

First Noscapine Glycoconjugates inspired by Click Chemistry

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1. General methods

All of the reactions were executed using anhydrous solvents under an argon atmosphere in one-hour oven-dried glassware at 100°C. All reagents and solvents were of pure analytical grade. Thin-layer chromatography (TLC) was performed on 60 F254 silica gel, pre-coated on aluminum plates and revealed with either a UV lamp ($\lambda_{max} = 254$ nm) or a specific color reagent (iodine vapors) or by spraying with methanolic H₂SO₄ solution and subsequent charring by heating at 100°C. ¹H and ¹³C NMR were recorded at 300 and 75 MHz, respectively. Chemical shifts given in ppm downfield from internal TMS; *J* values in Hz. Mass spectra recorded using electrospray ionization mass spectrometry (ESI-MS). Infrared spectra recorded as Nujol mulls in KBr plates. Single-crystal Xray data collected on Xcalibur Eos (Oxford) CCD-diffractometer.

2. Synthesis of novel tosyl sugar (5i-j):

1,2-*O*-isopropylidene-3-*O*-Propyl-6-*O*-tosyl- α -D-glucofuranose (5i). A stirring solution of 3-1,2-*O*-isopropylidene-*O*-Propyl- α -D-glucofuranose (1.34 g, 5.1 mmol) in pyridine (15 mL) was treated with *p*-toluene sulphonyl chloride (1.0 g, 5.1 mmol) at 0 °C under anhydrous condition followed by stirring for 12 h at 5-10 °C afforded viscous liquid (1.3 g, 62%); ¹H NMR (300 MHz, CDCl₃): δ 7.79 (d, *J* = 8.1 Hz, 2H), 7.34 (d, *J* = 7.8 Hz, 2H), 5.86 (d, *J* = 3.6 Hz, 1H), 4.52 (d, *J* = 3.6 Hz, 1H), 4.28-3.96 (m, 5H), 3.57 (dd, *J* = 6.6 and 15.6 Hz, 1H), 3.43 (dd, *J* = 6.6 and 15.6 Hz, 1H), 2.52-2.44 (m, 4H), 1.62-1.30 (m, 8H), 0.90 (t, *J* = 7.2 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃): 144.9, 132.5, 129.9, 128.0, 111.8, 105.0, 82.7, 81.9, 79.0, 72.1, 67.5, 26.7, 26.1, 22.8, 21.6, 10.5 ppm.

3-*O*-iso-pentyl-1,2-*O*-isopropylidene-6-*O*-tosyl- α -D-glucofuranose (5j). A stirring solution of 3-*O*-isopentyl-1,2-*O*-isopropylidene- α -D-glucofuranose (1.6 g, 5.7 mmol) in pyridine (15 mL) was treated with *p*-toluene sulphonyl chloride (1.08 g, 5.7 mmol) at 0 °C under anhydrous

condition followed by stirring for 12 h at 10 °C afford the title compound as viscous liquid (1.5 g, 60%); ¹H NMR (300 MHz, CDCl₃): δ 7.79 (d, *J* = 8.1 Hz, 2H), 7.34 (d, *J* = 8.1 Hz, 2H), 5.85 (d, *J* = 3.6 Hz, 1H), 4.52 (d, *J* = 3.6 Hz, 1H), 4.26-3.95 (m, 5H), 3.64 (dd, *J* = 6.6, 15.6 Hz, 1H), 3.49 (dd, *J* = 6.6, 15.9 Hz, 1H), 2.88 (d, *J* = 5.57 Hz, 1H) 2.44 (s, 3H), 1.69-1.23 (m, 13H), 0.91 (d, *J* = 6.6 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃): 144.9, 132.5, 129.9, 128.0, 111.8, 105.0, 82.8, 81.9, 79.0, 72.2, 68.9, 67.5, 38.4, 26.7, 26.2, 24.9, 22.5, 21.6 ppm.

3. Synthesis of novel sugar azides (6i-k):

6-Azido-6-deoxy-1,2-*O*-isopropylidene-3-*O*-propyl- α -D-glucofuranose (6i): Reaction of compound **5j** (1.34 g, 3.2 mmol) with NaN₃ (0.62 g, 9.6 mmol) in DMF (15 mL) at 80 °C afforded compound **6i** as viscous liquid (0.81 g, 88 % yield); IR (KBr) cm⁻¹: 3453, 2964, 2935, 2878, 2103, 1633, 1455, 1080; ¹H NMR (300 MHz, CDCl₃): δ 5.91 (d, *J* = 3.6 Hz, 1H), 4.57 (d, *J* = 3.9 Hz, 1H), 4.10 (m, 2H), 4.00 (s, 1H), 3.66-3.43 (m, 4H), 2.74 (s, 1H), 1.65-1.56 (m, 2H), 1.49 (s, 3H), 1.32 (s, 3H), 0.94 (t, *J* = 7.5 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃): δ 111.8, 105.0, 82.8, 81.9, 79.7, 72.0, 68.9, 54.5, 26.7, 26.2, 22.9, 10.5 ppm.

6-Azido-6-deoxy-3-*O*-isopentyl-1,2-*O*-isopropylidene- α -D-glucofuranose (6j): Reaction of compound **5j** (1.5 g, 3.3 mmol) with NaN₃ (1.67 g, 10.0 mmol) in DMF (15 mL) at 80 °C afforded compound **6j** as viscous liquid (0.98 g, 92 % yield); IR (KBr) cm⁻¹: 3472, 2958, 2872, 2104, 1711, 1633, 1466, 1081; ¹H NMR (300 MHz, CDCl₃): δ 5.90 (d, *J* = 3.6 Hz, 1H), 4.56 (d, *J* = 3.6 Hz, 1H), 4.10-3.98 (m, 3H), 3.72-3.43 (m, 4H), 2.62 (s, 1H), 1.72-1.61 (m, 2H), 1.49, 1.32 (each s, 6H), 0.92-0.90 (m, 7H); ¹³C NMR (75 MHz, CDCl₃): δ 111.8, 105.1, 82.9, 81.9, 79.8, 68.9, 68.8, 54.6, 38.4, 26.8, 26.2, 25.0, 22.5, 22.4 ppm.

4. General procedure for the synthesis of glycosyl epoxides (5k-m): A solution of orthogonally protected sugar **4k-m** having one free hydroxyl group (1.0 mmol) in anhydrous

DMF (15 mL) was cooled to 0 °C and sodium hydride (2.0 equiv.) was added portionwise. The reaction mixture was stirred at 0 °C under argon atmosphere for 20 minutes. Epichlorohydrin (1.2 mmol) was added at 0 °C and allowed to stir for 12 hour at room temperature. Upon completion of the reaction (monitor by TLC), excess of sodium hydride was quenched by adding water under inert atmosphere, the solvent was removed under reduced pressure, extracted with ethyl acetate. The organic layer was washed with brine solution, separated, dried over anhydrous Na₂SO₄, filtered, concentrated under vacuum, which on flash chromatography (ethyl acetate: hexane) afforded desired glycosyl epoxide **5k-m**.

1,2:5,6-Di-*O*-isopropylidene-3-*O*-(oxirane-2-ylmethoxy)- α -D-glucofuranose (5k**):** Reaction of compound 1,2:5,6-Di-*O*-isopropylidene- α -D-glucofuranose (1.0 g, 3.8 mmol) with epichlorohydrin (0.41 ml, 5.0 mmol) in presence of NaH (0.26, 11.4 mmol) in DMF (10 mL) 12 h afforded compound **5k** as colourless liquid (1.02 g, 85 % yield); ¹H NMR (300 MHz, CDCl₃): δ 5.86 (d, J = 4.5 Hz, 1H), 4.56 (dd, J = 3.6, 13.8 Hz, 1H), 4.34-4.27 (m, 1H), 4.11-3.84 (m, 5H), 3.66-3.44 (m, 1H), 3.13 (m, 1H), 2.82-2.77 (m, 1H), 2.64-2.61 (m, 1H), 1.49, 1.42, 1.34, 1.31 (each s, 12H); ¹³C NMR (75 MHz, CDCl₃): δ 111.7, 108.9, 105.1, 82.9, 82.5, 81.0, 72.2, 70.5, 67.3, 50.4, 44.0, 26.8, 26.7, 26.1, 25.2 ppm.

1,2:3,4-Di-*O*-isopropylidene-6-*O*-(oxirane-2-ylmethoxy)- α -D-galactopyranose (5l**):** Reaction of compound 1,2:3,4-Di-*O*-isopropylidene- α -D-galactopyranose (1.5 g, 5.7 mmol) with epichlorohydrin (0.60 ml, 7.4 mmol) in presence of NaH (0.39 g, 17.1 mmol) in DMF (10 ml) for 12 h afforded compound **5l** as colourless liquid (1.49 g, 82 % yield); ¹H NMR (300 MHz, CDCl₃): δ 5.53 (d, J = 4.5 Hz, 1H), 4.60 (d, J = 7.5 Hz, 1H), 4.32-4.26 (m, 2H), 4.24-3.98 (m, 1H), 3.97-3.60 (m, 3H), 3.53-3.40 (m, 1H), 3.17 (m, 1H), 2.78 (t, J = 4.2 Hz, 1H), 2.62 (d, J =

5.7 Hz, 1H), 1.58-1.33 (merged four s, 12H); ^{13}C NMR (75 MHz, CDCl_3): δ 109.2, 108.5, 96.4, 71.0, 70.6, 70.4, 50.4, 44.3, 26.0, 26.7, 24.8, 24.4 ppm.

5. General procedure for the synthesis of glycosyl azido alcohols from epoxides (6k-m): A solution of the compounds **5k-m** in EtOH/H₂O (1:1) treated with NaN₃ and NH₄Cl at 65 °C for 8 h. Upon completion of the reaction, the solvent was removed under reduced pressure, extracted with ethyl acetate and water. The organic layer was dried over anhydrous Na₂SO₄, filtered, concentrated under vacuum, followed by flash chromatography (ethyl acetate: hexane) afforded the desired glycosyl azido alcohol **6k-m** in good yields.

3-*O*-(3-Azido-2-hydroxypropoxy)-1,2:5,6-di-*O*-isopropylidene- α -D-glucofuranose (6k):

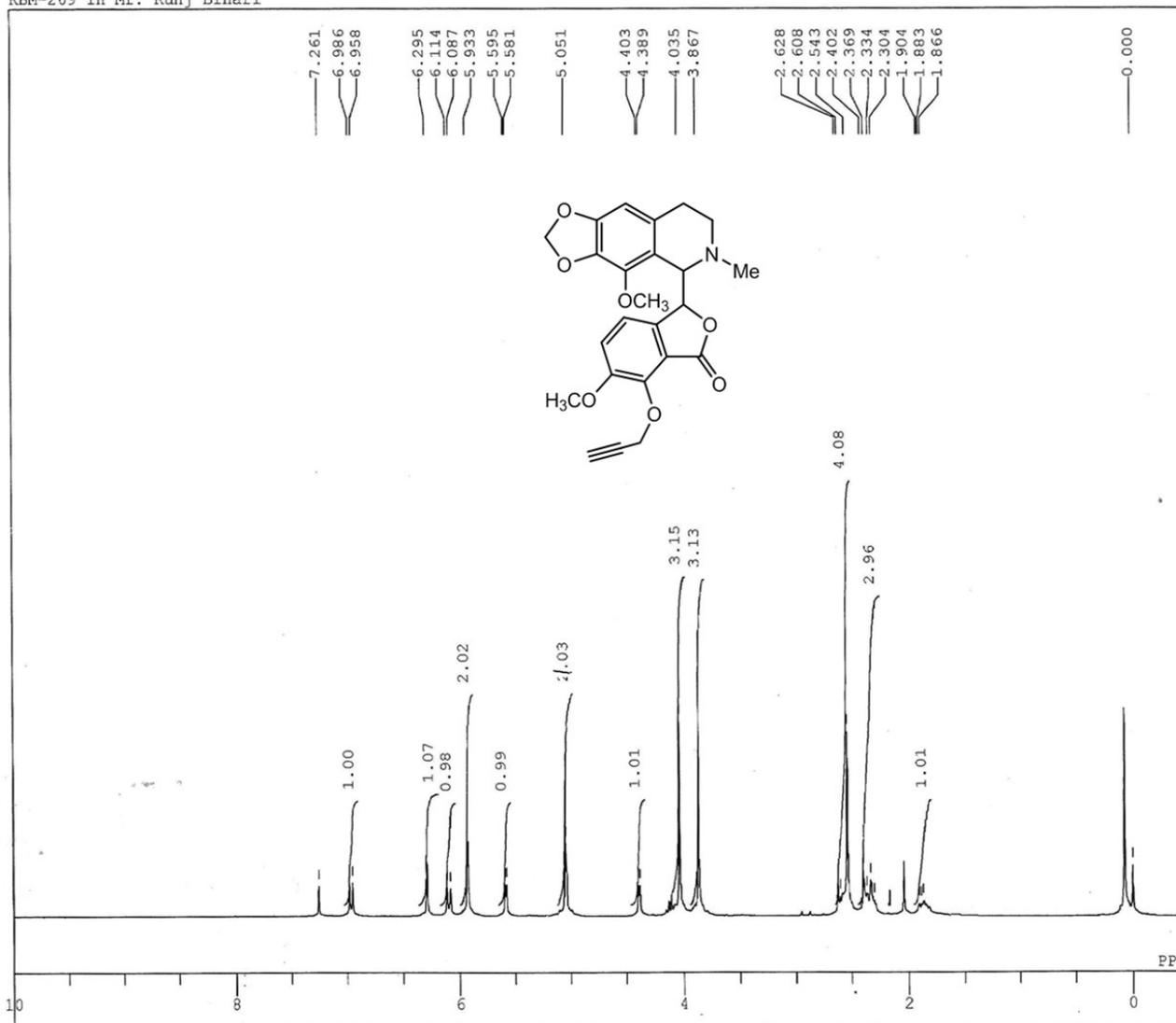
Reaction of compound **5k** (1.0 g, 3.1 mmol) with NaN₃ (0.31 g, 4.7 mmol) in presence of NH₄Cl (0.16 g, 3.1 mmol) in EtOH/H₂O (1:1, 10 mL) at 65 °C for 12 h afforded the title compound **6k** as viscous liquid (0.96 g, 86 % yield); IR (KBr) cm⁻¹: 3456, 2988, 2936, 2103, 1667, 1456, 1075; ^1H NMR (300 MHz, CDCl_3): δ 5.90 (d, J = 3.3 Hz, 1H), 4.56 (s, 1H), 4.32-4.29 (m, 1H), 4.15-3.75 (m, 6H), 3.59-3.32 (m, 3H), 1.49, 1.44, 1.37, 1.32 (each s, 12H); ^{13}C NMR (75 MHz, CDCl_3): δ 111.9, 109.4, 105.4, 84.3, 82.8, 82.3, 81.1, 72.7, 71.3, 70.1, 68.7, 67.7, 53.0, 52.6, 26.7, 26.0, 25.0 ppm.

