

Electronic Supplementary Information

**A triazine-based covalent organic framework-palladium hybrid for
one-pot silicon-based cross-coupling of silanes and aryl iodides**

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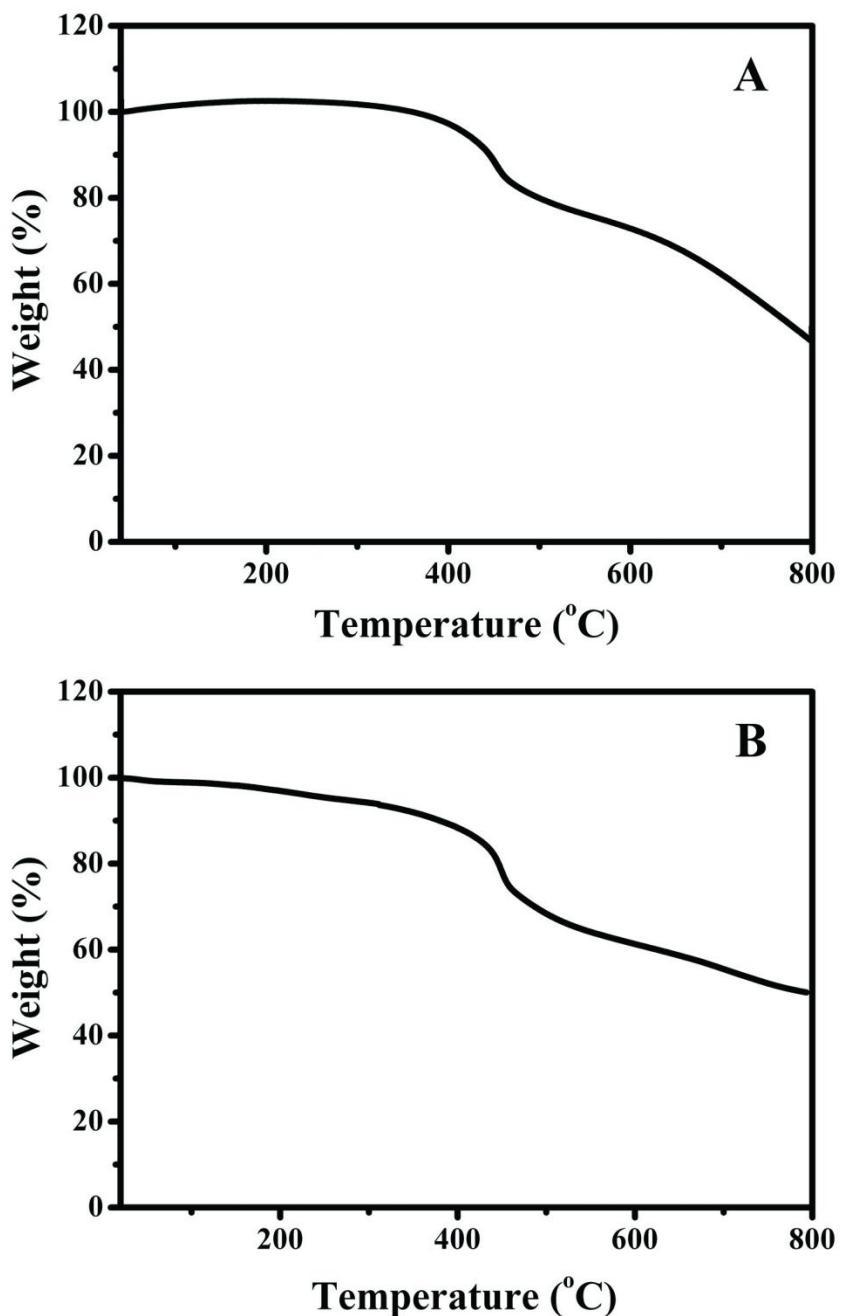


Figure S1. Thermogravimetric analysis (TGA) data of (A) **COF-SDU1** and (B) **Pd(II)/COF-SDU1**.

