

Supporting Information

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4 **Synthesis of Ni/Au/Co Trimetallic Nanoparticles and Their Catalytic Activity for Hydrogen**

5 **Generation from Alkaline Sodium Borohydride Aqueous Solution**

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- 3 Table S1 Comparison of diffraction peaks of Ni₄₅Au₄₅Co₁₀ TNPs with the JCPDS data ([Mⁿ⁺]= 0.66 mM, R_{NaBH4}=5, R_{PVP}=100).
- 5 Table S2 comparison of catalytic activity of NPs with different composition ([Mⁿ⁺]= 0.66 mM, R_{NaBH4}=5, R_{PVP}=100).
- 7 Fig. S1 Comparison of XRD patterns of Ni₄₅Au₄₅Co₁₀ TNPs with Ni, Co and Au (Au, JCPDS: 89-3697; Co, JCPDS: 15-0806; Ni, JCPDS: 88-2326; [Mⁿ⁺]= 0.66 mM, R_{NaBH4}=5, R_{PVP}=100).
- 9 Fig. S2 Catalytic activities of first run and repeated batches (2-7) during the long-time activity test of Ni₄₅Au₄₅Co₁₀ TNPs ([Mⁿ⁺]= 0.66 mM, R_{NaBH4}=5, R_{PVP}=100).
- 11 Fig. S3 Hydrogen volume of first run and repeated batches (2-7) during the long-time activity test of Ni₄₅Au₄₅Co₁₀ TNPs ([Mⁿ⁺]= 0.66 mM, R_{NaBH4}=5, R_{PVP}=100).
- 13 Fig. S4 TEM micrographs of Ni₄₅Au₄₅Co₁₀ TNPs before and after seven catalytic runs ([Mⁿ⁺]= 0.66 mM, R_{NaBH4}=5, R_{PVP}=100).
- 15 Fig. S5 Linear fit of LnK to 1/T of Ni₅₀Au₁₀Co₄₀ catalyst for hydrogen generation from NaBH₄.
- 16 Fig. S6 XPS core level spectra of Au, Co and Ni recorded from Ni₅₀Au₁₀Co₄₀ TNPs ([Mⁿ⁺]= 0.66 mM, R_{NaBH4}=5, R_{PVP}=100).

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Table S1

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Faces	Au/degree	Co/degree	Ni/degree	Peak of Ni ₄₅ Au ₄₅ Co ₁₀ TNPs
111	38.179	44.20	45.50	38.179< 38.42 <44.20 (45.50)
200	44.375	51.52	53.05	44.375< 44.68 < 51.52 (53.05)

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Table S2

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NPs with varied composition		UV-Vis peak /nm	Catalytic activity /mol-H ₂ /(h·mol-M)
MNPs	Au	520	150
	Ni	-	110
	Co	-	80
BNPs	Ni ₅₀ Au ₅₀	492	800
	Co ₅₀ Au ₅₀	492	380
	Ni ₅₀ Co ₅₀	-	80
TNPs	Ni ₅₀ Au ₁₀ Co ₄₀	506	790
	Ni ₅₀ Au ₃₀ Co ₂₀	508	760
	Ni ₃₀ Au ₃₀ Co ₄₀	492	530
	Ni ₃₅ Au ₃₅ Co ₃₀	500	500
	Ni ₄₀ Au ₄₀ Co ₂₀	501	1030
	Ni ₄₅ Au ₄₅ Co ₁₀	509	1170

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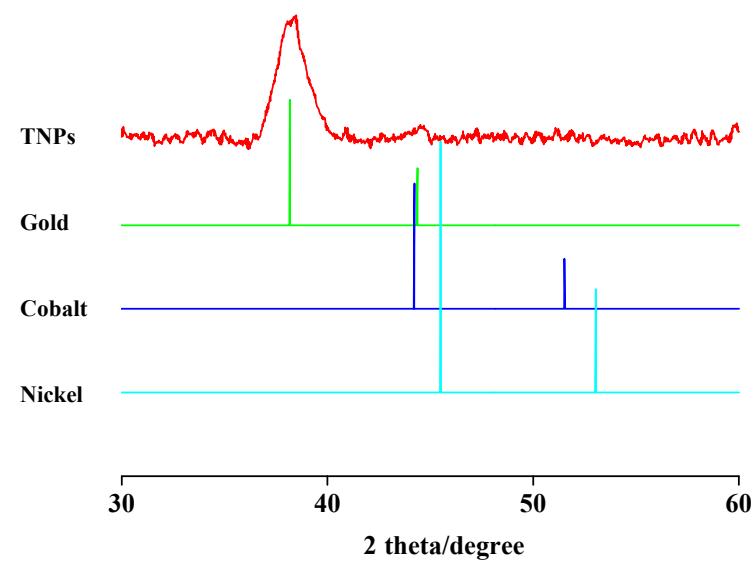


