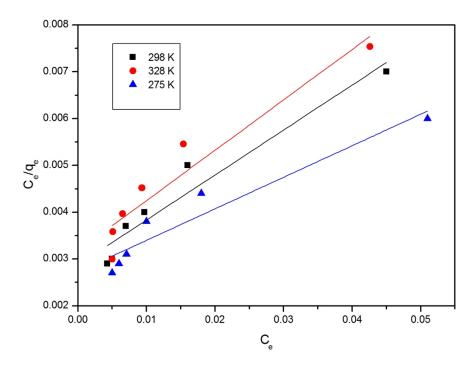
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Supplementary Material

Adsorption isotherms, kinetics and thermodynamic

1. Adsorption Isotherms



 $\label{eq:Fig.S1} \textbf{Fig. S1} \ \ \text{Langmuir adsorption isotherm for } Cr^{6^+} \ \text{adsorption onto the surface of } \alpha\text{-MnO}_2\text{-NH}_2\text{-RGO}$ at different temperatures.

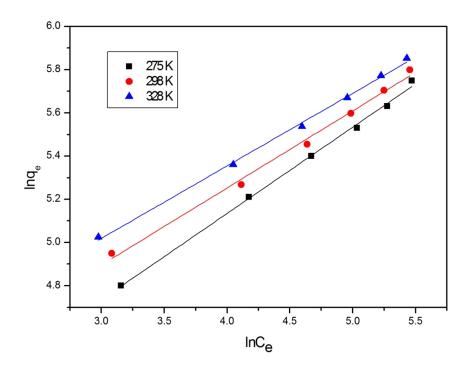


Fig. S2. Freundlich adsorption isotherm for Cr^{6+} adsorption onto the surface of α -MnO₂-NH₂-RGO at different temperatures.

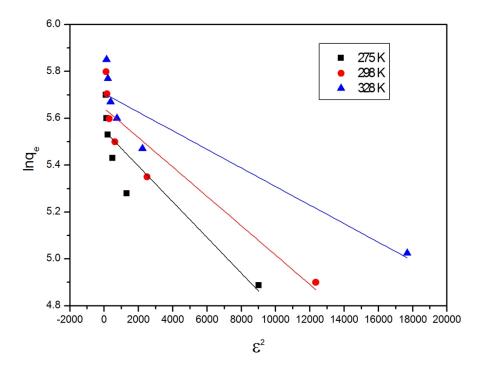


Fig. S3. D-R adsorption isotherm for Cr^{6+} adsorption onto the surface of α -MnO₂-NH₂-RGO at different temperatures.

3. Adsorption Kinetics

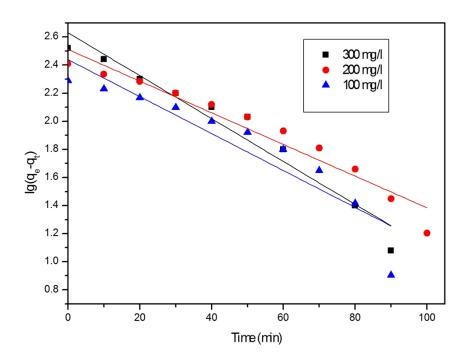


Fig. S4. Plot of the pseudo-first-order kinetic model for Cr^{6+} adsorption onto the surface of α - MnO_2-NH_2-RGO at different initial solution concentrations.

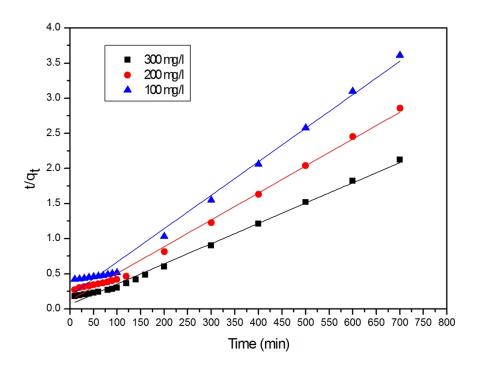


Fig. S5. Plot of the pseudo-second-order kinetic model for Cr^{6+} adsorption onto the surface of α -MnO₂-NH₂-RGO at different initial solution concentrations.

4. Adsorption Thermodynamic

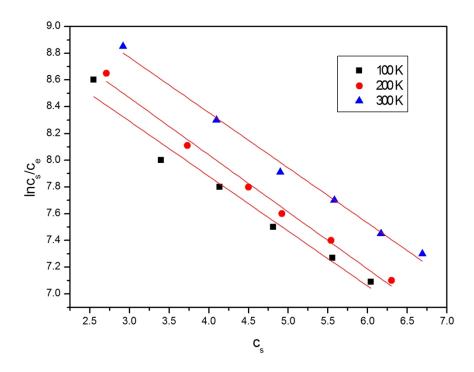


Fig. S6. Plots of ln C_s/C_e versus C_s at various temperatures.

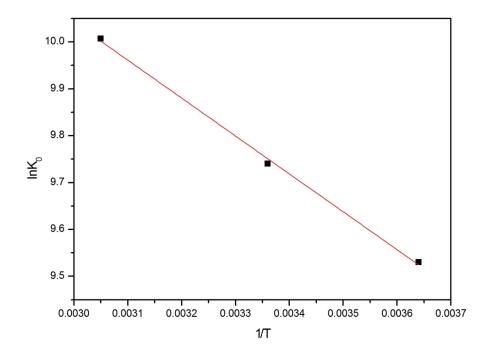


Fig. S7. Plots of lnK_0 versus 1/T

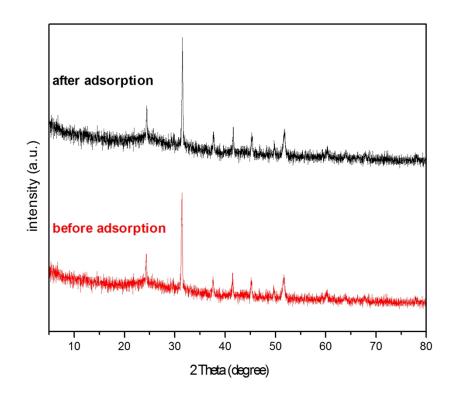


Fig. S8. The XRD of composite before and after adsorption