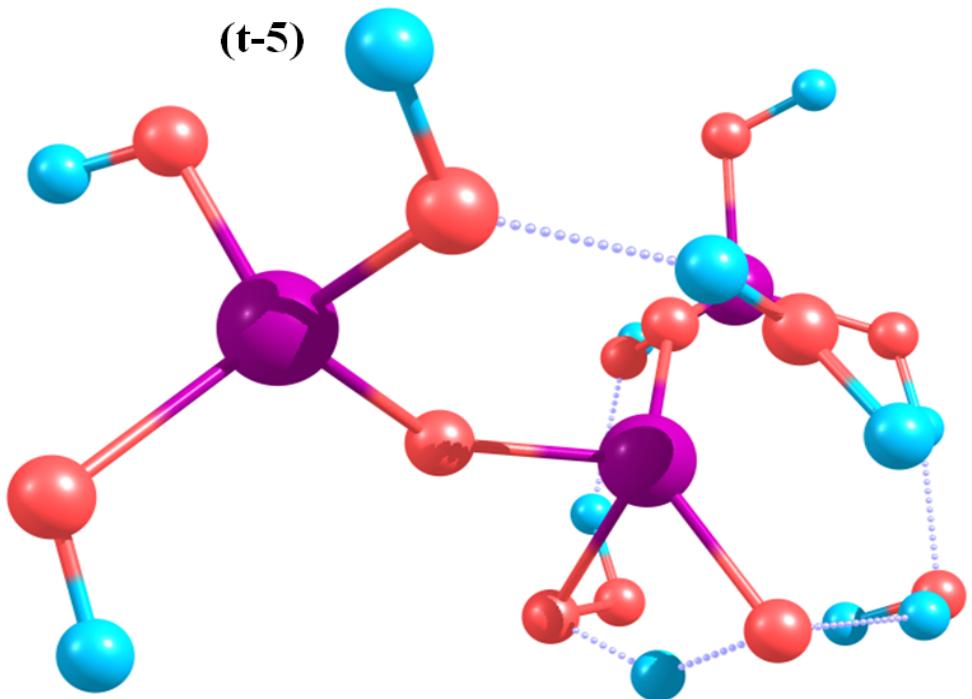
(t-3)



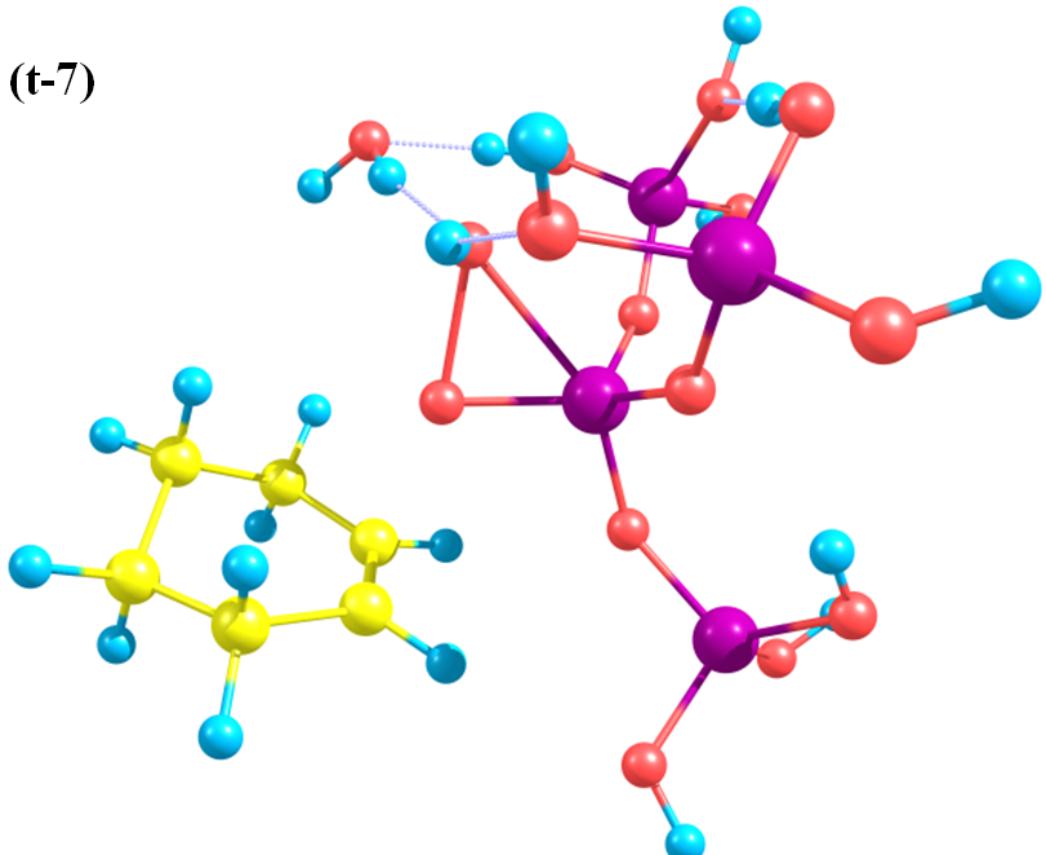
(t-4)