Fig. S1

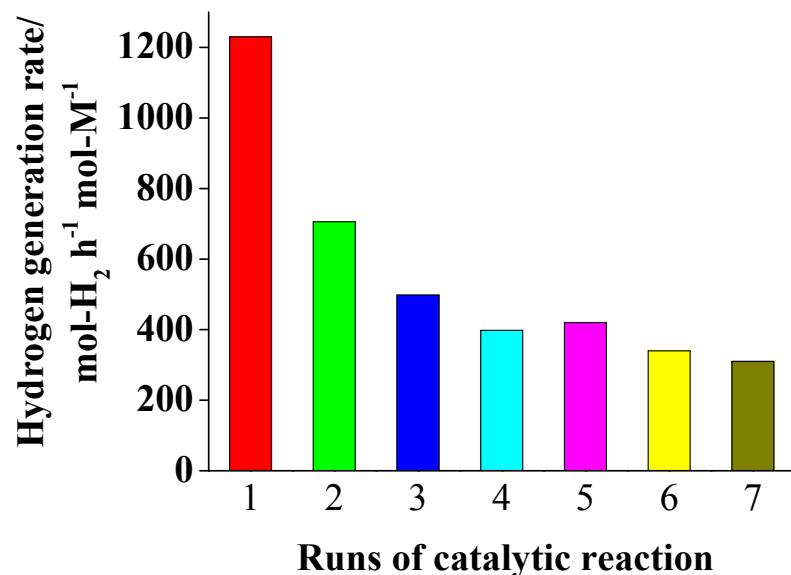


Fig. S2

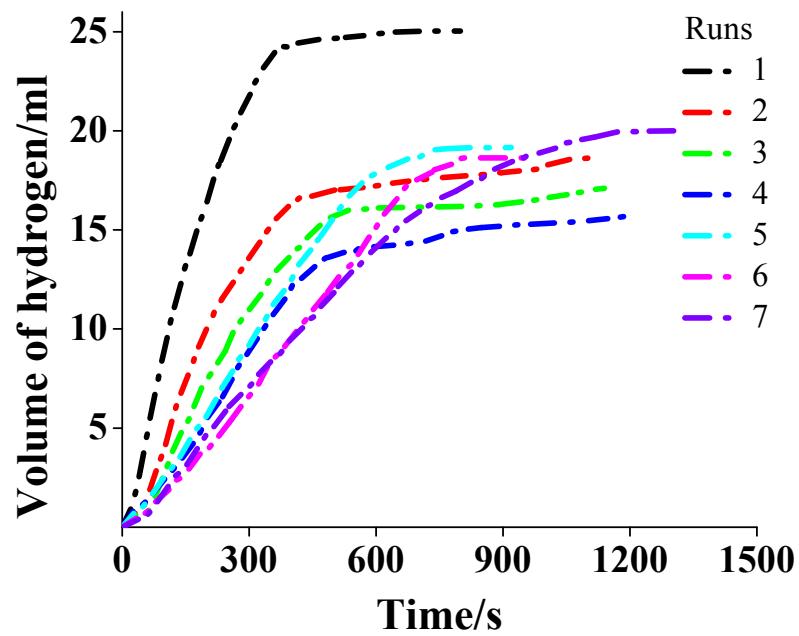


Fig. S3

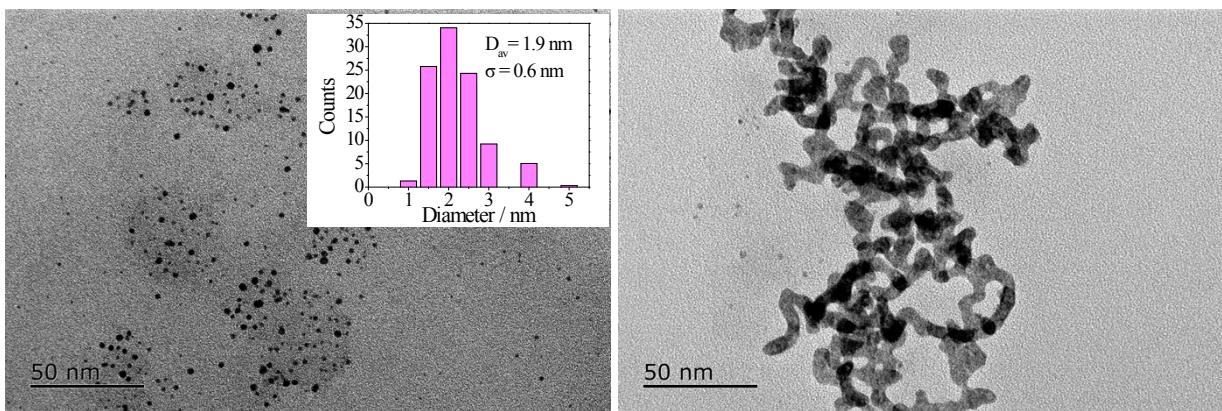


