

Supplementary Materials

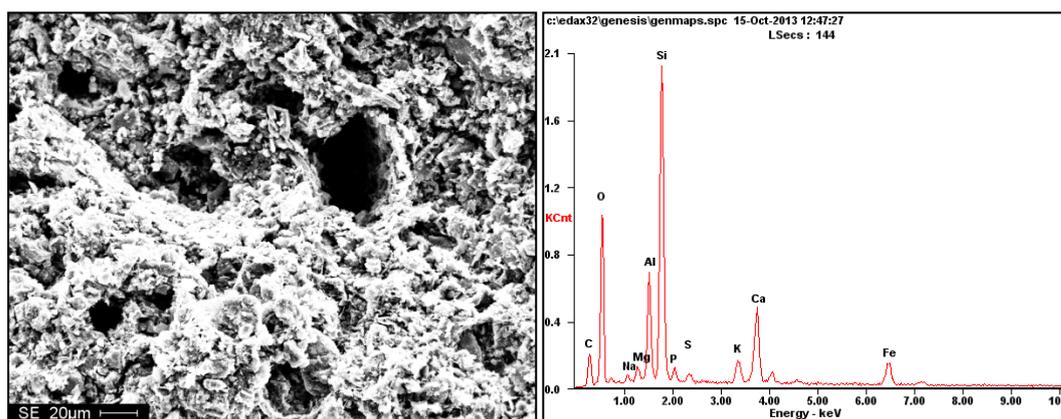
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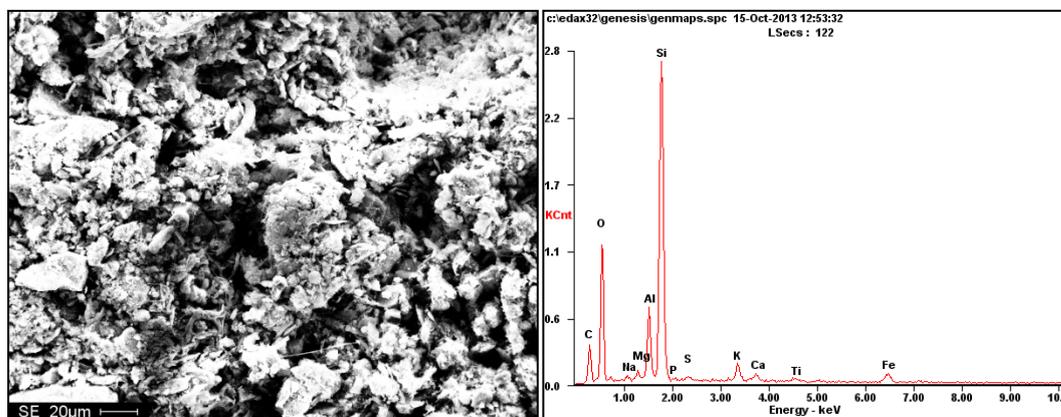
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1. SEM and EDAX



(a) SW



(b) HNO₃-SW

Fig.S1 SEM and EDAX spectrum of SW (a) and HNO₃-SW(b).

