

Electronic Supplementary Information

**Simple route for gram synthesis of less defective few layered
graphene and its electrochemical performance[†]**

Nazish Parveen, Mohd Omaish Ansari and Moo Hwan Cho*

School of Chemical Engineering, Yeungnam University, Gyeongsan-si, Gyeongbuk 712-749,
South Korea

*Email: mhcho@ynu.ac.kr

Digital photographs of gram synthesis of FLGN

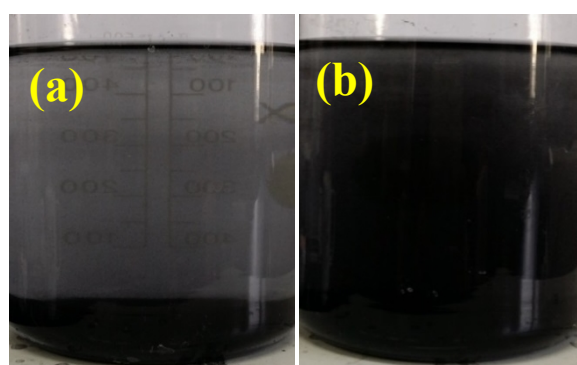


Fig. S1 (a) Exfoliated FLGN suspension at the bottom of electrolyte solution and (b) Dispersed FLGN in electrolyte solution.

I_D/I_G and I_G/I_{2D} ratio bar graph of FLGN

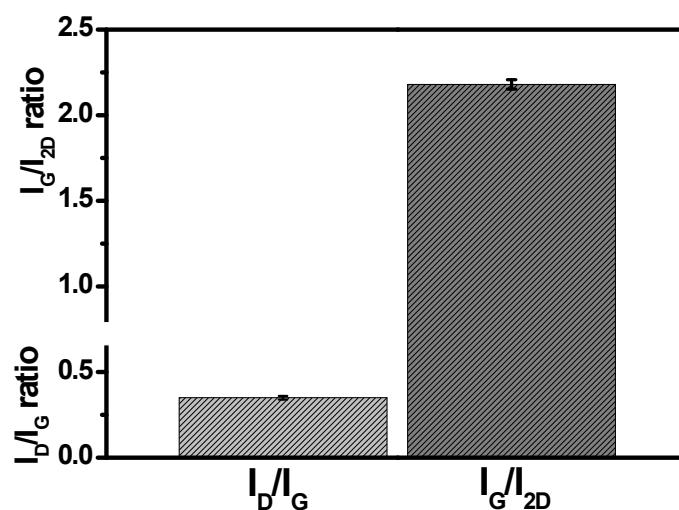


Fig. S2 I_D/I_G and I_G/I_{2D} ratio bar graph of FLGN.

AFM image of FLGN

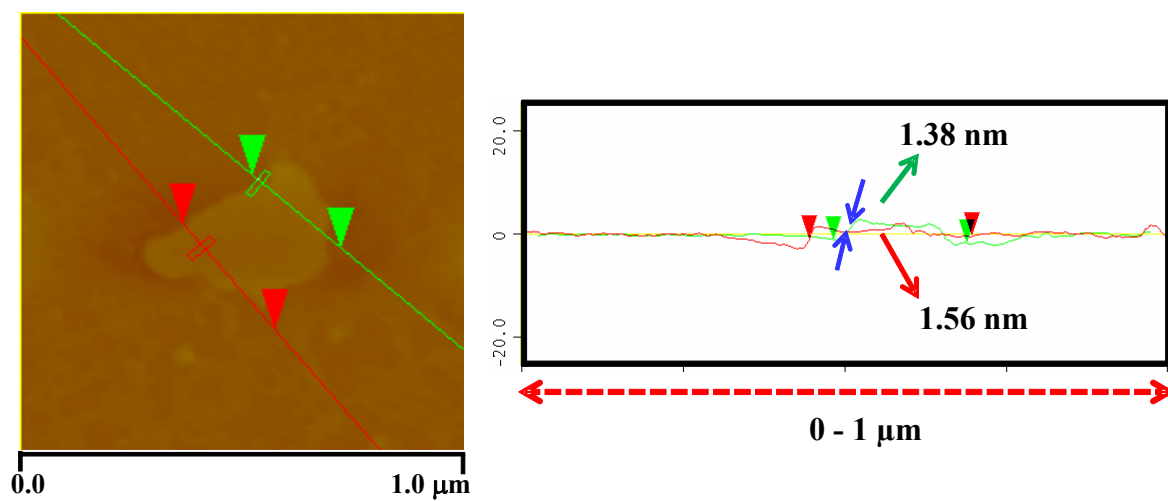


Fig. S3 AFM image of FLGN.