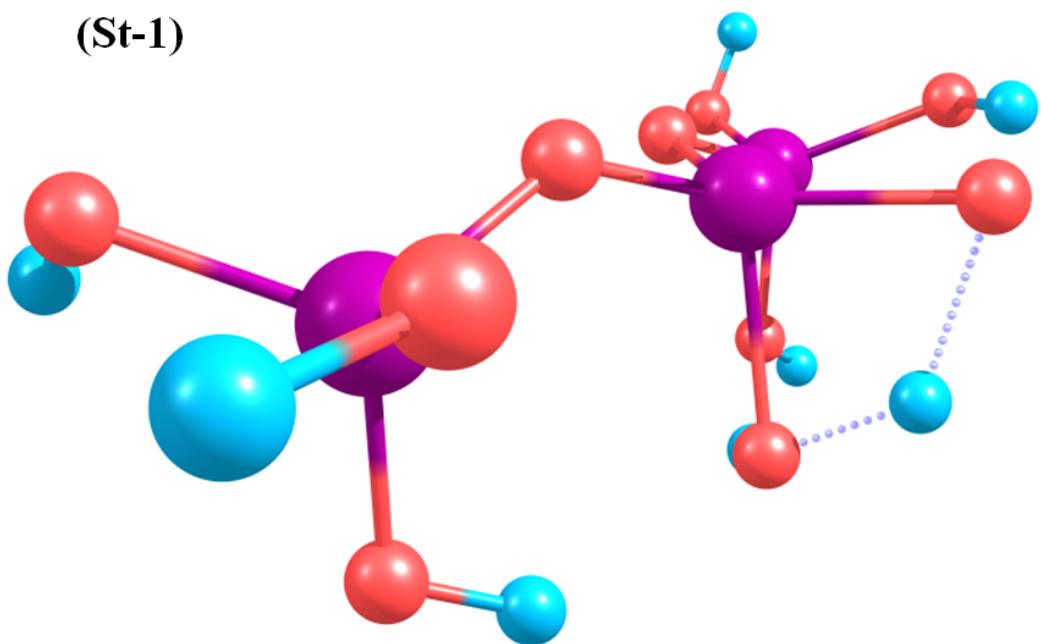
(t-5)



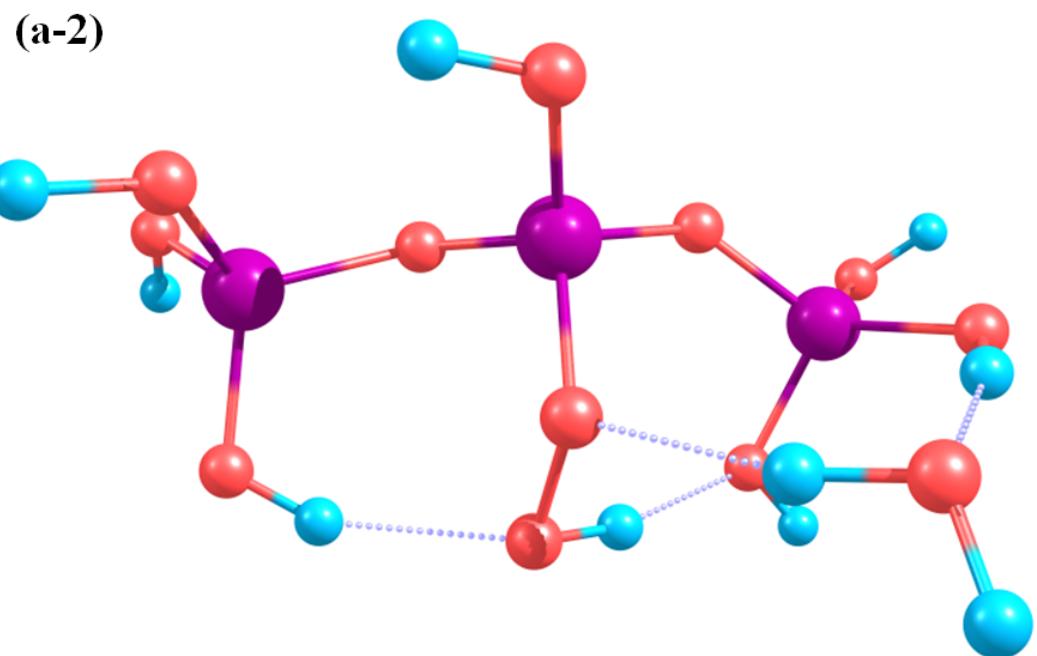
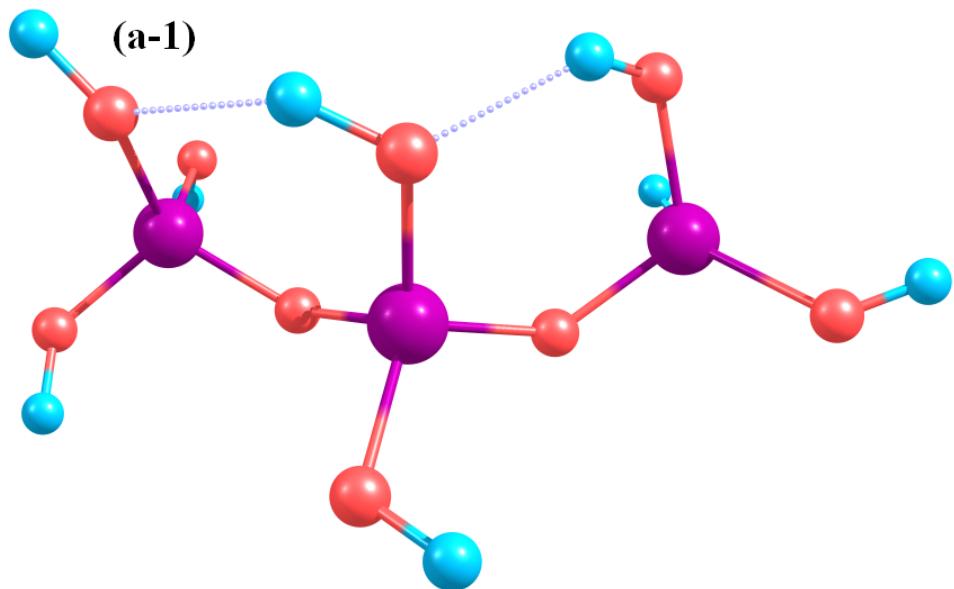
(t-7)