2. Specific surface area and pore distribution

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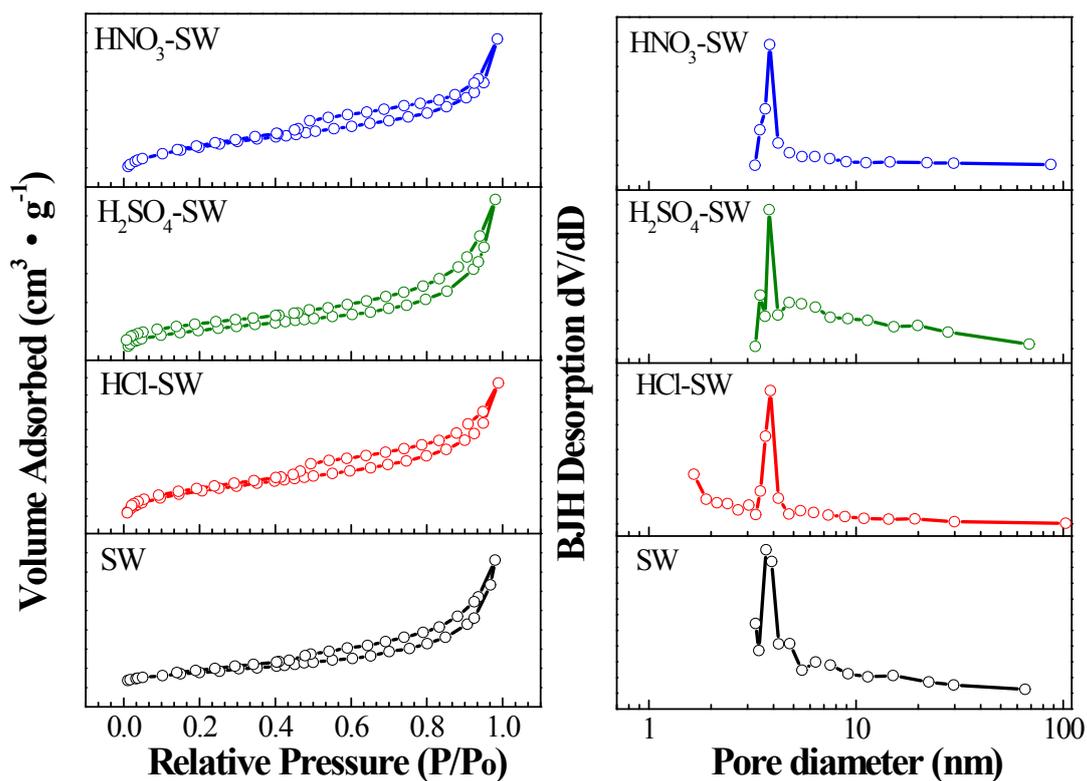


Fig.S2 Nitrogen adsorption/desorption isotherms and pore distribution of SWs.

