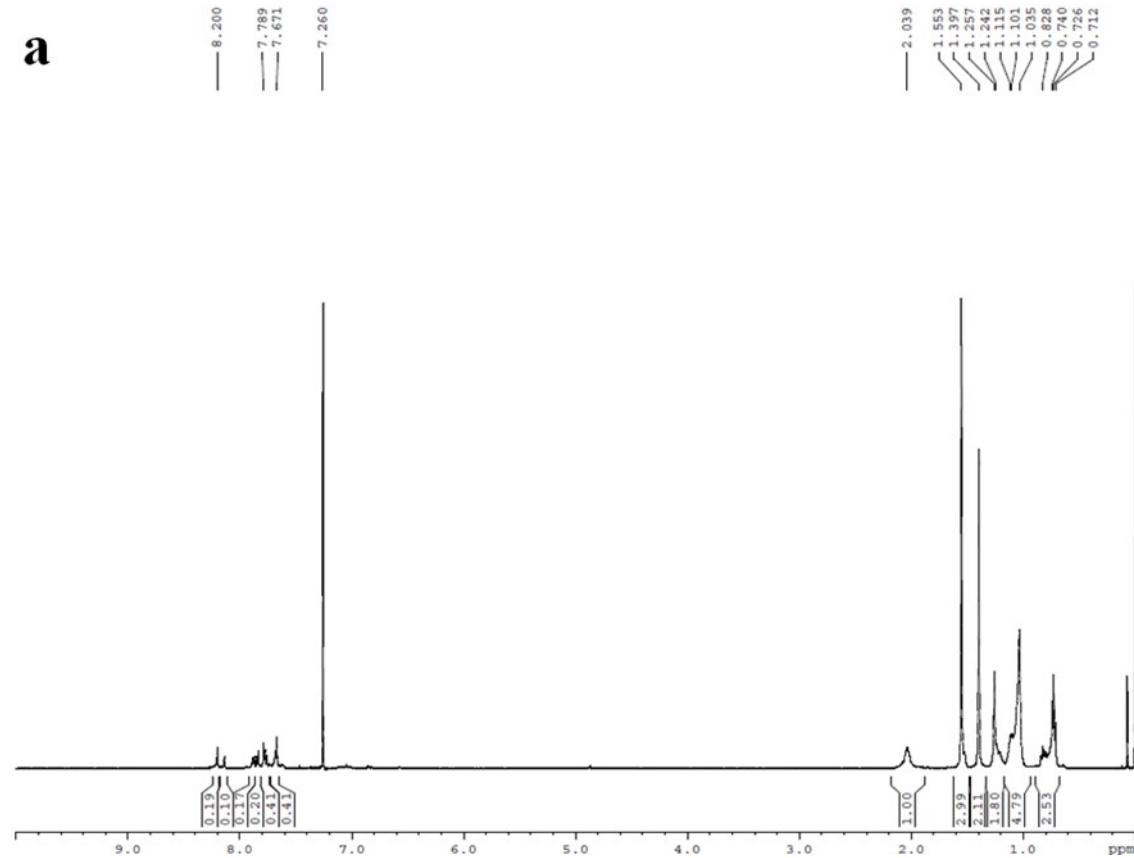


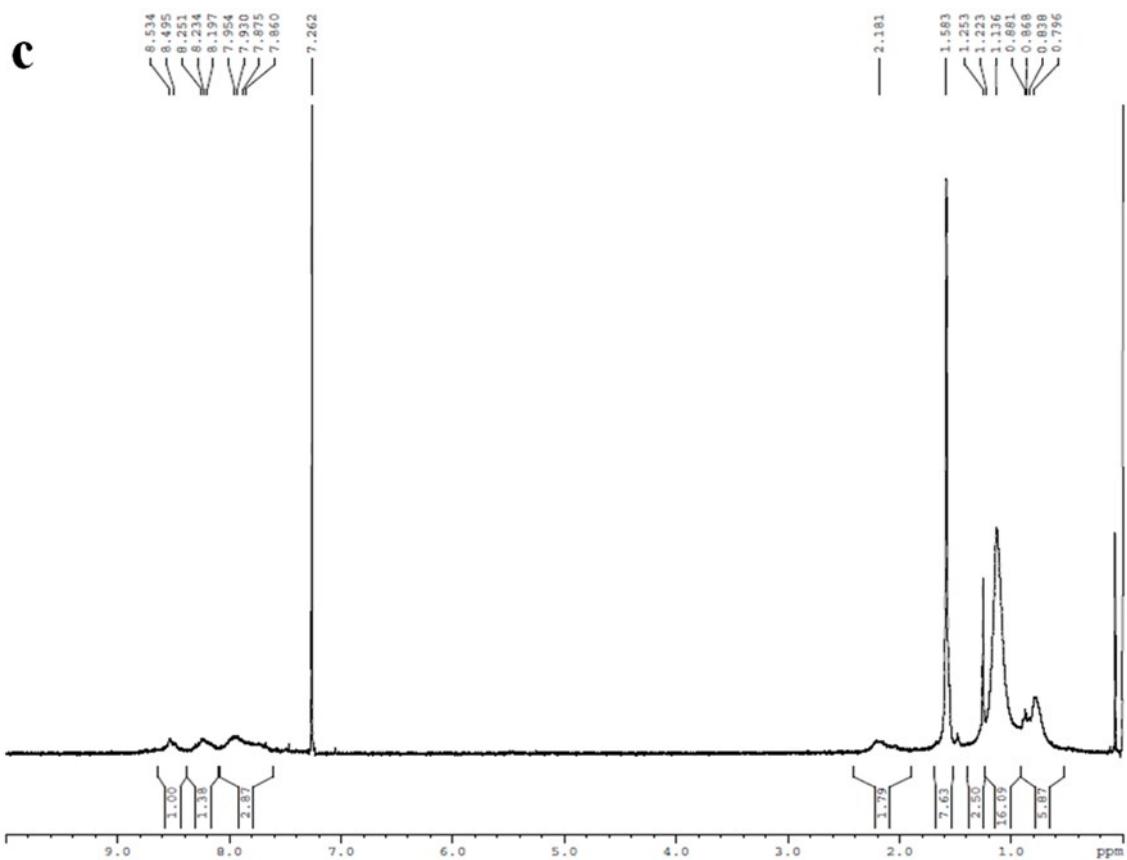
