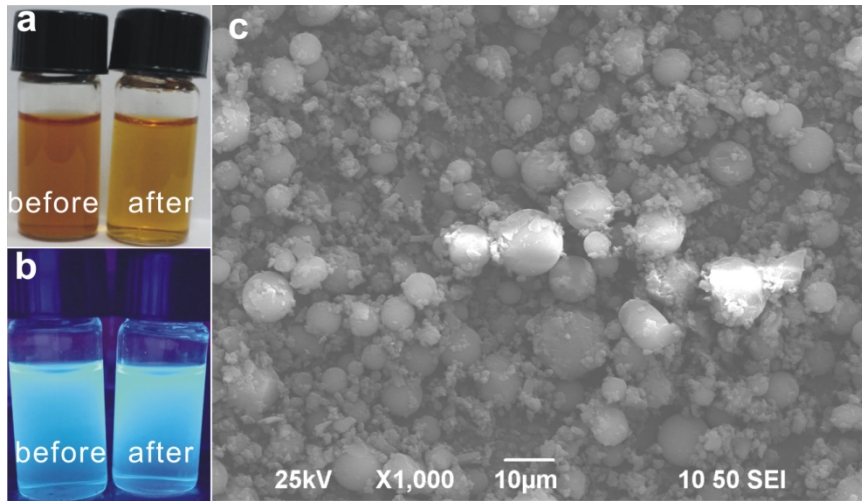


**Fluorescent carbon quantum dots, capacitance and catalysis active porous carbon microspheres  
from beer**

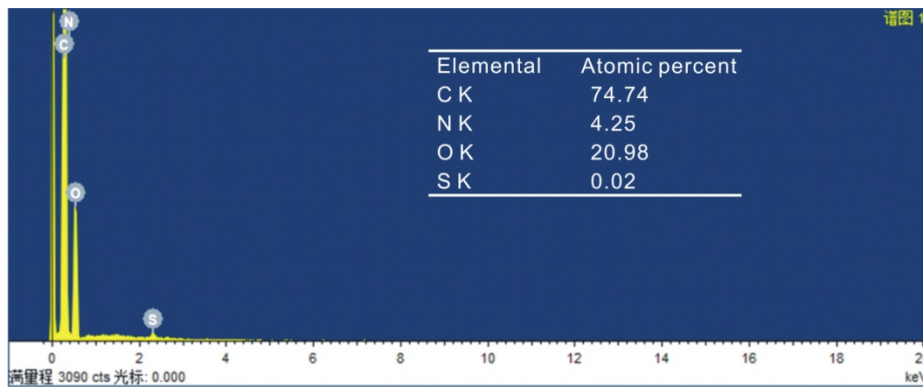
**Zhiyong Gao,<sup>a</sup> Xiaorui Wang,<sup>a</sup> Jiuli Chang,<sup>a</sup> Dapeng Wu,<sup>a</sup> Lan Wang,<sup>a</sup> Xiao Liu,<sup>a</sup> Fang Xu,<sup>a</sup>  
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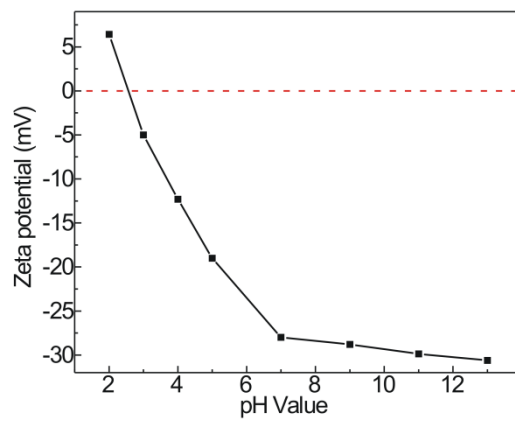
*E-mail: [kjiang512@163.com](mailto:kjiang512@163.com)*



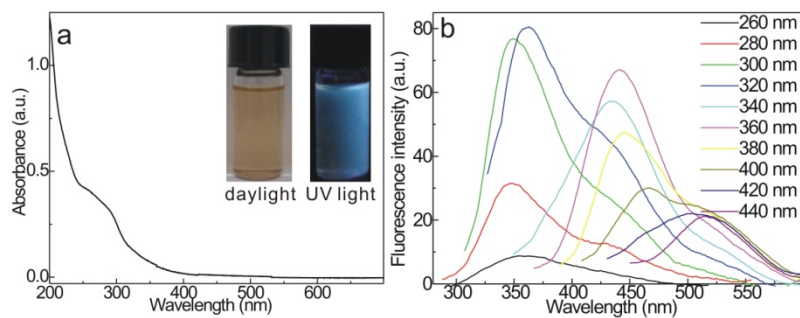
**Fig. S1** CQDs solution before and after hydrothermal treatment under natural light (a) and UV lamp (b).  
(c) SEM of the formed precipitate.