6-*O*-(3-Azido-2-hydroxypropoxy)-1,2:3,4-di-*O*-isopropylidene- α -D-galactopyranose (6l):

Reaction of compound **5l** (1.5 g, 4.7 mmol) with NaN₃ (0.46 g, 7.1 mmol) in presence of NH₄Cl (0.25 g, 4.7 mmol) in EtOH/H₂O (1:1, 10 mL) at 65 °C for 12 h afforded compound **6l** as viscous liquid (1.41 g, 84 % yield); IR (KBr) cm⁻¹: 3434, 2989, 2935, 2103, 1643, 1455, 1383, 1257, 1070; ^1H NMR (300 MHz, CDCl_3): δ 5.53 (d, J = 3.9 Hz, 1H), 4.62 (d, J = 8.1 Hz, 1H), 4.33-4.22 (m, 2H), 3.98-3.88 (m, 2H), 3.70-3.48 (m, 4H), 3.35-3.34 (m, 2H), 3.07 (s, 1H), 1.54, 1.45,

1.38, 1.33 (each s, 12H); ^{13}C NMR (75 MHz, CDCl_3): δ 109.4, 108.7, 96.2, 72.9, 72.6, 71.1, 70.6, 70.5, 70.4, 69.9, 69.3, 66.5, 53.2, 25.9, 24.8, 24.4 ppm.

C:\Kunj B.Mishra\KBM-269_1H.als
 KBM-269 1H Mr. Kunj Bihari



JEOL AL300 FTNMR
 CHEMISTRY DEPARTMENT
 Banaras Hindu University
 VARANASI-221005

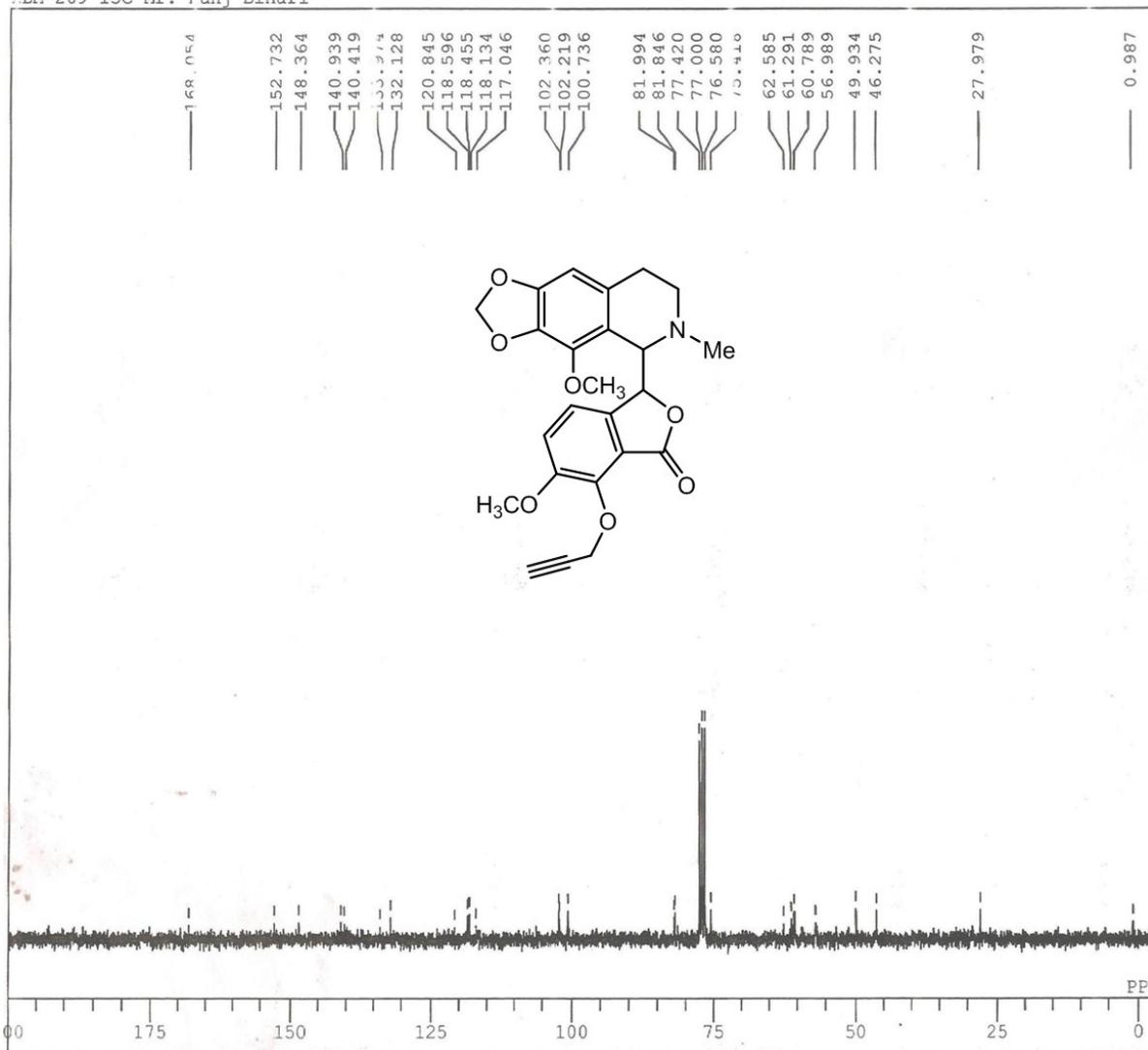
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 Shishir Singh

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 RGAIN 17

Figure 1: ¹H

NMR of compound 3

C:\Kunj B.Mishra\KBM-269_13C.als
 KBM-269 13C Mr. Kunj Bihari



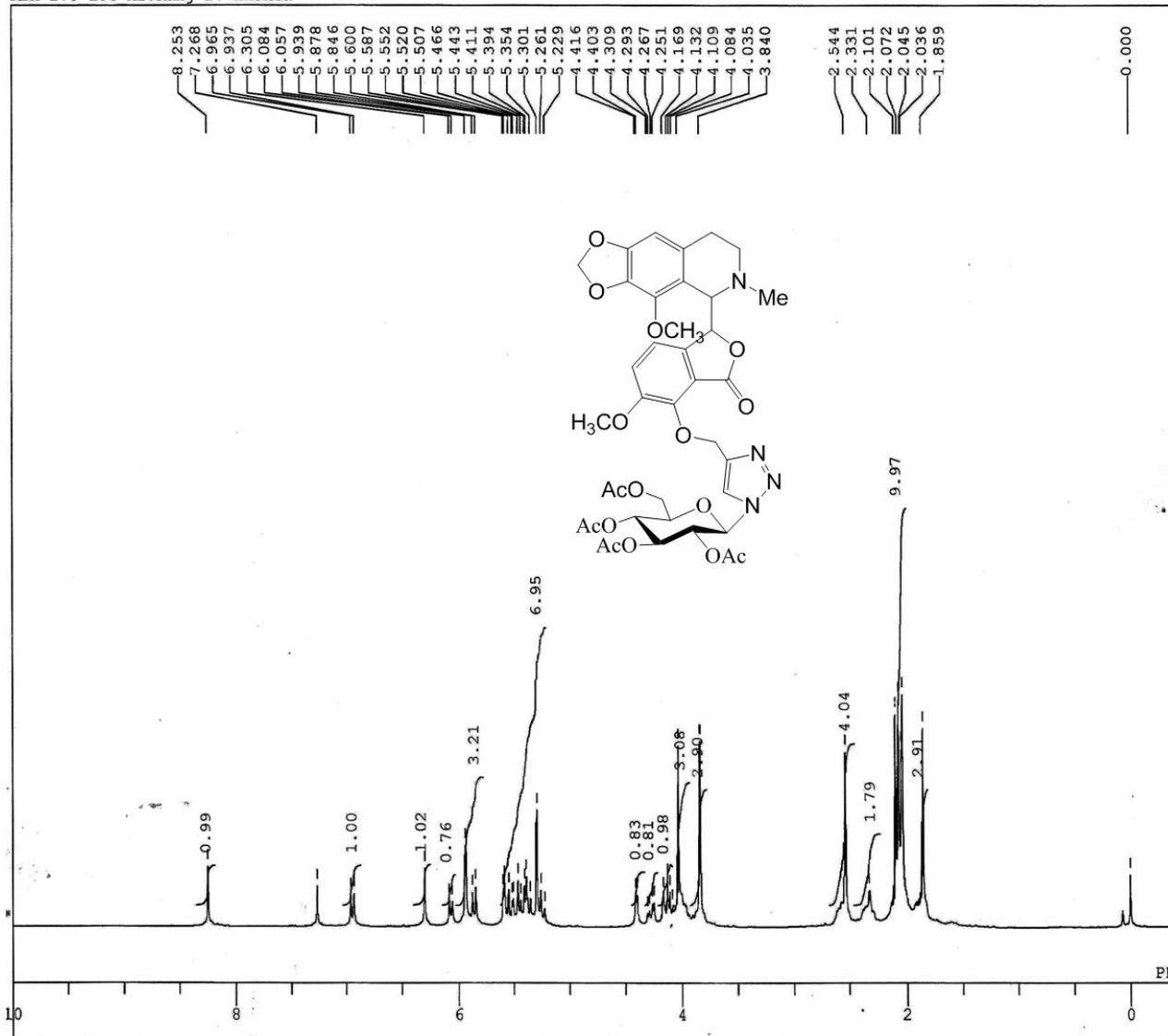
JEOL AL300 FTNMR
 CHEMISTRY DEPARTMENT
 Banaras Hindu University,
 VARANASI-221005

Operator : Nagendra Kumar
 Shishir Singh

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 PW1 5.9 us
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 EXREF 77.00 ppm
 BF 1.20 Hz
 RGAIN 24

Figure 2: ^{13}C NMR of compound 3

C:\Kunj B.Mishra\KBM-270_1H.als
 KBM-270 13C Mr.Kunj B. Mishra



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 Banaras Hindu University
 VARANASI-221005

