

***β*-Substituted Triarylborane Appended Porphyrin: photophysical  
properties and anion sensing**

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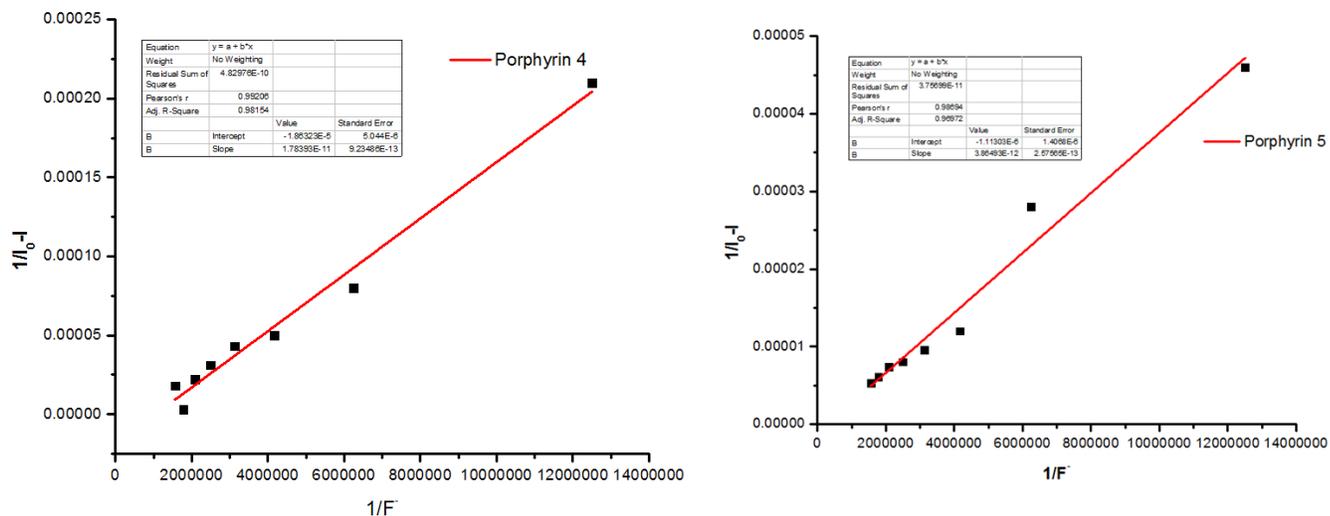


Figure. S1. Benesi-Hildebrand plot of porphyrin 4 and 5 by