## Supporting Information for

Significant effect of alkyl chain length on fluorescent
thermochromism of 9,10-bis(p-alkoxystyryl)anthracenes

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## 1 Experimental section

## Materials and measurements

Anthracene, paraformaldehyde, 4-hydroxybenzaldehyde, n-butyl bromide, n-octyl bromide and $n$-hexadecyl bromide were purchased from Tianjin Guangfu Fine Chemical Reagents Company (China) and used as received. Other common reagents and solvents were purchased as analytical grade from Tianjin Jiangtian Chemical technology Co., LTD (China) and used without further purification. Tetrahydrofuran (THF) was distilled from sodium/benzophenone before use. Deionized water was employed throughout this study. 9,10-Bis(chloromethyl) anthracene ${ }^{1}$ (1), 9,10bis(diethylphosphorylmethyl)anthracene ${ }^{2}(2)$ and alkoxy-substituted benzaldehydes ${ }^{3}$ (3, 4 and 5) were prepared according to the literature methods. 9,10-Bis $(p-$ alkoxystyryl)anthracenes ( $\mathbf{D S A}-\mathbf{p n}_{\mathrm{n}}, \mathrm{n}=4,8,16$ ) were successfully synthesized by Wittig-Horner reactions of corresponding alkoxy-substituted benzaldehydes and 9,10bis(diethoxyphosphorylmethyl)anthracene ${ }^{4}$, described in Scheme 1.

NMR spectra were recorded on a Bruker AVANCE III spectrometer ( 600 MHz for ${ }^{1} \mathrm{H}$ NMR and 150 MHz for ${ }^{13} \mathrm{C} \mathrm{NMR}$ ) with $\mathrm{CDCl}_{3}$ as solvent and tetramethylsilane (TMS) as the internal standard. High-resolution mass spectra (HRMS) were measured on a Bruker Daltonics micrOTOF-Q II instrument. Photoluminescence spectra (PL) were measured on an Edinburgh FLS920 spectrometer with a slit width of 1.0 nm for both excitation and emission. TGA (thermal gravimetric analysis) was recorded on an STA 409 PC with a heating rate of $10^{\circ} \mathrm{C} / \mathrm{min}$ from room temperature to $800^{\circ} \mathrm{C}$ in a stream of $\mathrm{N}_{2}(40 \mathrm{~mL} / \mathrm{min})$. Quantitative analysis of DSA-pn4, DSA-pn8 and DSA-pn16 were conducted on a Baker Norton pharmaceuticals detector with a baseline $\mathrm{C}_{18}{ }^{-}$ column at 365 nm under $1 \mathrm{~mL} / \mathrm{min}$ of mobile phase by gradient elution with the ratio $(\mathrm{V}: \mathrm{V})$ of methylene chloride and methanol as $80: 20,50: 50,30: 70$, data was collected and analyzed with N2000 chromatography data system. Differential scanning calorimetry (DSC) curves were obtained with a Perkin Elmer Diamond DSC at heating rate of $10{ }^{\circ} \mathrm{C} / \mathrm{min}$ under $\mathrm{N}_{2}$ atmosphere. Wide-angle X-ray diffraction (WAXD) measurements were performed by using a BRUKER X-ray diffractometer (D8 ADVANCE) at a scanning rate of $4^{\circ} / \mathrm{min}$. THF aqueous solution of certain concentration was made up by addition of water into the THF solutions of samples under vigorous stirring at room temperature. The concentrations of all samples were adjusted to $1.0 \times 10^{-5} \mu \mathrm{M}$ by adding water. Thermochromic experiments: The powder sample placed on a microslide was heated by an electronic hot plate to the colortransition temperature, the fluorescence images under a 365 nm UV lamp were taken immediately. Then the heated sample was exposed in air and naturally cooled to room temperature.



Scheme 1 Synthesis and structures of DSA-pn $n_{n}$

## 9,10-Bis(chloromethyl)anthracene (1)

To a mixture of anthracene $(8.90 \mathrm{~g}, 50 \mathrm{mmol})$ and paraformaldehyde $(7.50 \mathrm{~g}, 250$ $\mathrm{mmol})$ in dioxane $(100 \mathrm{~mL})$ was dropwise added concentrated hydrochloric acid (200 mL ) at room temperature under $\mathrm{N}_{2}$ atmosphere. Then the solution was heated and slowly stirred under gentle reflux for 3 h . After this, the reaction mixture was allowed to stand for 16 h . The crude product was obtained by filtration and washed with 1,4dioxane and water until $\mathrm{pH}=7$. Finally, its recrystallization in toluene yields a yellow solid ( $8.44 \mathrm{~g}, 60 \%$ yield). ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ): 8.40-8.38 (m, 4 H ), $7.67-7.66(\mathrm{~m}, 4 \mathrm{H}), 5.61(\mathrm{~s}, 4 \mathrm{H})$.