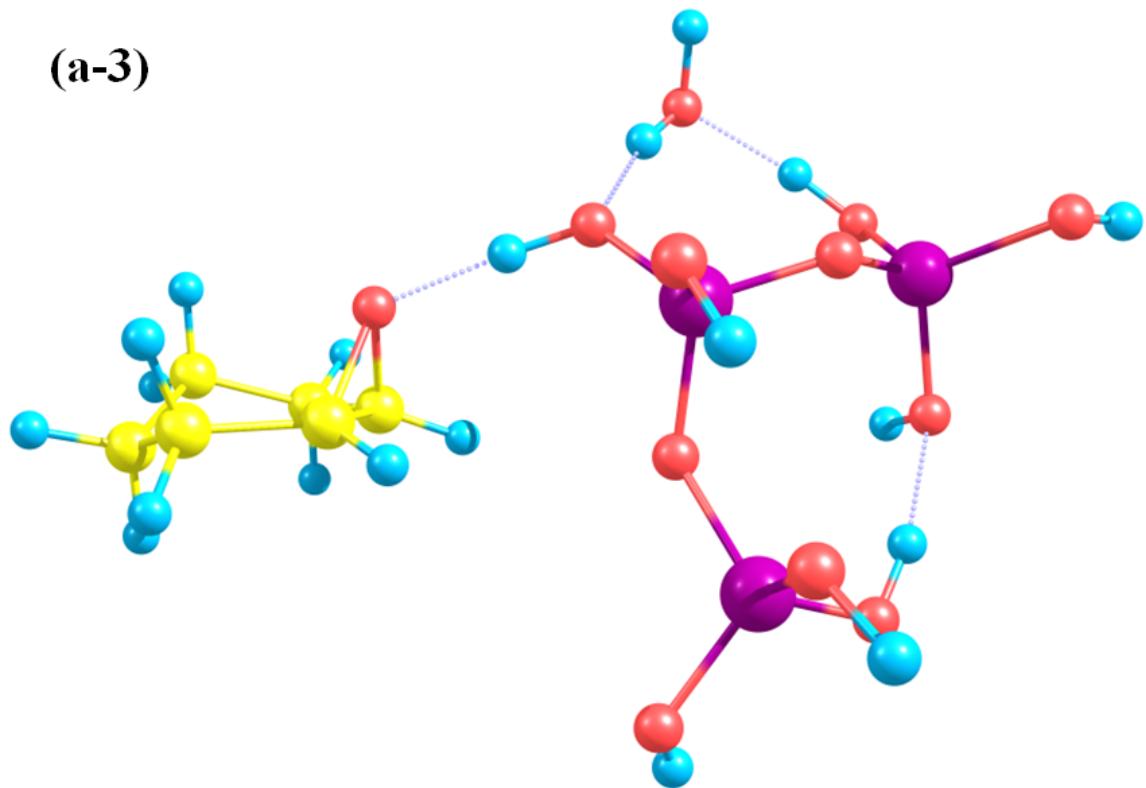
(St-1)



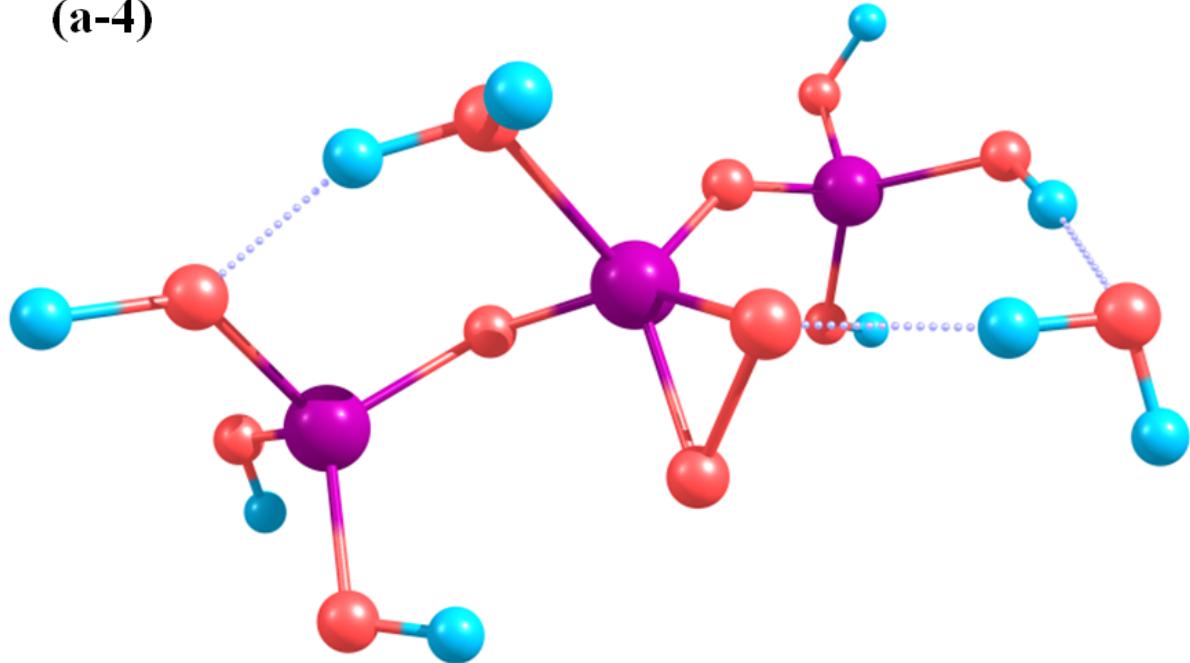
S17- Optimized structures for various reactants, products and intermediates involved during the silica catalyzed olefin epoxidation. (Nomenclature of each reactant, products and intermediates 1-1 to a-9 is according to as mentiond in the text)



