

Supplementary Information

Yolk-shell nanospheres with soluble amino-polystyrene as reservoir for Pd NPs

Guojun Lan^{a,b}, Xiaoming Zhang^b, Xiaomin Zhang^b, Mingrun Li^b, Ying Li^{a}, Qihua Yang^{b*}*

^a Institute of Industrial Catalysis, Zhejiang University of Technology, Hangzhou 310014 (China)

^b State Key Laboratory of Catalysis, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, 457 Zhongshan Road, Dalian 116023 (China)

* To whom correspondence should be addressed Email Address: yangqh@dicp.ac.cn;

liying@zjut.edu.cn; URL: <http://www.hmm.dicp.ac.cn>;

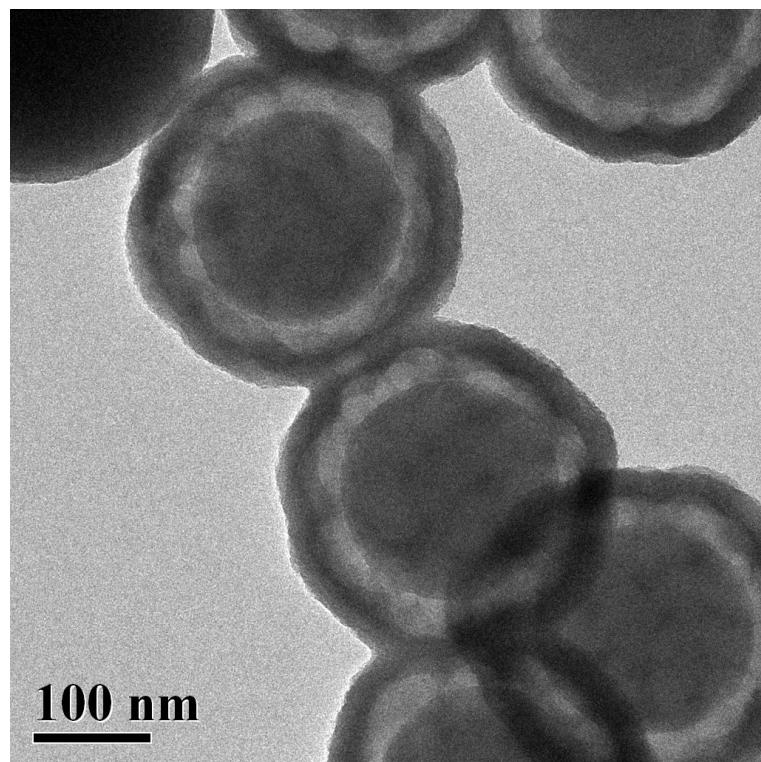


Figure S1. TEM image of PS @mesoSiO₂ YSNs

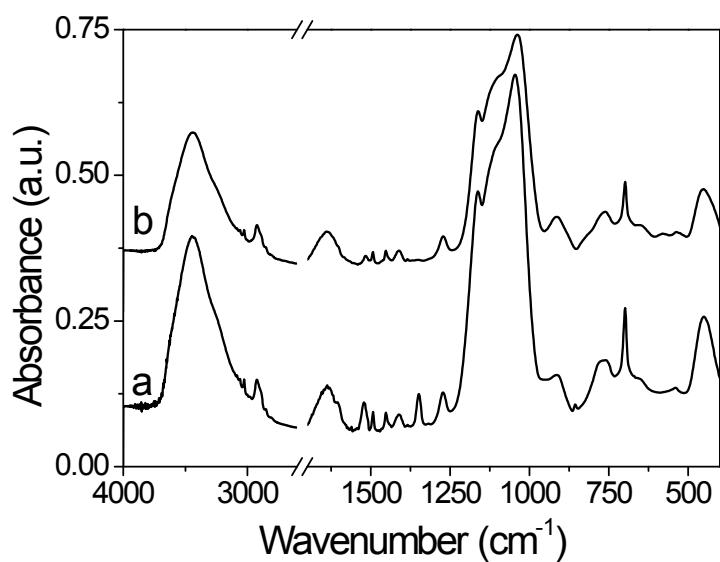


Figure S2. IR spectra for PS@mesoSiO₂-S YSNs (a) before nitration, (b) after nitration with 0.5mL HNO₃.