Operator : Nagendra Kuma
 Shishir Singh

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 BF 1.20 Hz
 RGAIN 17

Figure 3: ¹H NMR of compound 8a

C:\Kunj B.Mishra\KBM-270 13C.als
 KBM-270 13C Mr.Kunj B. Mishra

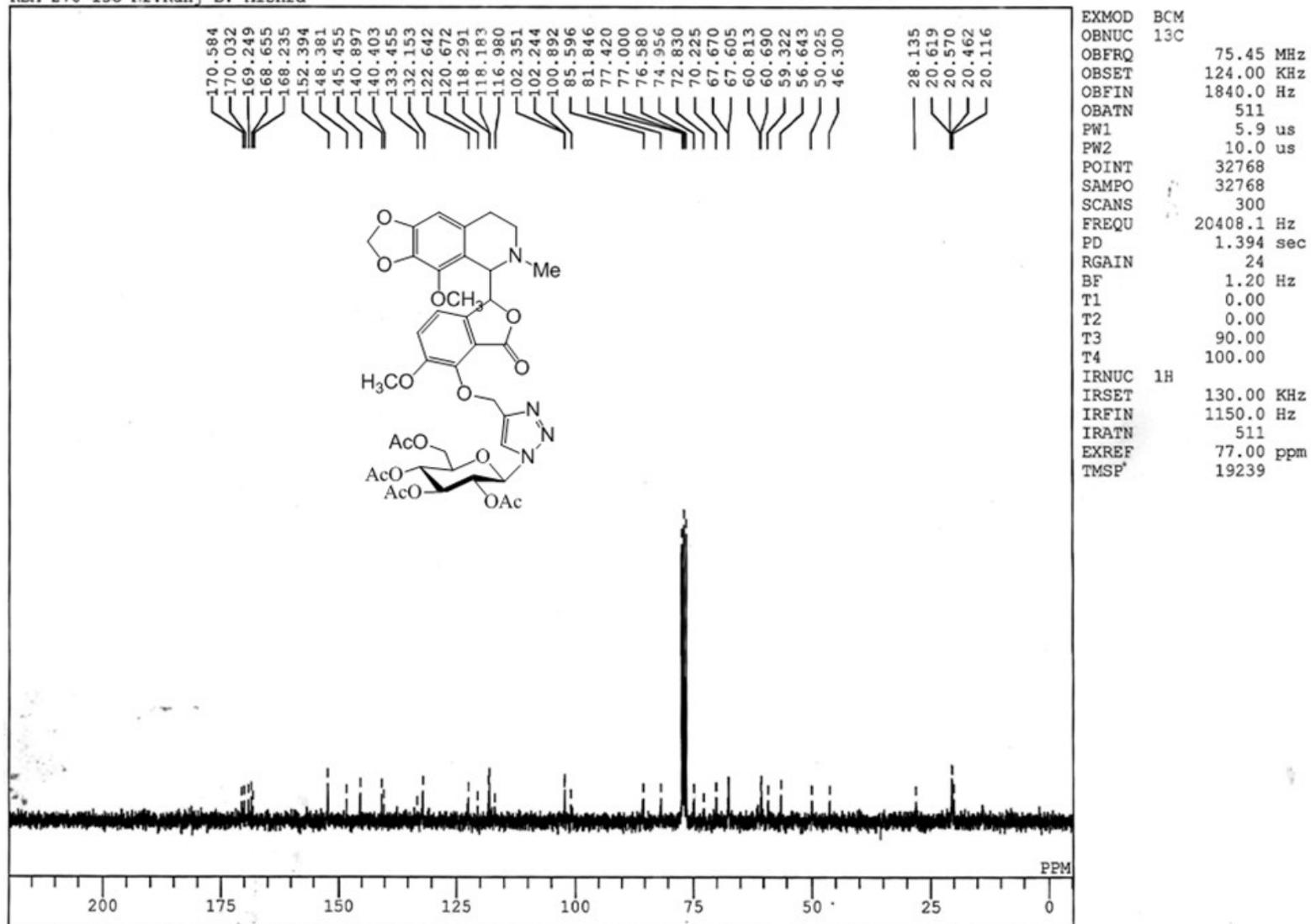


Figure 4: ¹³C NMR of compound 8a

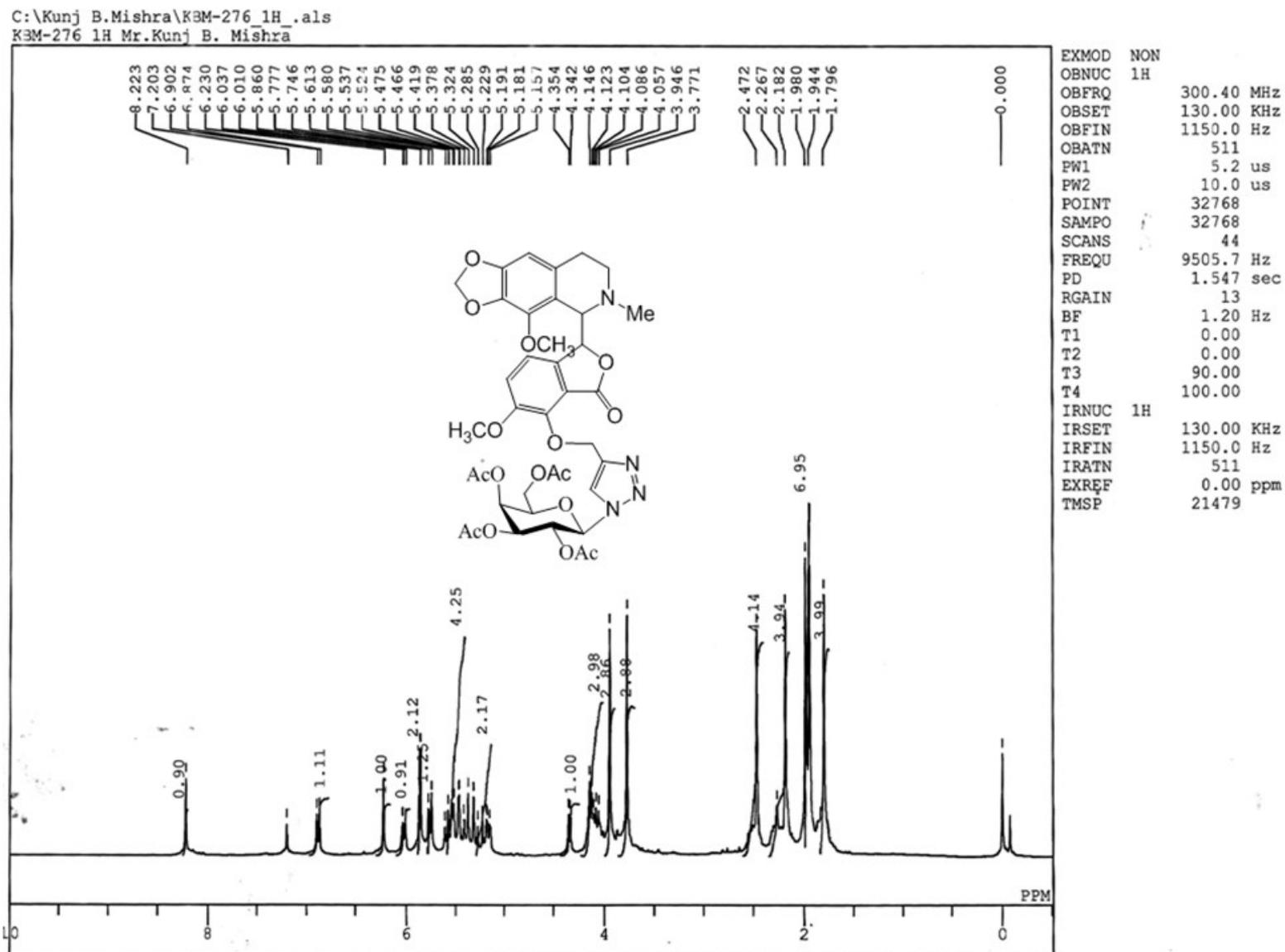


Figure 5: ¹H NMR of compound 8b

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 KBM-276 13C Mr.Kunj B. Mishra