## 9,10-Bis(diethylphosphorylmethyl)anthracene (2)

A solution of compound $\mathbf{1}(7.80 \mathrm{~g}, 28 \mathrm{mmol})$ in triethyl phosphite ( 30 ml ) was stirred vigorously at gentle reflux for 12 h . After cooling down to room temperature, it was purified by flash chromatography (ethyl acetate/petroleum ether, $1 / 1, \mathrm{v} / \mathrm{v}$ ) to give 8.16 g compound $\mathbf{2}$ as a yellow solid ( $59 \%$ yield). ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{ppm}$ ): 8.38$8.37(\mathrm{~m}, 4 \mathrm{H}), 7.58-7.56(\mathrm{~m}, 4 \mathrm{H}), 4.21(\mathrm{~d}, 4 \mathrm{H}, J=19.8 \mathrm{~Hz}), 3.91-3.86(\mathrm{~m}, 4 \mathrm{H}), 3.82-$ $3.78(\mathrm{~m}, 4 \mathrm{H}), 1.06$ (t, $12 \mathrm{H}, J=7.2 \mathrm{~Hz})$.

## 4-(n-Butoxy)benzaldehyde (3)

To a solution of 4-hydroxybenzaldehyde ( $3.00 \mathrm{~g}, 24 \mathrm{mmol}$ ) in dry DMF ( 30 mL ), $\mathrm{K}_{2} \mathrm{CO}_{3}(3.18 \mathrm{~g}, 30 \mathrm{mmol})$ and KI (catalytic amount) were added. The mixture was stirred and heated to $80^{\circ} \mathrm{C}$, followed by slow addition of $n$-butyl bromide ( $2.4 \mathrm{ml}, 27$ $\mathrm{mmol})$. The mixture was continually stirred for 8 h at this temperature. After cooling to room temperature, it was poured into brine ( 50 mL ) and extracted with dichloromethane $(3 \times 30 \mathrm{~mL})$. The organic phases were combined, washed by NaOH aqueous solutions $(10 \%, 3 \times 30 \mathrm{~mL})$ and deionized water $(3 \times 30 \mathrm{~mL})$. Then the organic
phases were dried over $\mathrm{MgSO}_{4}$ and concentrated to afford a brownish liquid ( 3.30 g , 65 \%).
4-( $n$-Octyl)benzaldehyde (4) and 4-( $n$-hexadecyl)benzaldehyde (5) were synthesized by the same method.

## 9,10-Bis(4-( $n$-butoxy)styryl)anthracenes (DSA-pn4)

To the solution of compounds $\mathbf{2}(0.60 \mathrm{~g}, 1.26 \mathrm{mmol})$ and $\mathbf{3}(0.40 \mathrm{~g}, 2.76 \mathrm{mmol})$ in 40 mL of dry THF, potassium $t$-butoxide ( $0.30 \mathrm{~g}, 2.76 \mathrm{mmol}$ ) was added. The obtained suspension was stirred at room temperature for 8 h . After removing the solvent under reduced pressure, the residue was recrystallized in $\mathrm{THF} / \mathrm{MeOH}(1 / 10, \mathrm{v} / \mathrm{v})$. The yellow solids were collected by filtration, washed with methanol, then purified by flash chromatography (petroleum ether/methylene chloride, $5 / 1, \mathrm{v} / \mathrm{v}$ ) to afford 0.43 g

