

Two-photon Absorption of Polyfluorene Aggregates Stabilized by Insulin Amyloid Fibrils

P. Hanczyc^{*1,2}, A. Justyniarski³, D. A. Gedefaw², M. R. Andersson^{2,4}, M. Samoc³, C. Müller²

1. University of California, Santa Barbara, Center for Polymers & Organic Solids, 2520A Physical Sciences Building North, Santa Barbara, United States.

2. Chalmers University of Technology, Department of Chemistry and Chemical Engineering, Kemivägen 10, Gothenburg, 41296, Sweden.

3. Advanced Materials Engineering and Modelling Group, Faculty of Chemistry, Wrocław University of Technology, 50-370, Wrocław, Poland.

4. Ian Wark Research Institute, University of South Australia, Mawson Lakes, South Australia 5095, Australia.

Corresponding author: hanczyc@physics.ucsb.edu

Material

1. Polyfluorene derivative with polyethylene glycol side chains (PFO)

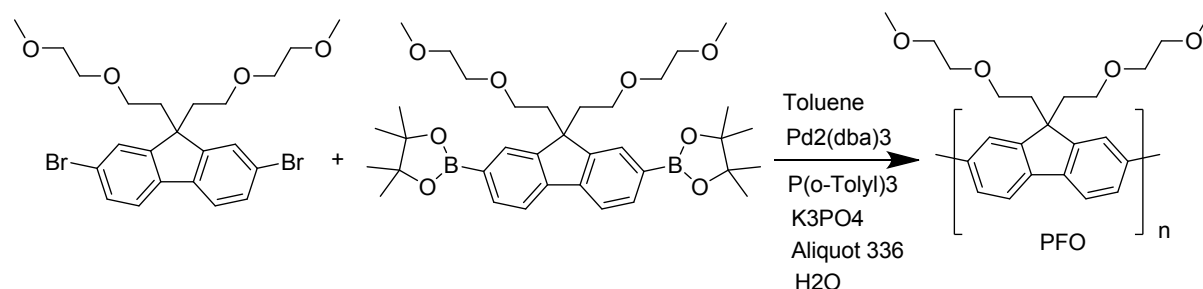


Figure S1. Synthesis of polyfluorene derivative (PFO)

¹H NMR (400 MHz, CDCl₃, δ): 7.8 (3H, m), 3.3 (9H, m), 2.9 (2H, broad singlet)

¹³C NMR (100 MHz, CDCl₃, δ): 150.0, 140.6, 139.5, 126.7, 121.6, 120.3, 71.8, 70.0, 57.2, 59.0, 51.7, 40.0,

