

Supporting Information for

Carbon Coated Manganese Monoxide Octahedrons Negative-electrode for Lithium-ion Batteries with Enhanced Performances

Huili Cao,^a Xinzhen Wang,^a Hongbo Gu,^b Jiurong Liu,^{*a}
Liqiang Luan,^c Wei Liu,^{*c} Yiran Wang,^d and Zhanhu Guo.^{*d}

^aKey Laboratory for Liquid–Solid Structural Evolution and Processing of Materials, Ministry of Education and School of Materials Science and Engineering, Shandong University, Jinan, Shandong 250061, China

^bDepartment of Chemistry, Tongji University, Shanghai, 200092, China

^cState Key Laboratory of Crystal Materials, Shandong University, Jinan 250100, China

^dIntegrated Composites Lab (ICL), Department of Chemical and Biomolecular Engineering, University of Tennessee, Knoxville, TN 37996 USA

*Corresponding author: E-mail: jrliu@sdu.edu.cn; Tel.: +86-531-88390236

zguo10@utk.edu; Tel: +1-865- 974-2933

This file includes:

- XRD pattern of the as-prepared Mn_3O_4 octahedrons.
- SEM images of the Mn_3O_4 octahedrons prepared with 0.5 g PVP addition at 150 °C over a reaction time of **(a)** 30, **(b)** 45, **(c)** 60, and **(d)** 90 min.
- SEM images of the Mn_3O_4 octahedrons prepared **(a)** without PVP, and with the addition of **(b)** 0.2, **(c)** 1.0 and **(d)** 1.5 g.
- TEM image of as prepared $\text{MnO}@C$ octahedrons.

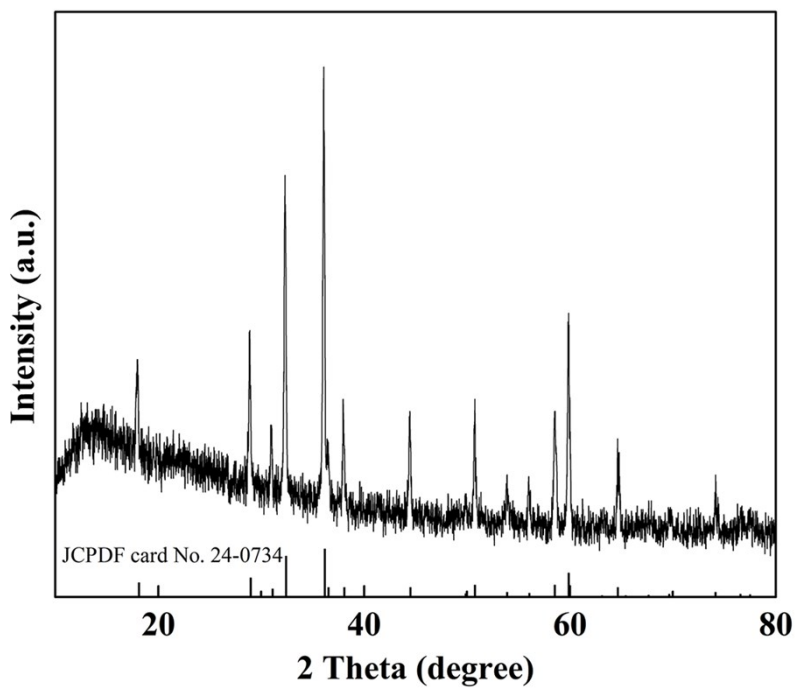


Figure S1. XRD pattern of the as prepared Mn_3O_4 octahedrons.

