## Supporting Information

## A Further Electrochemical Investigation on Solution to High Energetical Power Source: Isomerous Compound 0.75Li<sub>1.2</sub>Ni<sub>0.2</sub>Mn<sub>0.6</sub>O<sub>2</sub>·0.25LiNi<sub>0.5</sub>Mn<sub>1.5</sub>O<sub>4</sub>

Zhuo Zheng,<sup>a</sup> Zhen-Guo Wu,<sup>a</sup> Yan-Jun Zhong,<sup>a</sup> Chong-Heng Shen,<sup>b</sup> Wei-Bo Hua,<sup>a</sup> Bin-Bin Xu,<sup>b</sup>
Chong Yu,<sup>c</sup> Ben-He Zhong<sup>a</sup> and Xiao-Dong Guo\*<sup>a</sup>
<sup>a</sup> College of Chemical Engineering, Sichuan University, No.24 South Section 1, Yihuan Road,
Chengdu, 610065, China.
<sup>b</sup> Department of Chemistry, College of Chemistry and Chemical Engineering, Xiamen University,
Fujian 361005, China.
<sup>c</sup> National Key Laboratory for Nuclear Fuel and Materials, Nuclear Power Institute of China,
Chengdu, 610041, China.
\*Corresponding author. Email: xiaodong2009@scu.edu.cn

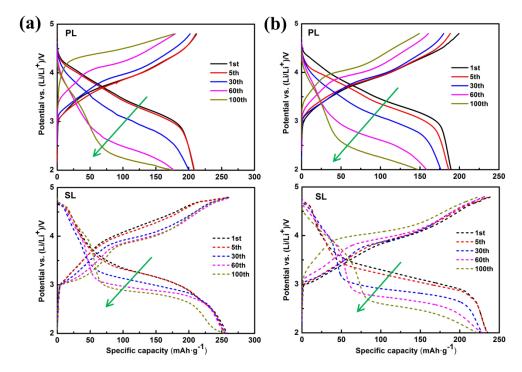


Fig. S1 Charge and discharge profiles of PL and SL samples upon the cycling (a) at

0.5 C rate, (b) at 1 C rate

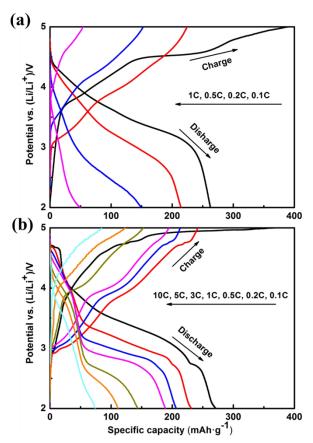


Fig. S2 Charge and discharge curves of (a) PL and (b) SL samples at various rates between 2.0 and 5.0 V

Samples .	Charge capacity (mAh g <sup>-1</sup> )		Discharge capacity (mAh g <sup>-1</sup> )	
	<4.4 V	>4.4 V	< 3.5 V	> 3.5 V
PL	124	219	125	137
SL	107	240	162	131

 Table S1
 Capacity contribution of first cycle in different voltage range of PL and SL samples