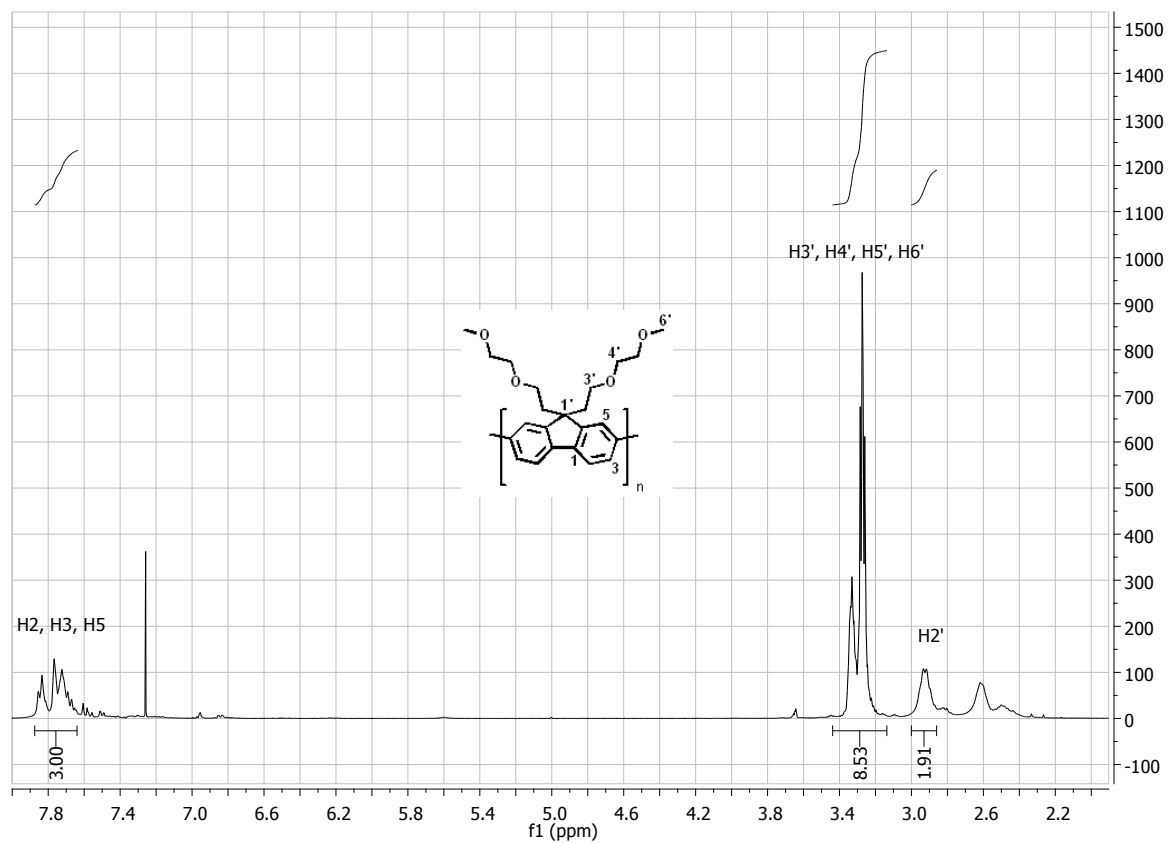


Figure S2. ^1H NMR spectrum (400 MHz, CDCl_3) of PFO.