3. GC/MS and HRMS

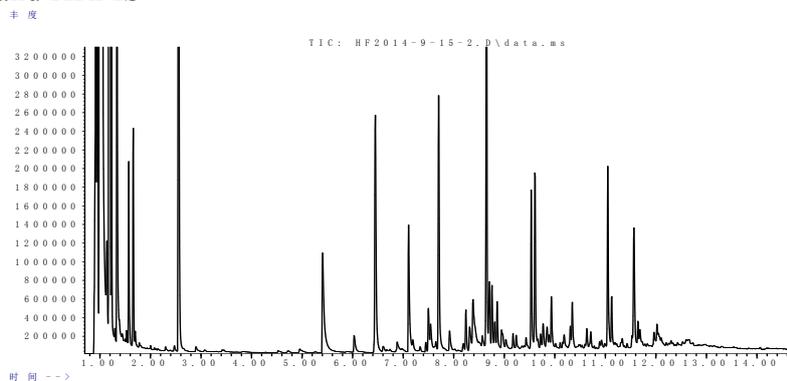
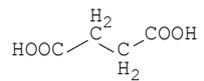
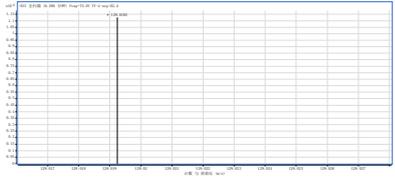
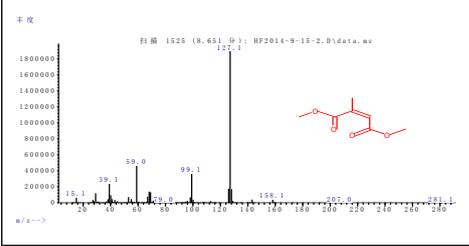
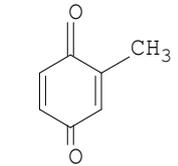
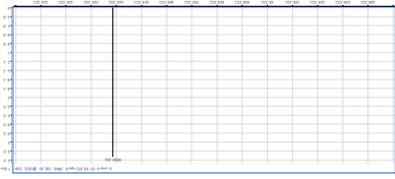
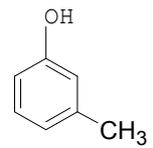
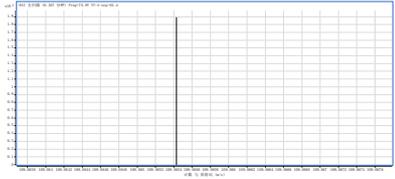
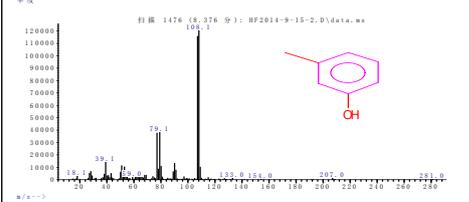
