

Supporting Information

Manganese oxide nanostructures: low-temperature selective synthesis and thermal conversion

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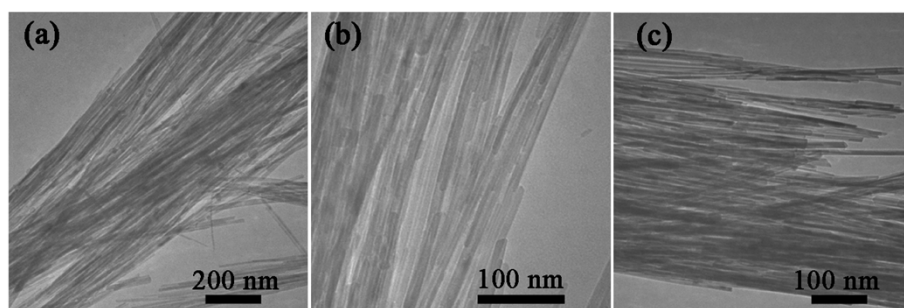


Figure S1. TEM images of the γ -MnOOH prepared at different temperatures: (a) RT, (b) 40 °C, and (c) 80 °C.

Figure S1 shows the TEM images of γ -MnOOH prepared at different temperatures. We can see that the morphologies of γ -MnOOH synthesized at other temperatures (RT, 40 °C, and 80 °C) also have the bundle-like structures.

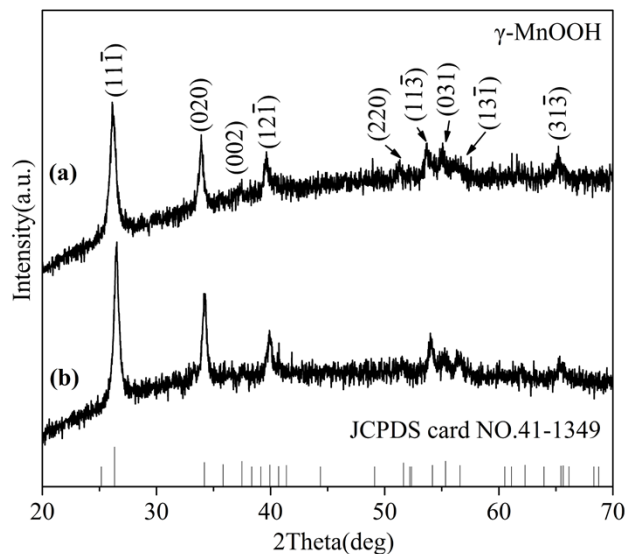


Figure S2. XRD patterns of the products prepared by other caustic alkalis: (a) KOH and (b) $\text{LiOH}\cdot\text{H}_2\text{O}$.

Figure S2 shows XRD patterns of the products prepared by other caustic alkalis. From the XRD patterns, the $\gamma\text{-MnOOH}$ are also obtained when comparative experiments are performed by substituting NaOH with the same mass of KOH and $\text{LiOH}\cdot\text{H}_2\text{O}$.

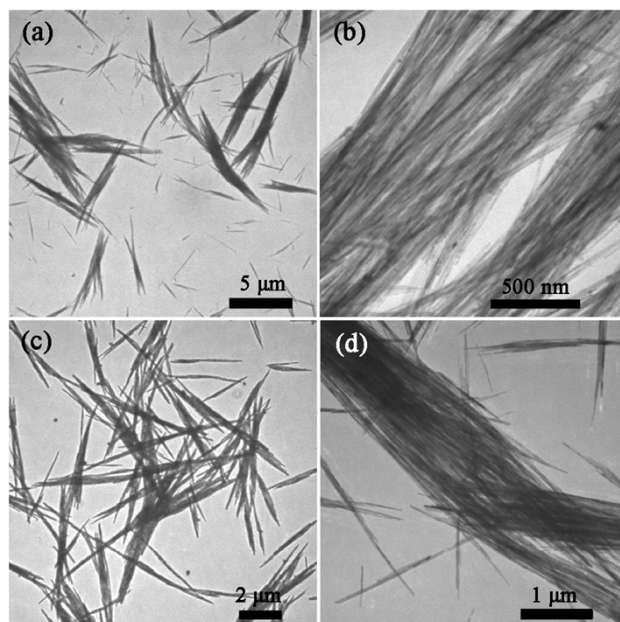


Figure S3. TEM images of the products prepared by other caustic alkalis: (a, b) KOH

and (c, d) $\text{LiOH}\cdot\text{H}_2\text{O}$.

Figure S3 shows the TEM images of the products prepared by other caustic alkalis. Bundle-like $\gamma\text{-MnOOH}$ structures are also obtained when comparative experiments are performed by substituting NaOH with the same mass of KOH and $\text{LiOH}\cdot\text{H}_2\text{O}$.