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Ni/Mn ratio and morphology-dependent crystallographic facet structure and electrochemical properties of high-voltage spinel $LiNi_{0.5}Mn_{1.5}O_4$ cathode material

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Figure captions

- Figure S1. SEM image for spherical (a) and cubic (b) MnCO₃.
- Figure S2. SEM image for spherical LNMO samples: (a) S-1.45, (b) S-1.50 and (c) S-1.55.
- Figure S3. SEM image for cubic LNMO samples: (a) C-1.20, (b) C-1.55 and (c) C-Cr.
- Figure S4. XPS of the Mn 2p3/2 peak of the as-synthesized LNMO samples.
- Figure S5. Voltage profiles of LNMO/Li half cells under various current densities at room temperature.
- **Figure S6**. Voltage profiles of LNMO/Li half cells for C-Cr and C-1.55 samples under 5-C rate at room temperature.
- Figure S7. Cyclic voltammetry (CV) curves of the C-1.55 electrode at scan rates from 0.1 to 0.5 mV s⁻¹. The oxdation (charging) peaks are labelled (a) and (b) while the reduction (discharging) peaks are labelled (c) and (d).
- Figure S8. HTEM and SAED images for cubic C-Cr sample.
- Figure S9. The 1st and 5th cycle of the CV curves for the C-Cr sample.
- Table S1. Summary of the CV results obtained at different scanning rates and the Li⁺ diffusion coefficients determined for the C-1.55 electrode.^[a]





Fig. S3











Figure S8.







Scanning rate (mv/s)	Potential (V)					
	E _{O1}	E ₀₂	E _{R1}	E _{R2}	ΔE_1	ΔE_2
0.1	4.745	4.799	4.640	4.686	0.105	0.113
0.2	4.759	4.827	4.620	4.671	0.139	0.156
0.3	4.780	4.853	4.594	4.663	0.186	0.190
0.4	4.786	4.859	4.588	4.659	0.198	0.200
0.5	4.796	4.879	4.577	4.650	0.219	0.229
$D_{Li}^{+}(\times 10^{-11} \text{ cm}^2/\text{s})$	3.305	6.653	4.834	2.477		

^[a] E_0 : anodic peak potential, E_R : cathodic peak potential, ΔE : the separation between E_0 and E_R . The subscript numbers 1 and 2 denote the redox couple at lower and higher potential, respectively.