TEM image of FLGN

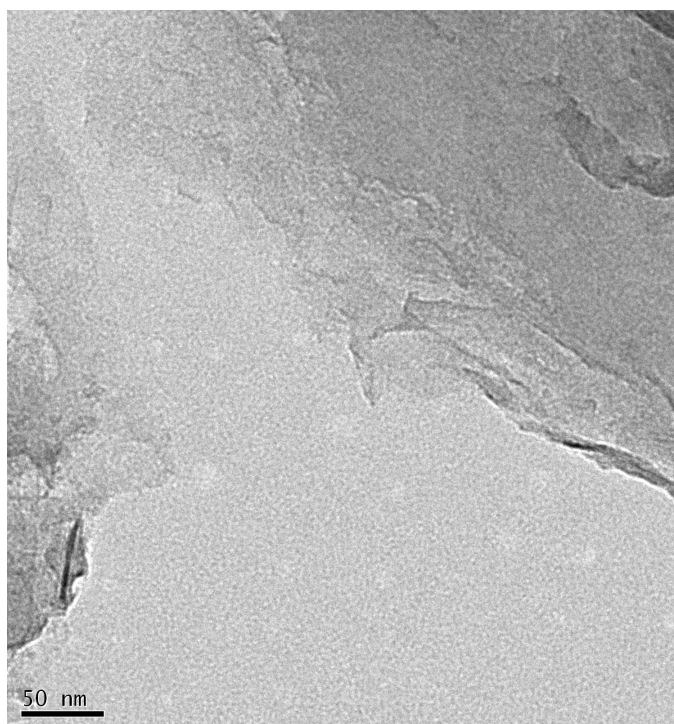


Fig. S4 TEM image of FLGN.

HRTEM images of FLGN

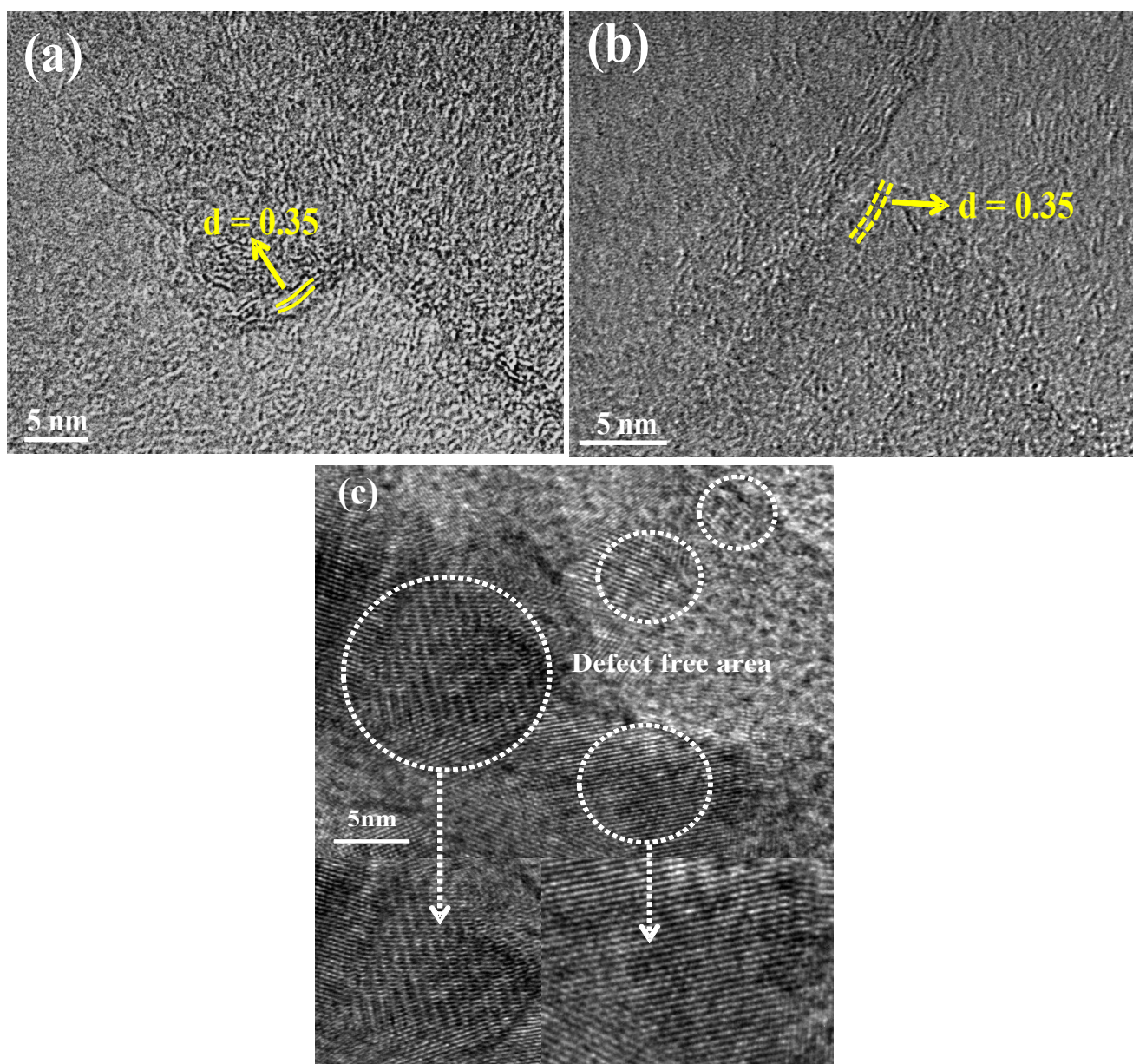


Fig. S5 HRTEM images of FLGN (a, b and c).