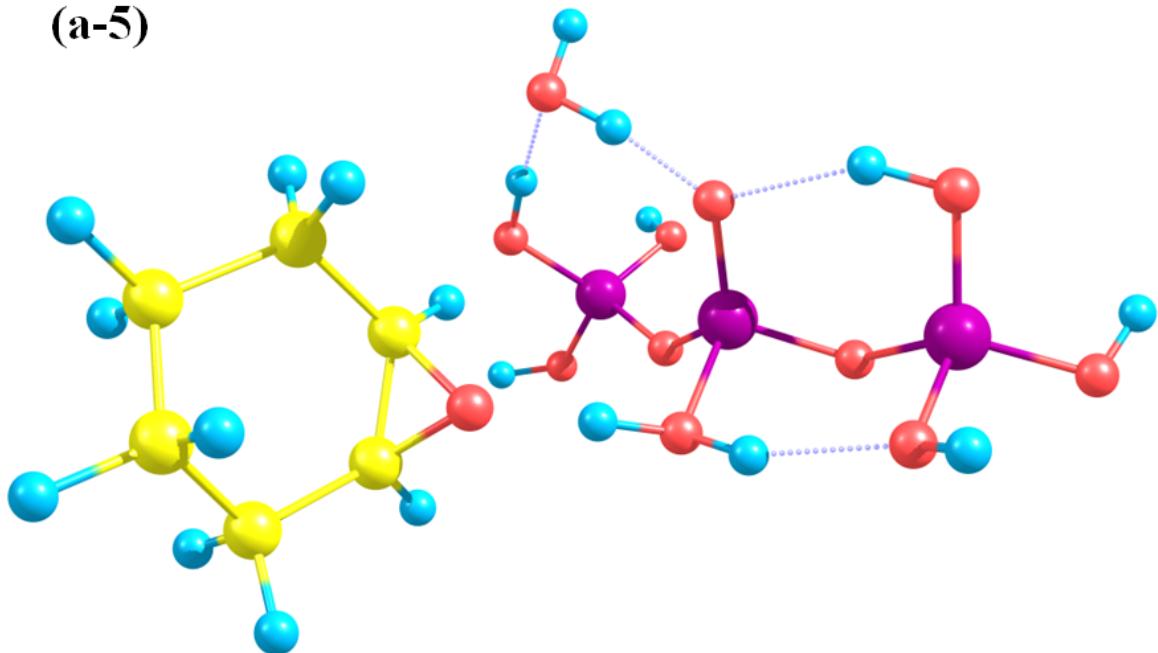
(a-3)



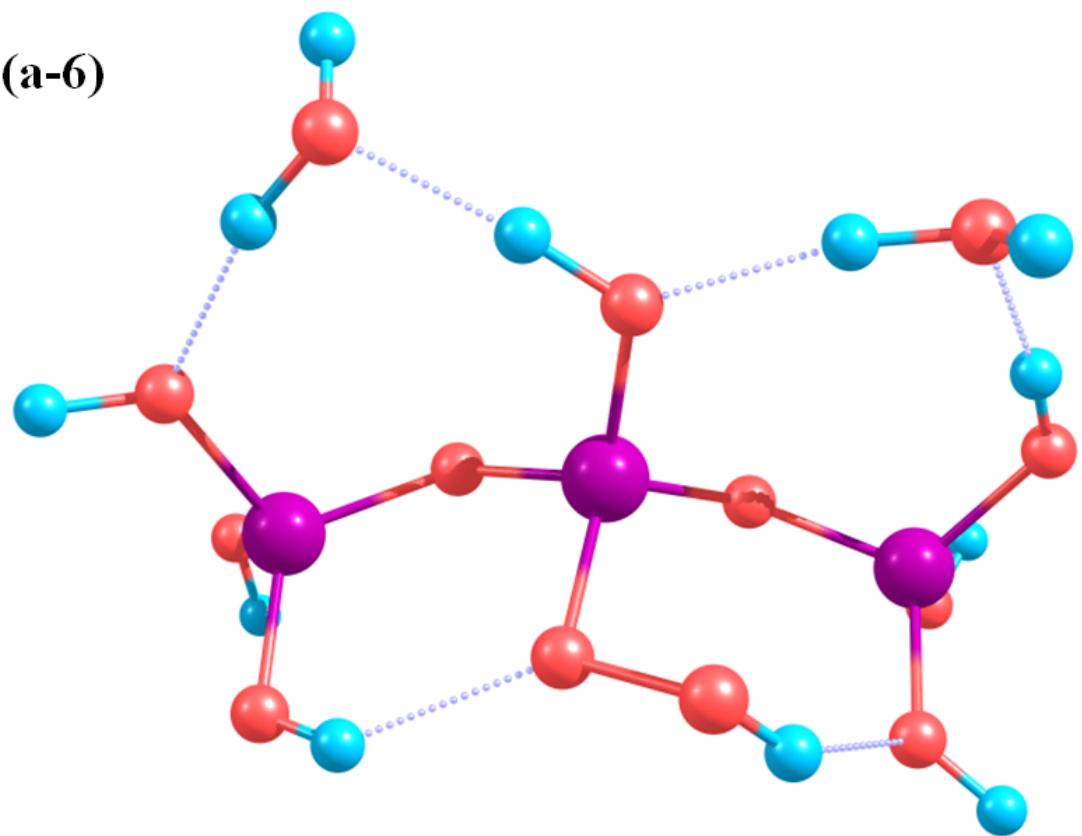
(a-4)