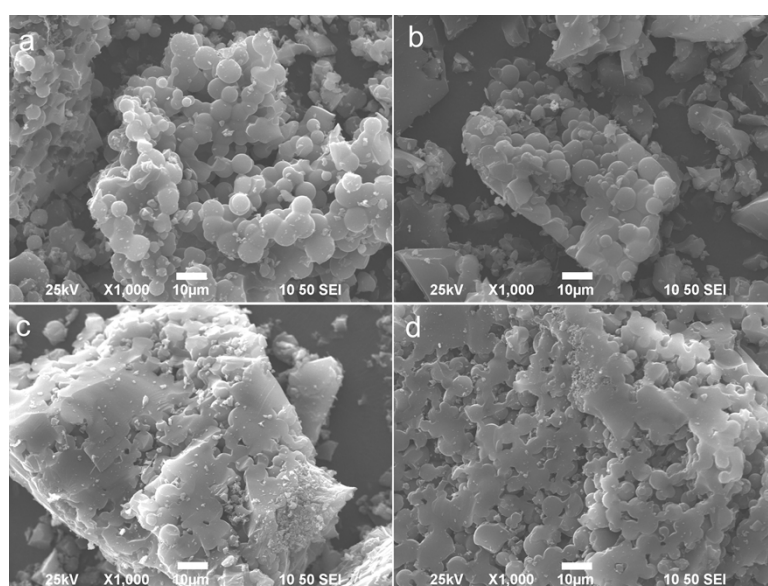
**Fig.S2** EDS of NCQDs.



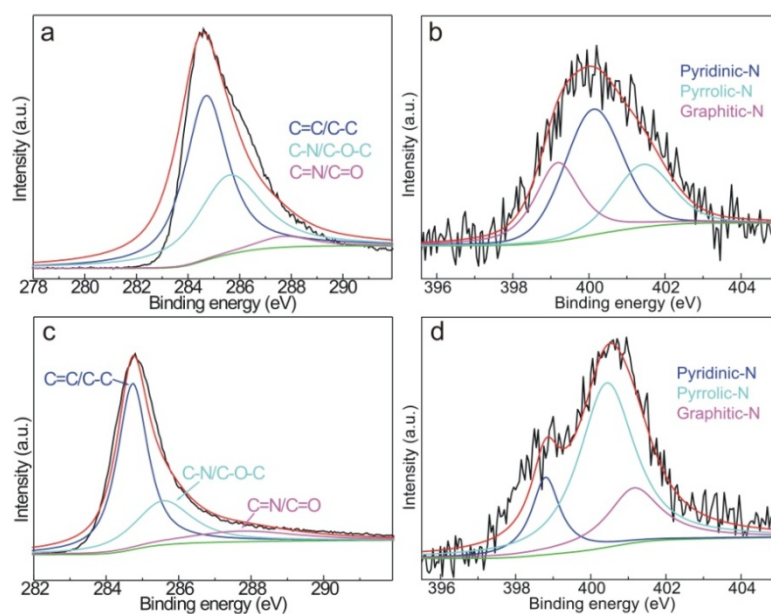
**Fig. S3** Zeta of CQDs aqueous solution at different pH media.



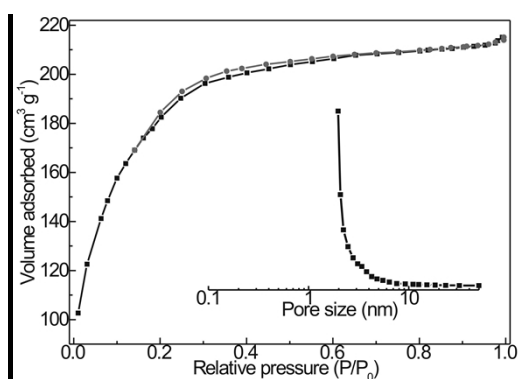
**Fig. S4** (a) UV-vis absorption spectrum and (b) excitation wavelength dependent fluorescence spectra of maltose hydrothermal synthesized carbon quantum dots. Inset in panel a: image of maltose derived carbon quantum dots solution under daylight and UV light.



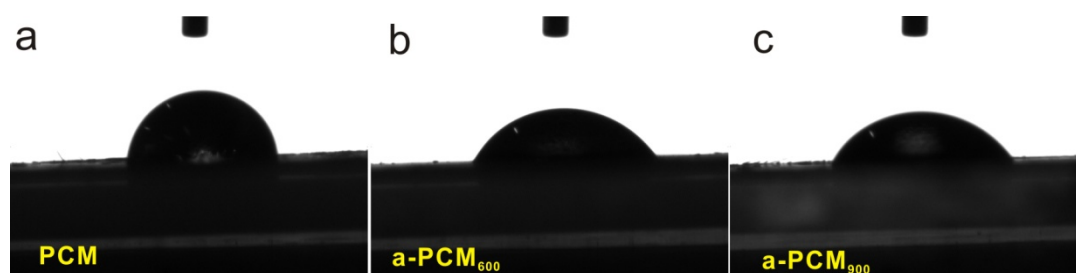
**Fig. S5** SEM of (a) a-PCM<sub>600</sub>, (b) a-PCM<sub>700</sub>, (c) a-PCM<sub>800</sub>, and (d) a-PCM<sub>900</sub>.