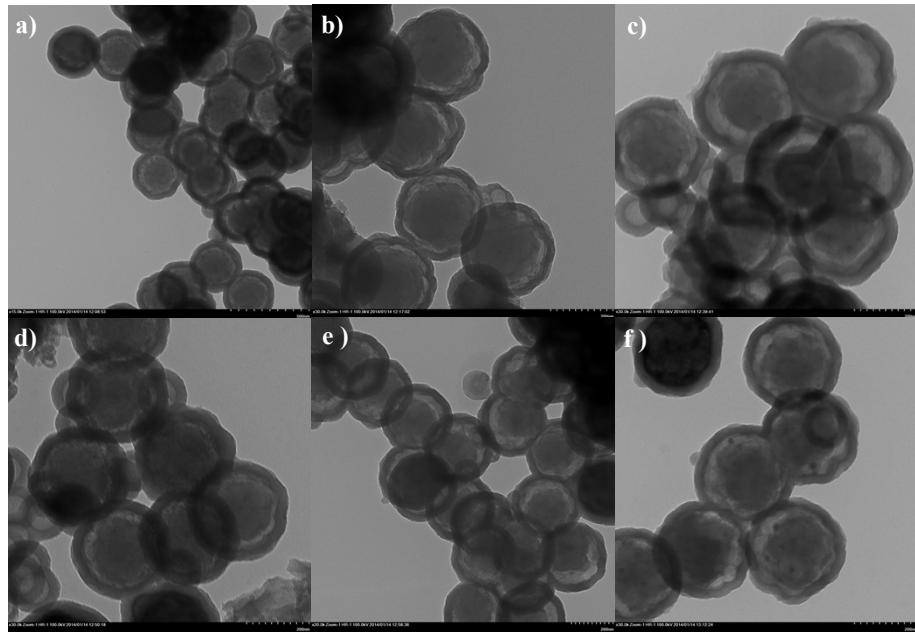


Figure S3. TEM images of $\text{PS-NH}_2@\text{mesoSiO}_2\text{-L}$ after treatment in (a) ethyl acetate, (b) dioxane, (c) acetonitrile, (d) N, N dimethyl amide, (e) thionyl chloride and (f) dichloromethane for 19 h.

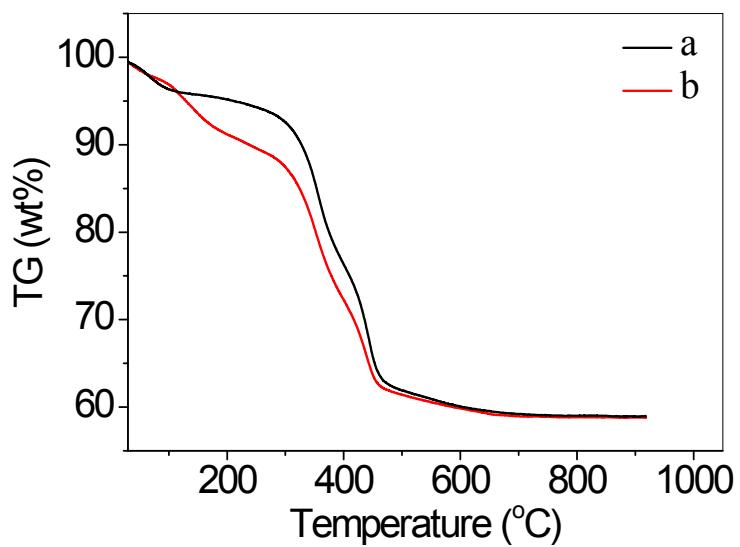


Figure S4. TG curves of (a) $\text{PS-NH}_2@\text{mesoSiO}_2\text{-S}$, (b) $\text{PS-NH}_2@\text{mesoSiO}_2\text{-S}$ after treatment with TEA/toluene=1 under the air.

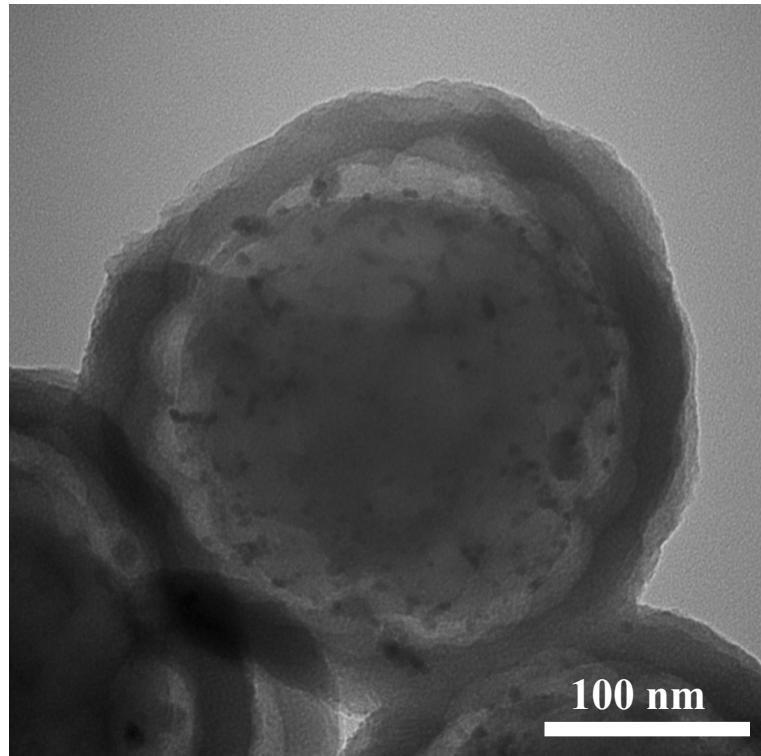


Figure S5. TEM image of 2wt%Pd/ PS-NH₂@mesoSiO₂-L after 1 cycle.

Table S1. Textural properties of Pd/PS-NH₂@mesoSiO₂ catalysts.

samples	S_{BET} (m ² ·g ⁻¹)	Pore volume (cm ³ ·g ⁻¹)	Pore diameter (nm)	N content (wt%)
2wt%Pd/PS-NH ₂ @mesoSiO ₂ -S YSNs	540	0.45	1.3/2.4	1.17
2wt%Pd/PS-NH ₂ @mesoSiO ₂ -L YSNs	507	0.41	1.3/2.4	3.38
1wt%Pd/PS-NH ₂ @mesoSiO ₂ -L YSNs	521	0.42	1.3/2.4	3.44