Supporting Information

2 Application of surface complexation modeling on modification of hematite surface with cobalt

3 cocatalysts: a potential tool for preparing homogeneously distributed catalysts

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9 Table S1 The properties of Co-loaded hematite samples	prepared in this study
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In Sample con	Initial Co ²⁺	equilibrated pH (± 0.2)	Hematite added (g)	loaded Co (mol/g)	RhB	Photocurrent
	concentration				degradation	density
	(M)				rate (min ⁻¹)	$(\mu A/cm^2)$
A	10-2	10	1	1.00 x 10 ⁻³	2.48 x 10 ⁻³	9.03
В	10-2	10	3	3.33 x 10 ⁻⁴	6.34 x 10 ⁻³	8.66
С	10-2	10	6	1.67 x 10 ⁻⁴	4.60 x 10 ⁻³	8.69
D	10-2	10	8	1.25 x 10 ⁻⁴	1.45 x 10 ⁻²	8.95
Е	10-2	10	10	1.00 x 10 ⁻⁴	6.01 x 10 ⁻³	7.32
F	10-3	7	1	5.28 x 10 ⁻⁴	6.08 x 10 ⁻³	9.22
G	10-3	7	3	1.76 x 10 ⁻⁴	5.16 x 10 ⁻³	6.76
Н	10-3	7	6	8.80 x 10 ⁻⁵	1.32 x 10 ⁻²	5.98
Ι	10-3	7	8	6.60 x 10 ⁻⁵	1.88 x 10 ⁻²	5.46
J	10-3	7	10	5.28 x 10 ⁻⁵	1.11 x 10 ⁻²	4.99
Κ	10-3	4	1	2.00 x 10 ⁻⁵	2.53 x 10 ⁻²	3.06
L	10-4	4	1	2.00 x 10 ⁻⁶	2.88 x 10 ⁻²	2.01
М	10-5	7	1	1.00 x 10 ⁻⁶	3.63 x 10 ⁻²	2.00
Ν	10-5	4	1	5.00 x 10 ⁻⁷	2.67 x 10 ⁻²	1.88
0	0	7	1	0	2.95 x 10 ⁻²	1.22



17 Figure S1 The XPS spectra of Co₃O₄ loaded hematite samples. It is worth mentioning that although

18 we can identify the phase of loaded Co cocatalysts, the intensity of this Co_3O_4 phase is too weak to

19 be observed in XRD analysis, even in the sample A, which has the highest Co_3O_4 loading of 1.0

20 mmol-Co/g-hematite.



Figure S2 The zeta potential of samples collected from case A, case B and unmodified hematiteparticles. Condition: 0.1 g sample in 100 mL 1.0 mM NaCl solution.



2930 Figure S3 The representative photodegradation of RhB dye by different hematite samples.31



33 Figure S4 The representative photocurrent of different hematite samples.



Figure S5 The representative EIS spectra collected under illumination with an applied potential of0.4 V vs SCE.