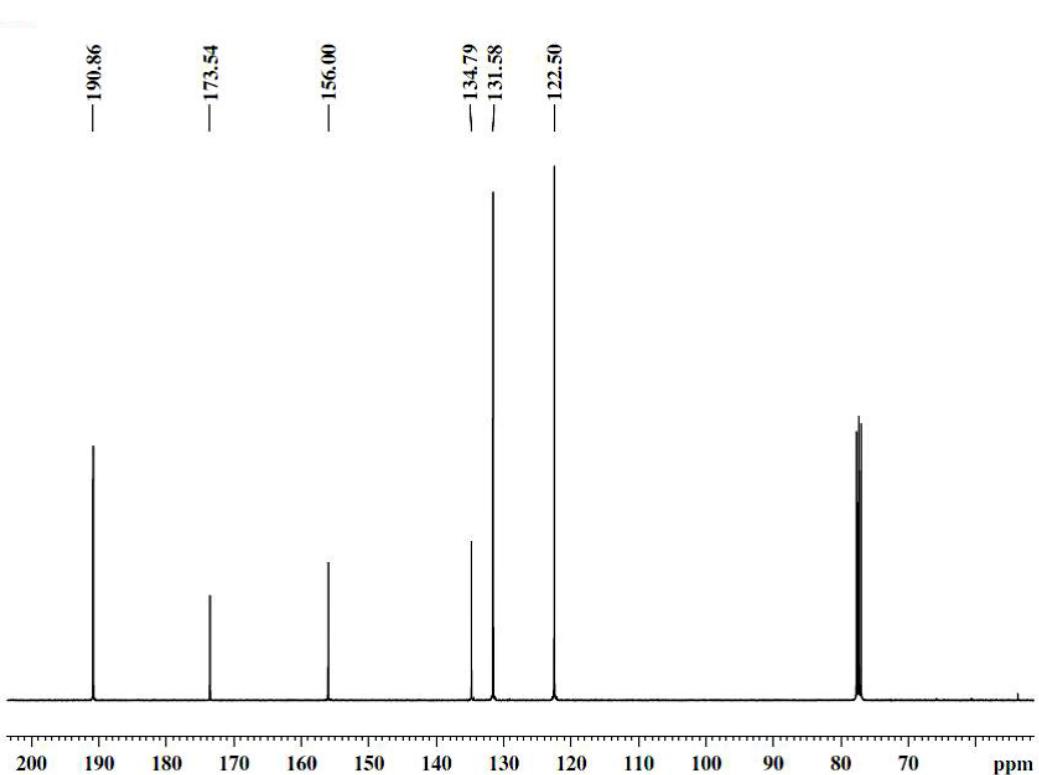


Figure S2. ¹³C CP/MAS NMR spectrum of trif.