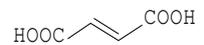
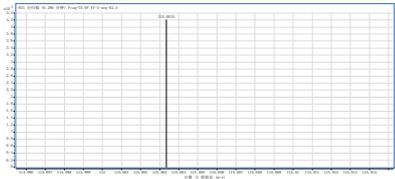
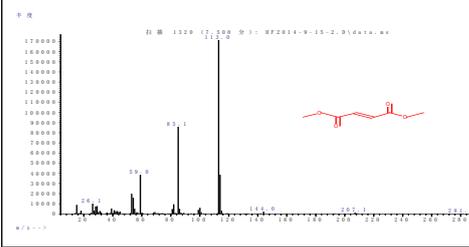
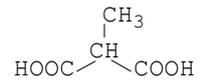
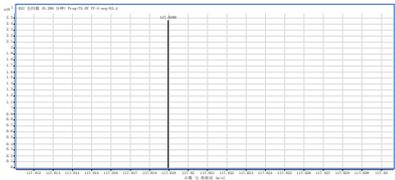
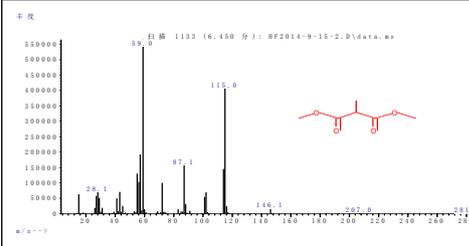
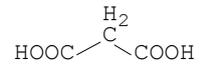
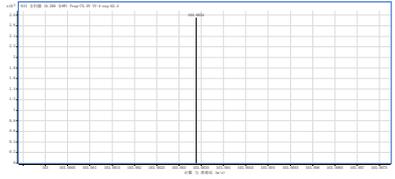
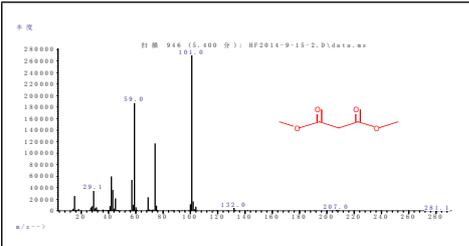
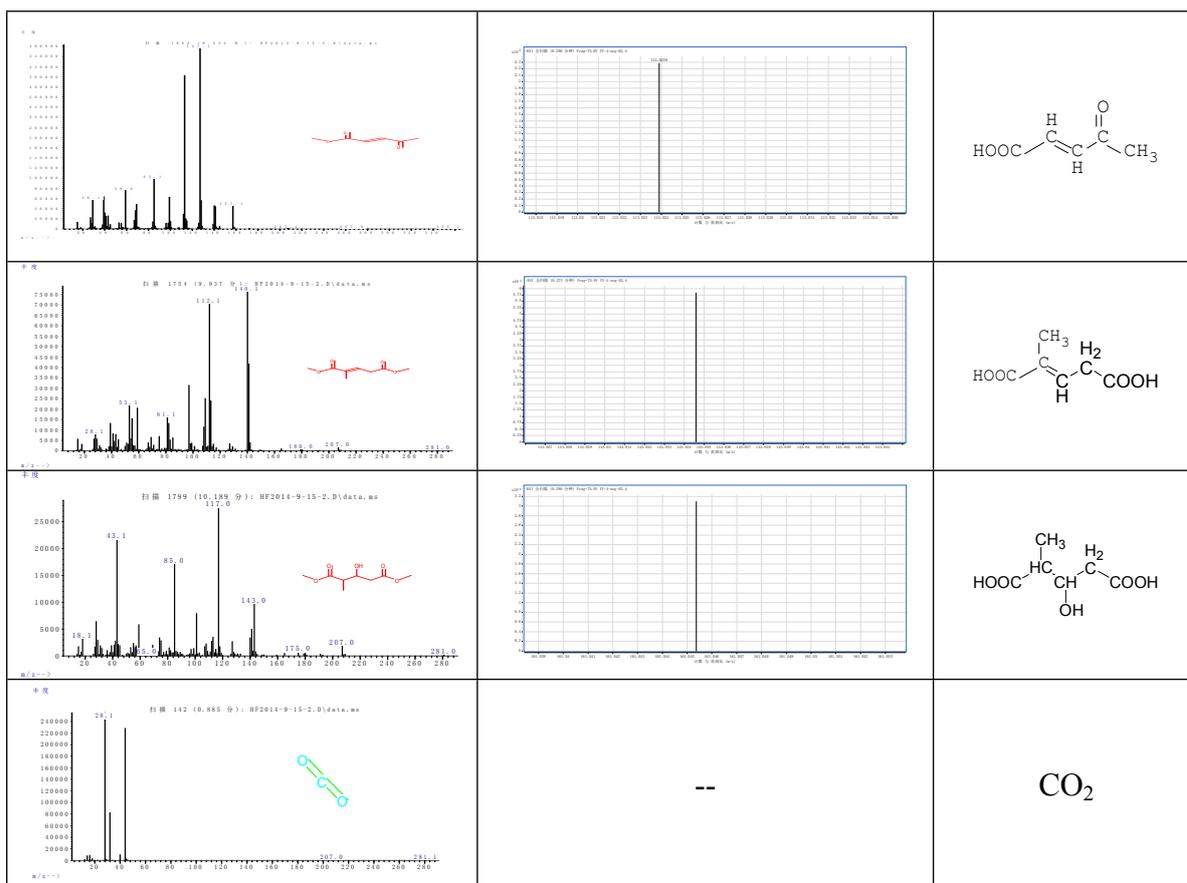