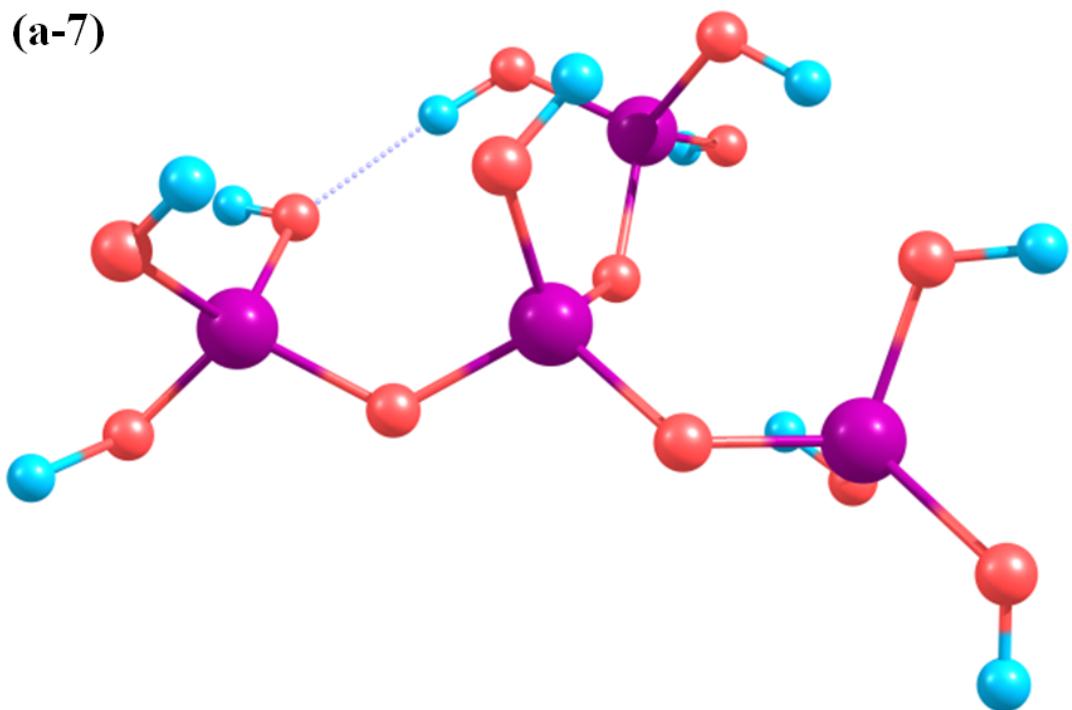
(a-5)



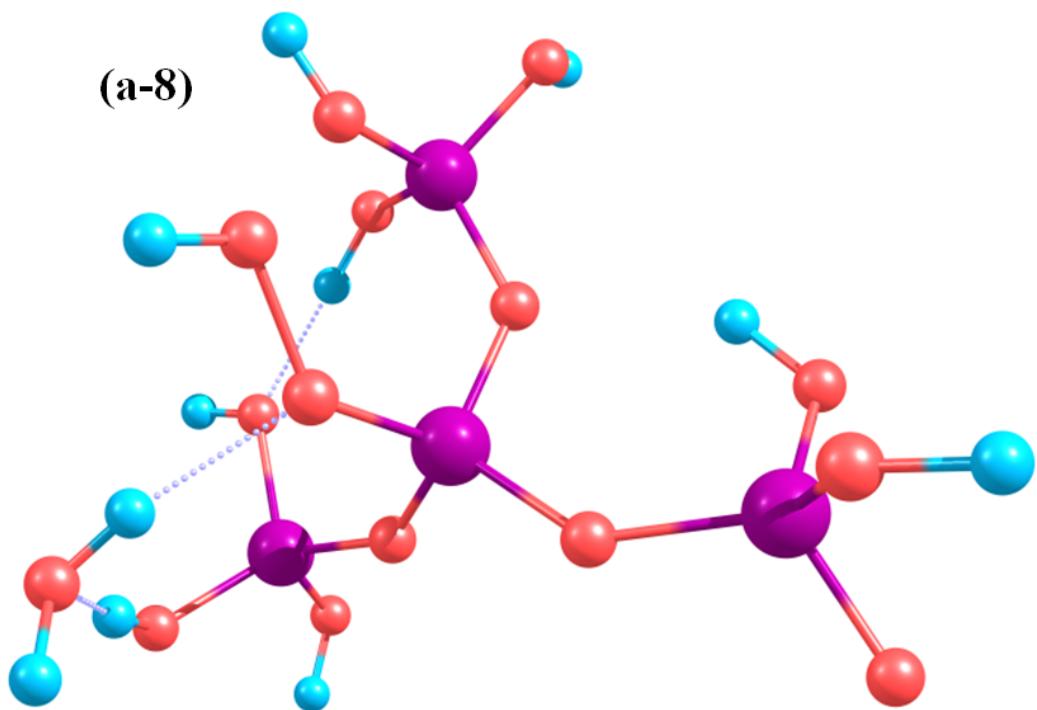
(a-6)

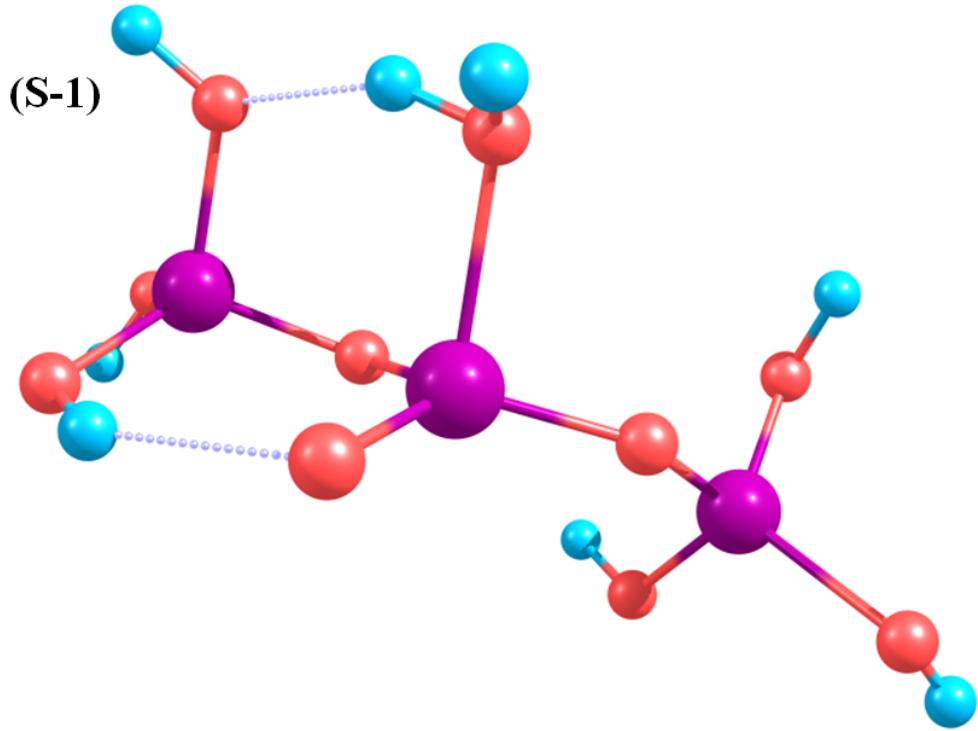


(a-7)