plotting  $1/(I_0 - I)$  as a function of  $1/[F]$

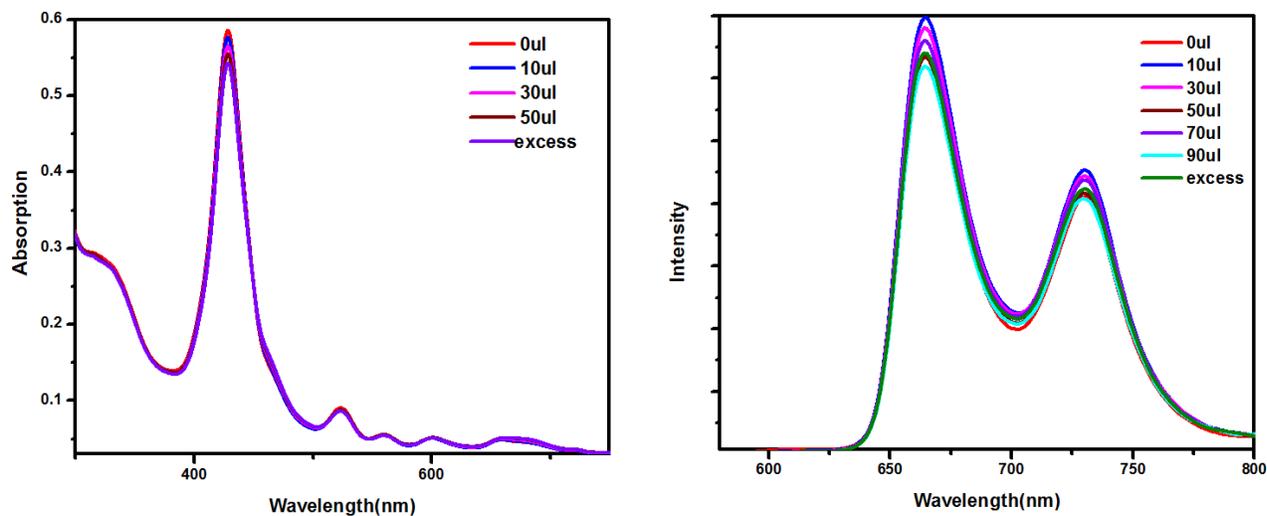
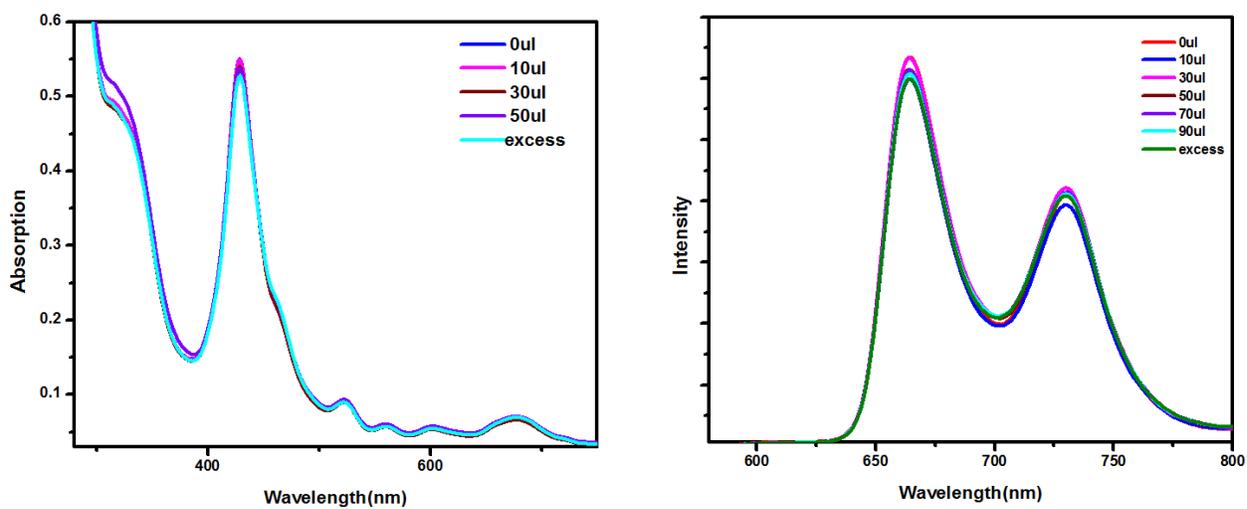
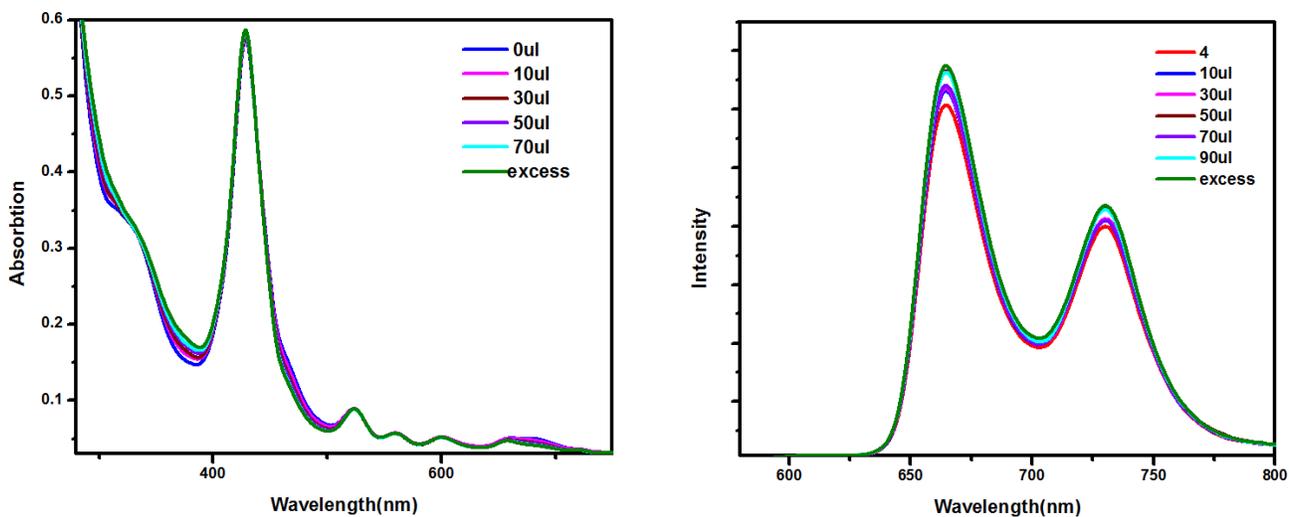


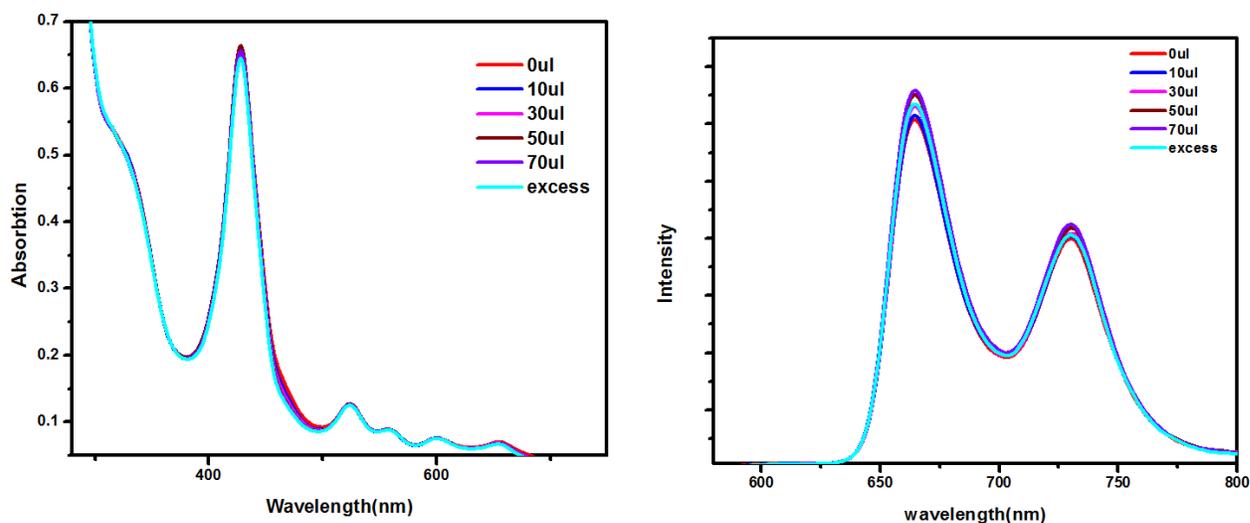
Figure S2. Absorption and Fluorescence titration