

Supplementary Information

Electrochemiluminescence Sensor for Dopamine with the Dual Molecular Recognition Strategy Based on Graphite-like Carbon Nitride Nanosheets/3,4,9,10-Perylenetetracarboxylic Acid Hybrids

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Table S1 Comparison of the prepared ECL sensor with different reported sensors for the determination of DA.

Table S2 Recoveries test of DA in hydrochloride injection sample.

Table S3 The ECL signal of the blank for twelve times.

Fig. S1 CVs of (A) the g-C₃N₄ NSs-PTCA/mercaptopropionic acid/APTES/AuNPs/GCE electrode and (B) the g-C₃N₄ NSs-PTCA/DA/mercaptopropionic acid/APTES/AuNPs/GCE electrode in the absence (a) and in the presence (b) of 3.0×10⁻⁹ M DA in 0.10 M PBS (pH 7.0).

CV mode was performed with potential scanning from 0 to 0.6 V. Scan rate: 100 mV s⁻¹.

Fig. S2 The ECL stability of the sensor to 0.3 nM DA in 0.10 M PBS (pH 7.0)

containing 0.10 M K₂S₂O₈.

Fig. S3 The storage stability of the sensor.

Fig. S4 The ECL response of the sensor to 3.0 nM (a) KCl, (b) Na₂SO₄, (c) ascorbic acid, (d) tyrosine, (e) glucose, (f) ascorbic acid, (g) epinephrine and (h) clenbuterol, respectively. Column j is the ECL response of the sensor toward 0.3 nM DA.

Fig . S5 The structure of epinephrine (a), clenbuterol (b) and DA (c).

Fig. S6 Calibration plots at a low concentration range of DA.

Table S1

Electrode	Method	Linear response range (mol L ⁻¹)	Limit of detection (mol L ⁻¹)	Reference
rGO/MWCNTs/AuNPs/GCE	ECL	2.0×10 ⁻⁷ ~7×10 ⁻⁵	6.7×10 ⁻⁸	1
Ru(bpy) ₃ ²⁺ /OMC@Nafion/ GCE	ECL	5.0×10 ⁻⁹ ~ 5.0×10 ⁻⁴	7.0×10 ⁻⁹	2
CNTs/P1 ^c /DA/DSP-QDs ^d	ECL	5.0×10 ⁻¹¹ ~ 1.0×10 ⁻⁸	2.6×10 ⁻¹¹	3
PNVs ^e /GCE	ECL	1.0×10 ⁻¹¹ ~2.0×10 ⁻¹⁰	3.15×10 ⁻¹²	4
PEI-MWNTs-AuNPs/GCE	DPV ^a	5.0×10 ⁻⁸ ~4.0×10 ⁻⁶	6.56×10 ⁻⁹	5
Ppy-RGO /GCE	DPV	1.0×10 ⁻⁸ ~1.0×10 ⁻⁵	1.0×10 ⁻⁹	6
Porous carbon/GCE	CV ^b	9.0×10 ⁻⁹ ~3.0×10 ⁻⁷	2.9×10 ⁻⁹	7
AuNF@g-C ₃ N ₄ -PANI/GCE	ECL	5.0×10 ⁻⁹ ~ 1.6×10 ⁻⁶	1.7×10 ⁻⁹	8
PTC-NH ₂ /GCE	ECL	5.0×10 ⁻⁹ ~ 1.1×10 ⁻⁶	1.6×10 ⁻⁹	9
g-C ₃ N ₄ -PTCA/DA/APTES/ AuNPs/GCE	ECL	3.0×10 ⁻¹² ~ 6×10 ⁻⁹	2.4×10 ⁻¹²	This work

^aDPV, differential pulse voltammetry.^bCV, cyclic voltammetric profiles.P1^c, a boronic acid-functionalized pyrene probeDSP-QDs^d, 3,3'- dithiodipropionic acid di(N-hydroxysuccinimide ester) (DSP)-functionalized CdTe QDsPNVs^e, peptide nanovesicles

Table S2

Sample	C_{original} (nM)	C_{Added} (nM)	C_{Detected} (nM) ^a	Recovery (%)	RSD (%)
1	0.020	1.02	1.07±0.05	103	4.7
2	0.020	0.167	0.198±0.003	107	1.5
3	0.020	0.033	0.052±0.001	97.0	1.9
4	0.020	0.017	0.036±0.001	94.1	2.8

^a Mean±SD, $n = 3$.