(a-8)





Xyz coords :

Silica trimer (a-1)			Si -0.481 -2.315 -1.515			Si 0.179 -5.135 -1.062		
Si -0.188	0.286	0.066	O 0.378	-3.694	-1.559	O -0.513	-5.003	0.452
O 0.840	0.515	1.347	Si 0.252	-5.289	-1.063	O -2.298	-2.479	-0.655
H 1.534	1.177	1.200	O -0.194	-6.115	-2.427	O -0.252	-6.480	-1.930
O 0.261	-0.888	-1.002	O -0.975	0.929	-1.938	O 1.820	-5.184	-0.816
Si -0.393	-2.320	-1.537	O -1.973	-2.622	-0.375	H 1.486	1.422	0.053
O -1.453	-2.911	-0.375	O -1.584	-1.820	-2.575	H 2.136	-4.628	-0.086
O -1.116	-1.931	-2.982	O 1.702	-5.905	-0.544	H -1.429	-4.678	0.466
H -1.333	-2.689	-3.550	O -0.802	-5.365	0.223	H 0.187	-7.299	-1.650
O 0.748	-3.485	-1.753	H 1.110	2.068	-0.492	H -2.215	-1.646	-0.061
Si 1.084	-4.951	-1.024	H 1.910	-5.722	0.386	H -0.893	0.256	-2.500
O 2.707	-5.181	-0.736	H -1.533	-4.720	0.203	H -2.079	0.506	0.861
H 3.004	-4.842	0.123	H -0.089	-7.079	-2.379	H -3.090	-2.345	-1.215
O 0.294	-5.032	0.438	H -2.023	-1.860	0.264			
H -0.504	-4.461	0.454	H -1.331	0.176	-2.476	TS_Slica_trimer_H ₂ O ₂ _activatio		
O 0.624	-6.075	-2.162	H -2.017	0.654	1.008	n(t-1)		
H 0.945	-6.974	-1.988	H -2.346	-2.229	-1.375	O -0.279	0.584	1.670
H -1.854	-2.188	0.154				Si -0.647	0.239	0.088
O -0.305	1.753	-0.710	Optimized_geometry_after_silic			O -2.280	-0.145	0.047
H -0.577	1.706	-1.641	a_oxoFormation (S-2)			O 0.081	-1.125	-0.506
O -1.632	-0.294	0.706				Si -0.334	-2.263	-1.651
H -1.968	0.170	1.490	O 0.856	0.858	0.530	O 0.458	-3.726	-1.784
			Si -0.208	0.051	-0.441	Si -0.322	-5.136	-2.213
TS_for_silica_to_oxoFormation			O -0.543	0.903	-1.810	O -1.051	-4.697	-3.680
(St-1)			O -1.596	-0.263	0.518	O -0.188	1.522	-0.862
O 0.658	1.494	0.146	O 0.211	-1.531	-0.787	O -1.809	-2.827	-0.985
Si -0.335	0.357	-0.524	Si -0.779	-2.305	-1.906	O -1.067	-1.949	-3.324
O -1.491	-0.072	0.636	O -1.377	-1.417	-3.051	O 0.712	-6.391	-2.524
O 0.406	-1.131	-0.765	O -0.343	-3.876	-2.022	O -1.410	-5.616	-1.056