spectra of triarylborane porphyrin 4 with TBACl in  $1.0 \times 10^{-6}$  M DCM solution at  $\lambda_{ex} = 428$  nm



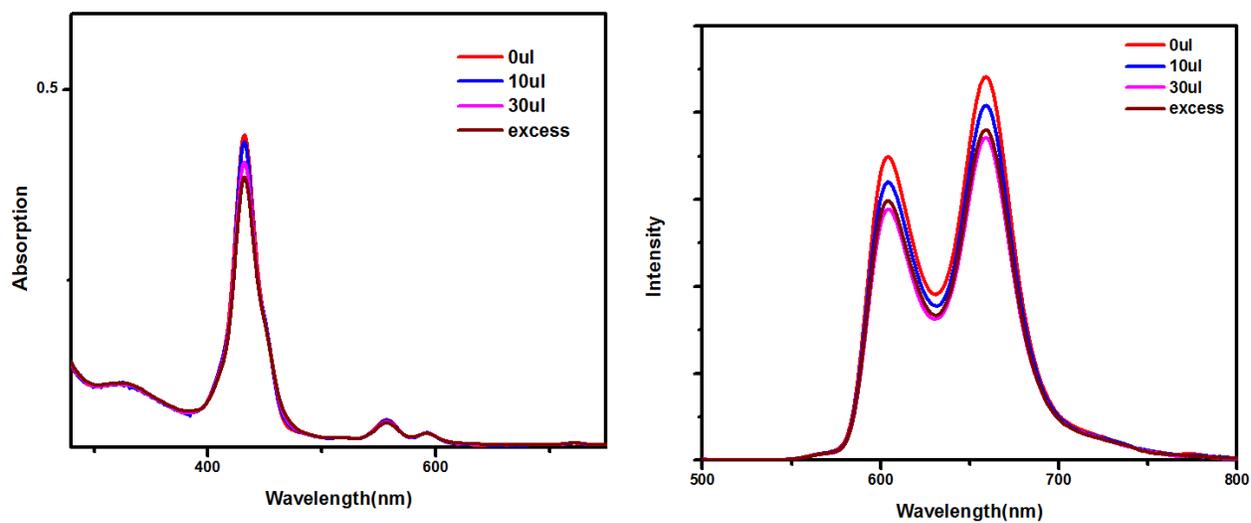
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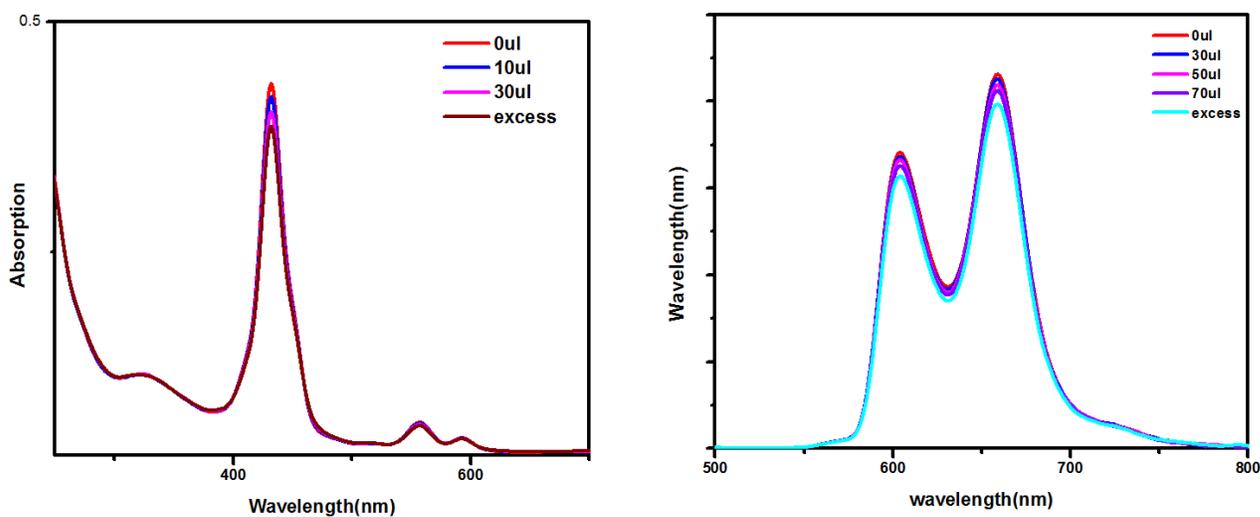