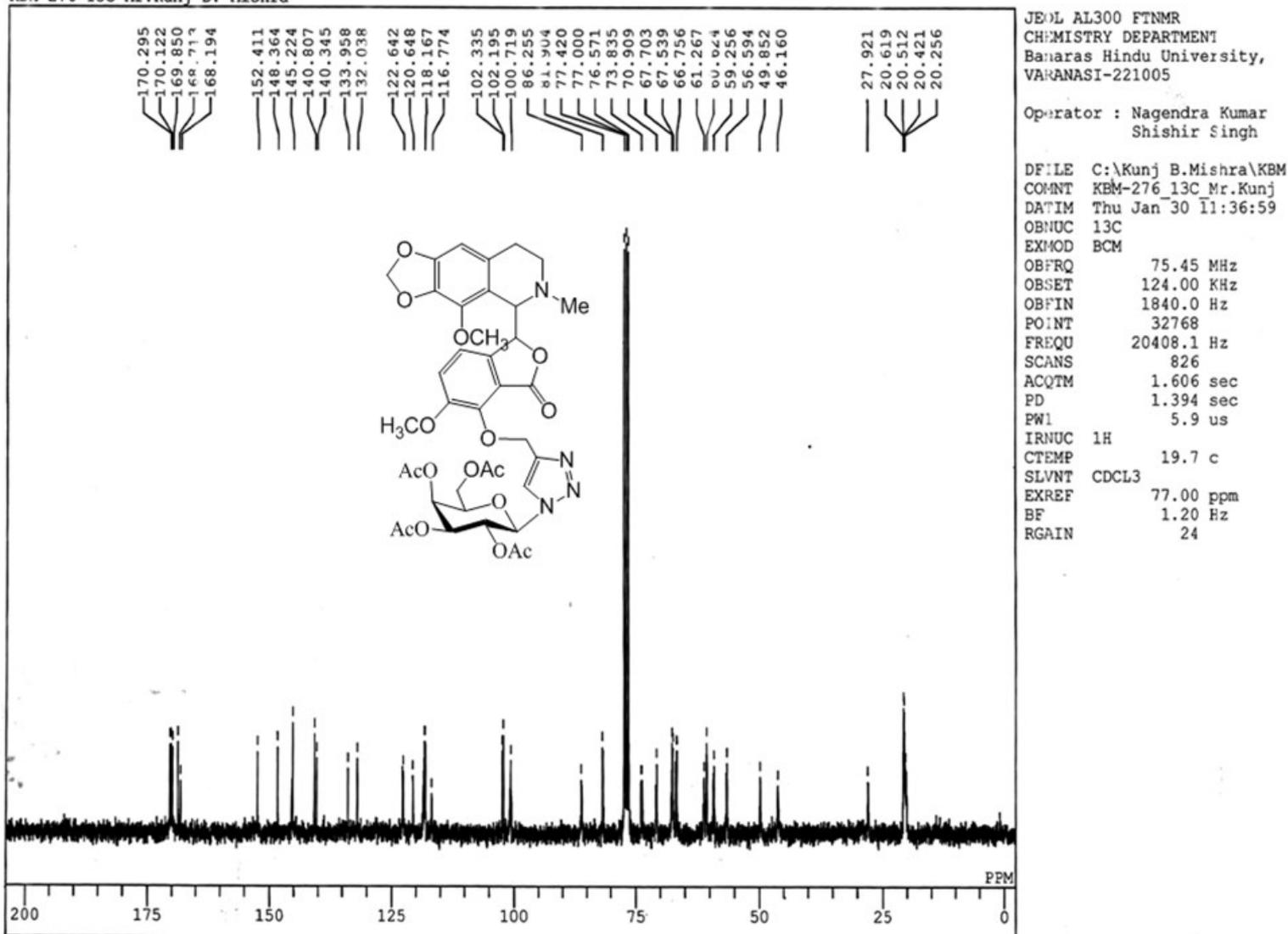


Figure 6: ¹³C NMR of compound 8b

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 KBM-285 1H Mr.KBM

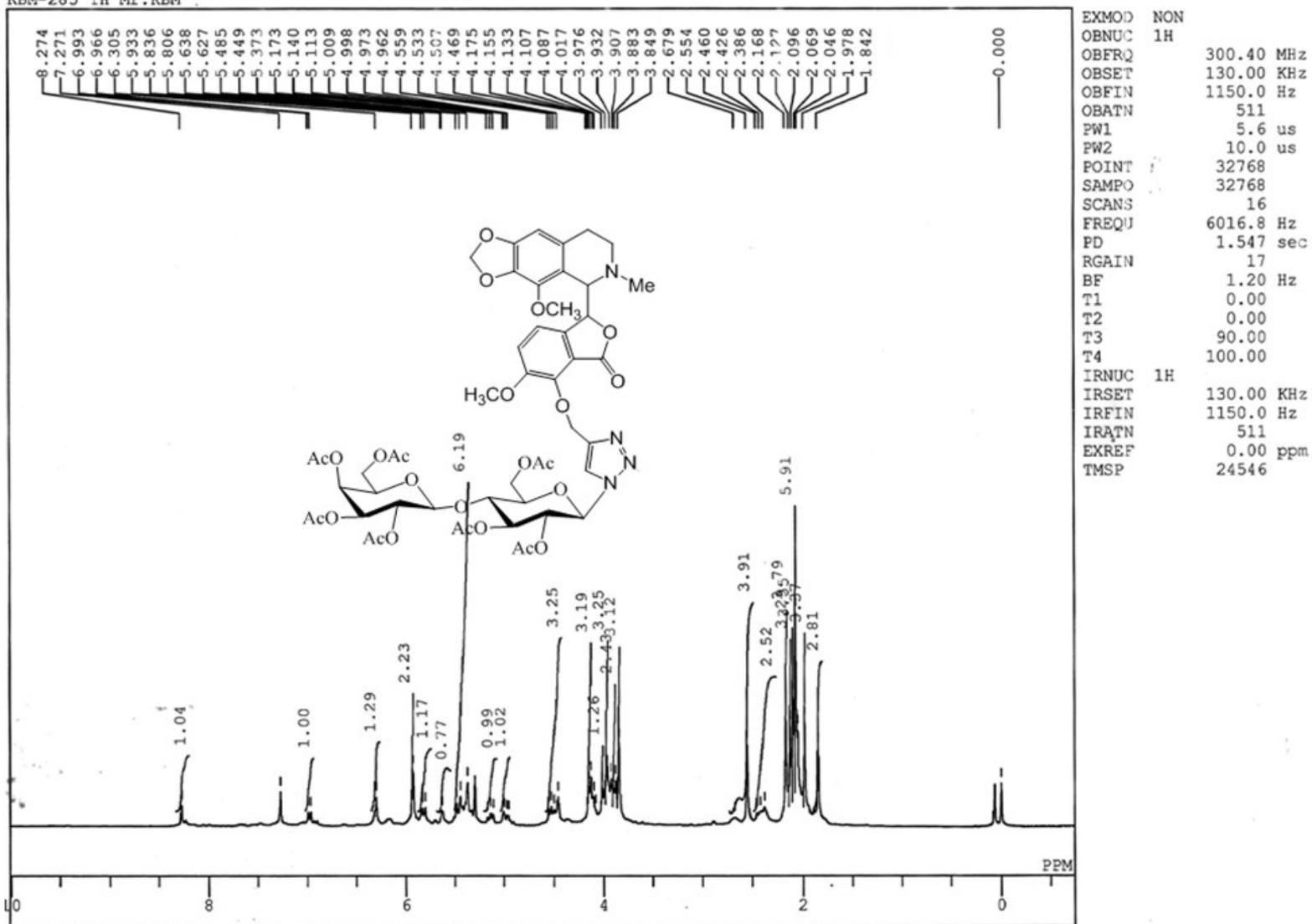


Figure 7: ¹H NMR of compound 8c

C:\Kunj B.Mishra\KBM-285CREV_13C.als
 KBM-285CREV 13C Mr. Kunj

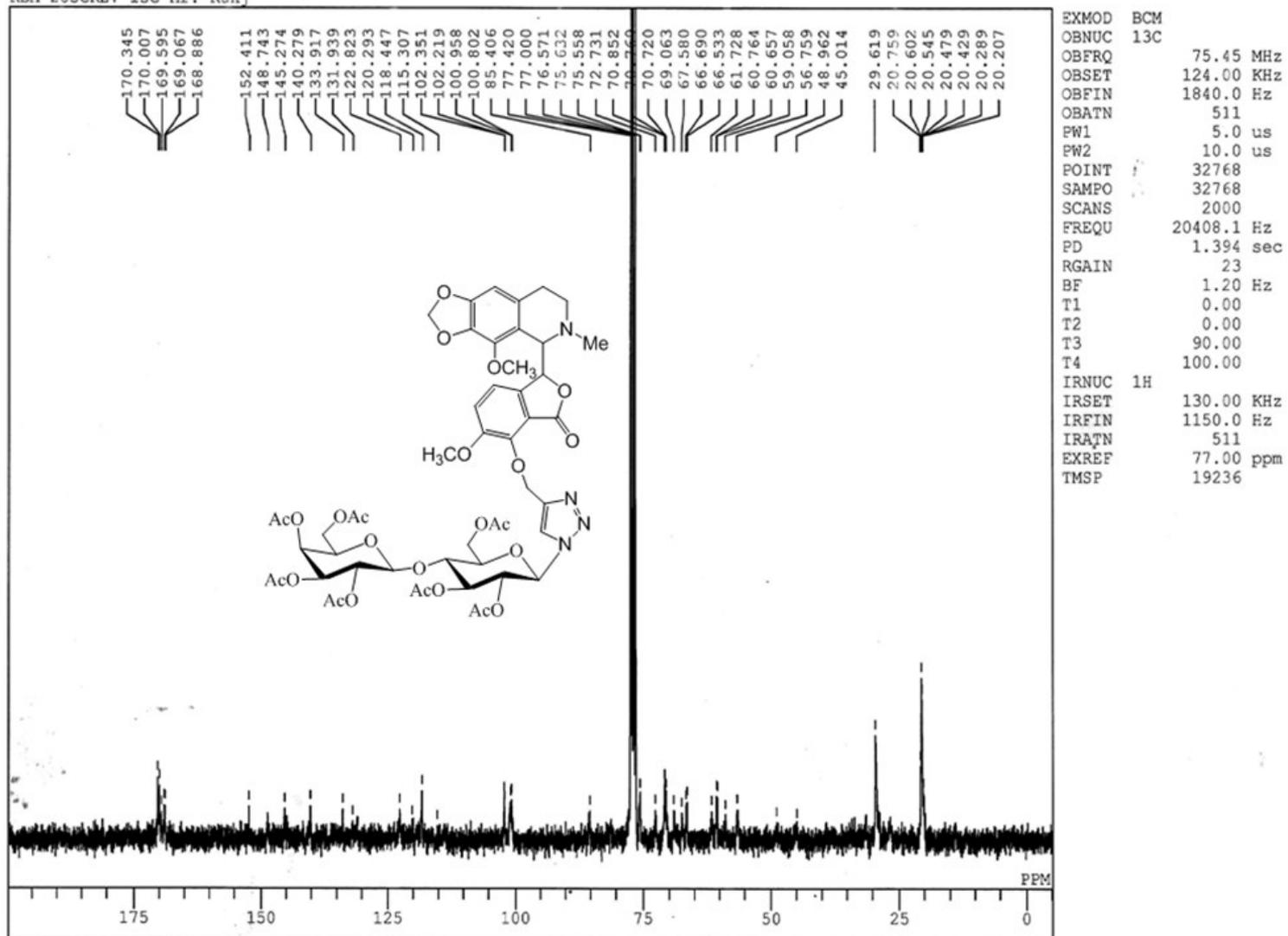
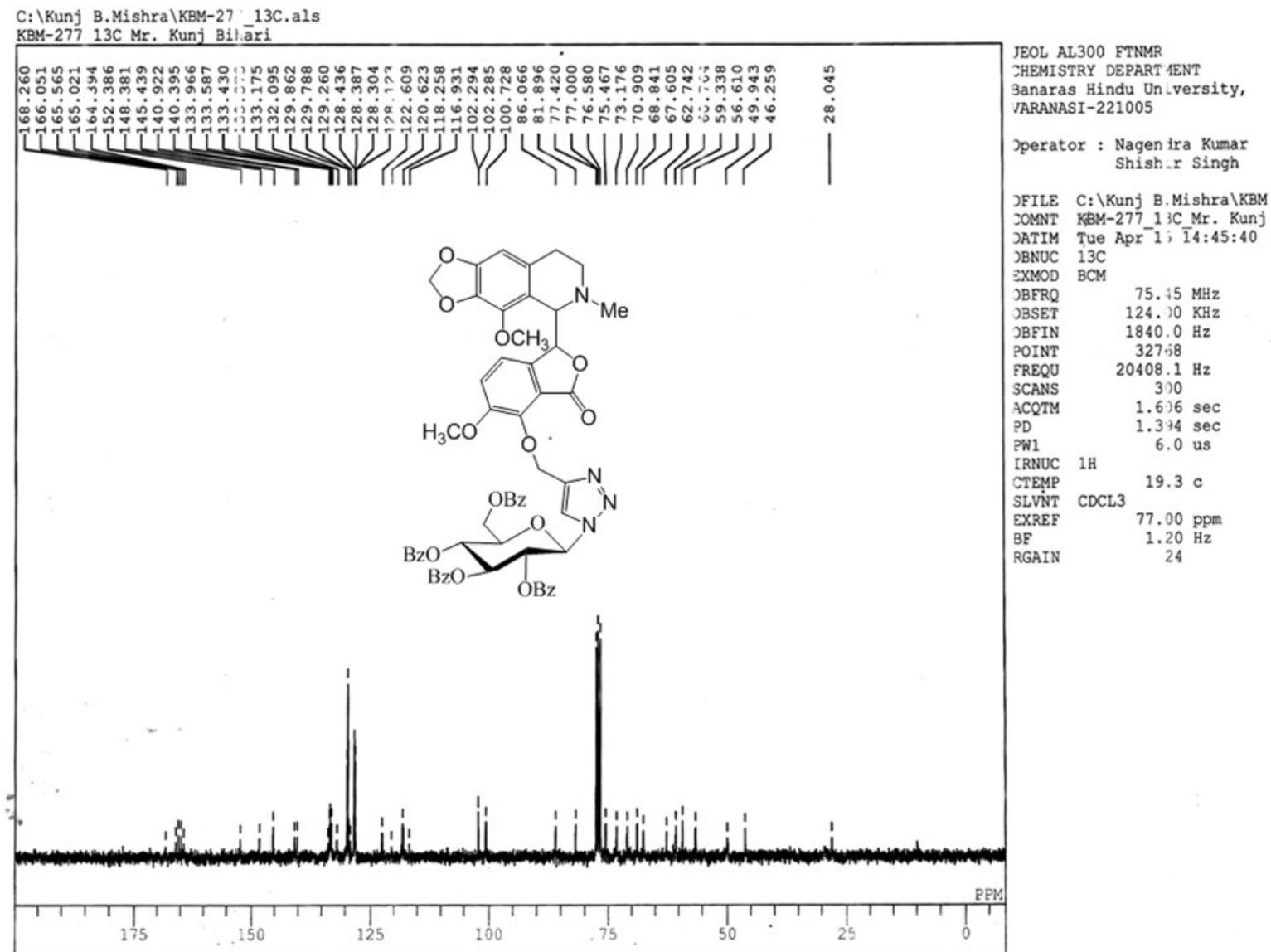