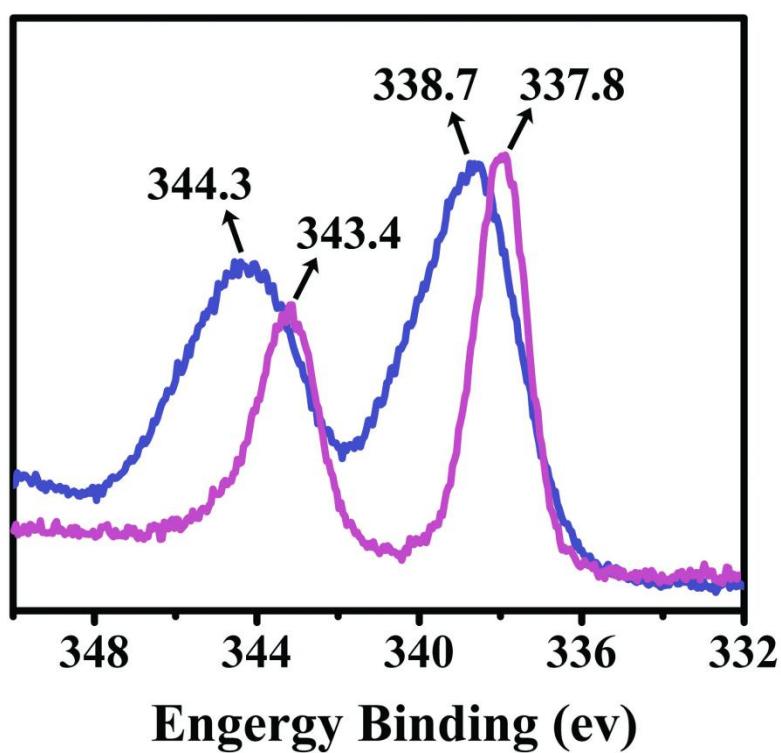


Figure S3. XPS spectrum of $\text{Pd}(\text{OAc})_2$ (blue line) and $\text{Pd}(\text{II})/\text{COF-SDU1}$ (pink line).

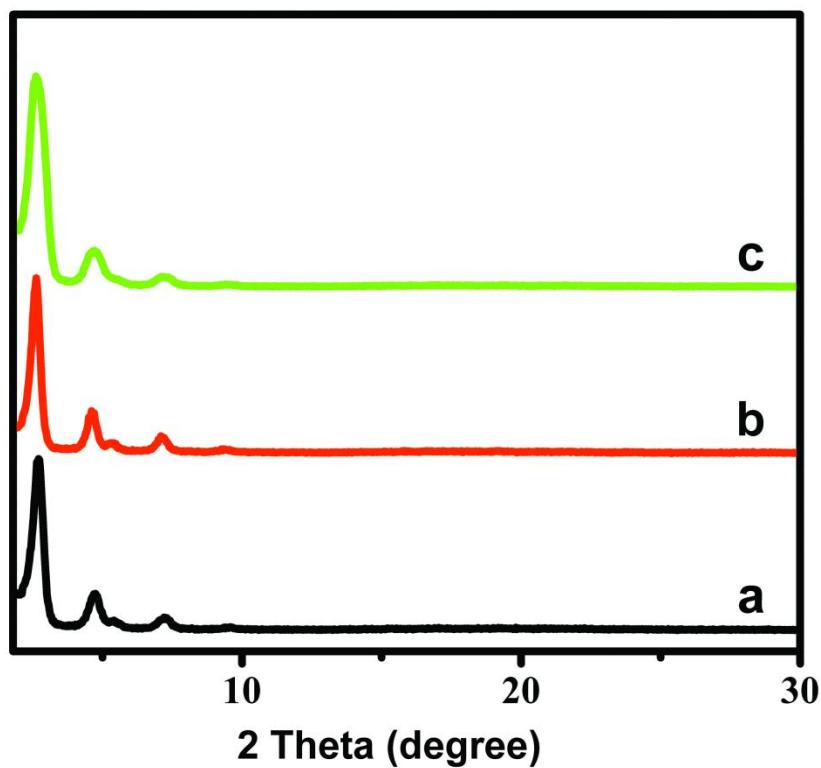


Figure S4. PXRD patterns of **Pd(II)/COF-SDU1** (a) before reused, (b) after reused first time and (c) three times.

