

Supplementary Informations

Figure Captions

SI Fig. 1 Plot of k against complex ion for Cis-[Co(dpq)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ under various temperatures; Cis-[Co(dpq)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ = 4 x 10⁻⁴ mol dm⁻³, μ = 1.0 mol dm⁻³, [Fe²⁺] = 0.01 mol dm⁻³

SI Fig. 2 Plot of k against complex ion for Cis-[Co(dpqc)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ under various temperatures; Cis-[Co(dpqc)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ = 4 x 10⁻⁴ mol dm⁻³, μ = 1.0 mol dm⁻³, [Fe²⁺] = 0.01 mol dm⁻³

SI Fig. 3 Plot of k against [β-CD] for Cis-[Co(dpq)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ at various temperatures; Cis-[Co(dpq)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ = 4 x 10⁻⁴ mol dm⁻³, μ = 1.0 mol dm⁻³, [Fe²⁺] = 0.01 mol dm⁻³

SI Fig. 4 Plot of k against [β-CD] for Cis-[Co(dpqc)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ at various temperatures; Cis-[Co(dpqc)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ = 4 x 10⁻⁴ mol dm⁻³, μ = 1.0 mol dm⁻³, [Fe²⁺] = 0.01 mol dm⁻³

SI Fig. 5 Eyring plot for Cis-[Co(dpq)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ in aqueous medium. [Complex] = 4 x 10⁻⁴ mol dm⁻³; [Fe²⁺] = 0.01 mol dm⁻³; [μ] = 1.0 mol dm⁻³.

SI Fig. 6 Eyring plot for Cis-[Co(dpqc)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ in aqueous medium. [Complex] = 4 x 10⁻⁴ mol dm⁻³; [Fe²⁺] = 0.01 mol dm⁻³; [μ] = 1.0 mol dm⁻³.

SI Fig. 7 Eyring plot for Cis-[Co(dpq)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ in [β-CD] medium. [Complex] = 4 x 10⁻⁴ mol dm⁻³; [Fe²⁺] = 0.01 mol dm⁻³; [μ] = 1.0 mol dm⁻³, [β-CD] = 1.8 x 10⁻³.

SI Fig. 8 Eyring plot for Cis-[Co(dpqc)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ in [β-CD] medium. [Complex] = 4 x 10⁻⁴ mol dm⁻³; [Fe²⁺] = 0.01 mol dm⁻³; [μ] = 1.0 mol dm⁻³, [β-CD] = 1.8 x 10⁻³.

SI Fig. 9 Isokinetic plot of the activation parameters for the reduction of Cis-[Co(dpq)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ by ion(II) in aqueous solutions. [Complex] = 4 x 10⁻⁴ mol dm⁻³; [Fe²⁺] = 0.01 mol dm⁻³; [μ] = 1.0 mol dm⁻³.

SI Fig. 10 Isokinetic plot of the activation parameters for the reduction of Cis-[Co(dpqc)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ by ion(II) in aqueous solutions. [Complex] = 4 x 10⁻⁴ mol dm⁻³; [Fe²⁺] = 0.01 mol dm⁻³; [μ] = 1.0 mol dm⁻³.

SI Fig. 11 Isokinetic plot of the activation parameters for the reduction of Cis-[Co(dpq)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ by ion(II) in β-cyclodextrin medium. [Complex] = 4 x 10⁻⁴ mol dm⁻³; [Fe²⁺] = 0.01 mol dm⁻³; [μ] = 1.0 mol dm⁻³.

SI Fig. 12 Isokinetic plot of the activation parameters for the reduction of Cis-[Co(dpqc)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ by ion(II) in β-cyclodextrin medium. [Complex] = 4 x 10⁻⁴ mol dm⁻³; [Fe²⁺] = 0.01 mol dm⁻³; [μ] = 1.0 mol dm⁻³.

