Supporting Information

Highly Monodisperse Magnetite/Carbon Composite Microspheres with

Mesoporous Structure as High-Performance Lithium-Ion Battery Anodes

Hyung-Seok Lim‡^a, Daun Kim‡^a, Jun-Ki Hwang^a, Yu-Jeong Kim^a, Yang-Kook Sun^b and Kyung-Do Suh^{*a}

^a Department of Chemical Engineering, College of Engineering, Hanyang University, Seoul, Republic of Korea,

133-791

^b Department of WCU Energy Engineering, College of Engineering, Hanyang University, Seoul, Republic of

Korea, 133-791

* Corresponding author. Tel.: +82-2-2220-0526, fax: +82-2-2220-4680.

E-mail: kdsuh@hanyang.ac.kr (K. -D. Suh)

‡ These authors contributed equally to this work.



Figure S1. OM images of the polymerization processes of highly monodisperse PS microspheres with a mesoporous structure; (a) PS seed microspheres, (b) swollen PS microspheres with second monomer mixture with heptane as a diluent and (c) porous PS microspheres after polymerization. (d) SEM image of highly monodisperse porous PS microspheres after dry.



Figure S2. SEM images of highly monodisperse PS microspheres with a mesoporous structure after sulfonation process; (a and c) SPS-H4 and (b and d) SPS-H7.



Figure S3. SEM images of porous carbon microspheres obtained from heat-treatment of SPS microspheres without Fe₃O₄ nanocrystals; (a) PC-H4 and (b) PC-H7.



Figure S4. Raman spectra of porous carbon microspheres obtained from heat-treatment of SPS microspheres without Fe₃O₄ nanocrystals; (black line) PC-H4 and (red line) PC-H7.



Figure S5. Voltage profiles of (a) PC-H4, (b) PC-H7 composite electrodes in the 0.02-3V voltage window at (first two cycles) 1/5 and (third and 100th cycles) 1C.



Figure S6. (a) Cycling performance of the composite electrode made with bare Fe₃O₄ nanoparticles in the 0.02-3V voltage window at 1/20 C (first two cycles) and 1C.
(b) The specific capacity of the composite electrode made with bare Fe₃O₄ nanoparticles as a function of the cycling rate (0.1-5 C).

List of Figures

- Figure S1. OM images of the polymerization processes of highly monodisperse PS microspheres with a mesoporous structure; (a) PS seed microspheres, (b) swollen PS microspheres with second monomer mixture with heptane as a diluent and (c) porous PS microspheres after polymerization. (d) SEM image of highly monodisperse porous PS microspheres after dry.
- Figure S2. SEM images of highly monodisperse PS microspheres with a mesoporous structure after sulfonation process; (a and c) SPS-H4 and (b and d) SPS-H7.
- Figure S3. SEM images of porous carbon microspheres obtained from heat-treatment of SPS microspheres without Fe₃O₄ nanocrystals; (a) PC-H4 and (b) PC-H7.
- Figure S4. Raman spectra of porous carbon microspheres obtained from heat-treatment of SPS microspheres without Fe₃O₄ nanocrystals; (black line) PC-H4 and (red line) PC-H7.
- Figure S5. Voltage profiles of (a) PC-H4, (b) PC-H7 composite electrodes in the 0.02-3V voltage window at (first two cycles) 1/5 and (third and 100th cycles) 1C.
- Figure S6. (a) Cycling performance of the composite electrode made with bare Fe₃O₄ nanoparticles in the 0.02-3V voltage window at 1/20 C (first two cycles) and 1C.
 (b) The specific capacity of the composite electrode made with bare Fe₃O₄ nanoparticles as a function of the cycling rate (0.1-5 C).