

Utility of a Heterogeneous Palladium Catalyst for the Synthesis of a Molecular Semiconductor *via* Stille, Suzuki, and Direct Heteroarylation Cross-Coupling Reactions

Seth M. McAfee, Jenny S. J. McCahill, Casper M. Macaulay, Arthur D. Hendsbee
and Gregory C. Welch*

Department of Chemistry, Dalhousie University, 6274 Coburg Road, P.O. Box 15000, Halifax,
Nova Scotia, Canada, B3H 4R2

* Corresponding Author
Email: gregory.welch@dal.ca
Phone Number: 1 (902) 494 4245
Fax Number: 1 (902) 494 1310

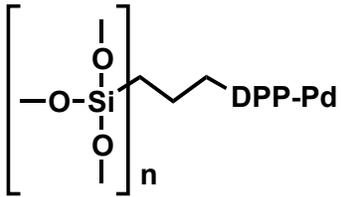
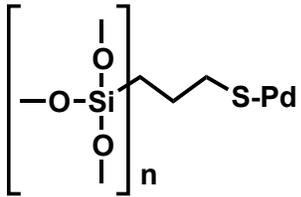
SUPPORTING INFORMATION

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SiliaCat® Heterogeneous Catalysts

Table S1: SiliaCat® Heterogeneous Catalyst Information.

Heterogeneous Catalyst	 DPP-Pd	 S-Pd
Loading	0.2 – 0.4 mmol/g	0.4 – 0.6 mmol/g
Pore Size	25 – 70 Å	
Surface Area	300 – 650 m ² /g	
Particle Size Distribution	60 – 250 μm	

Product information available online: <http://www.silicycle.com/ca/products/siliacat-heterogeneous-catalysts/siliacat-heterogeneous-catalysts-product-range>

Starting Materials Solution ¹H NMR Spectra

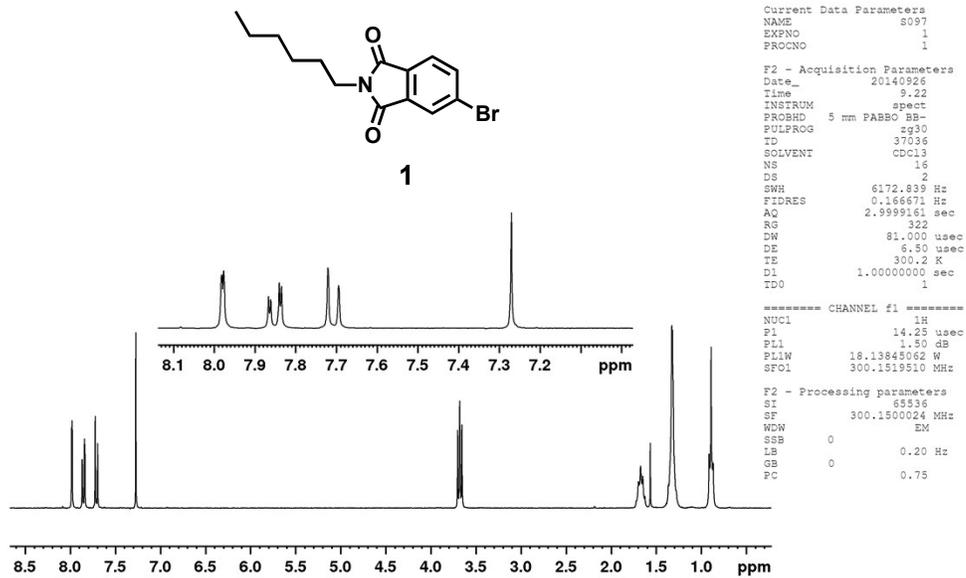


Figure S1: ¹H NMR spectra of starting material **1** in CDCl₃.

