# High performance polydimethylsiloxane pervaporation membranes with hyperbranched polysiloxane as cosslinker for separation of *n*-butanol from water

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### 1. PV experiments method

The PV experiments were carried out at various temperatures using a pervaporation cell with the effective membrane area of  $35.24~\rm cm^2$ . In pervaporation operation, the pressure of downstream side (permeate side) was maintained at  $200~\pm~10$ Pa. The permeated mass was collected in a condensation trap cooled by liquid nitrogen. The composition in feed mixture and the corresponding permeate was detected by gas chromatography (GC) SP-6800A (Shangdong, China) equipped with a packed column and a thermal conductivity detector (TCD).

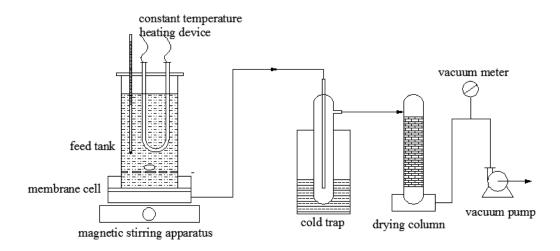


Fig.S1 Schematic diagram of the experimental equipment

### 2. Scanning electron microscope



Fig.S2 Surface morphology of HPSiO-c-PDMS membranes: (a) HPSiO-c-PDMS-1,

(b) HPSiO-c-PDMS-2, (c) HPSiO-c-PDMS-3

# 3. Contact angle measurements of HPSiO-c-PDMS membranes

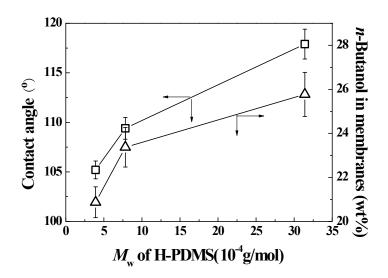


Fig.S3. Effect of H-PDMS molecular weight on *n*-butanol concentration in membranes and contact angle of HPSiO-c-PDMS membranes

## 4. GPC spectroscopy of PDMS

Table S1 Molecular weights amd PDI of H-PDMS and HPSiO

Samples	Molecular weight		PDIc
	$M_{ m w}{}^{ m a}$	$M_{ m n}{}^{ m b}$	
H-PDMS-1	39250	24369	1.61
H-PDMS-2	78263	49437	1.58
H-PDMS-3	314043	195815	1.60
HPSiO	3513	1417	2.48

 $<sup>^{\</sup>rm a}\,M_{\rm w}$ , weight average molecular weight;

 $<sup>^{\</sup>rm b}\,M_{\rm n}$ , number average molecular weight;

<sup>&</sup>lt;sup>c</sup> Polydispersity index (PDI) = $M_{\rm w}/M_{\rm n}$