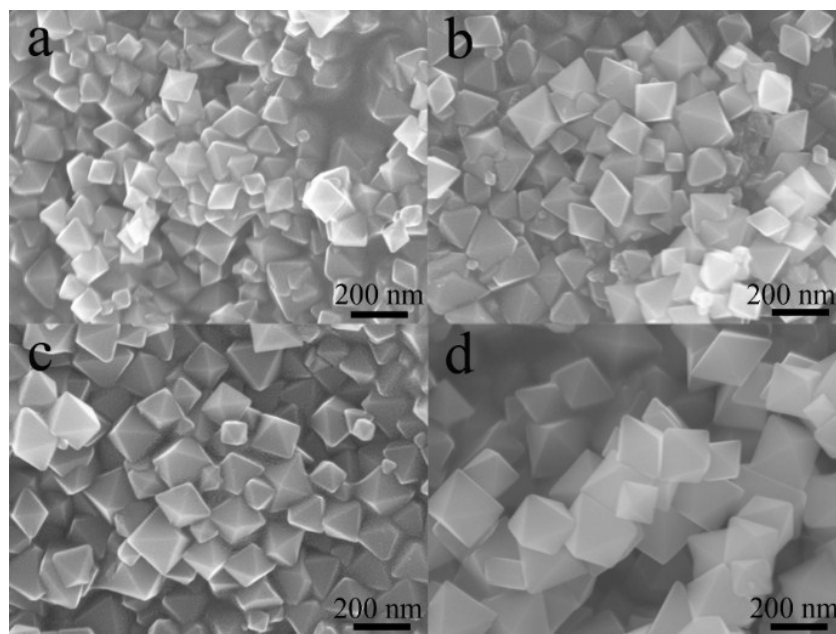


Figure S2. SEM images of the Mn_3O_4 octahedrons prepared with 0.5 g PVP addition at 150 °C over a reaction time of (a) 30, (b) 45, (c) 60, and (d) 90 min.

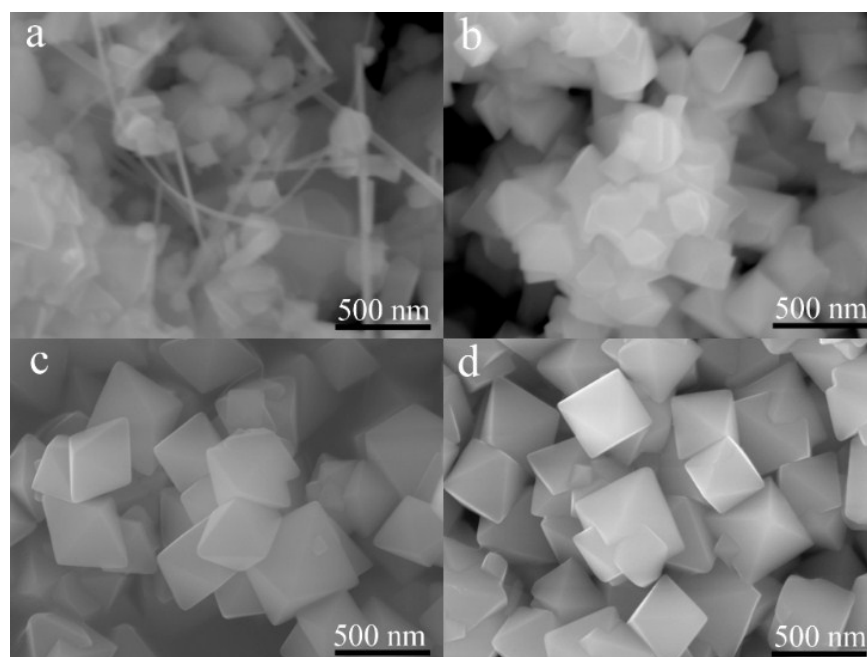


Figure S3. SEM images of the Mn_3O_4 octahedrons prepared (a) without PVP, and with the addition of (b) 0.2, (c) 1.0 and (d) 1.5 g.

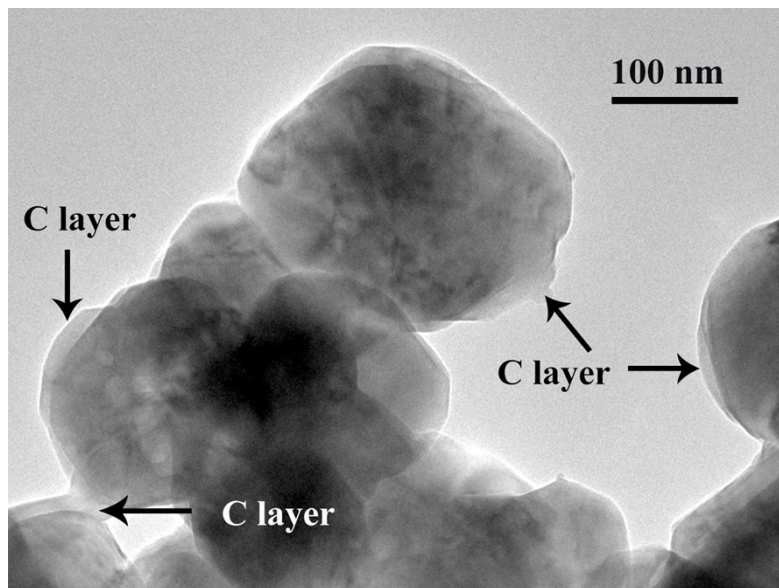


Figure S4. TEM image of as-prepared MnO@C octahedrons.