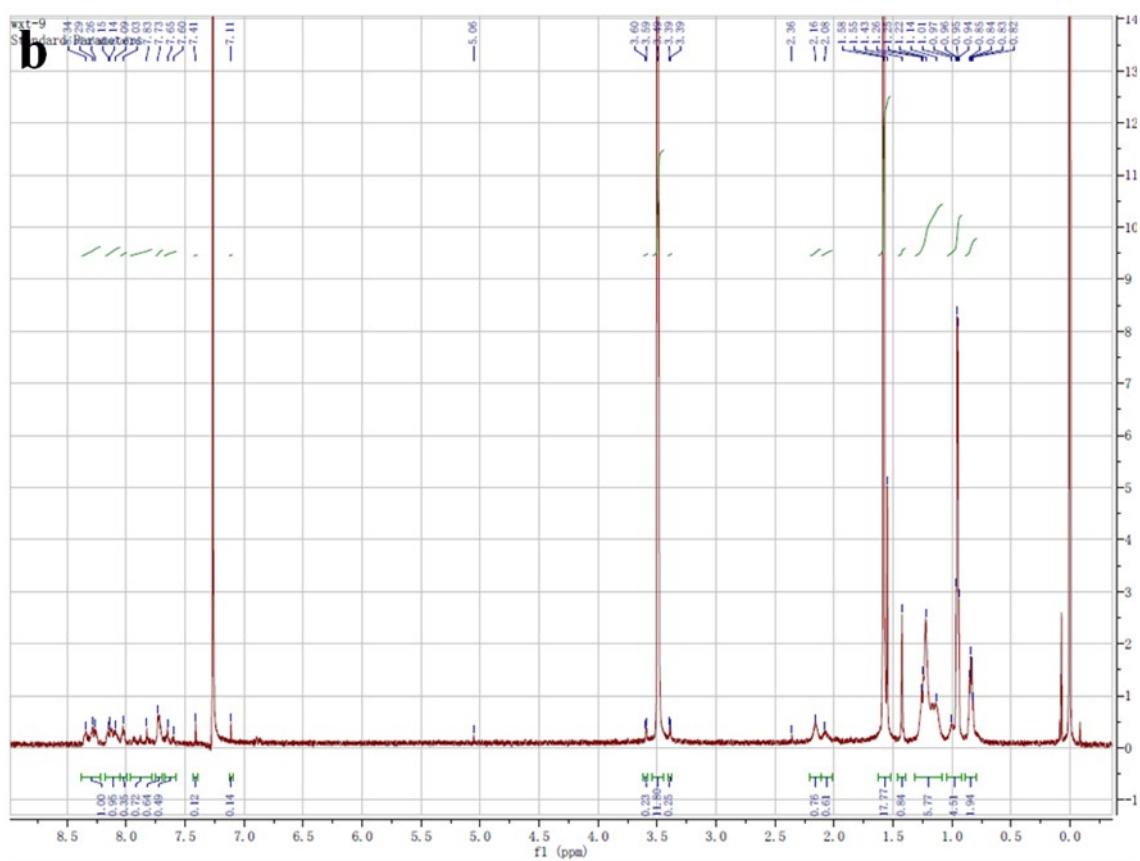
Supporting Information for