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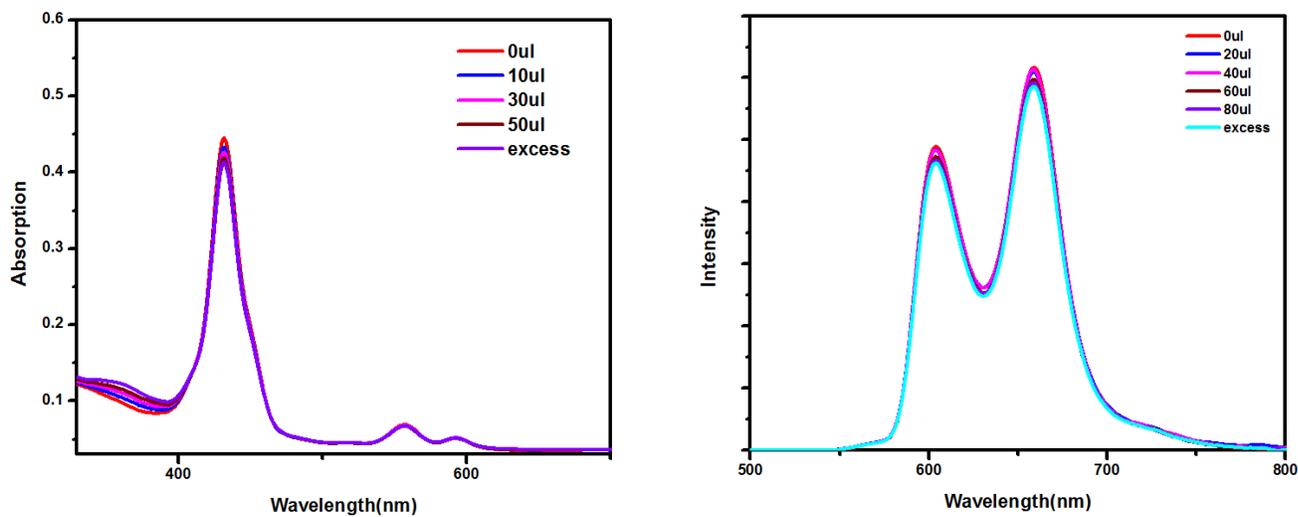
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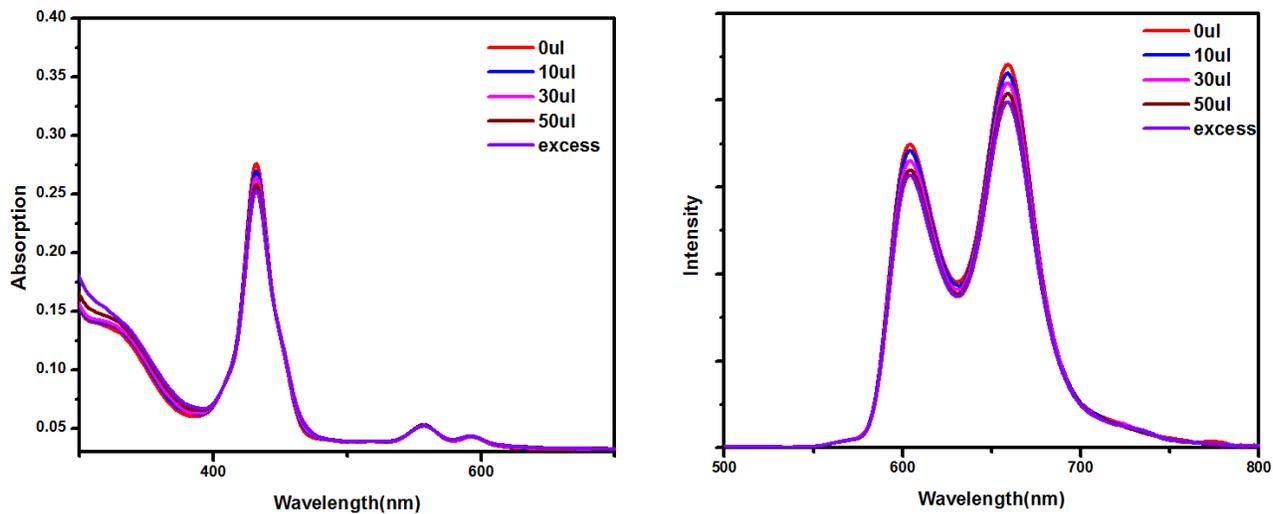
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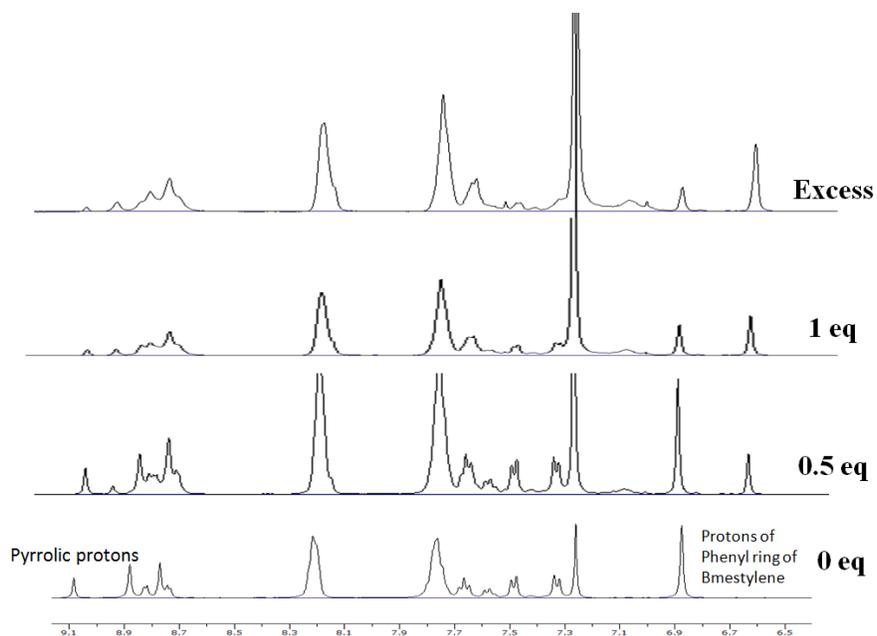