Figure 8: ¹³C NMR of compound 8c



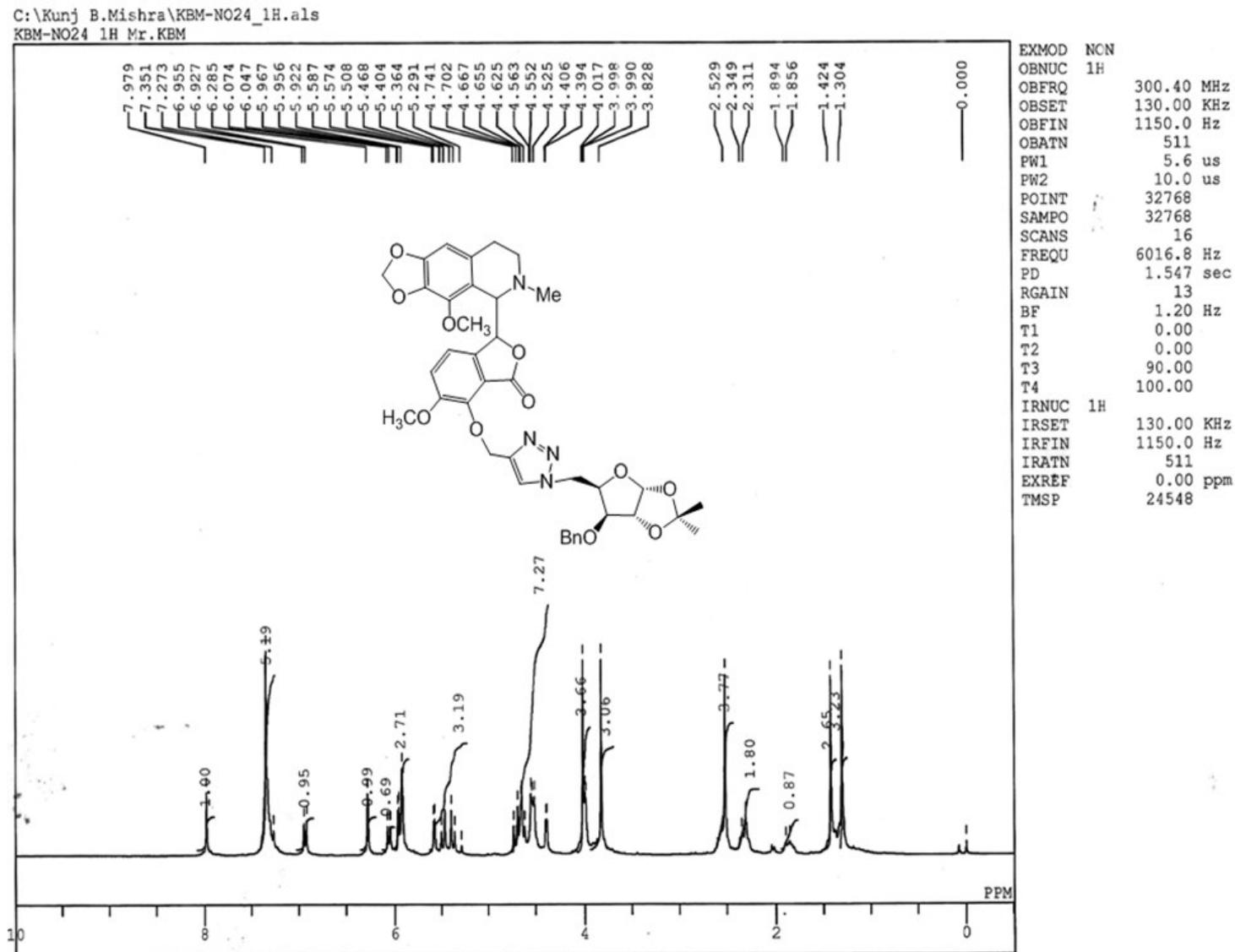
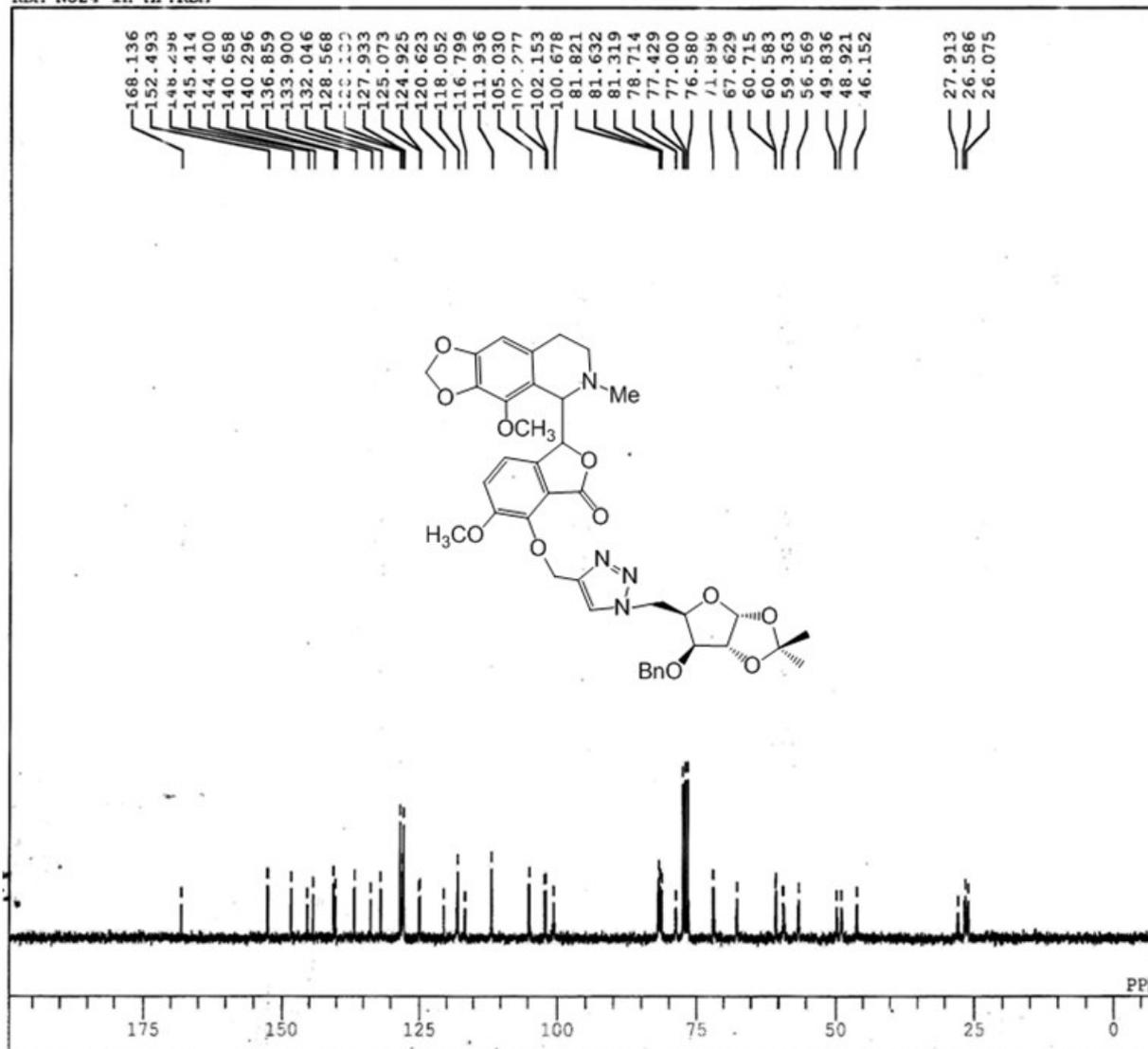


Figure 11: ¹H NMR of compound 8c

C:\Kunj B.Mishra\KBM-NO24_13C.als
KBM-NO24 1H Mr.KBM



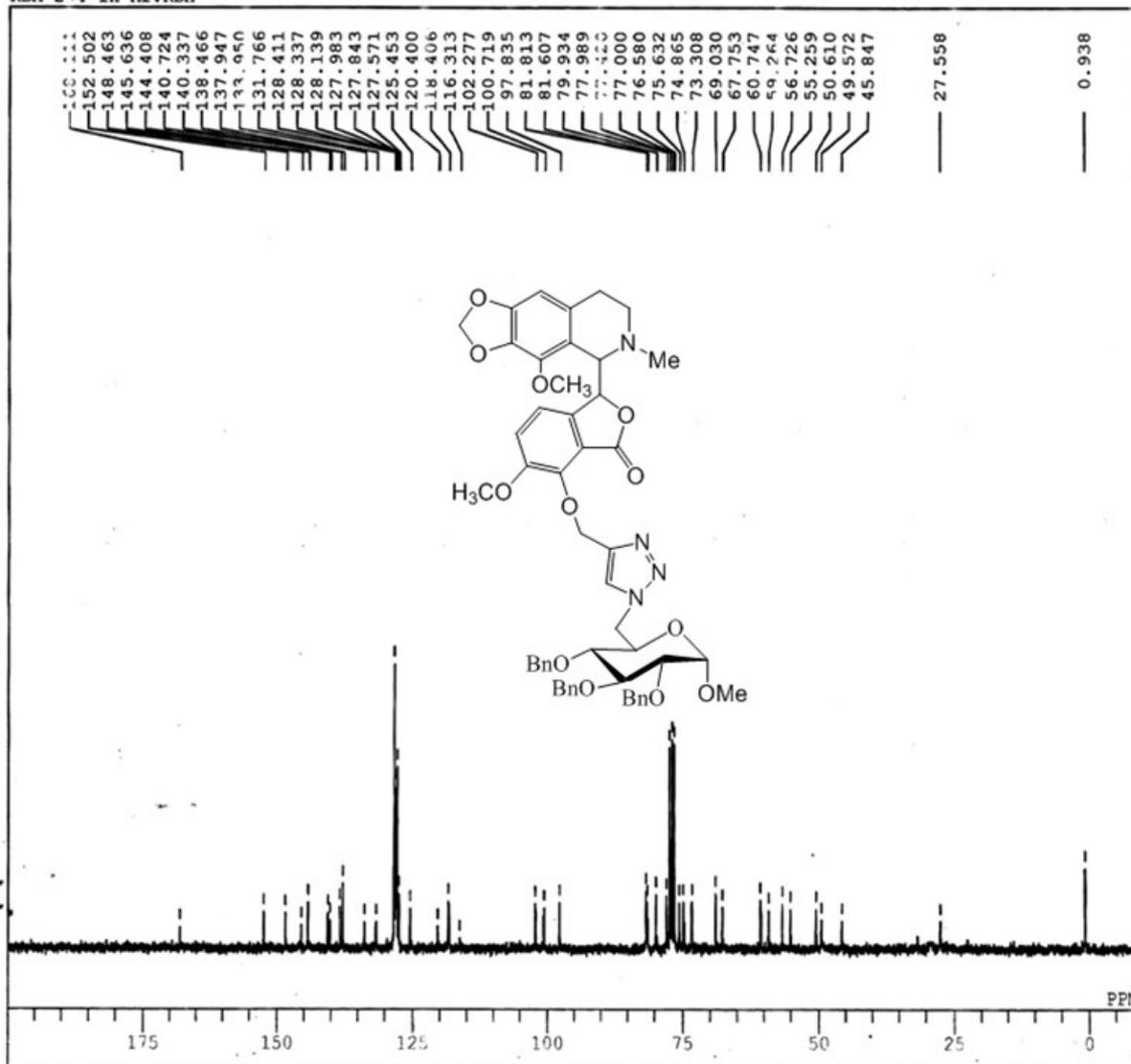
JEOL AI 300 FTNMR
CHEMISTRY DEPARTMENT
Banaras Hindu University,
VARANASI-221005