XPS survey spectra of FLGN

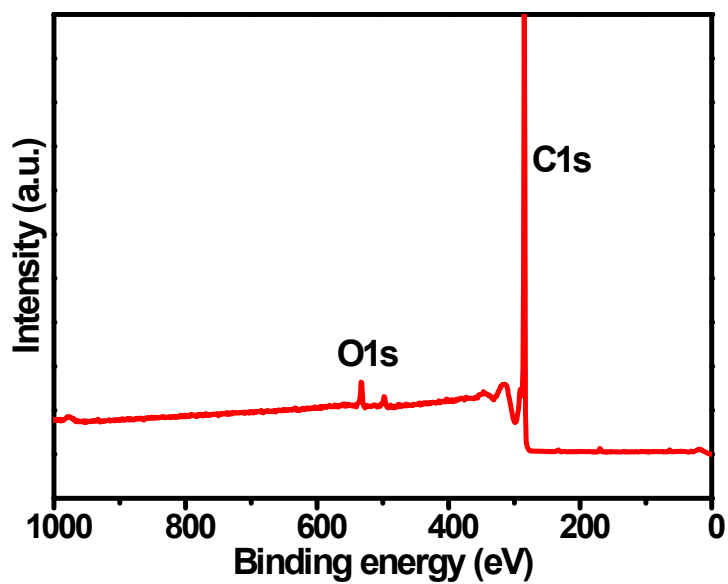


Fig. S6 XPS survey spectra of FLGN.

XRD pattern of graphite sheet and FLGN

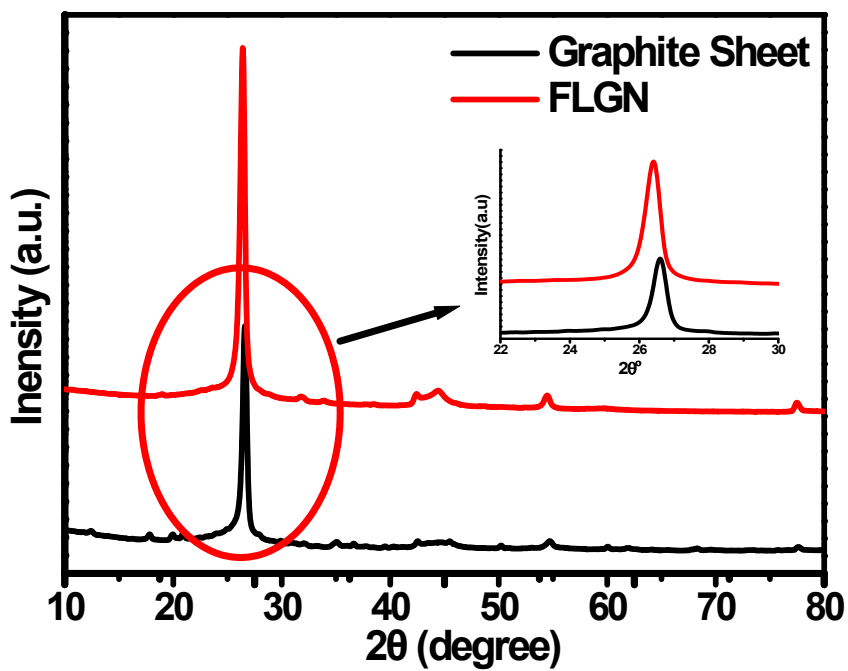


Fig. S7 XRD pattern of the FLGN.

Table S1: Summary of the inorganic salt based electrolytes investigated in the electrochemical exfoliation of graphite sheet

Electrolyte	Electrolyte concentration (M)	Voltage	Time	Results
NaClO₄	0.1	+10V	5 to 10 min	No exfoliation. (Intercalation of graphite electrode occurs)
NaOH	0.1	+10V	5 to 10 min	Poor exfoliation. (Product yield is very Low)
Na₂S₂O₃+ NaOH+ NaClO₄	0.1	+10V	3 to 5 min	Efficient exfoliation. (As described in the text)
NaOH+ Na₂S₂O₃	0.1	+10V	5 to 10 min	Poor exfoliation
Na₂S₂O₃+ NaClO₄	0.1	+10V	5 to 10 min	Poor exfoliation

Table S2: Comparison of the specific capacitances based on GN

Product	Specific capacitance	Scan rate	Reference
exfoliation of graphite (EG)	75 F/g	10 mA/g	33
KOH modified GN	136 F/g	10 mA/g	34
Pillared Graphene-Paper (GP)	138 F/g	10 mA/g	35
FLGN	147 F/g	10 mA/g	Present work