**Figure S8.** Absorption and Fluorescence titration spectra of metalated triarylborane porphyrin **5** with TBAI in  $1.0 \times 10^{-6}$  M DCM solution at  $\lambda_{\text{ex}} = 432$  nm



**Figure S9.** Absorption and Fluorescence titration

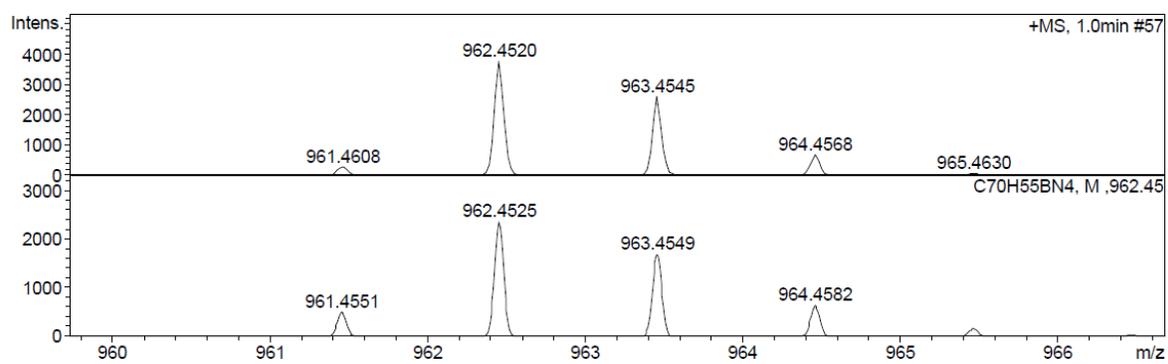
spectra of metalated triarylborane porphyrin **5** with TBACN in  $1.0 \times 10^{-6}$  M DCM solution at  $\lambda_{\text{ex}}$   
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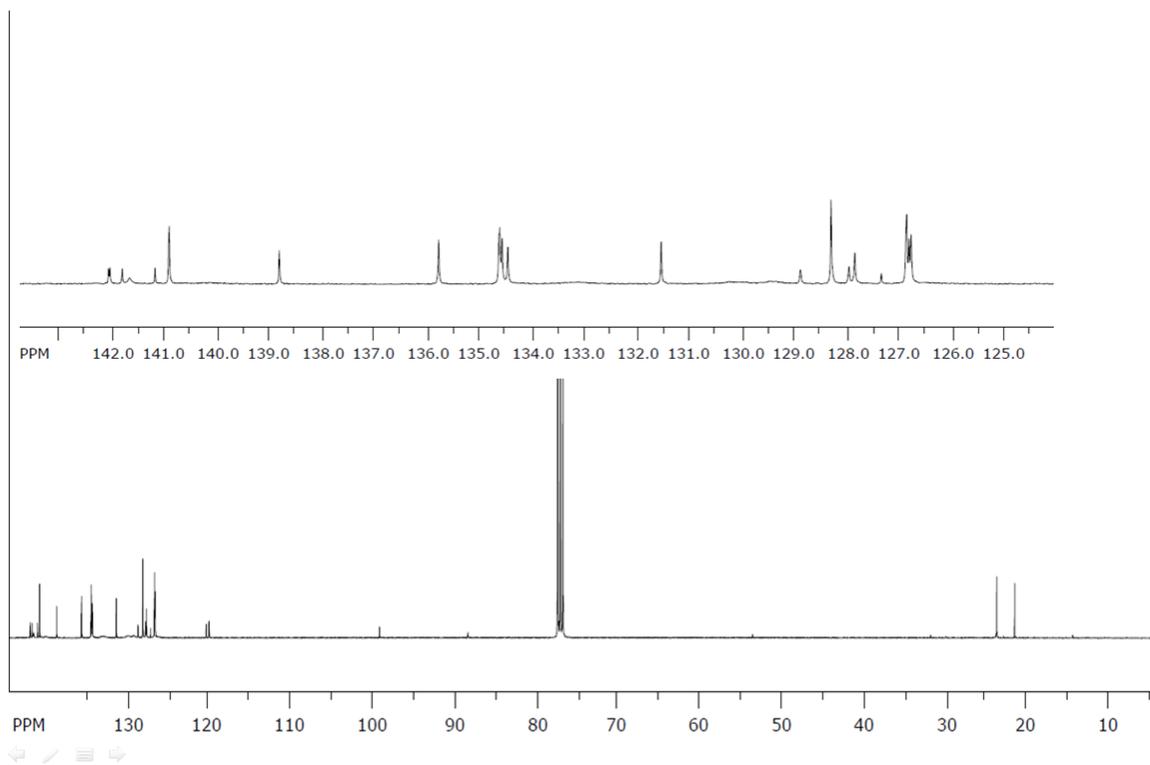
