Electronic Supplementary Information

Hydrothermal synthesis and enhanced photocatalytic activity of ternary

Fe₂O₃/ZnFe₂O₄/ZnO nanocomposite through cascade electron transfer

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Fig. S1 EDAX patterns of of a) α-Fe₂O₃, b) Fe₂O₃/ZnFe₂O₄ (Fe:Zn=90:10), c) Fe₂O₃/ZnFe₂O₄ (Fe:Zn=80:20), d) Fe₂O₃/ZnFe₂O₄/ZnO (Fe:Zn=70:30) and e)Fe₂O₃/ZnFe₂O₄/ZnO (Fe:Zn=60:40)



Fig. S2 Variation of percentage degradation of 100 ml of 20 mg L⁻¹ MG with different pH using 0.1 g of Fe₂O₃/ZnFe₂O₄/ZnO (Fe:Zn=70:30) nanocomposite under solar light irradiation.



Fig. S3 Variation of percentage degradation of 100 ml of MG with different concentration using 0.1 g of Fe₂O₃/ZnFe₂O₄/ZnO (Fe:Zn=70:30) nanocomposite under solar light irradiation.



Fig. S4 Photoluminescence emission spectra of a) α-Fe₂O₃ nanoparticle, b) Fe₂O₃/ZnFe₂O₄ (Fe:Zn=90:10), c) Fe₂O₃/ZnFe₂O₄ (Fe:Zn=80:20), d) Fe₂O₃/ZnFe₂O₄/ZnO (Fe:Zn=70:30), e)Fe₂O₃/ZnFe₂O₄/ZnO (Fe:Zn=60:40), f) ZnFe₂O₄/ZnO (Fe:Zn=50:50) nanocomposites at an excitation wavelength of 420 nm.

Table ST1 Elemental composition of nanocomposites obtained from EDAX analysis

Elements	a-Fe ₂ O ₃		Fe ₂ O ₃ /ZnFe ₂ O ₄		Fe ₂ O ₃ /ZnFe ₂ O ₄		Fe ₂ O ₃ /ZnFe ₂ O ₄ /ZnO		Fe ₂ O ₃ /ZnFe ₂ O ₄ /ZnO		ZnFe ₂ O ₄ /ZnO	
			(Fe:Zn=90:10)		(Fe:Zn=80:20)		(Fe:Zn=70:30)		(Fe:Zn=60:40)		(Fe:Zn=50:50)	
	Wt%	At%	Wt%	At%	Wt%	At%	Wt%	At%	Wt%	At%	Wt%	At%
0	36.66	66.89	28.25	58.08	23.02	51.91	16.08	41.05	20.24	47.83	20.24	47.83
Fe	63.34	33.11	63.80	37.93	59.71	38.57	61.38	44.88	49.06	33.35	39.06	26.35
Zn			07.95	03.99	17.26	09.52	22.53	14.07	30.71	18.82	40.71	25.82
Total	100	100	100	100	100	100	100	100	100	100	100	100