H	0.478	1.181	1.779	O	3.687	-3.097	-1.485	O	1.229	0.070	1.700		
H	-1.347	-2.831	-3.691	O	2.022	-2.145	-3.411	Si	0.035	-0.034	0.565		
H	0.912	-6.947	-1.753	O	1.907	-4.738	-2.945	O	-1.410	0.381	1.372		
H	-1.856	-4.809	-0.707	O	-0.752	-5.171	-0.357	O	-0.356	-1.554	0.058		
H	-1.600	-5.382	-4.093	O	0.410	-3.624	1.519	Si	-1.554	-1.959	-1.064		
H	-2.331	-2.116	-0.562	O	-3.131	-3.496	-3.757	O	-2.618	-1.356	-2.342		
H	0.266	1.191	-1.669	O	-0.704	-2.456	-4.319	O	-1.078	-1.026	-2.413		
H	-2.885	0.471	0.487	O	3.250	-3.469	1.158	O	0.385	0.983	-0.697		
O	1.105	-1.505	-2.869	H	2.433	-1.281	-3.240	O	-2.898	-1.378	0.182		
H	-0.002	-1.659	-3.601	H	-0.109	-4.321	1.967	O	-1.666	-3.582	-1.030		
O	1.361	-0.050	-2.783	H	0.236	-2.268	-4.103	Si	-1.275	-4.970	-1.882		
H	2.282	-0.078	-2.458	H	0.059	-5.086	-3.382	O	0.300	-4.684	-2.354		
				H	-3.279	-3.580	-4.712	O	-1.306	-6.290	-0.868		
Silica_trimer_product_H ₂ O ₂ _act				H	3.658	-3.238	-0.491	O	-2.356	-5.269	-3.089		
ivation (a-2)				H	2.597	-5.020	-3.568	O	-2.976	-3.041	-4.470		
O	0.506	0.672	1.961	H	3.554	-2.824	1.817	H	2.130	0.170	1.351		
Si	-0.069	0.228	0.472	H	2.267	-3.488	1.250	H	-2.352	-2.787	-5.170		
O	0.924	0.872	-0.687	H	-3.493	-2.361	-0.258	H	-2.127	-6.803	-0.949		
O	-1.683	0.683	0.388	H	-1.862	-0.615	-0.539	H	-2.610	-4.489	-3.668		
O	-0.172	-1.428	0.264	H	-4.974	-2.209	1.724	H	0.721	-5.435	-2.802		
Si	-1.124	-2.213	-0.830	H	-3.521	-2.991	2.320	H	-2.512	-0.678	0.816		
O	-0.913	-1.489	-2.393	H	-4.358	-0.056	2.668	H	-0.043	0.703	-1.533		
O	0.515	-1.193	-2.678	H	-4.019	-1.238	3.936	H	-1.440	1.195	1.897		
O	-0.708	-3.799	-0.927					H	-2.882	-2.340	-3.769		
Si	-0.056	-4.792	-2.093					H	-3.611	-0.965	-0.346		
O	-1.122	-5.255	-3.256	Alternate_mechanism_peroxo_f									
O	-2.742	-2.061	-0.551	romation_TS (t-3)									
O	-2.129	-3.020	-4.422	O	1.131	-0.052	1.898	TS_for_sharpless_epoxidation_					
O	1.208	-3.827	-2.694	Si	0.098	-0.044	0.610	mechanism (t-4)					
O	0.615	-6.148	-1.422	O	-0.273	-1.539	0.008	O	2.443	-0.846	0.222		
H	1.471	0.772	2.001	Si	-1.563	-1.984	-0.965	Si	1.019	-0.871	-0.623		
H	-1.933	-2.758	-5.337	O	-2.933	-1.323	0.025	O	0.217	0.577	-0.209		
H	0.059	-6.941	-1.485	O	0.730	0.977	-0.536	O	-0.105	-1.970	-0.132		
H	-1.568	-4.510	-3.753	O	-1.409	0.470	1.200	Si	-1.610	-2.220	-0.890		
H	1.794	-4.269	-3.330	O	-1.502	-0.952	-2.284	O	-2.726	-1.824	-2.199		
H	-2.962	-1.160	-0.242	O	-1.623	-3.597	-1.016	O	-1.072	-2.529	-2.469		
H	0.947	0.317	-1.500	Si	-1.217	-4.965	-1.906	O	1.340	-1.003	-2.247		
H	-1.868	1.632	0.472	O	-1.154	-6.286	-0.896	O	-2.169	-0.530	-0.065		
H	-1.771	-2.300	-3.849	O	-2.328	-5.286	-3.071	O	-2.349	-3.334	0.063		
H	0.898	-2.116	-2.775	O	0.328	-4.603	-2.424	Si	-2.889	-4.907	0.073		
				O	-3.167	-3.136	-4.403	O	-1.508	-5.811	0.326		
Cyclohexene_epoxidation_by_s				H	2.075	0.003	1.679	O	-3.936	-5.147	1.351		
econdary_salinol (t-2)				H	-2.627	-2.812	-5.141	O	-3.742	-5.280	-1.296		
C	-1.813	0.091	1.578	H	-1.966	-6.818	-0.920	O	-2.316	-4.727	-3.527		
C	-2.266	-0.797	0.458	H	-2.653	-4.514	-3.632	H	3.139	-1.413	-0.145		
C	-3.209	-1.767	0.612	H	0.769	-5.324	-2.900	H	-1.625	-5.371	-3.752		
C	-3.901	-2.035	1.914	H	-2.615	-0.546	0.583	H	-4.864	-5.189	1.069		
C	-3.704	-0.903	2.936	H	0.339	0.896	-1.423	H	-3.256	-5.133	-2.163		
C	-2.244	-0.429	2.958	H	-1.439	1.296	1.709	H	-1.685	-6.739	0.546		
O	-0.905	-2.693	0.924	H	-3.132	-2.417	-3.706	H	-1.372	0.086	0.022		
Si	-0.503	-3.529	-0.511	O	-2.971	-1.439	-2.322	H	0.591	-1.467	-2.696		
O	-1.527	-2.960	-1.714	H	-3.381	-1.056	-0.916	H	0.705	1.413	-0.270		
Si	-1.537	-3.471	-3.305					H	-1.824	-3.899	-3.260		
O	-0.922	-5.040	-3.316	Optimized_geometry_for_perox									
O	1.023	-3.113	-1.042	oFormation (a-4)						H	-2.804	-0.144	-0.705
Si	2.205	-3.272	-2.168					C	-4.909	-0.688	-1.679		