Table S3

Number	1	2	3	4	5	6
<i>I / a.u</i>	23	19	22	25	25	22
Number	7	8	9	10	11	12
<i>I / a.u</i>	22	22	24	22	24	24

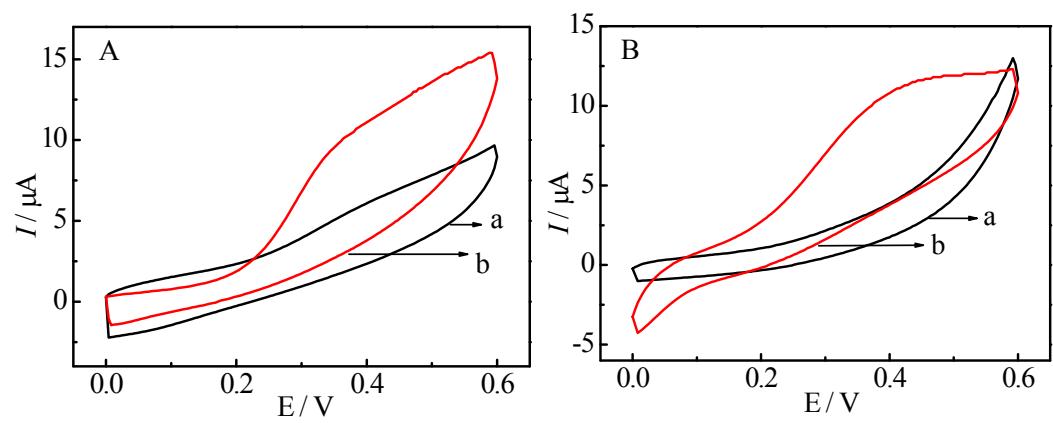


Fig. S1

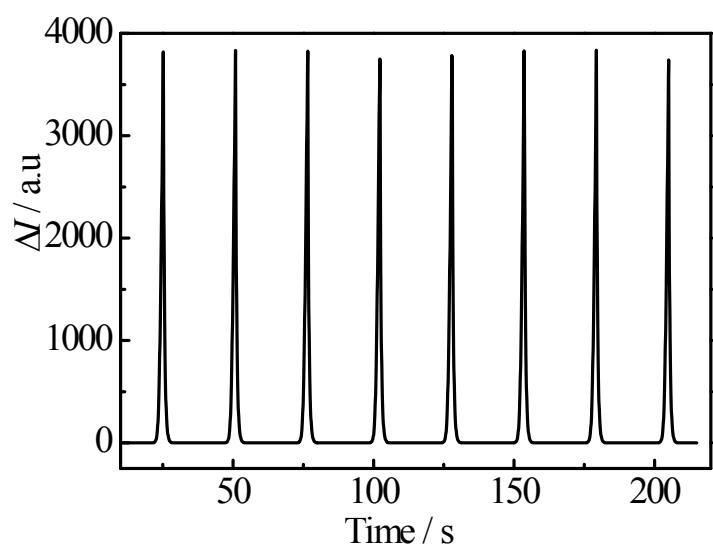


Fig. S2

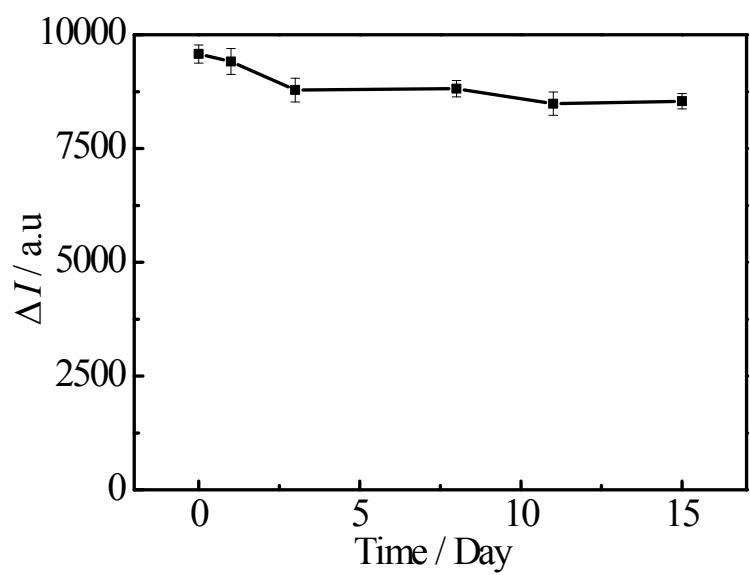


Fig. S3

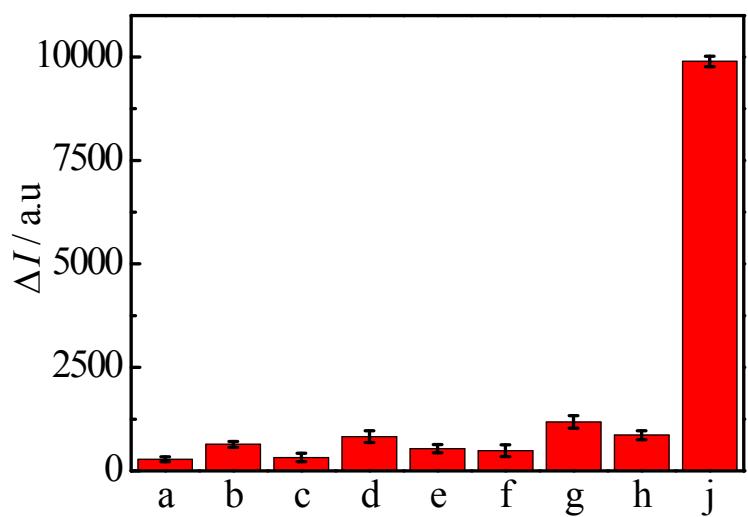


Fig. S4

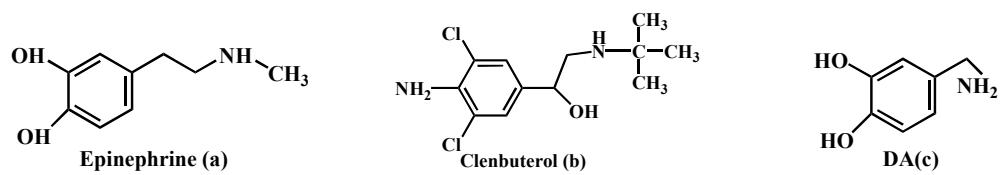


Fig .S5

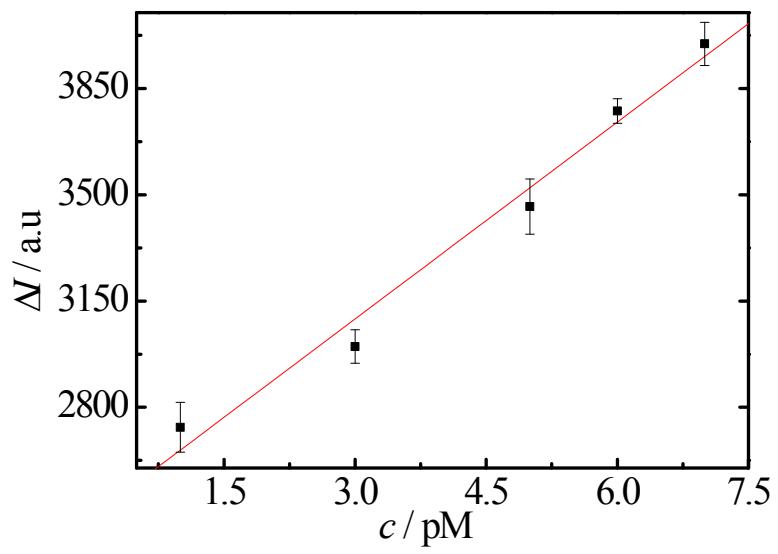


Fig. S6

As for the detection limit (LOD), it was calculated according to the IUPAC recommendation⁹

$$\text{LOD} = k S_b/m$$

Here, k is the numerical factor chosen in accordance with the desired confidence level. As suggested by Long and Winefordner,¹⁰ the use of $k = 3$ allows a confidence level of 99.86% for a normal distribution of the blank signals. S_b is the standard deviation of the blank signals ($n_B=12$), and m is the analytical sensitivity, which can be estimated by the slope of calibration plot at lower concentration ranges. Fig. S6 displayed the calibration curve between ΔI and DA concentration at lower concentration ranges of DA. The corresponding calibration equation was $\Delta I = 2.16 c_{\text{pM}} + 2442.5$.

The S_b of twelve times zero-dose was 1.7, and the detailed data were presented in Table S3. In addition, relative standard deviation (RSD) of 7.4% for twelve times was obtained. Therefore, the recalculated LOD of the proposed sensor according to IUPAC recommendation is 2.4 pM ($\text{LOD} = 3 \times 1.7 / 2.16 = 2.4$).

Limit of detection (LOQ) was calculated using the equations: $\text{LOD} = 10 S_b/m$,¹¹ and 7.9 pM of LOQ was obtained.

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