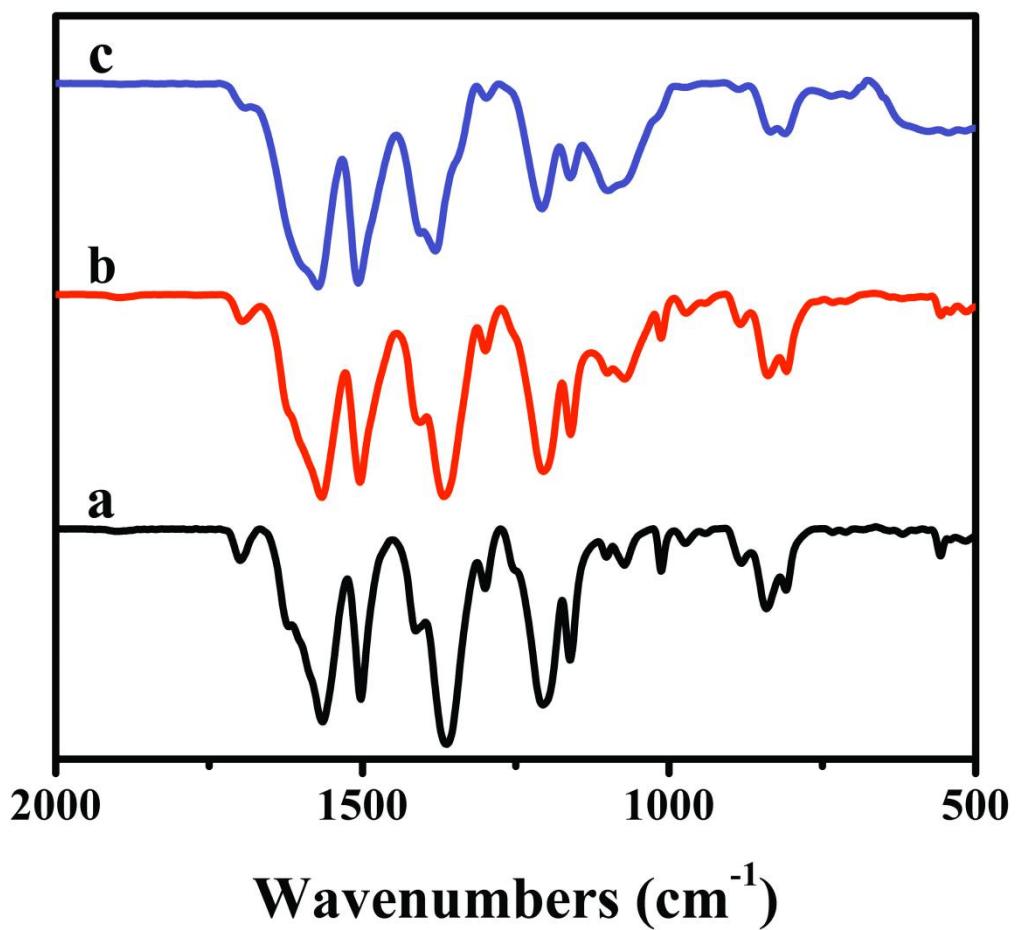


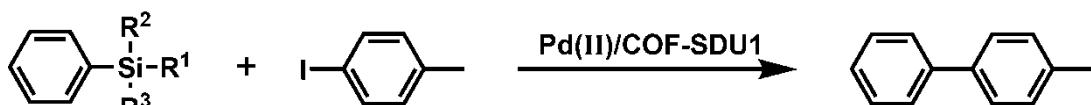
Figure S5. Fourier transforms infrared (FT-IR) spectra of **Pd(II)/COF-SDU1** (a) before reused, (b) after reused first time and (c) three times in the region of 500-2000 cm⁻¹ with 2 cm⁻¹ resolution.

Table S1 Catalytic activity test of **Pd(II)/COF-SDU1** towards oxidation of organosilanes to organo(alkoxy)silanes or organosilanols.

$\text{R}_{(4-n)}\text{SiH}_n + n \text{ ROH} \xrightarrow{\text{Pd(II)/COF-SDU1}} \text{R}_{(4-n)}\text{Si(OR)}_n + n \text{ H}_2$			1		
Entry	$\text{R}_{(4-n)}\text{SiH}_n$	ROH	1	t (h) ^a	Yield (%) ^{b,c}
1	PhSiH ₃	H ₂ O	1a	2	97.8
2	Me ₂ PhSiH	H ₂ O	1b	2	97.3
3	Ph ₂ SiH ₂	H ₂ O	1c	2	97.1
4	PhSiH ₃	CH ₃ OH	1d	3	98.0
5	PhSiH ₃	C ₂ H ₅ OH	1e	3	97.0

^a Reaction conditions: $\text{R}_{(4-n)}\text{SiH}_n$ (1 mmol), ROH (5 mmol), **Pd(II)/COF-SDU1** (20 mg, 8.5×10^{-3} mmol of Pd) and THF (2 mL), room temperature. ^b Isolated yield. ^c The values are the average of two independent experiments.

Table S2 Catalytic activity test of **Pd(II)/COF-SDU1** towards silicon-based cross-coupling reaction of the organo(alkoxy)silanes or organosilanols with aryl iodides.