C	-4.868	-1.948	-2.213	H	-7.273	-1.529	-4.056	H	1.544	-0.879	1.453
C	-5.076	0.480	-2.606	H	-4.247	-1.610	-4.587	H	-2.711	2.385	-5.627
C	-6.227	0.240	-3.629	H	-5.318	-2.948	-4.207	H	2.983	3.391	-4.683
C	-6.343	-1.236	-4.087	TS_for_conversion of Si=O to Si-(O-H) (OOH) species (t-5)				H	-0.480	2.823	-5.581
C	-5.115	-2.090	-3.688					H	0.479	5.337	-2.929
H	-4.908	-0.548	-0.593					H	-2.370	-1.951	-1.592
H	-4.866	-2.835	-1.579					H	-1.195	1.357	0.718
H	-4.138	0.613	-3.174					H	-1.824	-2.270	0.773
H	-5.243	1.415	-2.053	O	1.084	-0.263	1.095	H	-1.889	1.264	-4.920
H	-6.061	0.901	-4.492	Si	-0.442	-0.516	0.511	H	-2.446	-2.639	-2.994
H	-7.176	0.559	-3.172	O	-1.472	-0.592	1.809	H	-2.129	-0.453	-3.105
H	-6.482	-1.289	-5.177	O	-0.313	-1.965	-0.368	H	-1.305	3.788	-2.012
H	-7.239	-1.692	-3.639	O	-1.033	0.535	-0.599	Optimized_geometry_after_seco nd_TS_(a-3)			
H	-4.222	-1.737	-4.230	Si	-1.227	0.846	-2.230	C	-2.983	-0.028	3.319
H	-5.255	-3.142	-3.965	O	0.150	1.459	-2.873	C	-2.112	0.307	2.099
Optimized_geometry_after_shar pless_Mechanism (a-5)				Si	0.855	2.444	-4.013	C	-1.933	-0.871	1.172
C	-5.133	-1.886	-3.993	O	0.194	2.276	-5.507	C	-2.862	-2.020	1.213
C	-4.762	-1.763	-2.532	O	-2.691	0.684	-2.948	C	-4.019	-2.067	2.192
C	-5.337	-0.704	-1.675	O	-0.806	-1.013	-2.763	C	-4.283	-0.719	2.887
C	-6.330	0.287	-2.226	O	0.662	3.959	-3.291	O	-1.521	-2.121	1.812
C	-6.293	0.372	-3.761	O	2.490	2.166	-4.132	O	0.444	-3.365	0.509
C	-6.348	-1.026	-4.387	O	-2.475	2.243	-5.286	Si	0.083	-4.085	-0.942
O	-3.961	-0.557	-2.181	H	1.153	0.340	1.852	O	-0.703	-5.521	-0.696
Si	-1.228	-2.490	-0.547	H	-2.582	3.025	-4.704	O	-0.881	-3.058	-1.851
O	-2.244	-3.364	0.398	H	2.728	1.630	-4.906	Si	-1.559	-3.311	-3.350
Si	-3.057	-4.824	0.284	H	-0.815	2.237	-5.535	O	-0.598	-2.759	-4.576
O	-3.918	-4.928	-1.123	H	1.114	4.694	-3.737	O	1.523	-4.321	-1.722
O	0.131	-1.998	0.297	H	-0.572	-1.586	-1.966	Si	2.725	-3.557	-2.560
Si	1.222	-1.058	-0.547	H	-2.389	-0.328	1.627	O	3.542	-4.802	-3.296
O	1.612	-1.753	-1.997	H	0.310	-2.621	-0.016	O	-1.827	-4.973	-3.344
O	2.588	-0.660	0.296	H	-2.678	1.506	-4.654	O	-3.035	-2.555	-3.503
O	0.308	0.372	-0.713	H	-1.675	-1.298	-3.116	O	3.643	-2.540	-1.648
O	-0.924	-2.904	-2.055	H	-2.209	2.626	-1.755	O	2.049	-2.577	-3.780
O	-2.071	-0.863	-0.542	H	-2.870	1.853	-2.368	O	3.072	-2.745	0.952
O	-1.861	-5.965	0.503	H	-2.046	3.727	-2.734	H	2.224	-1.638	-3.597
O	-4.163	-4.943	1.528	H	-1.078	3.937	-2.663	H	-0.317	-2.960	1.026
O	-2.513	-4.506	-3.303	Optimized_geometry_for_OH_ OOHFormation (a-6)				H	0.376	-2.750	-4.385
H	3.323	-1.284	0.186	O	0.923	-1.449	0.971	H	-2.158	-5.354	-4.174
H	-2.005	-5.234	-3.695	Si	-0.323	-0.644	0.235	H	-2.987	-1.725	-4.004
H	-5.078	-4.836	1.225	O	0.037	-0.122	-1.299	H	3.529	-2.607	-0.649
H	-3.391	-4.874	-1.992	Si	-0.603	1.040	-2.292	H	4.305	-4.526	-3.827
H	-2.192	-6.858	0.685	O	-1.334	2.069	-1.141	H	3.482	-3.471	1.451
H	-1.361	-0.159	-0.652	H	0.815	-2.049	3.137	H	2.102	-2.965	0.914
H	0.815	-2.275	-2.307	H	0.697	-0.796	0.608	H	-2.955	-2.622	0.304
H	-1.819	-3.868	-2.889	H	-1.819	-1.521	-1.806	H	-1.404	-0.680	0.232
H	-2.851	-0.735	-1.244	H	-5.369	-0.882	-0.595	H	-4.920	-2.401	1.653
H	-5.369	-0.882	-0.595	H	-4.386	-2.666	-2.041	H	-3.798	-2.848	2.938
H	-6.156	1.273	-1.766	H	-7.329	-0.041	-1.891	H	-4.827	-0.053	2.195
H	-7.329	-0.041	-1.891	H	-5.373	0.888	-4.079	H	-4.945	-0.875	3.752
H	-7.139	0.984	-4.109	H	-6.408	-0.956	-5.483	H	-3.206	0.895	3.874
Tertiary_salinol_model_optimize d (a-7)				O	0.912	3.294	-3.852	Optimized_geometry_after_seco nd_TS_(a-3)			
O	2.539	3.630	-3.853	O	0.364	3.338	-5.398	C	-2.983	-0.028	3.319
O	-1.652	0.392	-3.393	O	0.209	4.414	-2.796	C	-2.112	0.307	2.099
O	-2.741	-1.852	-2.507	O	-1.878	1.890	-5.691	C	-1.933	-0.871	1.172
O	0.364	3.338	-5.398	O	-0.580	1.211	1.123	Optimized_geometry_after_seco nd_TS_(a-3)			

O	-4.318	-2.777	0.108
O	1.783	-6.234	0.422
O	2.319	-4.867	2.661
O	0.299	-0.663	-2.954
H	-0.563	1.493	1.673
H	-1.233	-3.603	-2.102
H	1.966	-4.518	3.496
H	-1.049	-5.390	1.851
H	2.174	-7.026	0.826
H	-0.185	0.292	-1.764
H	-3.409	0.330	-0.415
H	-3.849	-1.901	0.069
H	-4.815	-5.261	1.054
H	-3.760	-4.452	-2.446
H	1.277	-0.687	-3.124
H	0.055	-1.579	-2.685
H	2.776	-2.643	-2.038
H	3.052	-0.337	-1.113
H	4.751	-3.785	-2.731
H	4.524	-2.814	-4.187
H	6.439	-2.308	-1.842
H	6.843	-2.491	-3.551
H	7.010	-0.098	-2.910
H	5.656	-0.295	-4.030
H	5.517	-0.011	-0.970
H	4.892	1.129	-2.168