

## Supporting Information

### A terbium metal–organic framework with stable luminescent emission in wide pH range that acts as a quantitative detection material for nitroaromatics

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#### 1. Crystal data collections, structure refinements, bond lengths [Å] and angles [°] for Tb-MOF

Table S1. Crystal data and structure refinement for Tb-MOF

Compound	Tb-MOF
Formula	C <sub>20</sub> H <sub>18</sub> N <sub>2</sub> O <sub>12</sub> Tb
Formula weight	637.28
Crystal system	Trigonal
Space group	R-3
a (Å)	29.4783(15)
b (Å)	29.4783(15)
c (Å)	14.1301(8)
V (Å <sup>3</sup> )	10633.6(9)
Z	18
ρ <sub>calc</sub> (mg·mm <sup>-3</sup> )	1.791
F(000)	5634
μ(mm <sup>-1</sup> )	15.320
T(K)	293
Θ range	3.00-67.23

Reflections collected /unique	21999 / 4235 [R(int) = 0.0725]
Final R indices [ $I > 2\sigma$ ]	0.0399
Final wR indices ( $F^2$ ) [ $I > 2\sigma$ ]	0.0841
Goodness of fit	1.025
Max/min $\Delta\rho(e \text{ \AA}^{-3})$	1.054 and -0.813

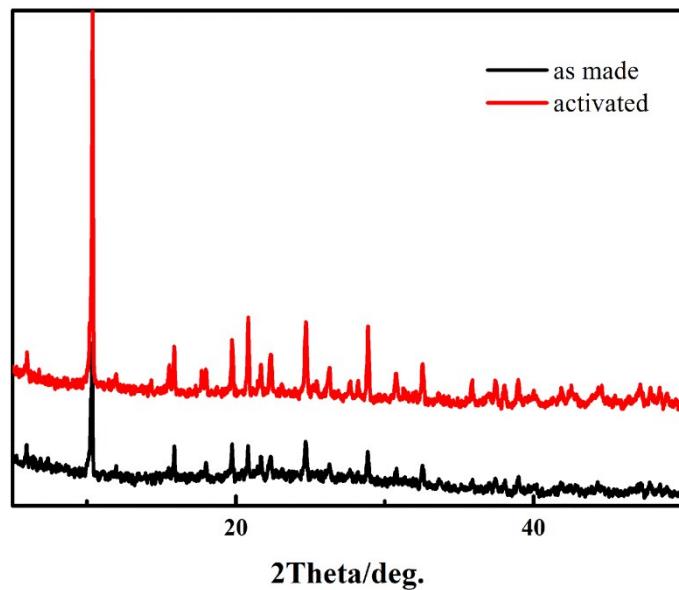
**Table S2. Selected bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ] for Tb-MOF**

Tb(1)-O(6)	2.271(4)	Tb(1)-O(1)	2.425(4)
Tb(1)-O(7)G	2.305(4)	Tb(1)-O(2)	2.481(4)
Tb(1)-O(4)	2.306(4)	Tb(1)-O(10)	2.506(4)
Tb(1)-O(5)G	2.388(4)		
Tb(1)-O(9)	2.421(4)		
O(6)-Tb(1)-O(7)G	151.95(15)	O(9)-Tb(1)-O(1)	129.46(13)
O(6)-Tb(1)-O(4)	95.35(14)	O(6)-Tb(1)-O(2)	77.69(14)
O(7)G-Tb(1)-O(4)	83.14(15)	O(7)G-Tb(1)-O(2)	127.07(13)
O(6)-Tb(1)-O(5)G	76.53(15)	O(4)-Tb(1)-O(2)	70.84(14)
O(7)G-Tb(1)-O(5)G	119.86(14)	O(5)G-Tb(1)-O(2)	73.55(14)
O(4)-Tb(1)-O(5)G	144.39(15)	O(9)-Tb(1)-O(2)	144.86(14)
O(6)-Tb(1)-O(9)	83.63(15)	O(1)-Tb(1)-O(2)	52.91(12)
O(7)G-Tb(1)-O(9)	80.38(15)	O(6)-Tb(1)-O(10)	77.10(15)
O(4)-Tb(1)-O(9)	141.10(14)	O(7)G-Tb(1)-O(10)	76.25(15)
O(5)G-Tb(1)-O(9)	73.29(15)	O(4)-Tb(1)-O(10)	68.98(13)
O(6)-Tb(1)-O(1)	129.23(14)	O(5)G-Tb(1)-O(10)	138.91(14)
O(7)G-Tb(1)-O(1)	78.30(13)	O(9)-Tb(1)-O(10)	72.95(13)
O(4)-Tb(1)-O(1)	80.31(13)	O(1)-Tb(1)-O(10)	141.93(13)
O(5)G-Tb(1)-O(1)	78.74(13)	O(2)-Tb(1)-O(10)	129.62(13)