**Figure S10.**  $^1\text{H}$  NMR titration spectra of triarylborane porphyrin **4** in  $\text{CDCl}_3$



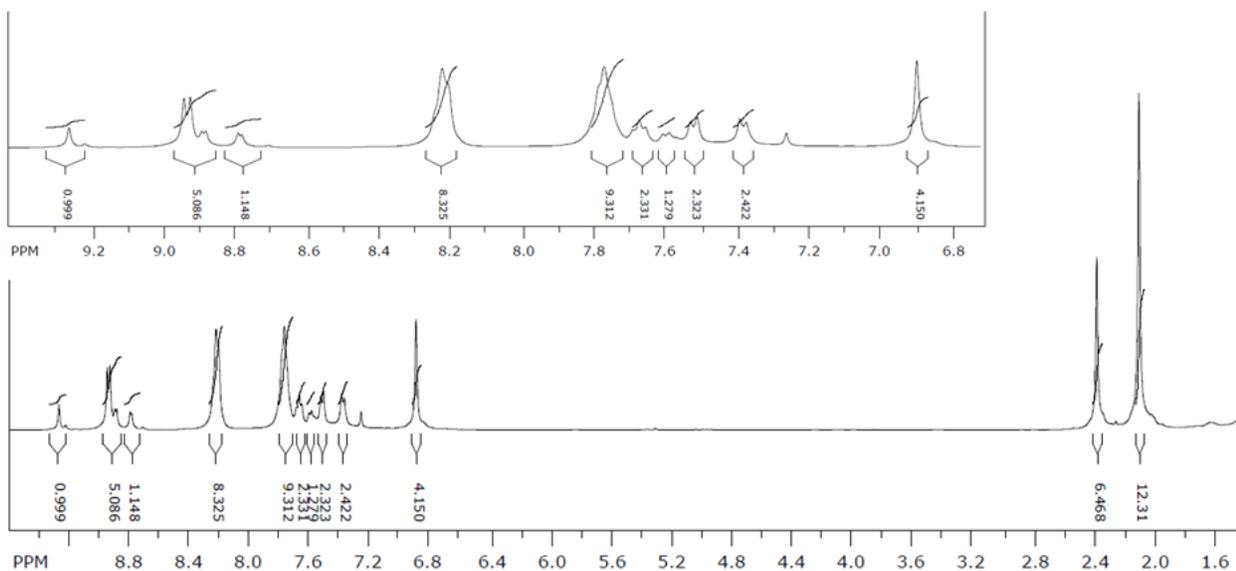
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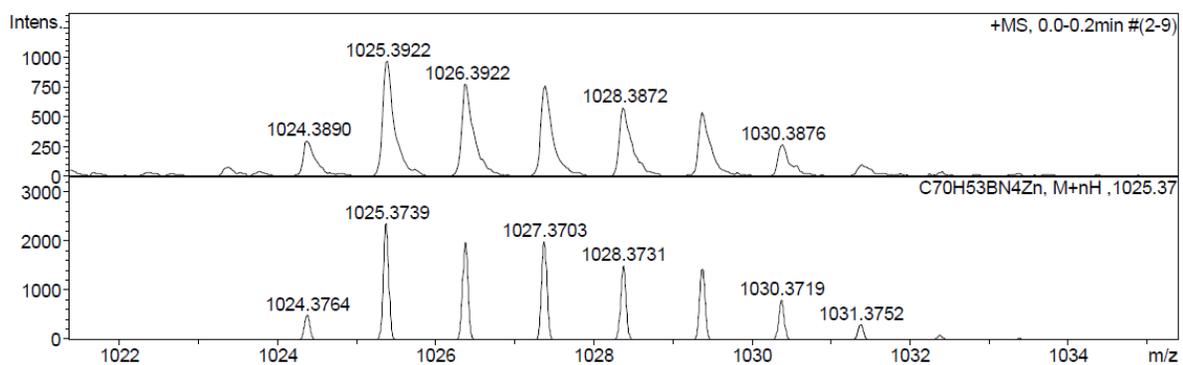
