Ultrasmall Cu₂O Nanocrystals: Facile Synthesis, Controllable

Assembly and Photocatalytic Properties

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Figure S1 XRD patterns of Cu₂O USNCs with different amount surfactants: (a) 0.2g PVP and 0.2ml OA; (b) 0.2g PVP and 2ml OA; (c) 0.5g PVP and 0.5ml OA; (d) 0.5g PVP and 2ml OA.



Figure S2. TEM images of Cu₂O USNCs: (a) 0.5g PVP and 0.5ml OA; (b) 0.5g PVP and 1ml OA; (c) 0.2g PVP and 0.2ml OA; (d) 0.2g PVP and 1ml OA.



Figure S3. SAED images of the Cu₂O USNCs



Figure S4 SEM image of the self-assembly Cu₂O superlattices with 3ml DEC in cyclohexane



Figure S5 TEM image of the Cu_2O nanoparticles.



Figure S6 TEM images of the self-assembly of Cu₂O nanoparticles in cyclohexane: (a) 0.5ml DEC; (b) 1.0ml DEC; (c) 1.5ml DEC and; (d) 3ml DEC.



Figure S7 XRD pattern of CuO sample by directly evaporating in air in a beaker



Figure S8 SEM and TEM image of the as-obtained Cu₂O-Au samples: (a) SEM image of the Cu₂O-Au; (b) TEM image of the single Cu₂O-Au colloidal spheres; (c) HRTEM image of the Au nanoparticle



Figure S9 UV–vis diffuse reflectance spectra of $Cu_2O(a)$ and $Cu_2O-Au(2)(b)$



Figure S10 UV–vis absorption spectra of MO aqueous solution with the Cu₂O–Au(2).