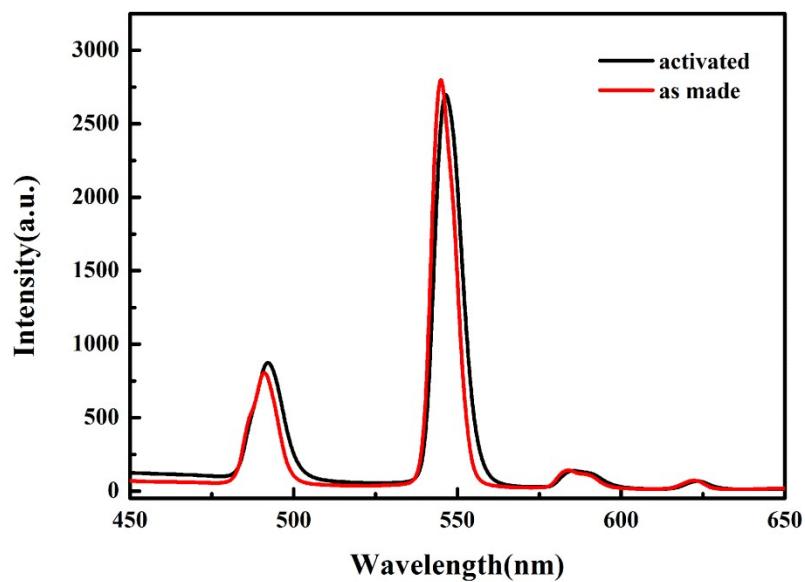
G: 2/3-x+y, 4/3-x, 1/3+z.

## 2. The contrast PXRD and luminescence emission spectra of as made and

activated samples.