**Borate Ester Endcapped Fluorescent Hyperbranched Conjugated Polymer for Trace Peroxide Explosive Vapor Detection**

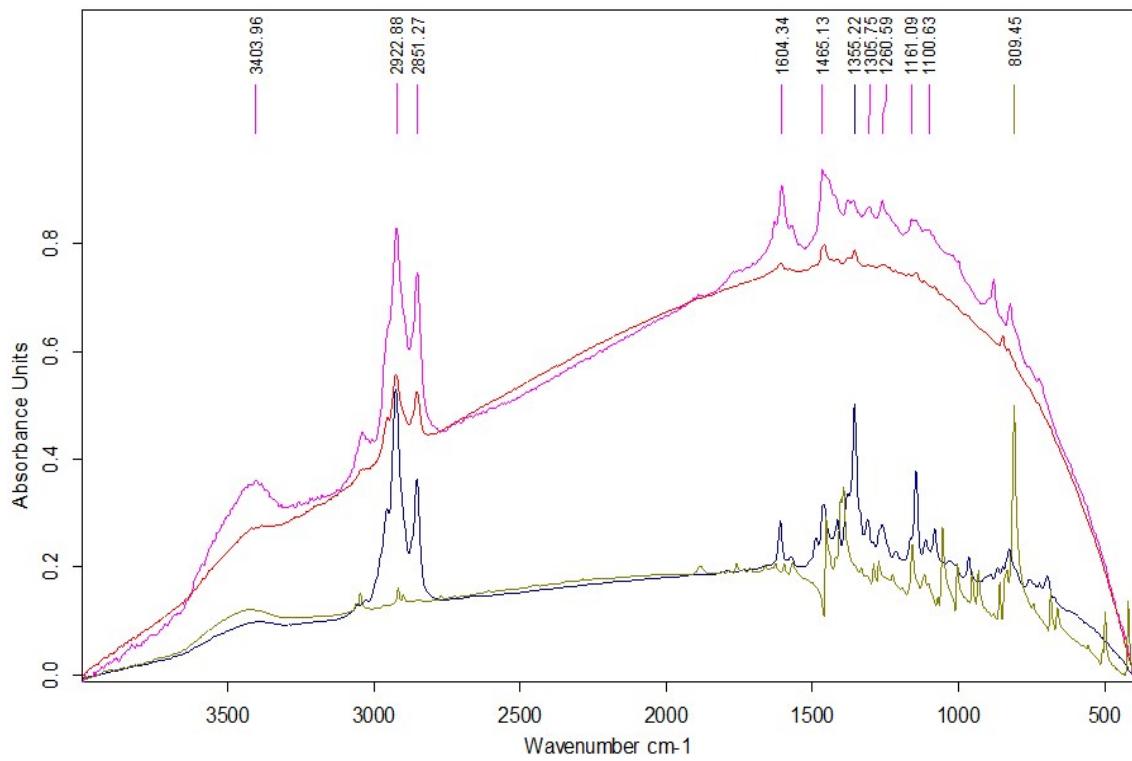
Lei Chen<sup>a,b</sup>, Yixun Gao<sup>a,b</sup>, Yanyan Fu<sup>a</sup>, Defeng Zhu<sup>a</sup>, Qingguo He<sup>a</sup> \*, Huimin Cao<sup>a</sup> and Jiangong Cheng<sup>\*</sup>