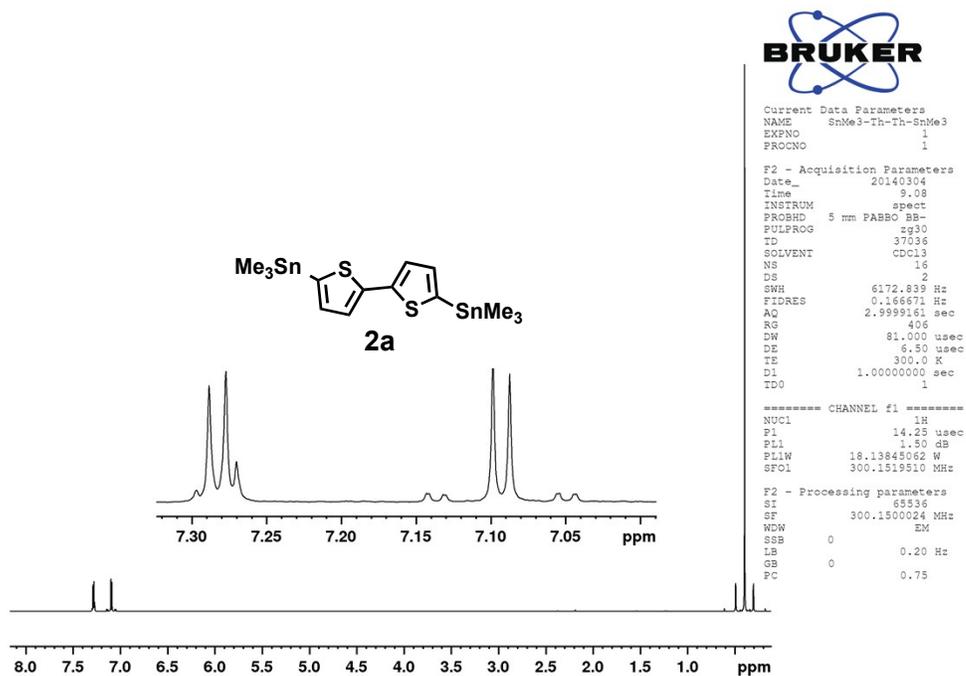


Figure S2: ^1H NMR spectra of starting material **2a** in CDCl_3 .

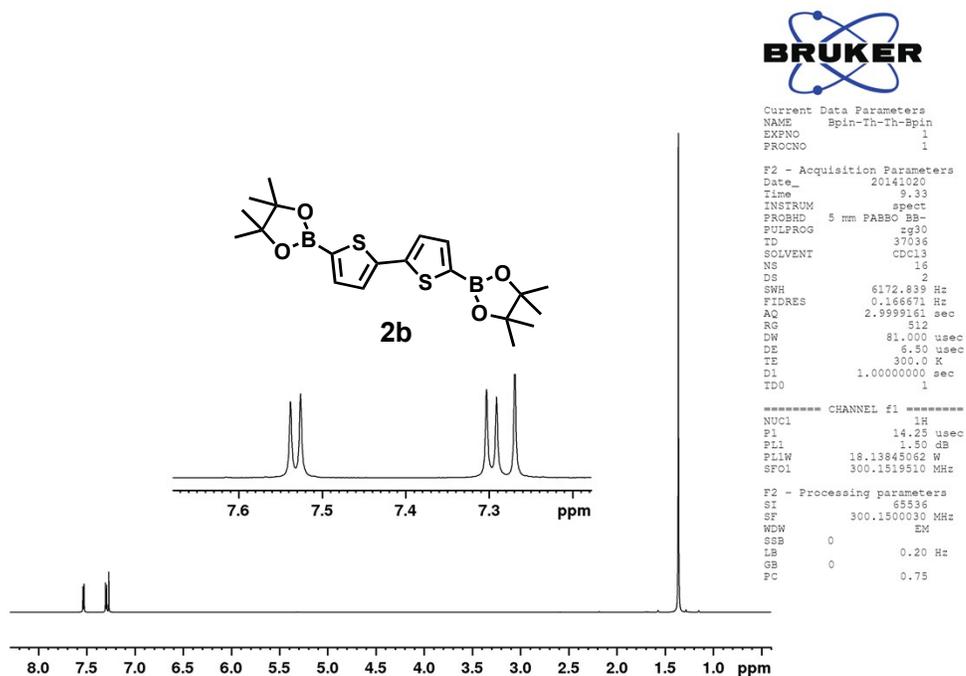


Figure S3: ^1H NMR spectra of starting material **2b** in CDCl_3 .