**Figure S13. HRMS mass spectrum of compound 4.**



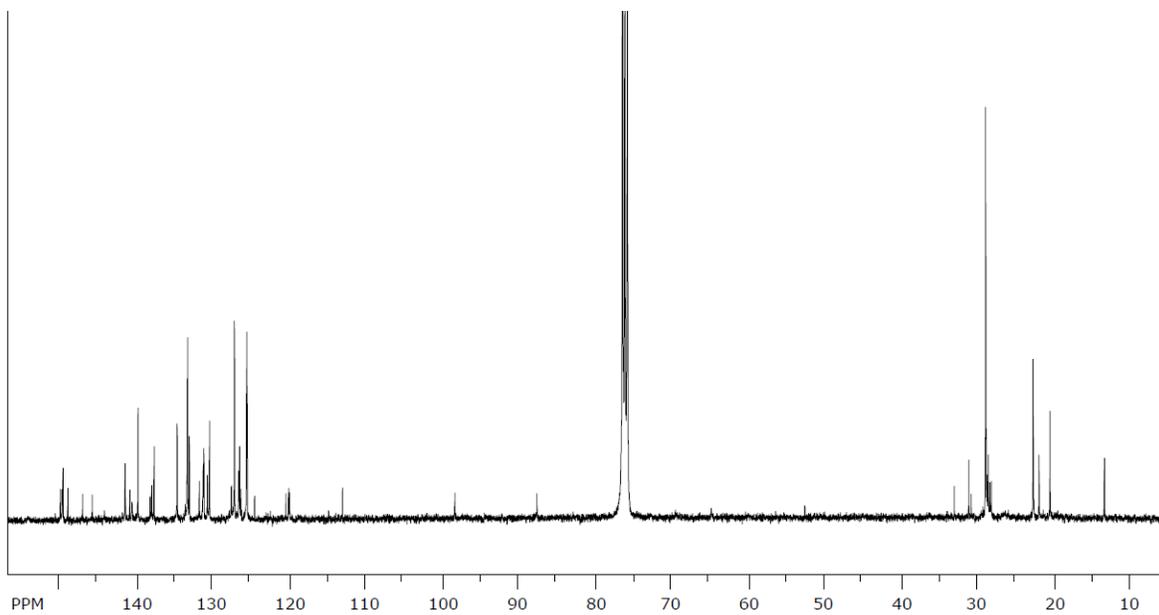
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**Figure S16.** HRMS mass spectrum of compound **5**.



**Figure S17.  $^{13}\text{C}$  NMR Spectra of 5**