**Fig. S6** High resolution C1s (a, c) and N1s (b, d) XPS spectra of PCM (a, b) and a-PCM<sub>600</sub> (c, d).



**Fig. S7** N<sub>2</sub> sorption isotherm and BJH pore size distribution of PCM.

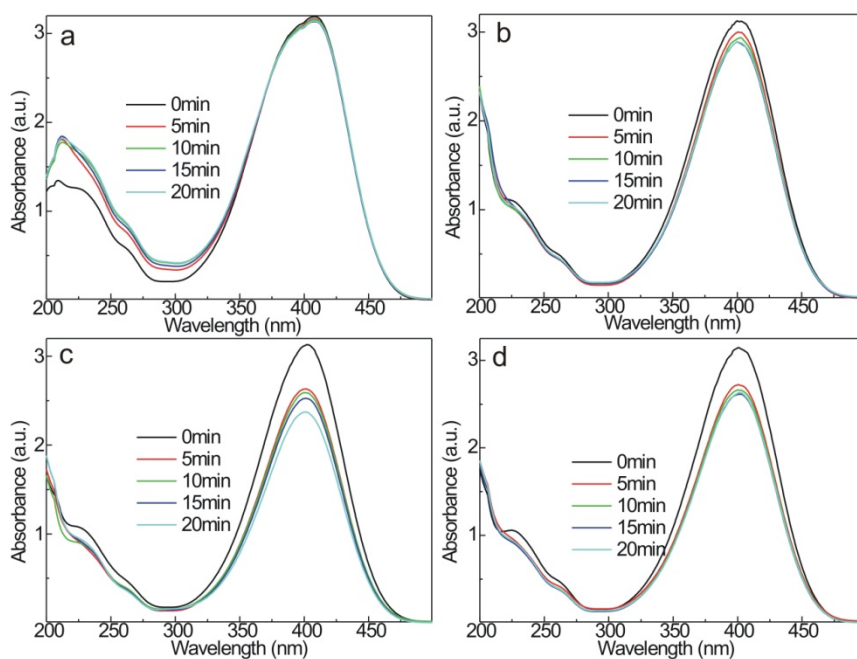


**Fig. S8** Contact angle tests of (a) PCM, (b) a-PCM<sub>600</sub> and (c) a-PCM<sub>900</sub>.

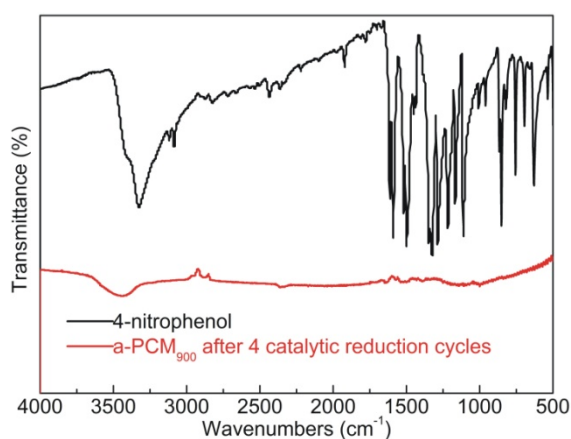
**Table S1** Surface and pore properties of PCM and a-PCMs.

Sample	$S_{\text{BET}}^{\text{a}}$ (m <sup>2</sup> g <sup>-1</sup> )	$S_{\text{micro}}^{\text{b}}$ (m <sup>2</sup> g <sup>-1</sup> )	$S_{\text{meso}}^{\text{c}}$ (m <sup>2</sup> g <sup>-1</sup> )	Pore volume <sup>d</sup> (cm <sup>3</sup> g <sup>-1</sup> )
PCM	488.2	317.6	170.6	0.246
a-PCM <sub>600</sub>	2120.7	1613.0	507.7	1.163
a-PCM <sub>700</sub>	1917.8	1467.3	450.5	1.028
a-PCM <sub>800</sub>	1779.5	1379.4	400.1	0.956
a-PCM <sub>900</sub>	1727.2	1214.6	512.6	0.947

<sup>a</sup> Specific surface area from multiple BET method. <sup>b</sup> Micropore surface area from t-plot method. <sup>c</sup> The t-method external surface area ( $S_{\text{meso}}=S_{\text{BET}}-S_{\text{micro}}$ ). <sup>d</sup> Total pore volume at  $p/p_0=0.99$ .



**Fig. S9** UV-vis absorption spectra during the catalytic reduction of 4-nitrophenol by (a) PCM, (b) a-PCM<sub>600</sub>, (c) a-PCM<sub>700</sub> and (d) a-PCM<sub>800</sub>.



**Fig. S10** FTIR of 4-nitrophenol and the a-PCM<sub>900</sub> catalyst after 4 cycles of catalytic reduction reaction.

From the FTIR spectra, characteristic absorption peaks belonging to 4-nitrophenol can not be observed in the catalyst, which excludes the adsorption of 4-nitrophenol onto catalyst. This result also manifests that decreased UV-Vis absorbance of 4-nitrophenol is caused by the catalytic reduction reaction.