Fig.S3 GC/MS spectrum of m-cresol degradation in CWPO by HNO₃-SW catalyst.

Table S1 Result of GC/MS and HRMS

| GC/MS | HRMS | Low molecular weight acids |
|-------|------|--------------------------------------------------------|
| | | <p>Low molecular weight acids</p> <chem>CC(=O)O</chem> |





4. Ion chromatogram

Ion chromatogram of the effluent by SWs was shown in Fig S4. The retention time of the peak before 3 min was monocarboxylic such as formic acid, acetic acid and propionic acid, while the retention time of the peak after 3 min was dicarboxylic acids. The result showed that more low molecular weight acids were generated by HNO₃-SW catalyst. It indicated that the catalytic activity of HNO₃-SW was better than SW.

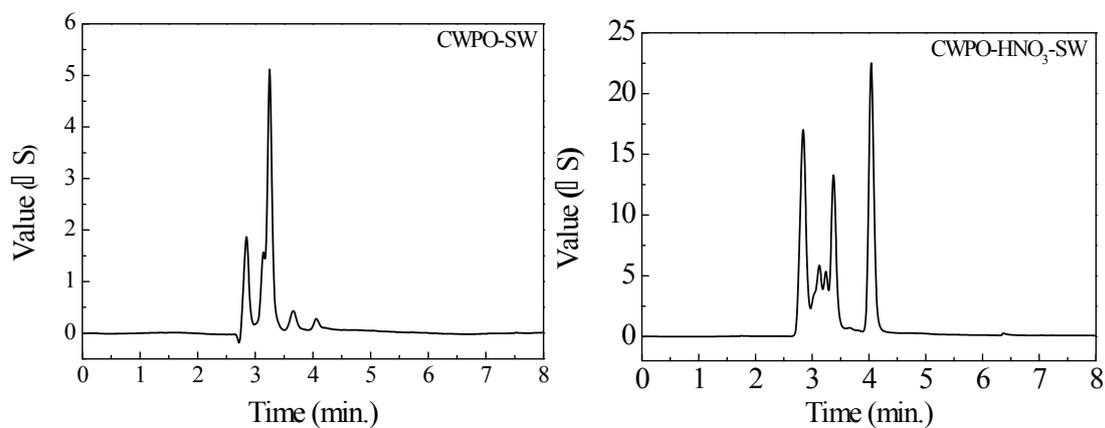


Fig.S4. Total ion chromatogram of the effluent by different SW catalysts: SW (a) and HNO₃-SW (b).

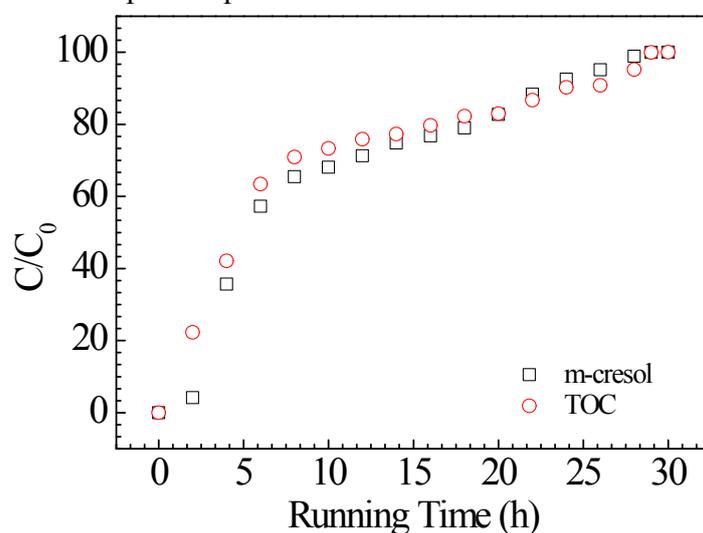
5. XRF

Table S2. XRF analysis of the four SW catalysts.

| Chemical composition | SW | HCl-SW | H ₂ SO ₄ -SW | HNO ₃ -SW |
|--------------------------------|-------|--------|------------------------------------|----------------------|
| C | 13.80 | 10.33 | 16.22 | 12.29 |
| SiO ₂ | 46.32 | 63.91 | 51.81 | 63.26 |
| Al ₂ O ₃ | 15.23 | 15.49 | 17.55 | 14.61 |
| CaO | 9.69 | 0.55 | 1.89 | 1.04 |
| Fe ₂ O ₃ | 5.06 | 1.68 | 3.92 | 2.39 |
| K ₂ O | 1.80 | 2.21 | 1.99 | 1.98 |
| P ₂ O ₅ | 2.06 | 0.35 | 1.73 | 0.35 |

6. Continuous adsorption experiment

The continuous adsorption experiment was carried out by HNO₃-SW catalyst without the addition of any hydrogen peroxide. The results were shown in Fig. S6. It took 29 hours to reach the adsorption equilibrium.

Fig.S6. Continuous adsorption experiment by HNO₃-SW.

7. BET

Table S3. BET surface area of SW catalysts

| Catalysts | S _{BET} (N ₂) | Total pore volume | Average pore diameter |
|------------------------------------|------------------------------------|----------------------------------|-----------------------|
| | m ² ·g ⁻¹ | cm ³ ·g ⁻¹ | nm |
| SW | 22.34 | 0.051 | 3.882 |
| HCl-SW | 80.74 | 0.083 | 3.860 |
| H ₂ SO ₄ -SW | 33.42 | 0.051 | 3.810 |
| HNO ₃ -SW | 52.94 | 0.049 | 3.826 |