

## Electronic Supplementary Information

### Au nanoparticles decorated WO<sub>3</sub> photoelectrode for enhanced photoelectrochemical properties

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#### Experimental details

##### 1. Synthesis of WO<sub>3</sub> nanoparticles

All of the reagents were of analytical grade and were used without further purification. WO<sub>3</sub> nanoparticles were prepared via hydrothermal method reported recently with some modification<sup>26</sup>. The typical experimental procedure was as follows: 0.125 g of H<sub>2</sub>WO<sub>4</sub> and 0.5 mL H<sub>2</sub>O<sub>2</sub> (30 wt%) were dissolved into 2 mL of distilled water and then stirred intensely at 100 °C for 30 min. After being cooled, 0.35 mL of HCl (37 wt%), 0.0315 g oxalic acid dehydrate, 7.5 mL ethanol and 7.5 mL distilled water were added to the above solution. The obtained solution was transferred to 25 mL Teflon-lined autoclave at a constant temperature of 180 °C for 4 h. The precipitate was collected and washed repeatedly with distilled water, and was then dried at 60 °C under vacuum followed by calcinations in air at 550 °C for 2 h.

##### 2. Fabrication of WO<sub>3</sub>/FTO photoelectrode

F: SnO<sub>2</sub> coated conductive glass flakes (FTO, 1.5\*2.5 cm<sup>2</sup>) were cleaned by distilled water, ethanol and acetone. Firstly, a paste was prepared by adding hydroxypropyl cellulose (Aldrich) into diethylene glycol with a concentration of 10 wt% and heated at 100 °C for 5 h. Then, WO<sub>3</sub> nanoparticles were dispersed into the paste with the quality ratio of 1:3, and absolute ethanol was added and grounded in an agate mortar to form viscous slurry. Finally, the slurry was dropped onto a piece of FTO glass by doctor blading method, and the film was heated up to 550 °C at a rate of 5 °C min<sup>-1</sup> and maintained for 30 min after drying naturally.

##### 3. Preparation of Au decorated WO<sub>3</sub> photoelectrode

The Au nanoparticles sol was prepared by reducing  $\text{HAuCl}_4$  using  $\text{NaBH}_4$  as reductant in the presence of polyvinyl pyrrolidone (PVP). Briefly, 0.01 g PVP was added into 10 ml of 1 mM  $\text{HAuCl}_4$  aqueous solution and was heated at  $100^\circ\text{C}$  in a round bottom flask for 30 min. Then, 1 ml of 0.04 M  $\text{NaBH}_4$  aqueous solution was added and kept reaction for 30 min, and finally Au nanoparticles sol was obtained.

For preparation of Au nanoparticles decorated  $\text{WO}_3$  photoelectrode, 20  $\mu\text{L}$  of the Au sol was dropped onto  $\text{WO}_3$  film, and was then calcined at  $300^\circ\text{C}$  at a rate of  $3^\circ\text{C min}^{-1}$  for 2 h. One drop-coating process was defined as one cycle. For comparison, a series of Au decorated  $\text{WO}_3$  film with 1, 5, 10, 15, 20 cycles were prepared, and the samples were denoted as  $\text{WO}_3\text{-1Au}$ ,  $\text{WO}_3\text{-5Au}$ ,  $\text{WO}_3\text{-10Au}$ ,  $\text{WO}_3\text{-15Au}$ ,  $\text{WO}_3\text{-20Au}$  respectively.

#### 4. Characterization

The phase structure of the products were characterized by X-ray diffraction (XRD) using a Bruker advance-D8 with  $\text{Cu K}\alpha$  radiation. The field emission scanning electron microscopy (FESEM) images were obtained on Carl Zeiss AG SUPR40. The transmission electron microscope (TEM) images were obtained on JEOL JSM-100, and the high-resolution transmission electron microscope (HRTEM) images were observed on FEI Tecnai G220. The UV-Vis optical absorption of the samples was carried out on a Perkin-Elmer Lambda 950 UV-Vis spectrometer.