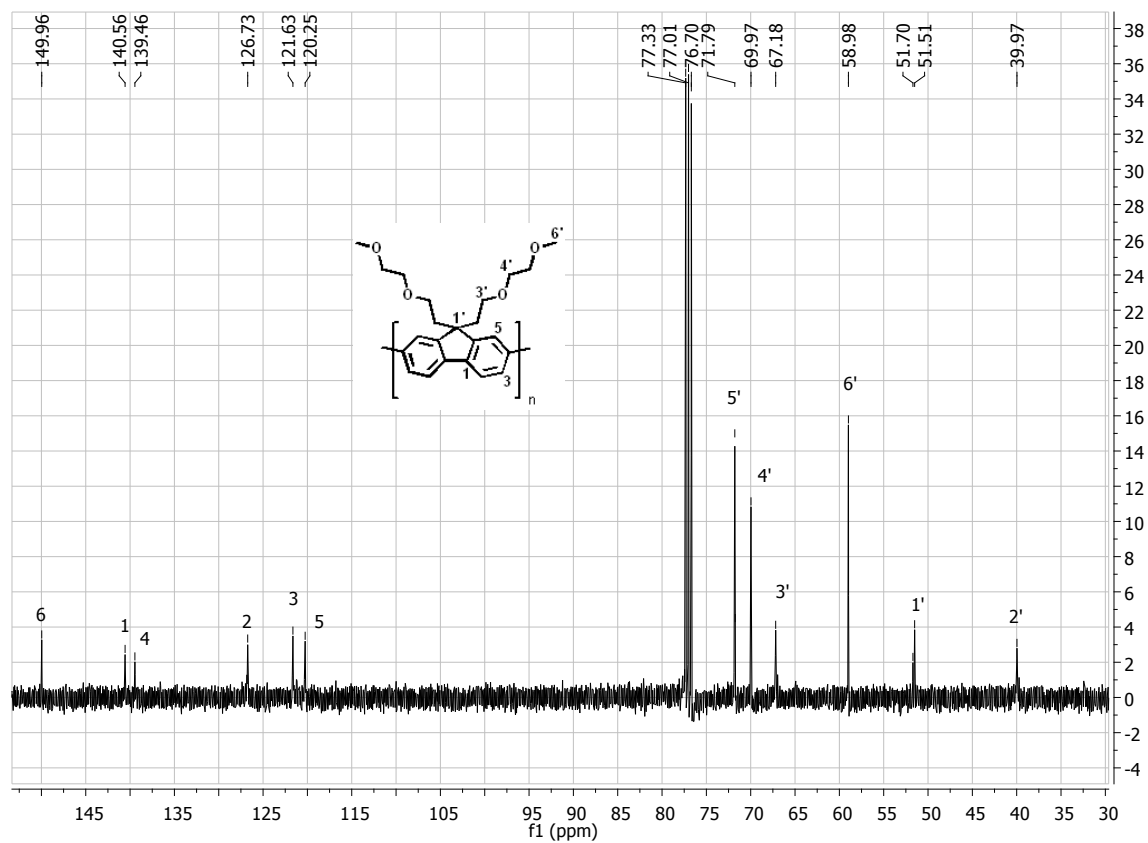


Figure S3. ^{13}C NMR spectrum (400 MHz, CDCl_3) of PFO.

2. Thioflavine T standard

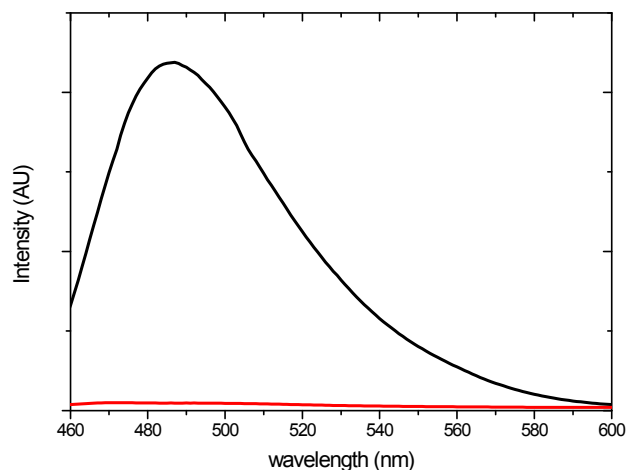


Fig S4. Emission of ThT T excited at 440 nm in presence of insulin fibrils (black solid) and insulin monomers (red solid)

3. Z-scan of PFO dissolved in ethanol and in mixture with acid water buffer (pH=2).

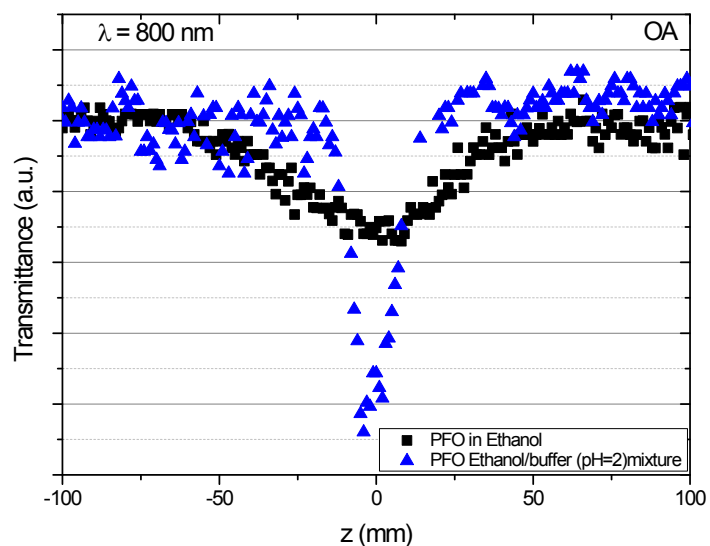


Fig. S5 Open-aperture Z-scan of aggregation PFO in ethanol/buffer mixture showing sample degradation (blue triangles) – exceedingly sharp dip in the OA trace at the focal plane position that cannot be fitted by standard Z-scan theory and stable PFO stock solution in ethanol for the sake of comparison (black squares).