**Figures and Tables**





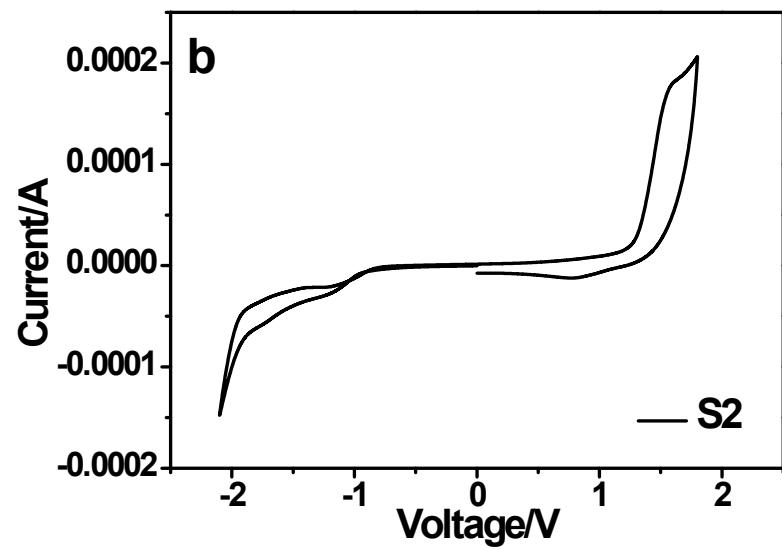
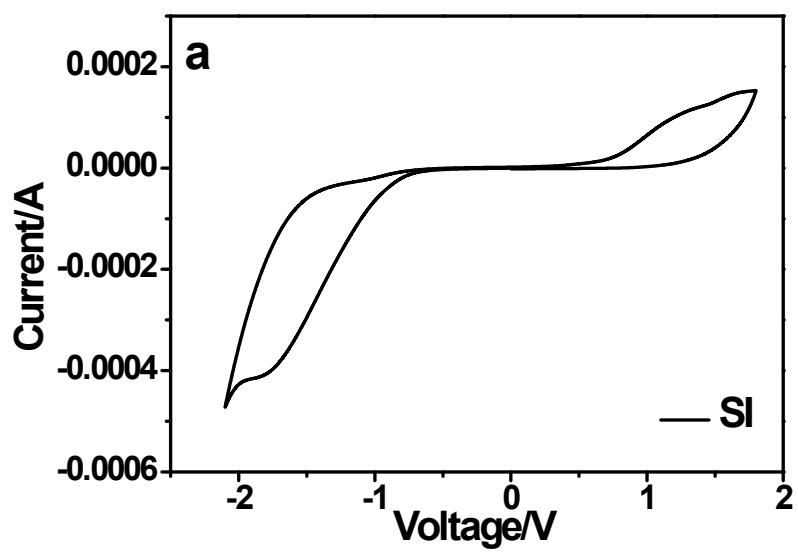
**Figure S1.** The <sup>1</sup>H-NMR spectrum of S1 (a), S2 (b), S3 (c).

