## **Supplementary Information**

## Facile Morphology-controllable Hydrothermal Synthesis and Color Tunable

## Luminescence Properties of NaGd(MoO<sub>4</sub>)<sub>2</sub>: Eu<sup>3+</sup>, Tb<sup>3+</sup> Microcrystals

Anming Li,<sup>a,b</sup> Dekang Xu,<sup>a</sup> Hao Lin,<sup>a</sup> Shenghong Yang,<sup>a</sup> Yuanzhi Shao,<sup>a</sup> Yueli Zhang,<sup>\*a</sup> and Zhenqiang Chen<sup>b</sup>

- <sup>a.</sup> State Key laboratory of Optoelectronic Materials and Technologies, School of Physics and Engineering, Sun Yat-sen University, Guangzhou 510275, China.
  E-mail: stszyl@mail.sysu.edu.cn
- <sup>b.</sup> Institute of Optoelectronic Engineering, Jinan University, Guangzhou 510632, China



**Fig. S1** TEM images, HRTEM images and SAED patterns of NaGd(MoO<sub>4</sub>)<sub>2</sub> samples with the morphology of bipyramids and tetragonal plates: (a) and (b) bipyramids, (c) and (d) tetragonal plates.



**Fig. S2** SEM images of the NaGd(MoO<sub>4</sub>)<sub>2</sub> samples synthesized at different Na<sub>2</sub>MoO<sub>4</sub>/Re(NO<sub>3</sub>)<sub>3</sub> molar ratios (4:1, 6:1 and 8:1) and different pH values (pH = 3.0, 4.0, 5.0 and 6.0).



**Fig. S3** XRD patterns of the NaGd(MoO<sub>4</sub>)<sub>2</sub> samples synthesized at different Na<sub>2</sub>MoO<sub>4</sub>/Re(NO<sub>3</sub>)<sub>3</sub> molar ratios (4:1, 6:1 and 8:1) and different pH values (pH = 3.0, 4.0, 5.0 and 6.0).



**Fig. S4** Excitation and emission spectra of  $NaGd(MoO_4)_2$ : 5% Tb<sup>3+</sup> microplates.



**Fig. S5** Spectral overlaps between emission spectrum of  $NaGd(MoO_4)_2$ : Tb<sup>3+</sup> and excitation spectrum of  $NaGd(MoO_4)_2$ : Eu<sup>3+</sup>.



**Fig. S6** Excitation and emission spectra of NaGd(MoO<sub>4</sub>)<sub>2</sub>: 1% Eu<sup>3+</sup>, 4% Tb<sup>3+</sup> microplates.



**Fig. S7** Emission spectra of NaGd(MoO<sub>4</sub>)<sub>2</sub>:  $Eu^{3+}$ ,  $Tb^{3+}$  microplates with different doping concentrations excited by near-UV light at 380 nm.