Operator : Nagendra Kumar
Shishir Singh

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CTEMP 19.3 c
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EXREF 77.00 ppm
BF 1.20 Hz
RGAIN 24

Figure 12: ¹³C NMR of compound 8e

C:\Kunj B.Mishra\KBM-284_13C.als
KBM-284 1H Mr.KBM



JEOL AL300 FTNMR
CHEMISTRY DEPARTMENT
Banaras Hindu University,
VARANASI-22100

Operator : Nagendra Kumar
Shi:hir Singh

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BF 1.20 Hz
RGAIN 24

Figure 14: ¹³C NMR of compound 8f

C:\Kunj B.Mishra\KBM-275_1H.als
 KBM-275 1H Mr. Kunj Bihari Mishra

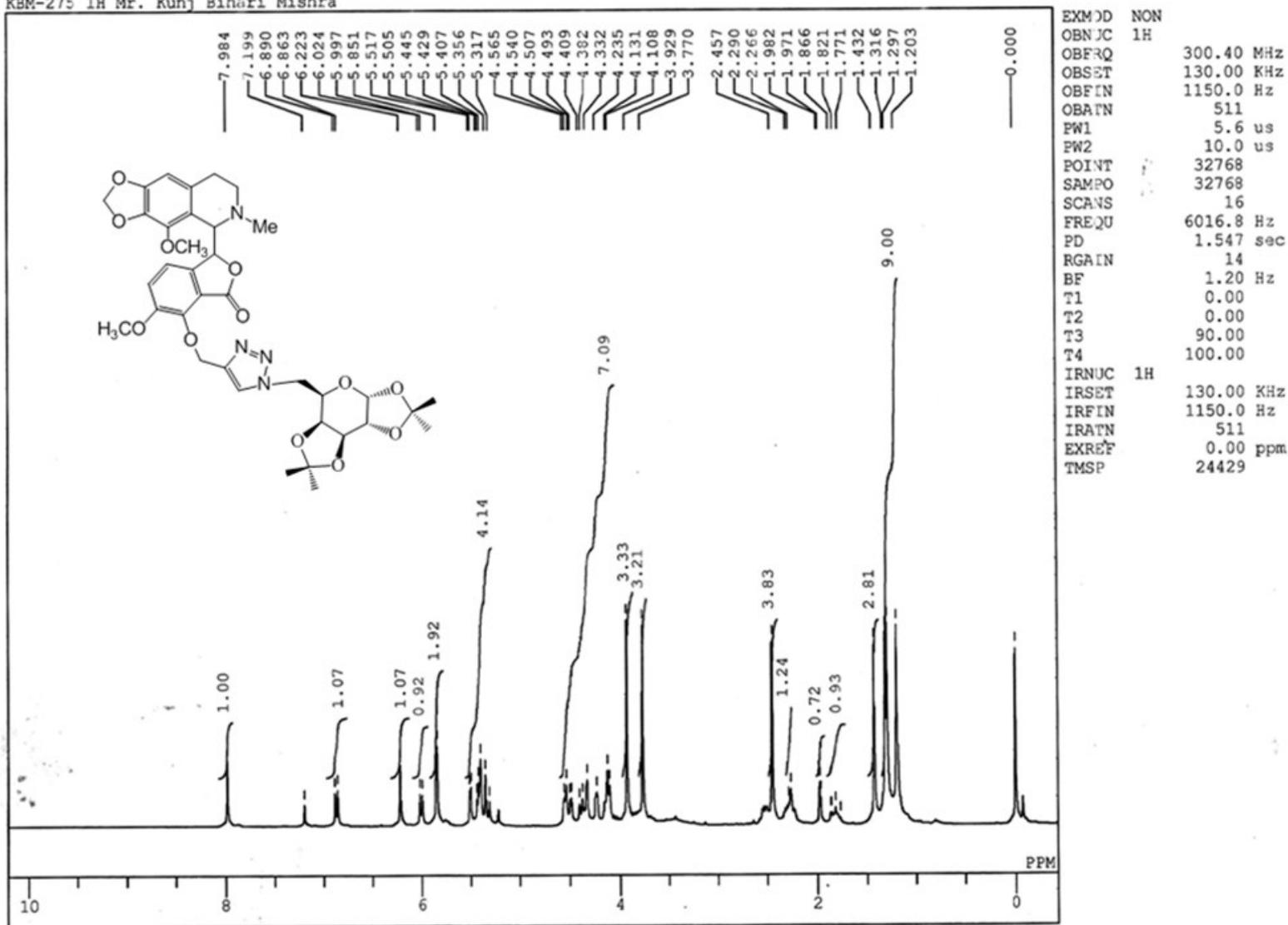


Figure 15: ¹H NMR of compound 8g

C:\Kunj B.Mishra\KBM-275_13C.als
 KBM-275 1H Mr. Kunj Bihari M.shra

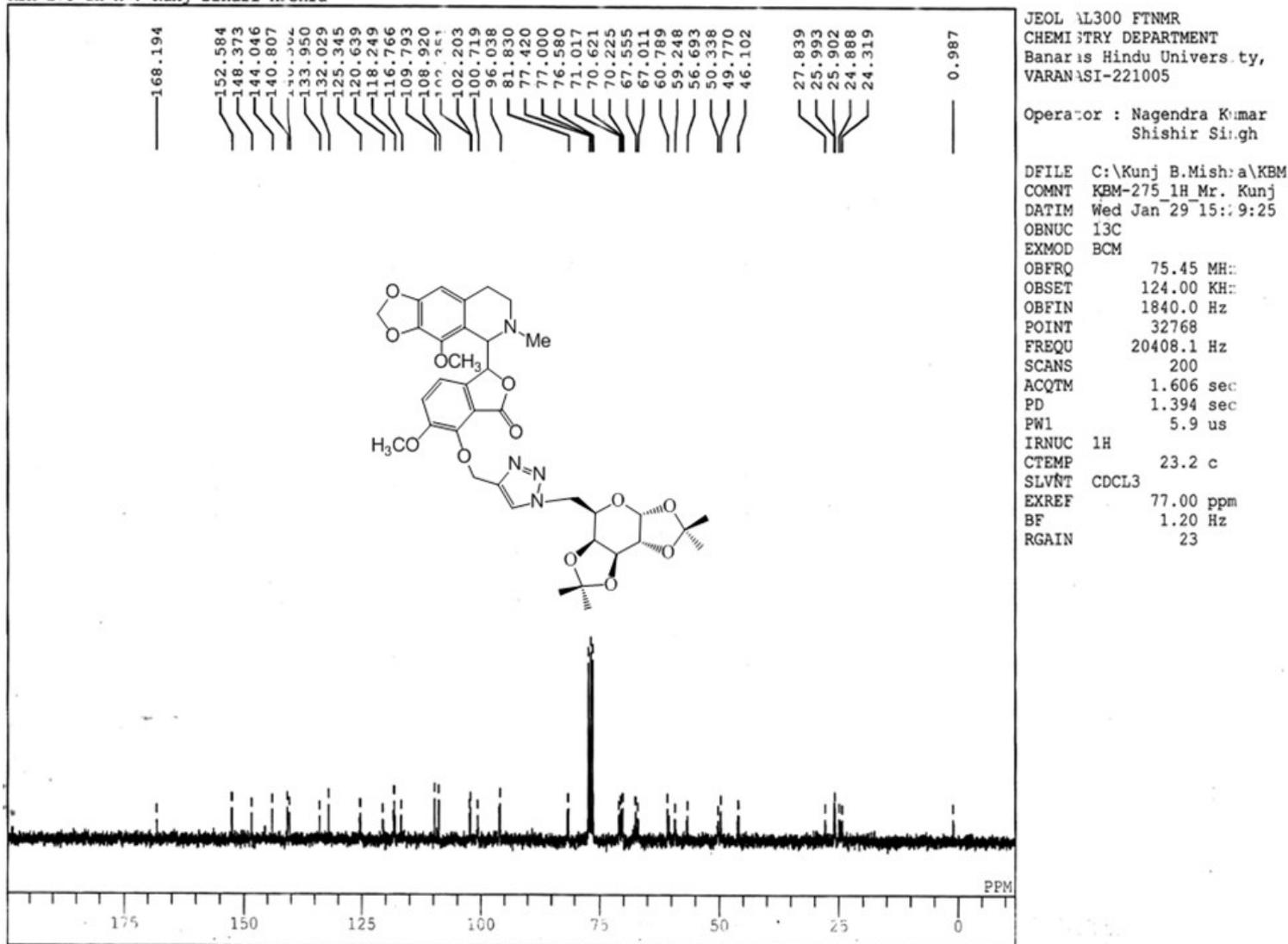


Figure 16: ¹³CNMR of compound 8g

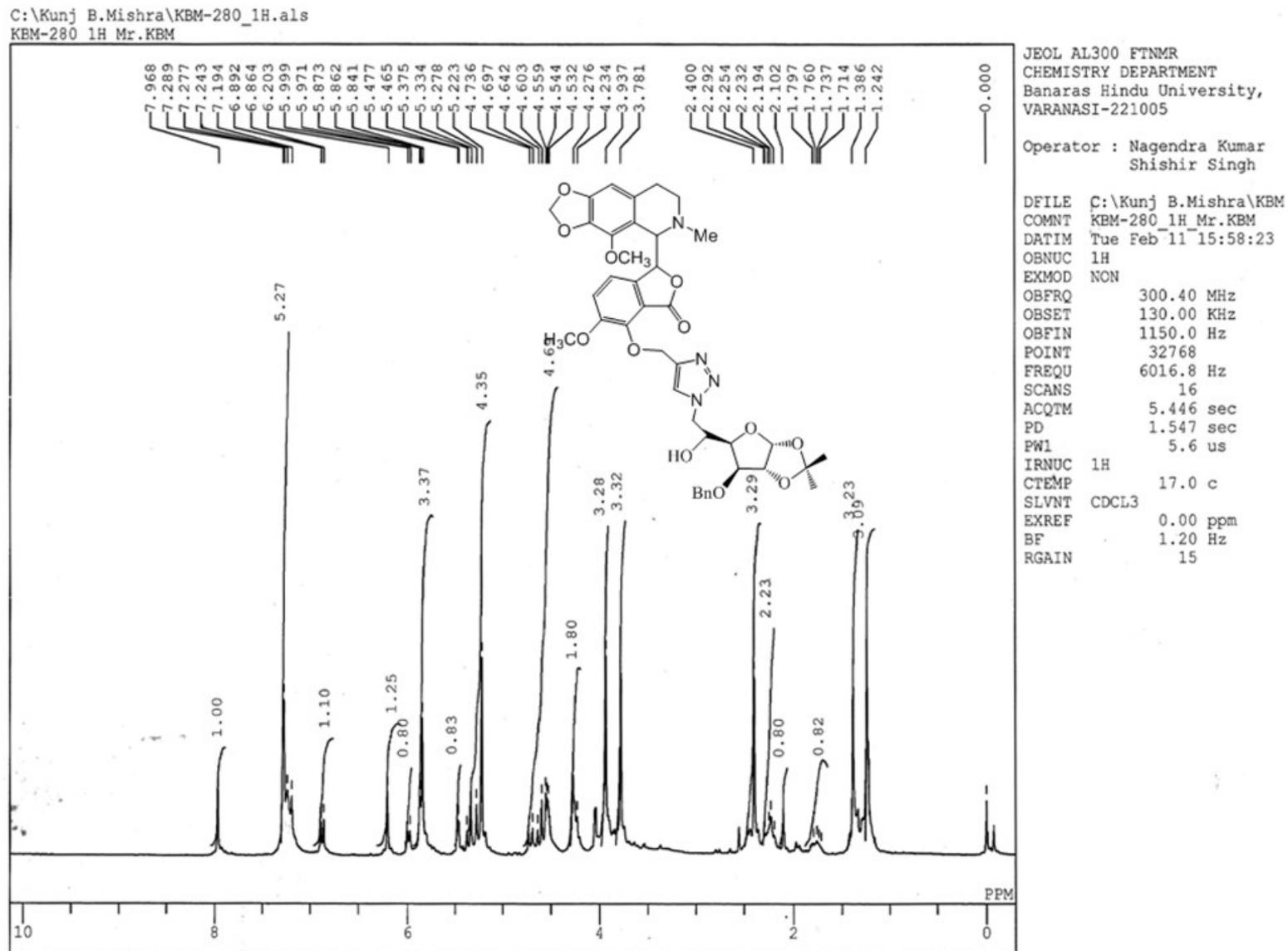


Figure 17: ¹H NMR of compound 8h

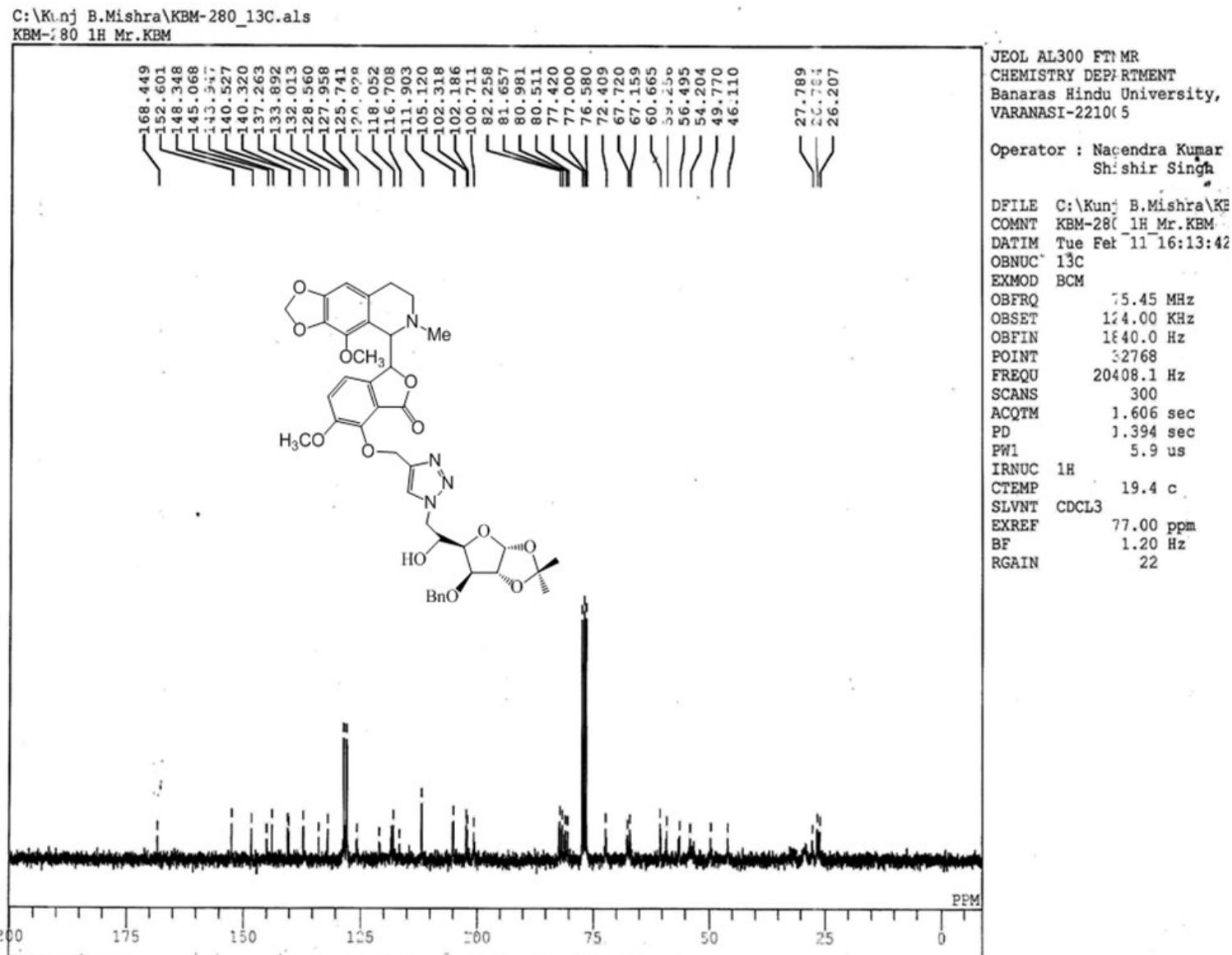


Figure 18: ¹³C NMR of compound 8h

C:\Kunj B.Mishra\KBM-333P2_1H.als
 KBM-333P2 1H Mr. Kunj Bihari

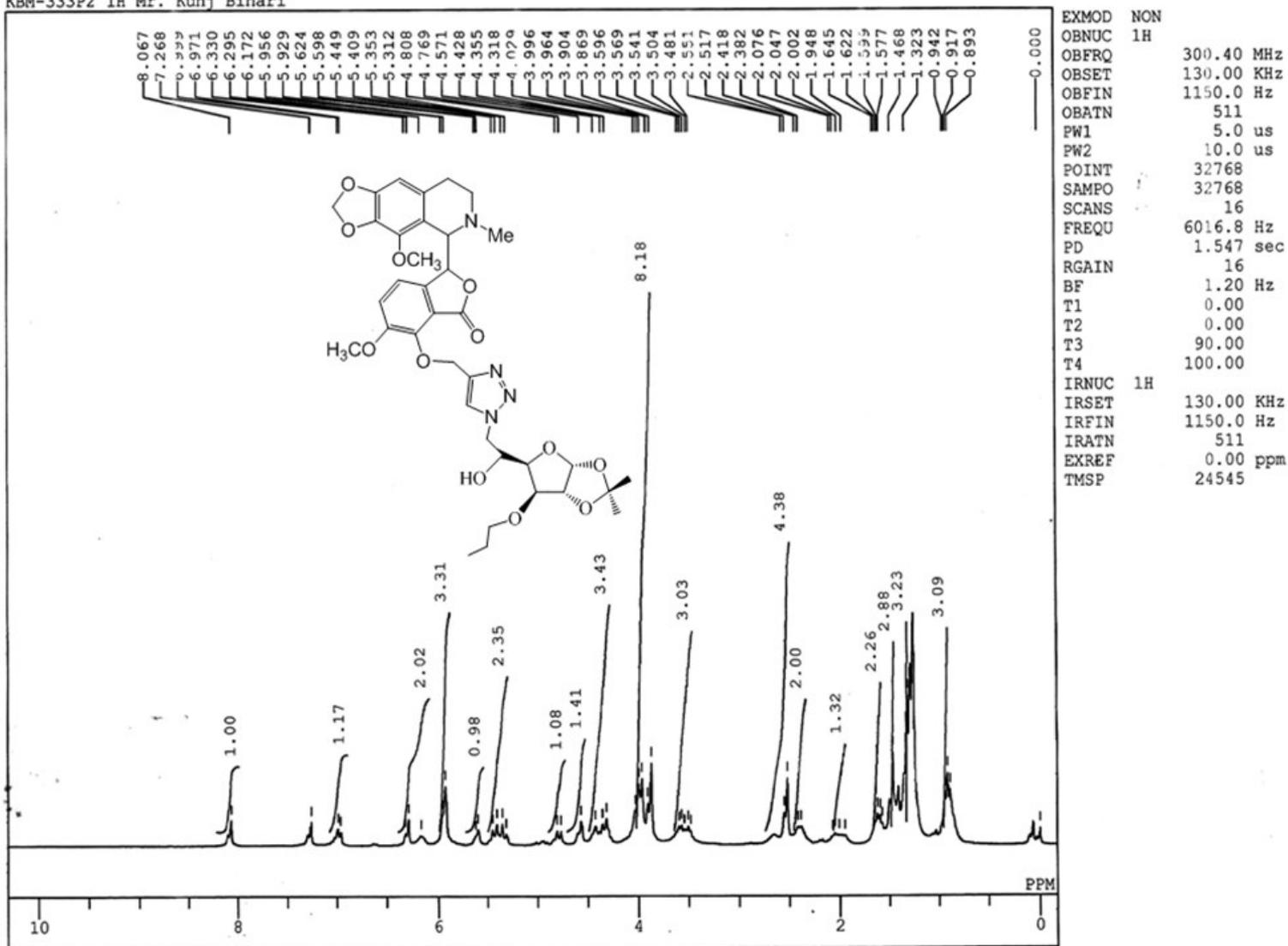
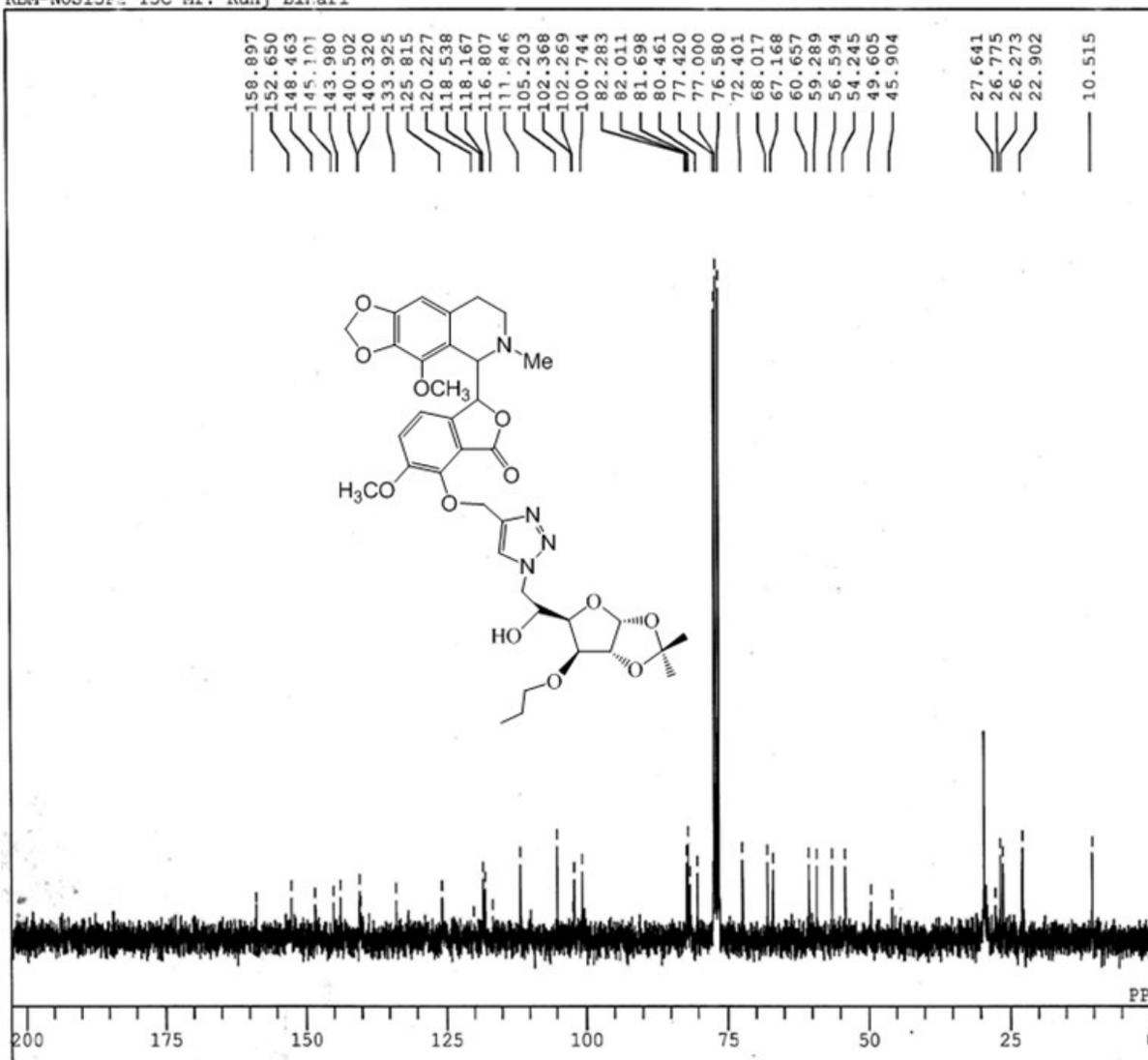


Figure 19: ¹H NMR of compound 8i

C:\Kunj B. Mishra\KBM-NOS13P2_13C.als
 KBM-NOS13P2_13C Mr. Kunj Bihari



JEOL AL300 FTNMR
 CHEMISTRY DEPARTMENT
 Banaras Hindu University,
 VARANASI-221005

Operator : Nagendra Kumar
 Shishir Singh

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 PW1 5.0 us
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 RGAIN 24