Fig. S4

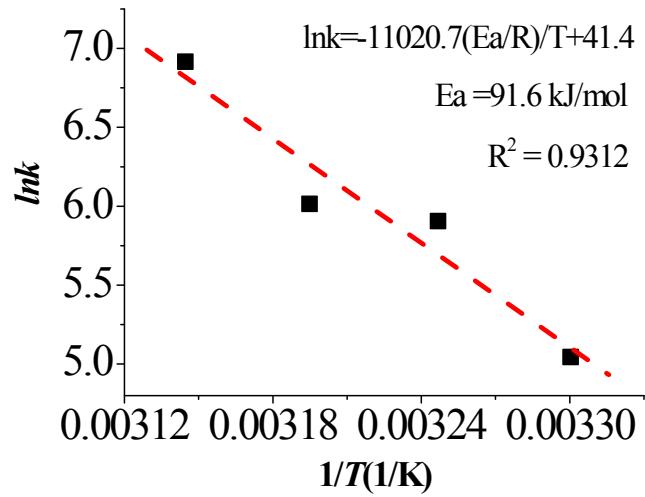


Fig. S5

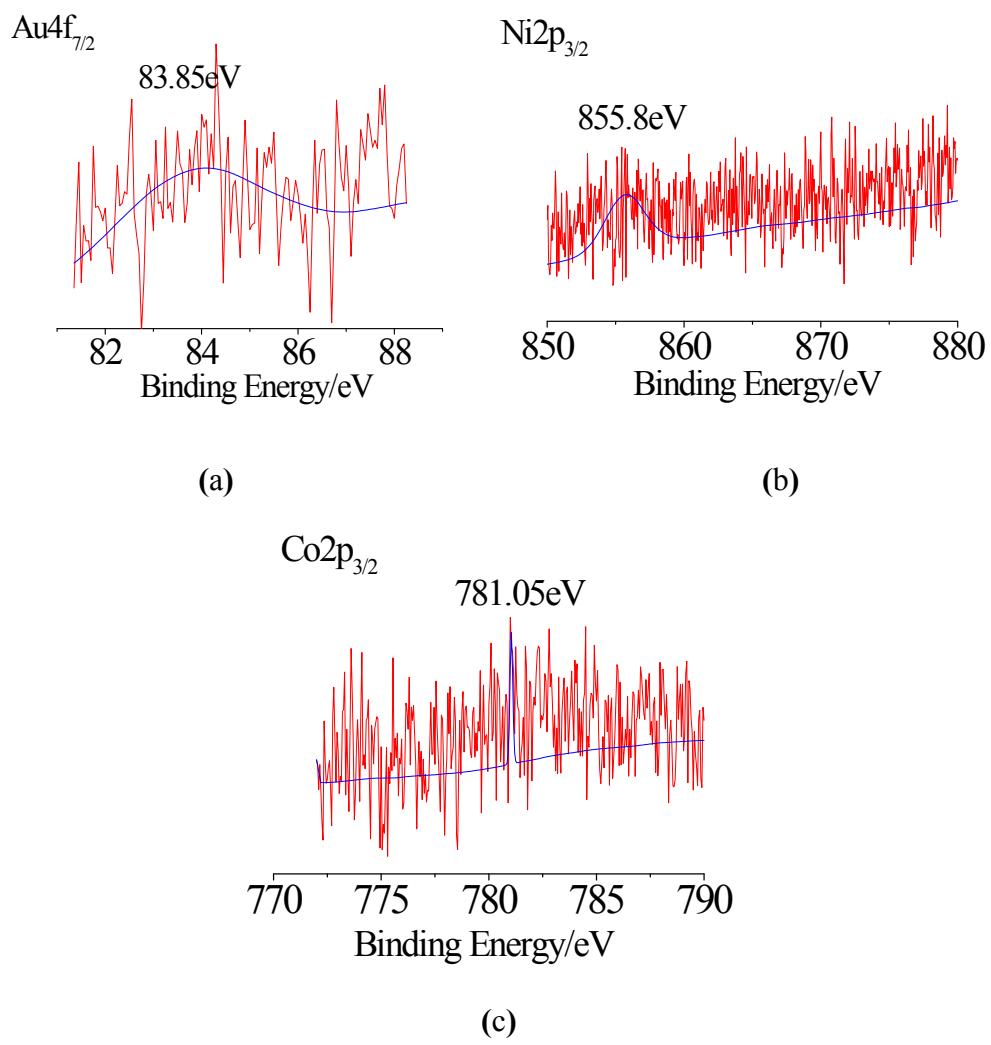


Fig. S6