Table Captions

SI Table 1. Second-order rate constants for the reduction of cobalt(III) complex ion by Fe²⁺ in aqueous solutions under various temperatures. Cis-[Co(dpq)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ = 4 x 10⁻⁴ mol dm⁻³, μ = 1.0 mol dm⁻³, [Fe²⁺] = 0.01 mol dm⁻³

SI Table 2. Second-order rate constants for the reduction of cobalt(III) complex ion by Fe²⁺ in aqueous solutions under various temperatures. Cis-[Co(dpqc)₂(C₁₂H₂₅NH₂)₂](ClO₄)₃ = 4 x 10⁻⁴ mol dm⁻³, μ = 1.0 mol dm⁻³, [Fe²⁺] = 0.01 mol dm⁻³

SI Table 3. Second-order rate constants for the reduction of cobalt(III) complex ion by Fe^{2+} in the presence of β -cyclodextrin under various temperatures. $\text{Cis-}[\text{Co}(\text{dpq})_2(\text{C}_{12}\text{H}_{25}\text{NH}_2)_2](\text{ClO}_4)_3 = 4 \times 10^{-4} \text{ mol dm}^{-3}$, $\mu = 1.0 \text{ mol dm}^{-3}$, $[\text{Fe}^{2+}] = 0.01 \text{ mol dm}^{-3}$

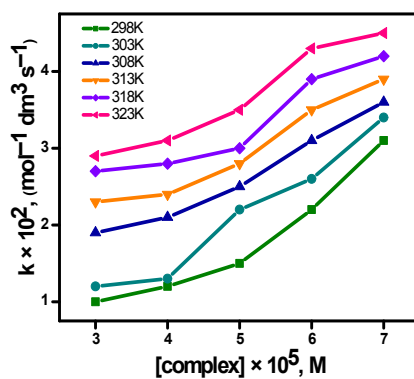
SI Table 4. Second-order rate constants for the reduction of cobalt(III) complex ion by Fe^{2+} in the presence of β -cyclodextrin under various temperatures. $\text{Cis-}[\text{Co}(\text{dpqc})_2(\text{C}_{12}\text{H}_{25}\text{NH}_2)_2](\text{ClO}_4)_3 = 4 \times 10^{-4} \text{ mol dm}^{-3}$, $\mu = 1.0 \text{ mol dm}^{-3}$, $[\text{Fe}^{2+}] = 0.01 \text{ mol dm}^{-3}$

SI Table 5. Activation parameters for the reduction of $\text{Cis-}[\text{Co}(\text{dpq})_2(\text{C}_{12}\text{H}_{25}\text{NH}_2)_2](\text{ClO}_4)_3$, $\mu = 1.0 \text{ mol dm}^{-3}$ in micelles medium

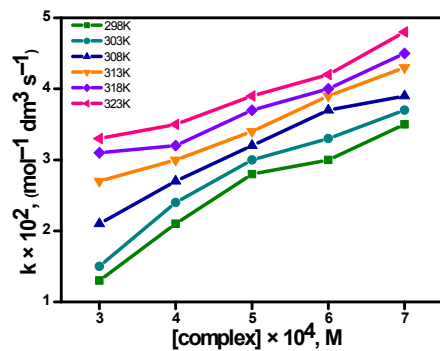
SI Table 6. Activation parameters for the reduction of $\text{Cis-}[\text{Co}(\text{dpqc})_2(\text{C}_{12}\text{H}_{25}\text{NH}_2)_2](\text{ClO}_4)_3$, $\mu = 1.0 \text{ mol dm}^{-3}$ in micelles medium

SI Table 7. Activation parameters for the reduction of $\text{Cis-}[\text{Co}(\text{dpq})_2(\text{C}_{12}\text{H}_{25}\text{NH}_2)_2](\text{ClO}_4)_3$, $\mu = 1.0 \text{ mol dm}^{-3}$ in β -CD medium

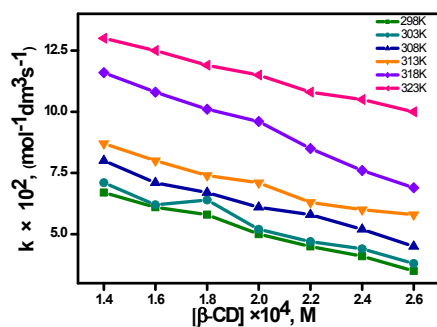
SI Table 8. Activation parameters for the reduction of $\text{Cis-}[\text{Co}(\text{dpqc})_2(\text{C}_{12}\text{H}_{25}\text{NH}_2)_2](\text{ClO}_4)_3$, $\mu = 1.0 \text{ mol dm}^{-3}$ in β -CD medium