DSA-pn4 as a yellow-green solid in 76 \% yield. m.p. 214-215 ${ }^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}): 8.41-8.40(\mathrm{~m}, 4 \mathrm{H}), 7.78(\mathrm{~d}, 2 \mathrm{H}, J=16.2 \mathrm{~Hz}), 7.62(\mathrm{~d}, 4 \mathrm{H}$, $J=8.4 \mathrm{~Hz}), 7.47-7.45(\mathrm{~m}, 4 \mathrm{H}), 6.99(\mathrm{~d}, 4 \mathrm{H}, J=8.4 \mathrm{~Hz}), 6.88(\mathrm{~d}, 2 \mathrm{H}, J=16.2 \mathrm{~Hz})$, $4.05(\mathrm{t}, 4 \mathrm{H}, J=6.6 \mathrm{~Hz}), 1.85-1.80(\mathrm{~m}, 4 \mathrm{H}), 1.56-1.54(\mathrm{~m}, 4 \mathrm{H}), 1.02(\mathrm{t}, 6 \mathrm{H}, J=7.2$ Hz ); ${ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}): 159.34,137.06,132.95,130.17,129.79$, 127.92, 126.69, 125.22, 122.91, 115.00, 68.01, 31.49, 19.42, 14.02. HRMS (APCl): calcd. for $\mathrm{C}_{38} \mathrm{H}_{38} \mathrm{O}_{2} \mathrm{H} 527.2945[\mathrm{M}+\mathrm{H}]^{+}$; found 527.2944. Purity: $96.95 \%$ (HPLC, $t_{\mathrm{R}}$ $=10.607 \mathrm{~min}$ ).
DSA-pn8 and DSA-pn16 were obtained by the same procedure.
DSA-pn8: Yield $67 \%$, m.p. $145-147{ }^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 8.42-$ $8.40(\mathrm{~m}, 4 \mathrm{H}), 7.78(\mathrm{~d}, 2 \mathrm{H}, J=16.8 \mathrm{~Hz}), 7.62(\mathrm{~d}, 4 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.47-7.46(\mathrm{~m}, 4 \mathrm{H})$, $6.99(\mathrm{~d}, 4 \mathrm{H}, J=8.4 \mathrm{~Hz}), 6.88(\mathrm{~d}, 2 \mathrm{H}, J=16.8 \mathrm{~Hz}), 4.04(\mathrm{t}, 4 \mathrm{H}, J=6.6 \mathrm{~Hz}), 1.86-1.82$ $(\mathrm{m}, 4 \mathrm{H}), 1.52-1.48(\mathrm{~m}, 4 \mathrm{H}), 1.40-1.33(\mathrm{~m}, 16 \mathrm{H}), 0.93(\mathrm{t}, 6 \mathrm{H}, J=6.6 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( $150 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}): 159.34,137.06,132.95,130.16,129.79,127.91,126.69$, 125.21, 122.90, 114.99, 68.34, 31.99, 29.54, 29.44, 29.41, 26.22, 22.83, 14.26.

HRMS (APCI): calcd. for $\mathrm{C}_{46} \mathrm{H}_{54} \mathrm{O}_{2} \mathrm{H} 639.4197[\mathrm{M}+\mathrm{H}]^{+}$; found 639.4195. Purity: 99.03 \% (HPLC, $\left.t_{\mathrm{R}}=12.557 \mathrm{~min}\right)$.

DSA-pn16: Yield $65 \%$, m.p. $124-125^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 8.41-$ $8.40(\mathrm{~m}, 4 \mathrm{H}), 7.78(\mathrm{~d}, 2 \mathrm{H}, J=16.2 \mathrm{~Hz}), 7.61(\mathrm{~d}, 4 \mathrm{H}, J=7.8 \mathrm{~Hz}), 7.47-7.45(\mathrm{~m}, 4 \mathrm{H})$, $6.99(\mathrm{~d}, 4 \mathrm{H}, J=7.8 \mathrm{~Hz}), 6.87(\mathrm{~d}, 2 \mathrm{H}, J=16.8 \mathrm{~Hz}), 4.03(\mathrm{t}, 4 \mathrm{H}, J=6.6 \mathrm{~Hz}), 1.85-1.81$ $(\mathrm{m}, 4 \mathrm{H}), 1.39-1.27(\mathrm{~m}, 52 \mathrm{H}), 0.89(\mathrm{t}, 6 \mathrm{H}, J=6.6 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ (ppm): 159.22, 136.93, 132.83, 130.03, 129.66, 127.78, 126.56, 125.08, 122.77, 114.87, 68.22, 31.95, 29.72, 29.70, 29.68, 29.64, 29.62, 29.45, 29.38, 29.31, 26.08, 22.71, 14.13.

HRMS (ESI): calcd. for $\mathrm{C}_{62} \mathrm{H}_{86} \mathrm{O}_{2} \mathrm{H} 863.6701$ [M + H] ${ }^{+}$; found 863.6710. Purity: $96.11 \%\left(\mathrm{HPLC}, t_{\mathrm{R}}=11.623 \mathrm{~min}\right)$.