Figure 20: ¹³C NMR of compound 8i

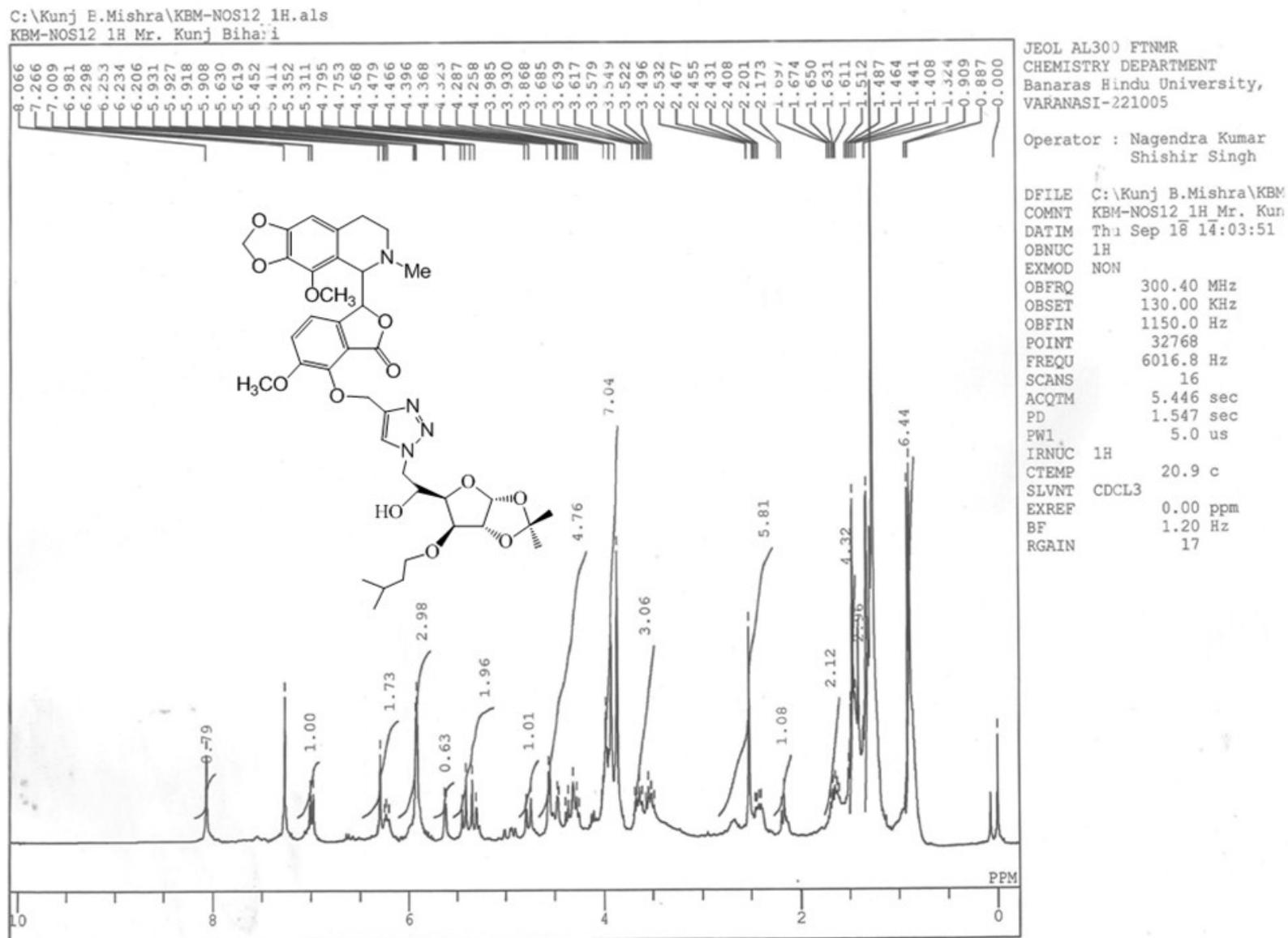


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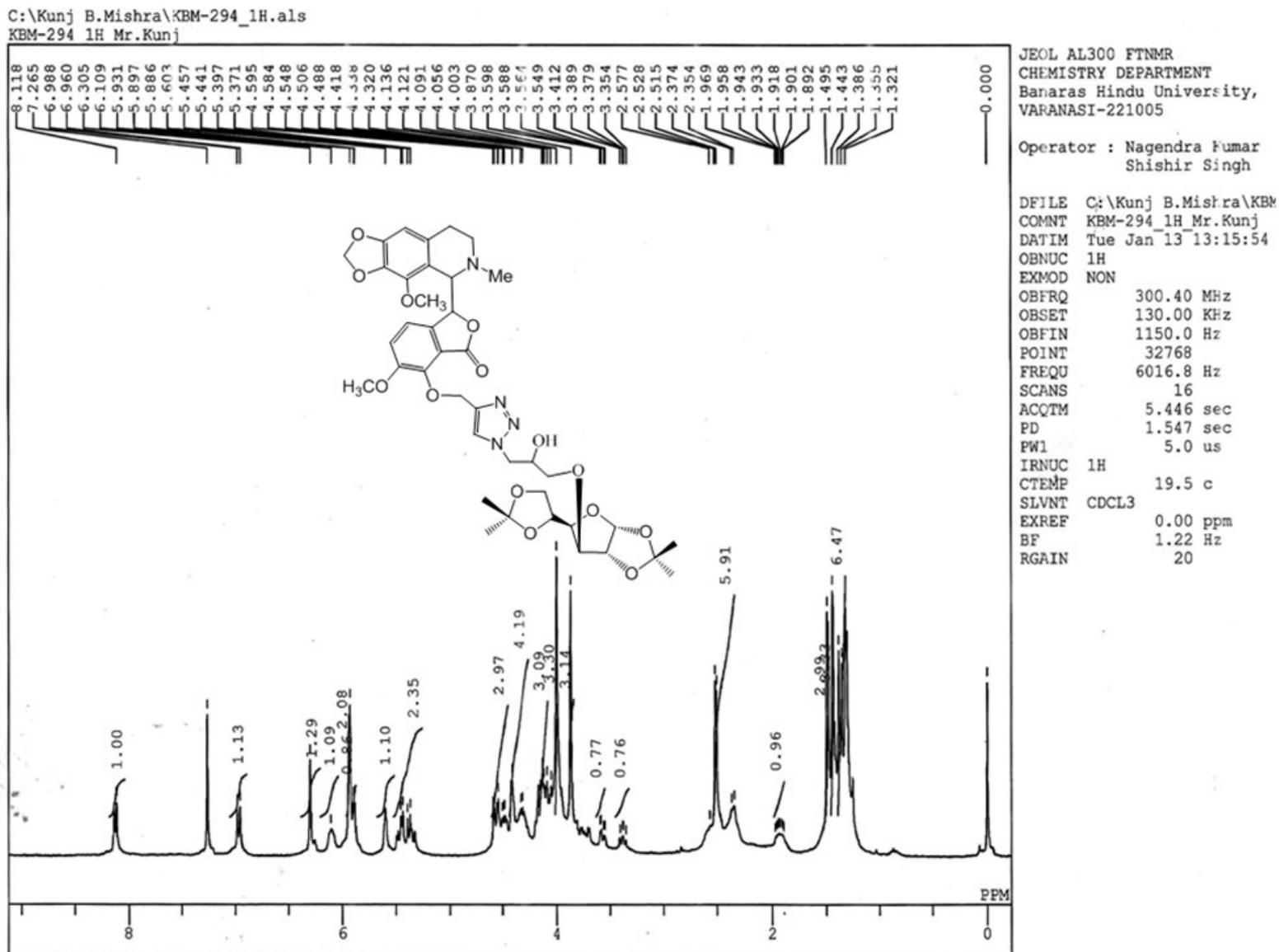


Figure 23: ¹H NMR of compound 8k

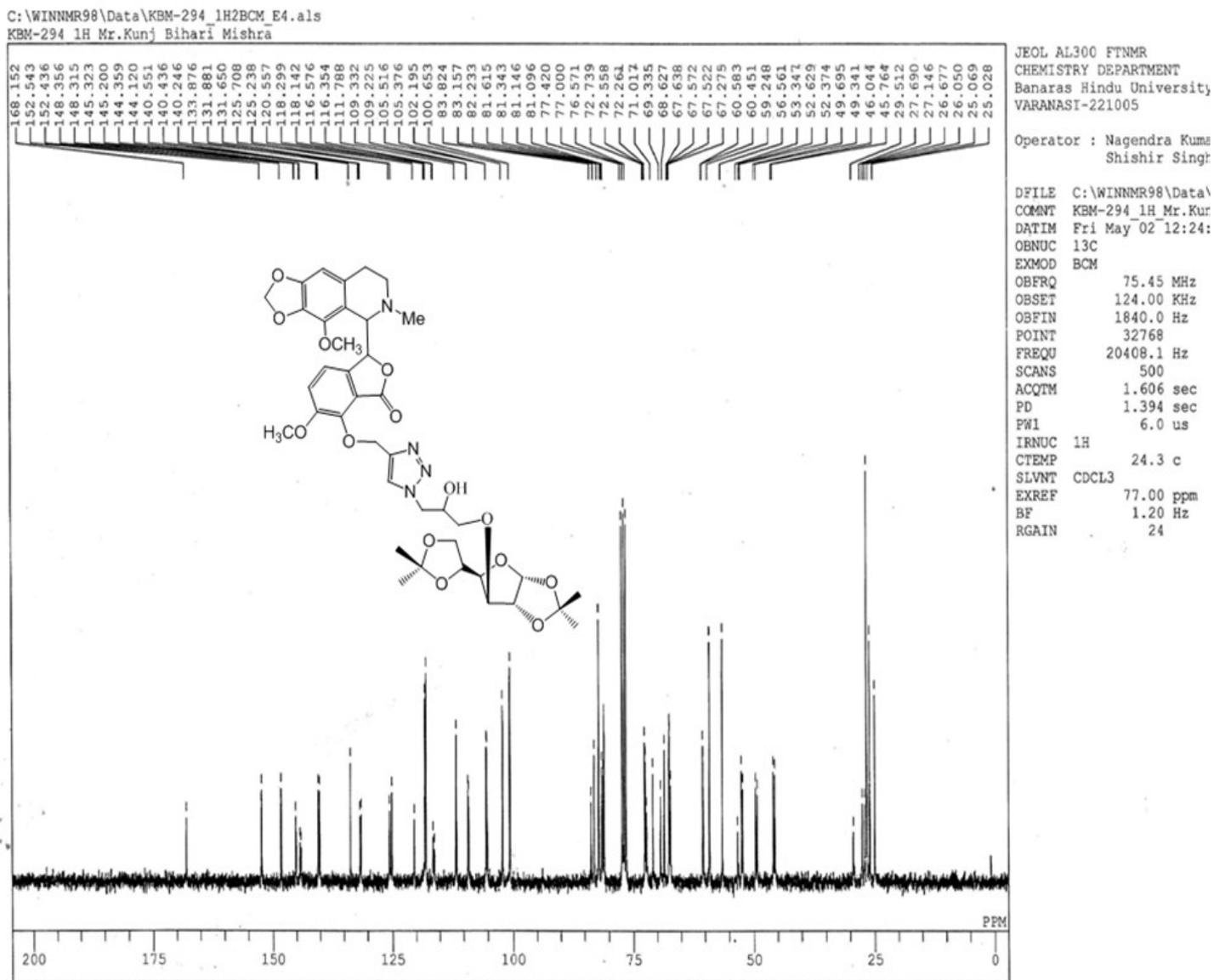
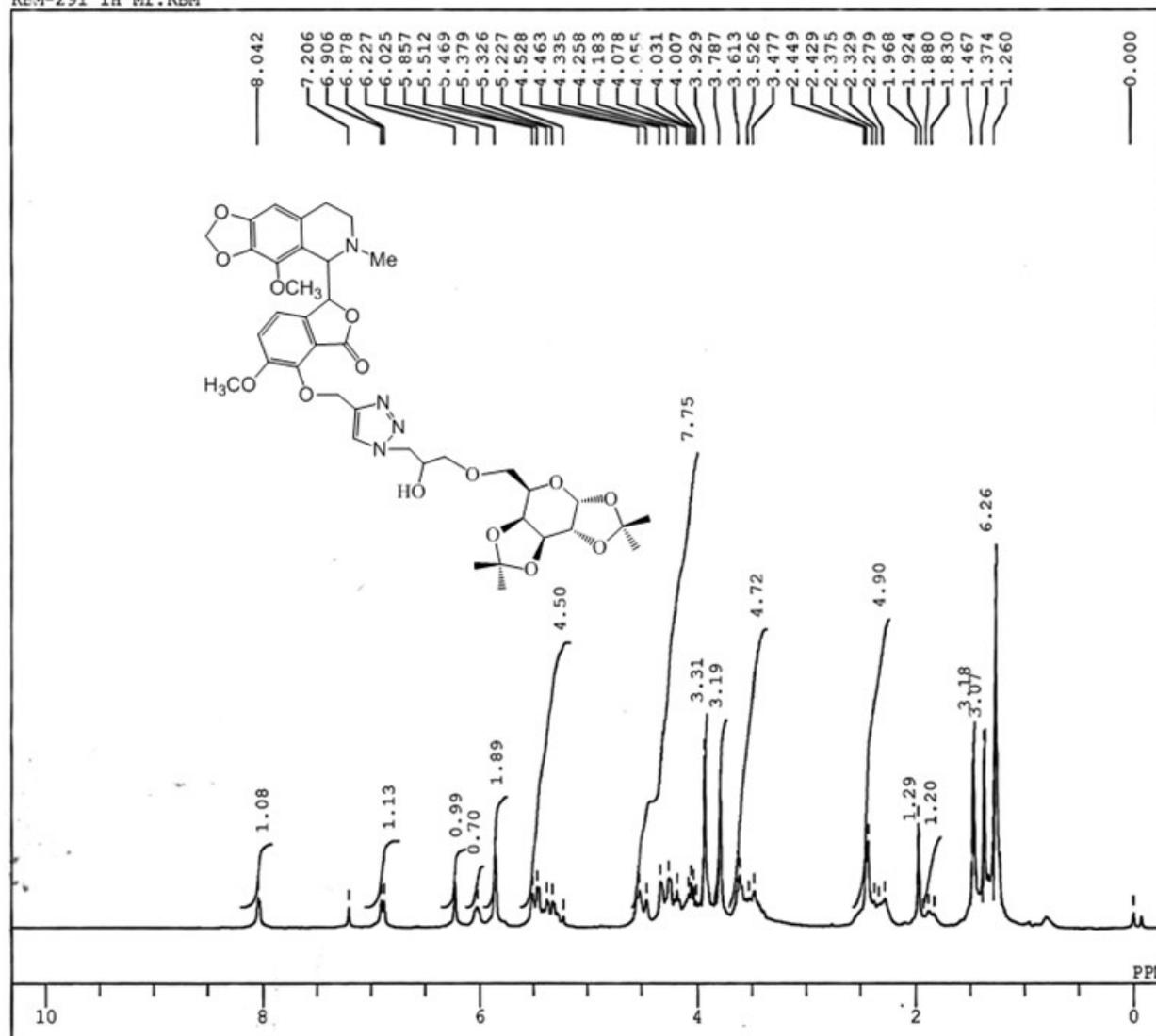


Figure 24: ¹³C NMR of compound 8k

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 KBM-291 1H Mr.KBM



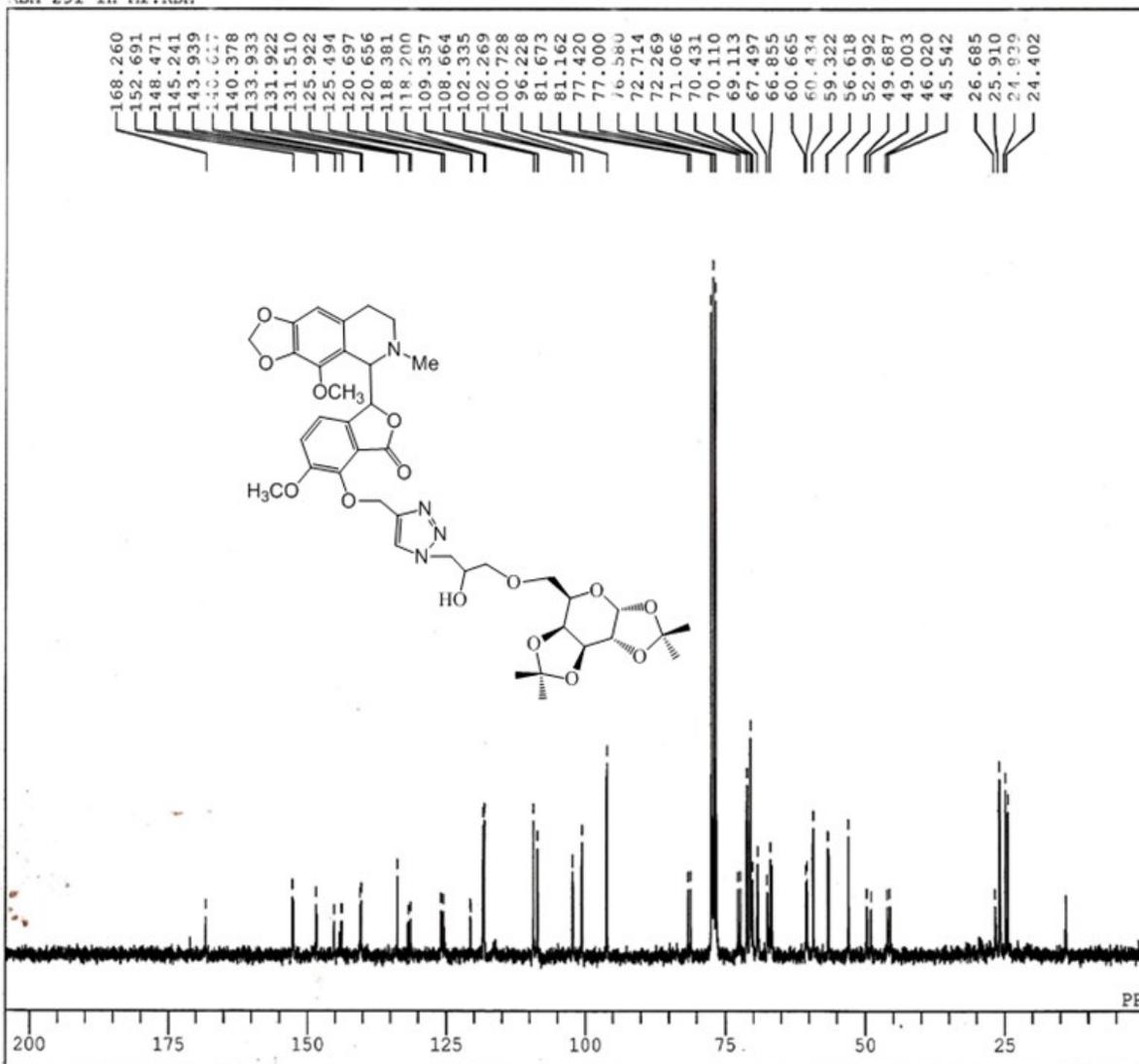
JEOL AL300 FTNMR
 CHEMISTRY DEPARTMENT
 Banaras Hindu University,
 VARANASI-221005