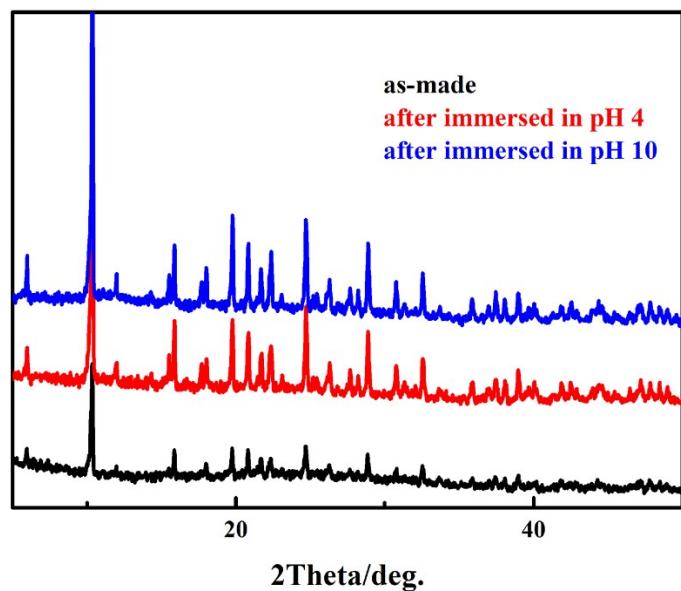


**Fig. S1** PXRD of as made samples and activated samples



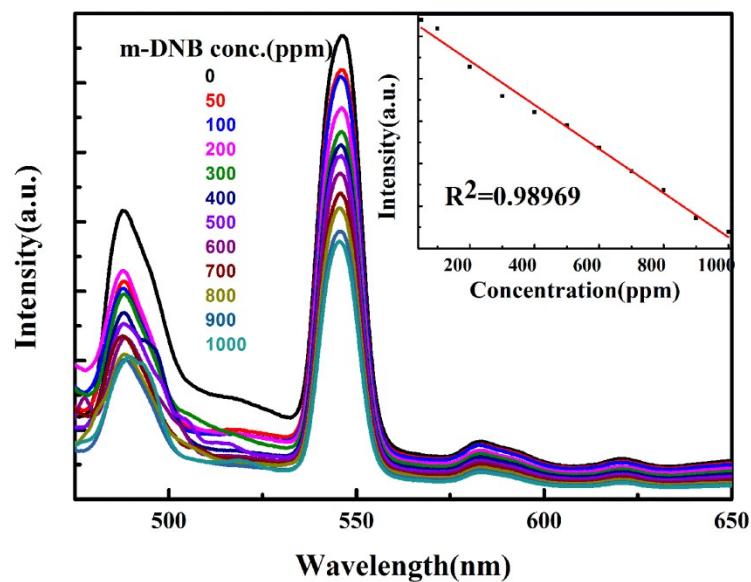
**Fig. S2** the luminescence emission spectra of as made and activated samples

### 3. The PXRD of samples immersed in different pH



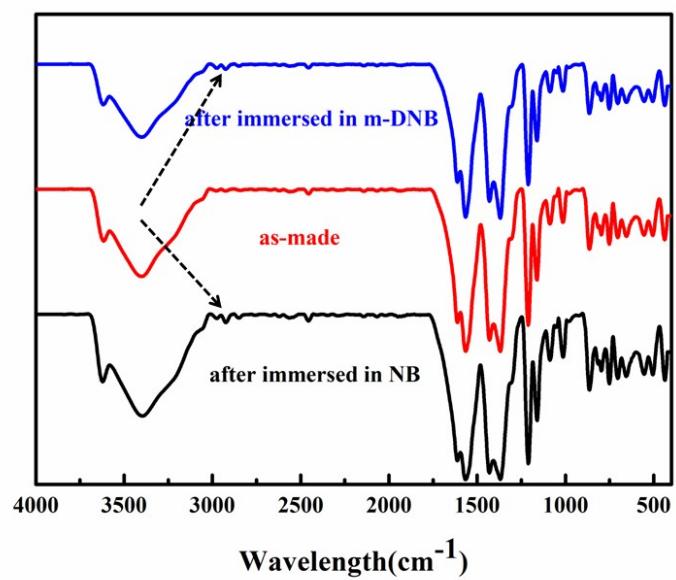
**Fig. S3** The PXRD of samples immersed in different pH

#### 4. The Concentration dependent luminescence for the Tb-MOF dispersed in m-DNB



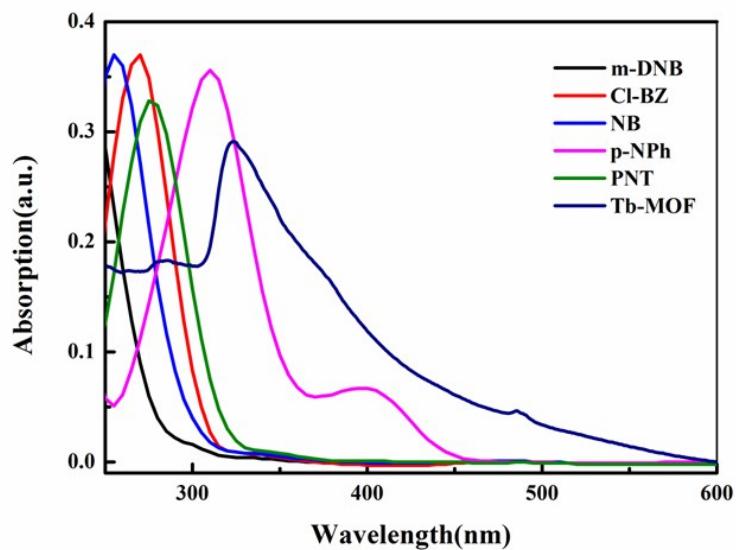
**Fig. S4** Concentration dependent luminescence for the Tb-MOF dispersed in dinitrobenzene (m-DNB)

#### 5. The IR spectra



**Fig. S5** Infrared spectra of the as-made sample of Tb-MOF (red), Tb-MOF after immersed in m-DNB (blue), Tb-MOF after immersed in NB.

## 6. The UV-vis spectra of analytes in ethanol



**Fig. S6** The UV-vis spectra of analytes in ethanol