Current Data Parameters
NAME JMc 8-75
EXPNO 1
PROCNO 1

F2 - Acquisition Parameters
Date_ 20141008
Time 9.47
INSTRUM spect
PROBHD 5 mm PABBO BB-
PULPROG zg30
TD 37036
SOLVENT CDCl3
NS 16
DS 2
SWH 6172.839 Hz
FIDRES 0.166671 Hz
AQ 2.9999161 sec
RG 456
DW 81.000 usec
DE 6.50 usec
TE 300.1 K
D1 1.0000000 sec
TDO 1

===== CHANNEL f1 =====
NUC1 1H
P1 14.25 usec
PL1 1.50 dB
PL1W 18.13845062 W
SFO1 300.1519510 MHz

F2 - Processing parameters
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SF 300.1500026 MHz
WDW EM
SSB 0
LB 0.20 Hz
GB 0
PC 0.75

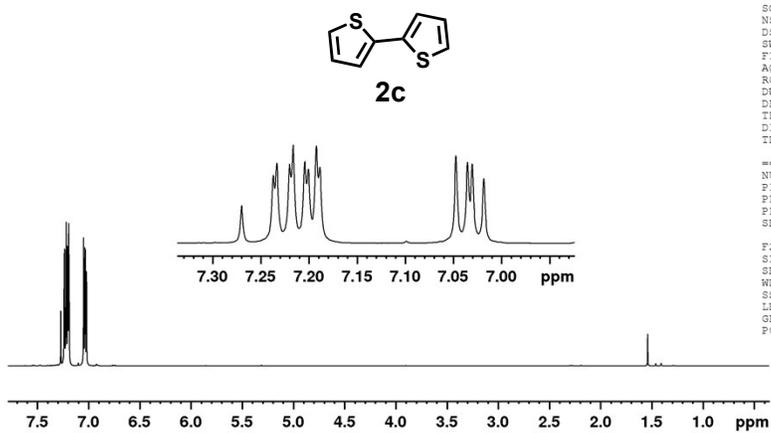


Figure S4: ^1H NMR spectra of starting material **2c** in CDCl_3 .