#### 5 Electrical and photoelectrochemical measurements

The photoelectrochemical measurements were carried out on a CHI 660D electrochemical workstation (shanghai chenhua, China) with a standard three-electrode electrochemical cell with a flat quartz window with Au decorated  $\text{WO}_3$  photoelectrode,  $\text{Ag/AgCl}$  and Pt foil using as working electrodes, reference electrode and counter electrode. All the electrodes were performed in a 0.5 M  $\text{Na}_2\text{SO}_4$  (pH=7) solution. The 500 W Xe lamp was used as irradiation source with a light intensity of  $100\text{ mW/cm}^2$ . By recording the wavelength-dependent photocurrent densities ( $J$ ) generated at specific wavelengths ( $\lambda$ ) and the monochromatic light intensities ( $J_{\text{light}}$ ) of the light source, the incident photon to electron conversion efficiency (IPCE) can be calculated using the following equation:

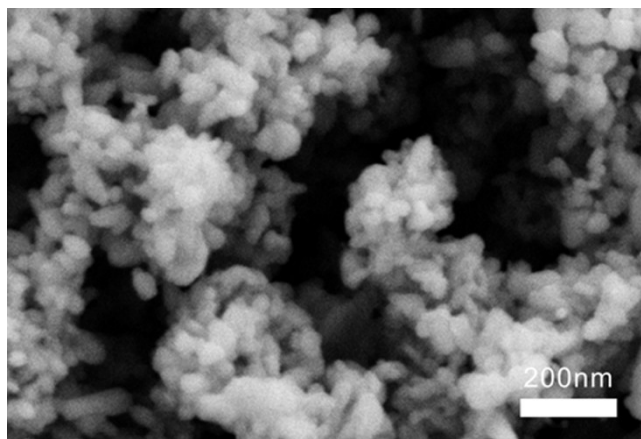
$$\text{IPCE} = 1240I(\text{mA} / \text{cm}^2) / [\lambda(\text{nm})J_{\text{light}}(\text{mW} / \text{cm}^2)]$$

Capacitance measurements were carried out at the AC frequency of 1 kHz in the dark.

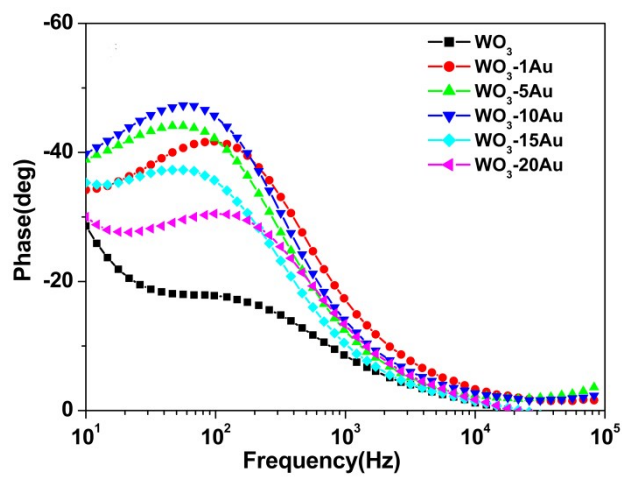
Electrochemical impedance spectroscopy (EIS) was applied to explore the conductivity of the resulting electrodes under illumination.

## References

- 1 H. He, S. P. Berglund, P. Xiao, W. D. Chemelewski, Y. Zhang and C. B. Mullins, *Journal of Materials Chemistry A*, 2013, **1**, 12826-12834.



**Fig. S1.** SEM image of  $\text{WO}_3$  nanoparticle.



**Fig. S2.** Bode phase diagram for  $\text{WO}_3$  and  $\text{WO}_3$ -Au electrode.

**Tab. S1.** Elemental ration of WO<sub>3</sub>-10Au

element	Quality percent	Atomic percent
O K	23.95	78.37
W M	74.85	21.31
Au M	1.20	0.32
total	100.00	

**Tab. S2.** The exact contents (wt %) of Au in the Au decorated WO<sub>3</sub> with different cycles

Sample	WO <sub>3</sub> -1Au	WO <sub>3</sub> -5Au	WO <sub>3</sub> -10Au	WO <sub>3</sub> -15Au	WO <sub>3</sub> -20Au
Contents (wt%)	0.47	0.65	1.20	1.40	2.01