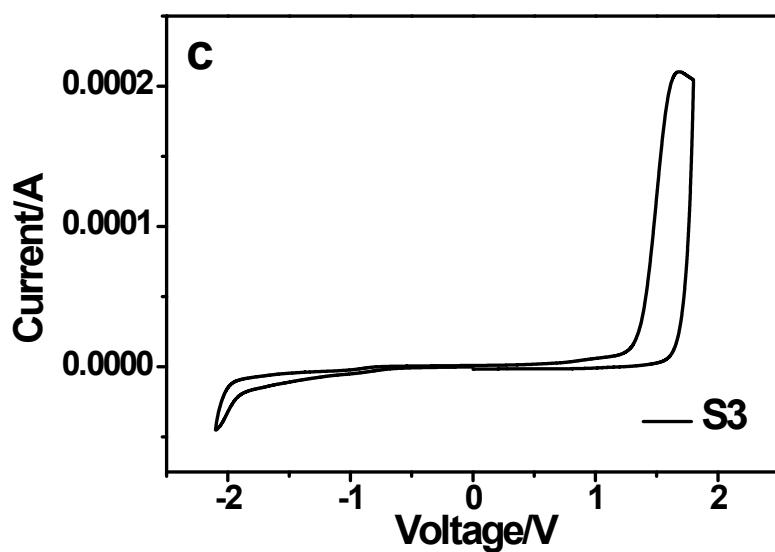


**Figure S2.** Fourier transform infrared spectra of S3, S2, S1 and the contrast (from up to down in the figure)

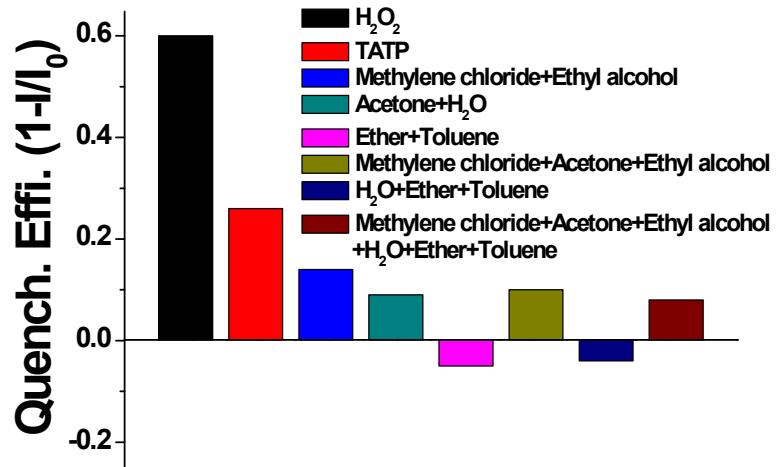
**Table S1.** The GPC results of S1-S3 polymer

	Mn (Da)	Mw (Da)	MP	Mz (Da)	Mz+1(Da)	Polydispersity
<b>S1</b>			78641			
	2875	3514	2739	4946	8140	1.222262
<b>S2</b>	4819	6981	5171	10520	15953	1.448601
<b>S3</b>	8278	27905	13575	73388	126930	3.370933





**Figure S3.** The CV curves of S1 (a), S2 (b), S3 (c) in  $\text{CH}_3\text{CN}$  solution at a sweep rate of 100mV/s.



**Figure S4.** Fluorescence responses of S1 film on ZnO nanorod array to some reaction mixtures saturated vapor after an exposure of 300s.

**Table S2** The quenching efficiency ( $1-I/I_0$ ) of S1 film on ZnO nanorod array to different vapor pressure of  $\text{H}_2\text{O}_2$  after an exposure of 300s.

Diluted Times	Vapor Pressure (ppm)	Quenching Efficiency
1	225	0.6
5	37.5	0.3
10	18.8	0.25
90	1	0.1