Product ^1H NMR Spectrum

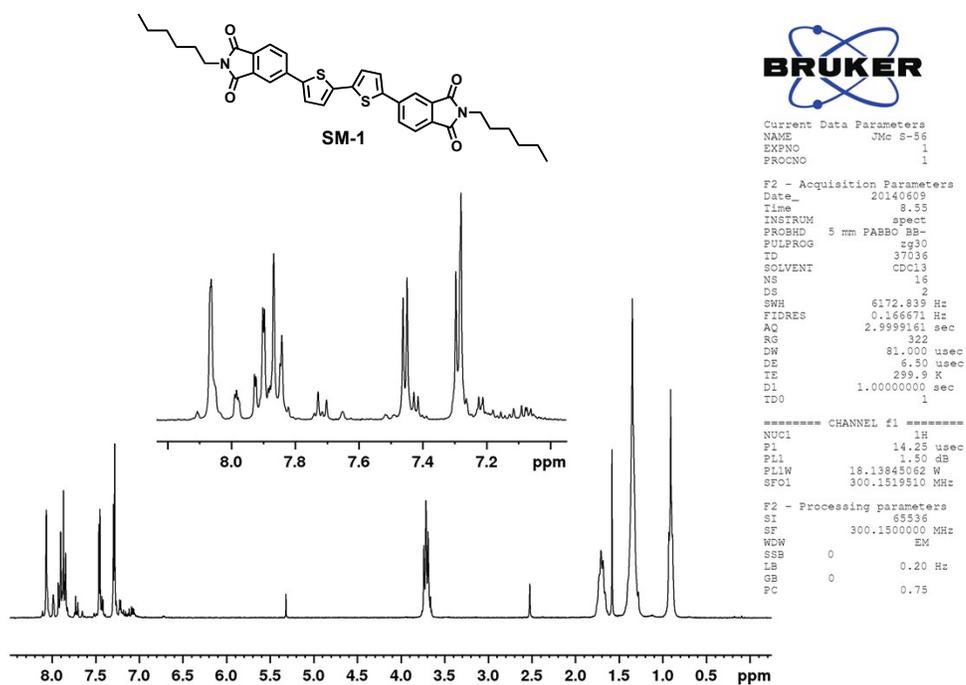
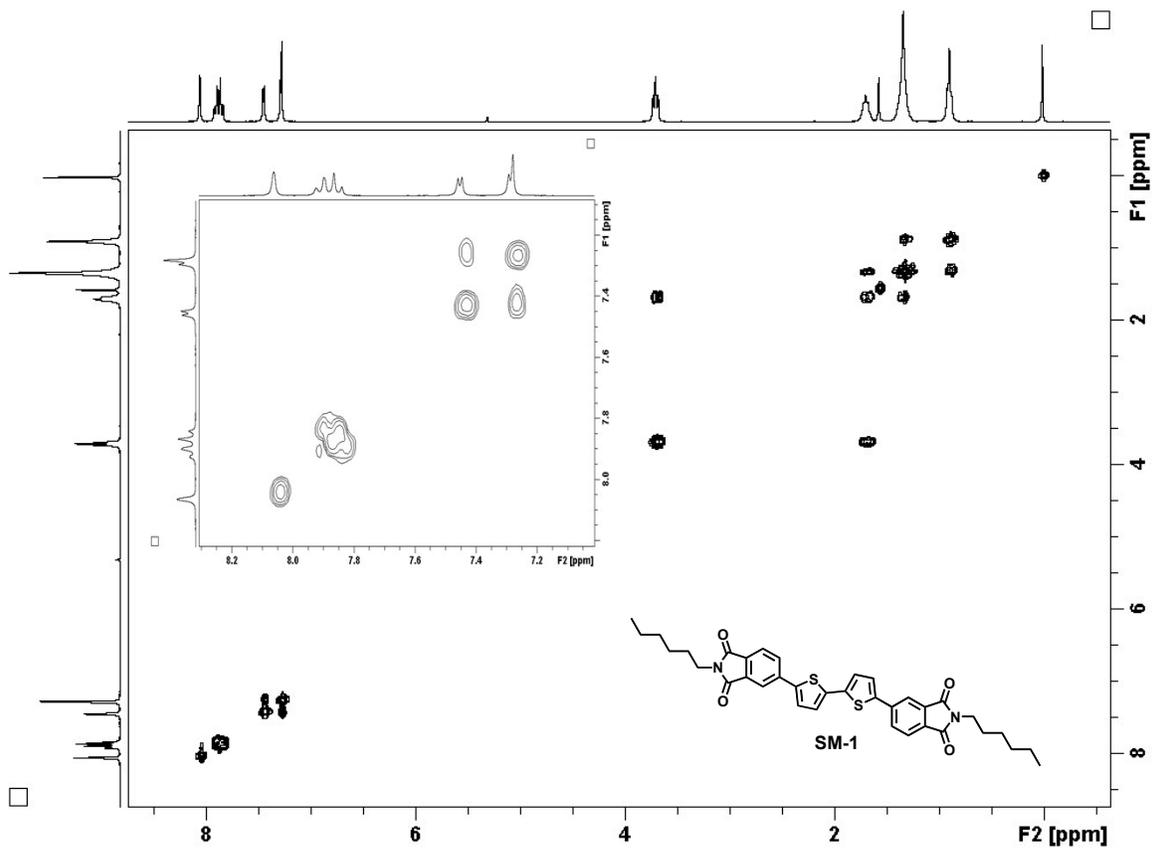


Figure S5: Aromatic region ^1H NMR spectra of **SM-1** in CDCl_3 with 5.0 mol % catalyst loading under ambient conditions *via* conventional heating methods.

Product 2-D ^1H NMR Spectrum



ICP-OES Pd Analysis

Table S1: Trace Pd Analysis for the Synthesis of **SM-1** at 5.0 mol % Catalyst Loading *via* Conventional Heating Method.

Catalyst	Material Yield		Pd Content (mg/L)
Pd(PPh ₃) ₄	0.209 g	82 %	26.25
SiliaCat® DPP-Pd	0.251 g	98 %	0.08*

* Detection limit of the instrument.

Samples were prepared from 10 mg of a given material, which was digested with 1.0 mL of 70 wt. % HNO₃ followed by 1.0 mL of 37 wt. % HCl. The contents were allowed to react in air for several hours and then loosely capped and set to stand overnight. Solid particulates were subsequently removed by filtering the solution through a 0.45 µm PTFE filter and sealed in a new vial.

The prepared samples were submitted to Dalhousie University Minerals Engineering Centre (MEC) for trace Pd analysis. The instrument used was a Varian Vista Pro ICP-OES equipped with an OneNeb nebulizer and a glass cyclonic chamber. The prepared samples were diluted 10-fold into 10 % HCl aqueous solution and measurements were taken with the spectral line 340.458 nm, the preferred line for Pd analysis.

Materials Engineering Centre
Manager - Daniel Chevalier
daniel.chevalier@dal.ca