					
Entry	R ¹	R ²	R ³	t (h) ^a	Yield (%) ^{b,c}
1	OCH ₃	OCH ₃	OCH ₃	7	97.8
2	OCH ₃	CH ₃	CH ₃	18	58.1
3	OH	CH ₃	CH ₃	18	66.5

^a Reaction conditions: organo(alkoxy)silanes or organosilanols (1 mmol), aryl halides (1.1 mmol), **Pd(II)/COF-SDU1** (20 mg, 8.5×10^{-3} mmol of Pd), TBAF (3 mmol) and THF (2 mL), 80 °C. ^b Isolated yield. ^c The values are the average of two independent experiments.

NMR signals:¹⁻⁶

Phenylsilanetriol (1a): ^1H NMR (300 MHz, CDCl_3): $\delta=7.74\text{-}7.55$ (m, 2H), 7.42-7.24 (m, 3H).

Dimethylphenylsilanol (1b): ^1H NMR (300 MHz, CDCl_3): $\delta=7.64\text{-}7.60$ (m, 2H), 7.46-7.37 (m, 3H), 3.15 (bs, 1H), 0.41 (s, 6H).

Diphenylsilanediol (1c): ^1H NMR (300 MHz, CDCl_3): $\delta=7.73\text{-}7.70$ (m, 4H), 7.49-7.33 (m, 6H), 2.17 (bs, 2H).

Phenyltrimethoxysilane (1d): ^1H NMR (400 MHz, CDCl_3): $\delta=7.71\text{-}7.69$ (m, 2H), 7.46-7.40 (m, 3H), 3.65 (s, 9H).

Phenyltriethoxysilane (1e): ^1H NMR (400 MHz, CDCl_3): $\delta=7.70\text{-}7.66$ (m, 2H), 7.40-7.30 (m, 3H), 3.90-3.83 (q, $J = 6.9$ Hz, 6H), 1.25-1.20 (t, $J = 7.2$ Hz, 9H).

4-methyl-bipheny (2a): ^1H NMR (300 MHz, CDCl_3): $\delta=7.53\text{-}7.49$ (m, 2H), 7.44-7.41 (m, 2H), 7.40-7.32 (m, 2H), 7.28-7.22 (m, 1H), 7.19-7.15 (m, 2H), 2.33-2.32 (d, $J = 3.0$ Hz, 3H). $R_f = 0.5$ (100% petroleum ether).

2-methyl-bipheny (2b): ^1H NMR (300 MHz, CDCl_3): $\delta=7.37\text{-}7.32$ (m, 2H), 7.29-7.24 m, 3H), 7.21-7.16 (m, 4H), 2.20 (s, 3H). $R_f = 0.5$ (100% petroleum ether).

4-methoxy-biphenyl (2c): ^1H NMR (300 MHz, CDCl_3): $\delta=7.57\text{-}7.51$ (m, 4H), 7.44-7.39 (m, 2H), 7.33-7.28 (m, 1H), 7.01-6.56 (m, 2H), 3.86 (s, 3H). $R_f = 0.5$ (100% petroleum ether).

4-fluoro-bipheny (2d): ^1H NMR (300 MHz, CDCl_3): $\delta=7.51\text{-}7.41$ (m, 4H), 7.40-7.25 (m, 3H), 7.09-7.02 (m, 2H). $R_f = 0.5$ (100% petroleum ether).

4-hydroxy-bipheny (2e): ^1H NMR (300 MHz, CDCl_3): $\delta=7.57\text{-}7.53$ (m, 2H), 7.51-7.47 (m, 2H), 7.45-7.39 (m, 2H), 7.34-7.28 (m, 1H), 6.94-6.89 (m, 2H). $R_f = 0.5$ (petroleum ether/n-hexane 1:1).

4-nitro-biphenyl (2f): ^1H NMR (300 MHz, CDCl_3): $\delta=8.33\text{-}8.28$ (m, 2H), 7.77-7.72 (m, 2H), 7.65-7.61 (m, 2H), 7.53-7.42 (m, 3H). $R_f = 0.5$ (petroleum ether/ethyl acetate 1:1).

References:

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