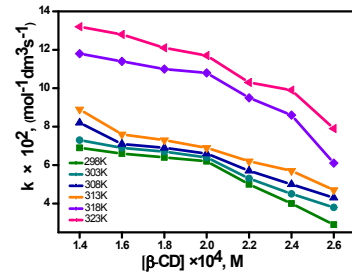
SI Fig. 1



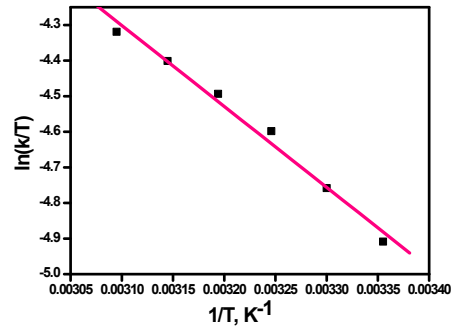
SI Fig. 2



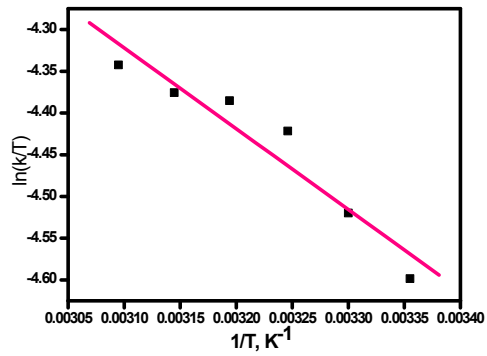
SI Fig. 3



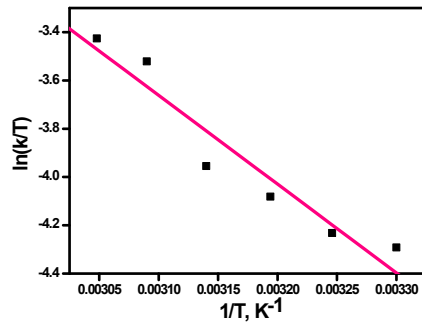
SI Fig. 4



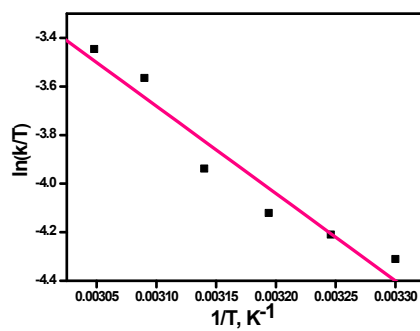
SI Fig. 5



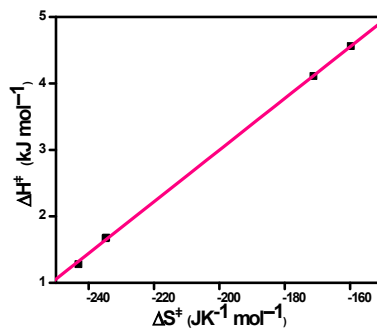
SI Fig. 6



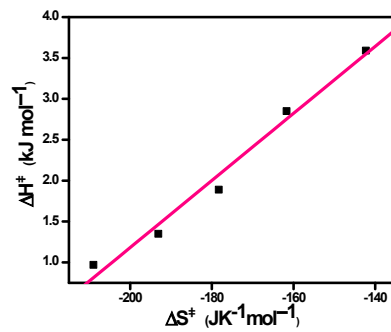
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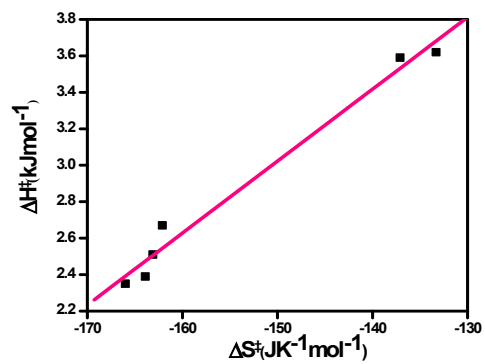
SI Fig. 8