Operator : Nagendra Kumar
 Shishir Singh

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 OBFIN 1150.0 Hz
 POINT 32768
 FREQU 9505.7 Hz
 SCANS 9
 ACQTM 3.447 sec
 PD 1.547 sec
 PW1 5.2 us
 IRNUC 1H
 CTEMP 23.6 c
 SLVNT CDCL3
 EXREF 0.00 ppm
 BF 1.20 Hz
 RGAIN 13

Figure 25: ¹H NMR of compound 8I

:\Kunj B.Mishra\KBM-291_13c.als
 <BM-291 1H Mr.KBM



JEOL AL300 FTNMR
 CHEMISTRY DEPARTMENT
 Banaras Hindu University,
 VARANASI-221005

Operator : Nagendra Kumar
 Shishir Singh

DFILE: C:\Kunj B.Mishra\KBM
 COMMENT: KBM-291_1H_Mr.KBM
 DATETIME: Mon May 05 23:39:37
 OBSERVED: 13C
 EXMOL: BCM
 OBSERVED: 75.45 MHz
 OBSERVED: 124.00 KHz
 OBSERVED: 1840.0 Hz
 POINT: 32768
 FREQUENCY: 20408.1 Hz
 SCANS: 2000
 ACQUISITION: 1.606 sec
 PD: 1.394 sec
 PULSE: 6.0 us
 IRNUC: 1H
 CTEMP: 22.0 c
 SLVNT: CDCL3
 EXREF: 77.00 ppm
 BF: 1.20 Hz
 RGAIN: 23

Figure 26: ¹³CNMR of compound 81

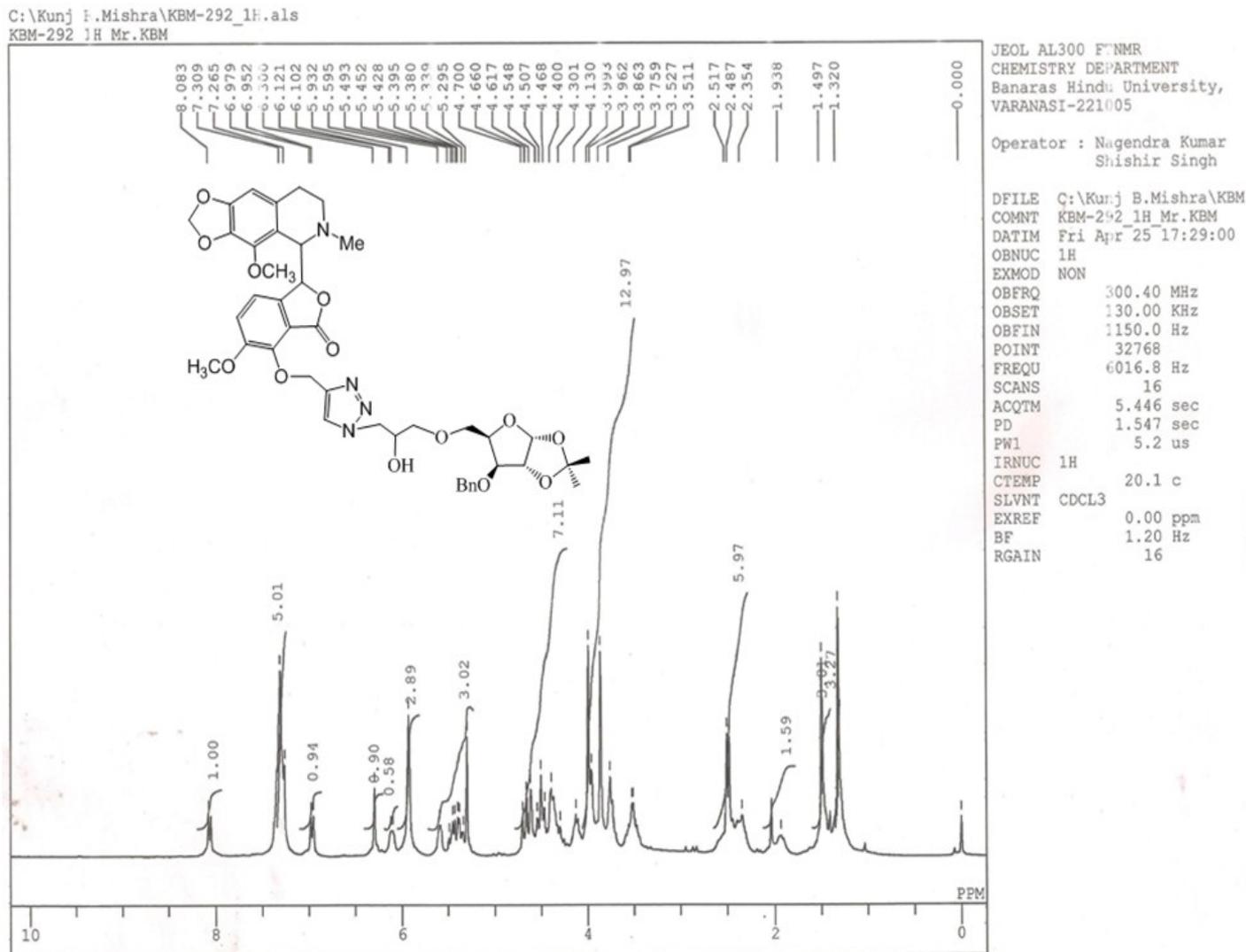


Figure 27: ¹H NMR of compound 8m

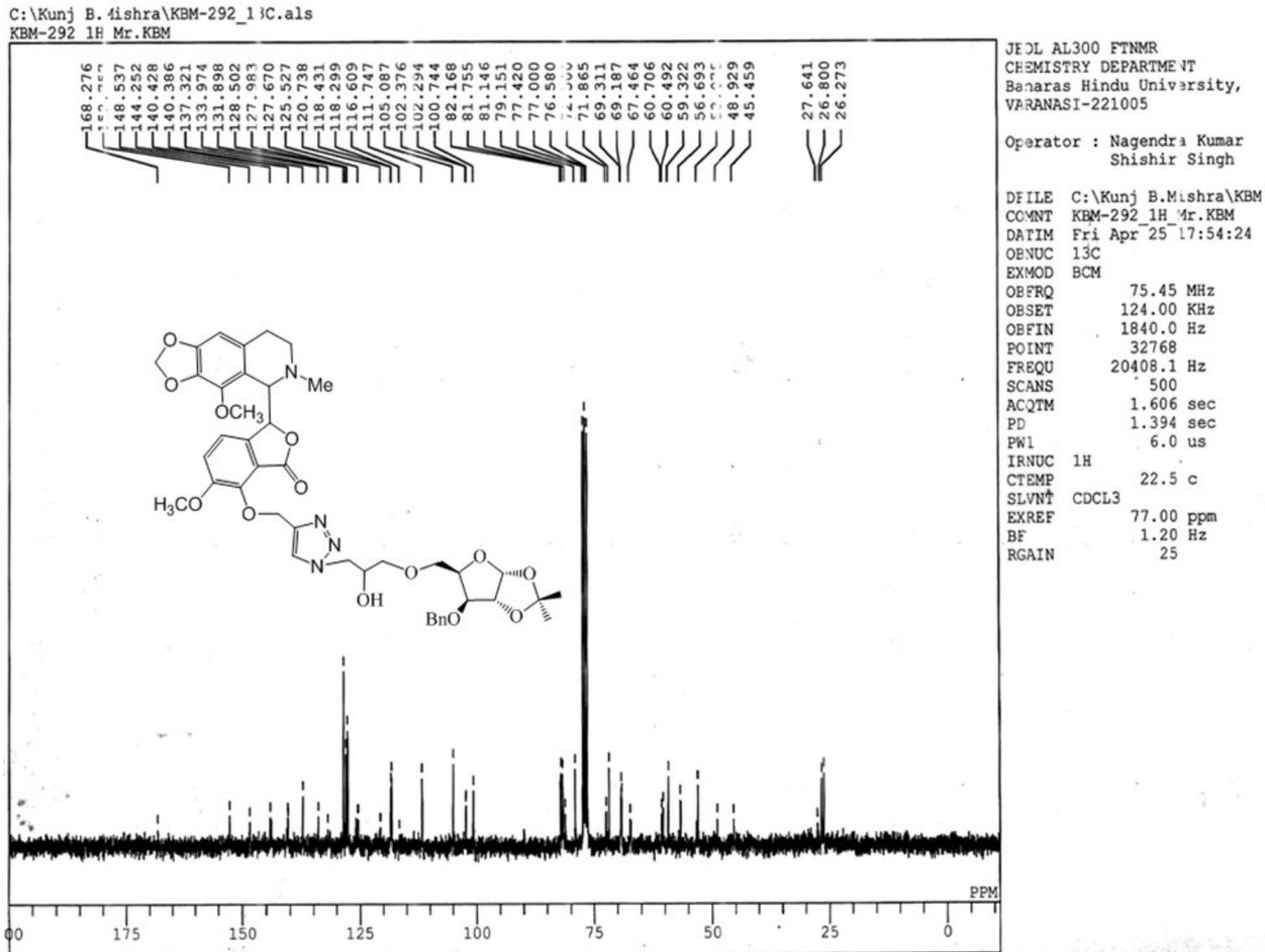


Figure 28: ¹³C NMR of compound 8m

Single Crystal X-ray data

Single-crystal X-ray data of compound **5a** was collected on Xcalibur Eos (Oxford) CCD diffractometer using graphite monochromated MoK α radiation ($\lambda = 0.71073 \text{ \AA}$). The data integration and reduction were processed with CrysAlis Pro software. The structures were solved by the direct method and then refined on F^2 by the full matrix least-squares technique with the SHELX-97 set of software using the WinGX (version 1.80.05) program package. All non-hydrogen atoms were refined anisotropically and hydrogen atoms were treated as riding atoms using SHELX default parameters. Molecular structures have drawn using ORTEP software given in Figure S-1. Further information on the crystal structure determination (excluding structure factors) has been deposited in the Cambridge Crystallographic Data Centre as supplementary publications no. **949197**. Copies of the data can be obtained free of charge upon application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: (+44) 1223-336-033. e-mail: deposit@ccdc.cam.ac.uk) or via internet.

Table S1. Crystallographic refinement data for compound **3a**

Compound	3
Empirical Formula	C ₂₄ H ₂₃ NO ₇
Formula Weight	437.44
Crystal System	Monoclinic
Space group	P 21
<i>a</i> (Å)	8.5515(14)
<i>b</i> (Å)	11.720(3)
<i>c</i> (Å)	11.059(2)
β (°)	100.720(17)
<i>V</i> (Å ³)	1089.0(4)
<i>Z</i>	2
Density (calc)	1.334
F(000), F ^o (000)	460.0, 460.26
μ (mm ⁻¹)	0.099
Crystal Size [mm]	0.09 x 0.11 x 0.14
Temperature (K)	293
Radiation	MoK α 0.71073
θ Min-Max [°]	3.3, 29.190
<i>h, k, l</i>	-10:11; -11:16; -17:12
Tot.,UniqData, R(int)	4714, 3367, 0.106

Obs. data [$I > 2.0 \sigma(I)$]	1873
Nref, Npar	5882, 289
R1, wR2, S	0.0976, 0.2378, 1.191
CCDC	1022189

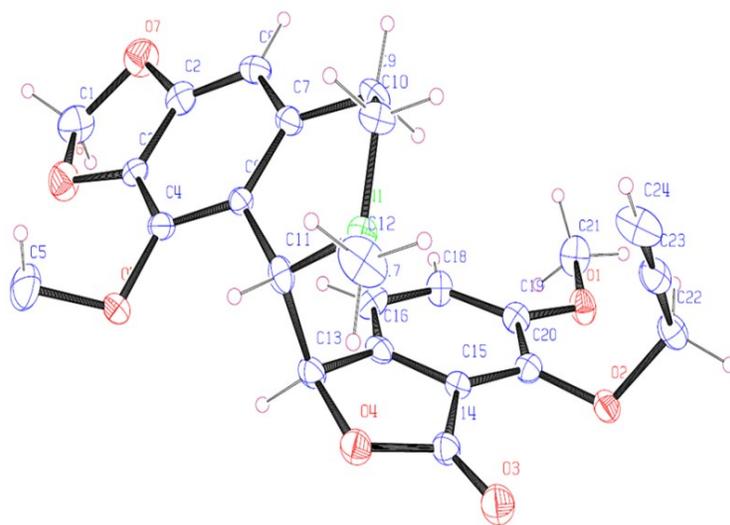


Figure S1. Molecular structure of **3a**. Thermal ellipsoids of C, N, and O are set at 40 % probability