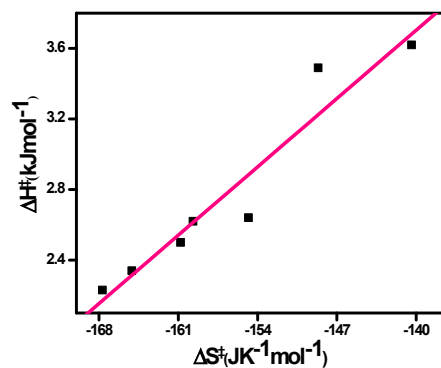
SI Fig. 9



SI Fig. 10



SI Fig. 11



SI Fig. 12

SI Table 1

[Complex] $\times 10^4, \text{mol dm}^{-3}$	$k \times 10^2, \text{dm}^3 \text{mol}^{-1} \text{s}^{-1}$					
	298K	303K	308K	313K	318K	23K
3.0	1.0	1.2	1.9	2.3	2.7	2.9
4.0	1.2	1.3	2.1	2.4	2.8	3.1
5.0	1.5	2.2	2.5	2.8	3.0	3.6
6.0	2.2	3.1	3.4	3.5	3.9	4.5
7.0	3.1	3.4	3.9	4.3	4.5	4.9

SI Table 2

[Complex] $\times 10^4, \text{mol dm}^{-3}$	$k \times 10^2, \text{dm}^3 \text{mol}^{-1} \text{s}^{-1}$					
	298K	303K	308K	313K	318K	323K
3.0	1.6	1.8	2.0	2.7	3.3	3.5
4.0	2.1	2.4	2.4	3.0	3.5	3.9
5.0	2.8	3.0	3.2	3.4	3.7	4.0
6.0	3.0	3.3	3.3	3.9	4.0	4.4
7.0	3.8	4.1	4.6	5.1	5.6	5.9

SI Table 3

[β -cyclodextrin] $\times 10^4, \text{mol dm}^{-3}$	$k \times 10^2, \text{dm}^3 \text{mol}^{-1} \text{s}^{-1}$					
	298K	303K	308K	313K	318K	323K
1.4	6.7	7.1	8.0	8.7	11.6	13.0
1.6	6.1	6.2	7.1	8.0	10.8	12.5
1.8	5.8	6.4	6.7	7.4	10.1	11.9
2.0	5.0	5.2	6.1	7.1	9.6	11.5
2.2	4.5	4.7	5.8	6.3	8.5	11.2
2.4	4.0	4.4	5.2	6.0	7.6	10.5
2.6	3.5	3.8	4.5	5.8	6.9	10.0

SI Table 4

[β -cyclodextrin] $\times 10^4, \text{mol dm}^{-3}$	$k \times 10^2, \text{dm}^3 \text{mol}^{-1} \text{s}^{-1}$					
	298K	303K	308K	313K	318K	323K
1.4	6.9	7.3	8.2	8.9	11.8	13.2
1.6	6.6	6.9	7.1	7.6	11.4	12.8
1.8	6.4	6.7	6.9	7.3	11.0	12.2
2.0	6.2	6.4	6.6	6.9	10.8	11.5
2.2	5.0	5.3	5.7	6.2	9.5	11.4
2.4	4.2	4.5	5.0	5.7	8.6	11.0
2.6	2.9	3.8	4.3	4.7	6.1	10.5

SI Table 5

[Complex] $\times 10^4,$ mol dm^{-3}	$\Delta H^\ddagger \text{ kJmol}^{-1}$	$\Delta S^\ddagger \text{ JK}^{-1}$
3.0	1.11	-204.5
4.0	2.27	-174.7
5.0	2.61	-167.8
6.0	3.70	-140.4
7.0	4.08	-130.9

SI Table 6

[Complex] $\times 10^4$, mol dm ⁻³	ΔH^\ddagger kJmol ⁻¹	ΔS^\ddagger JK ⁻¹
3.0	0.97	-209.0
4.0	1.35	-193.1
5.0	1.89	-178.3
6.0	2.85	-161.7
7.0	3.59	-142.3

SI Table 7

[β -cyclodextrin] $\times 10^3$, mol dm ⁻³	ΔH^\ddagger kJmol ⁻¹	ΔS^\ddagger JK ⁻¹
1.4	2.35	-166.0
1.6	2.39	-163.9
1.8	2.51	-163.1
2.0	2.67	-162.1
2.2	3.59	-137.1
2.4	3.62	-133.3

SI Table 8

$[\beta\text{-cyclodextrin}] \times 10^3,$ mol dm^{-3}	ΔH^\ddagger kJmol^{-1}	ΔS^\ddagger JK^{-1}
1.4	2.23	-167.7
1.6	2.34	-165.1
1.8	2.50	-160.8
2.0	2.62	-159.7
2.2	2.64	-154.8
2.4	3.49	-148.7
2.6	3.62	-140.4