## 2 References

1. W. Liu, et al. Chem. Commun., 2013, 49, 6042
2. H. Zhang, et al. Chem. Mater., 2009, 21, 5125.
3. C. Wang, et al. Chem. Eur. J., 2011, 17, 3322.
4. X. Zhang, et al. Chem. Commun., 2012, 48, 10895

## 3 Table S1, Figs. S1-S5

Table S1 Emission wavelength (nm) and intensity of DSA-pn $\mathbf{n}_{\mathbf{n}}$ from 298 K to 498 K

|  | DSA-pn4 |  | DSA-pn8 |  | DSA-pn16 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Temperature/K | Peak Wavelength/nm | Peak Intensity | Peak Wavelength/nm | Peak <br> Intensity | Peak Wavelength/nm | Peak <br> Intensity |
| 298 | 520 | 403.8 | 521 | 12210.0 | 505 | 3934.0 |
| 308 | 522 | 386.5 | 522 | 12160.0 | 505 | 3890.0 |
| 318 | 521 | 372.0 | 523 | 11950.0 | 505 | 3852.0 |
| 328 | 521 | 398.6 | 523 | 11800.0 | 506 | 3767.0 |
| 338 | 521 | 411.3 | 522 | 11610.0 | 508 | 3757.0 |
| 348 | 522 | 407.0 | 524 | 11320.0 | 512 | 3627.0 |
| 358 | 521 | 413.0 | 529 | 11420.0 | 515 | 3613.0 |
| 368 | 519 | 405.5 | 527 | 11260.0 | 518 | 3574.0 |
| 378 | 519 | 388.4 | 518 | 10470.0 | 524 | 3360.0 |
| 388 | 515 | 356.4 | 520 | 9012.0 | 528 | 2667.0 |
| 398 | 515 | 324.6 | 517 | 8005.0 | 532 | 1226.0 |
| 408 | 518 | 294.3 | 517 | 7574.0 | 573 | 315.4 |
| 418 | 519 | 278.6 | 519 | 7781.0 | 570 | 271.2 |
| 428 | 520 | 302.4 | 518 | 7829.0 | 576 | 233.6 |
| 438 | 521 | 316.4 | 541 | 5094.0 | 576 | 212.1 |
| 448 | 524 | 303.6 | 562 | 302.5 | 575 | 198.8 |
| 458 | 523 | 288.1 | 568 | 273.6 | 574 | 171.9 |
| 468 | 524 | 269.0 | 568 | 246.0 | 577 | 163.8 |
| 478 | 522 | 255.1 | 569 | 221.4 | 576 | 156.7 |
| 488 | 567 | 33.6 | 569 | 200.2 | 577 | 148.9 |
| 498 | 566 | 20.8 | 569 | 177.7 | 577 | 140.6 |

Fig. S1 Maximum fluorescence intensity vs temperature of DSA-pn $n_{\mathbf{n}}$ upon heating




Fig. S2 DSC curves of DSA-pn $\mathbf{n}_{\mathbf{n}}$ pristine solids



Fig. S3 Powder X-ray diffraction patterns of DSA-pn ${ }_{\mathbf{n}}$ at room temperature and 478 K for DSA-pn4, 428K for DSA-pn8, 398K for DSA-pn16



Fig. $\mathbf{S 4}$ TGA curves for $\mathbf{D S A}-\mathbf{p n}_{\mathbf{n}}$


Fig. S5 Molecular stacking modes of DSA-pn4, DSA-pn8 and DSA-pn16 in crystals


$4{ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, HRMS spectra and HPLC


Fig. S6 ${ }^{1} \mathrm{H}$ NMR spectrum of DSA-pn4


Fig. S7 ${ }^{13} \mathrm{C}$ NMR spectrum of DSA-pn4

| Acquisition Parameter |  |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- |
| Source Type | APCl | Ion Polarity | Positive | Set Nebulizer | 1.6 Bar |
| Focus | Active | Set Capillary | 4000 V | Set Dry Heater | $200{ }^{\circ} \mathrm{C}$ |
| Scan Begin | $350 \mathrm{~m} / \mathrm{z}$ | Set End Plate Offset | -500 V | Set Dry Gas $^{\text {Scan End }}$ | $2000 \mathrm{~m} / \mathrm{z}$ |



Fig. S8 HRMS spectrum of DSA-pn4


| Peak |  | RT (min) | Height ( $\mu$ V) | Area ( $\boldsymbol{\mu}$ V*Sec) | Area\% |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 |  | 5.198 | 10.485 | 170825.857 | 3.05 |
| 2 |  | 10.607 | 295.756 | 5430554.000 | 96.95 |
| Total |  |  | 306.241 | 5601379.857 | 100.00 |

Fig. S9 HPLC spectrum of DSA-pn4


Fig. S10 ${ }^{1} \mathrm{H}$ NMR spectrum of DSA-pn8


Fig. S11 ${ }^{13} \mathrm{C}$ NMR spectrum of DSA-pn8


Fig. S12 HRMS spectrum of DSA-pn8


| Peak |  | RT (min) | Height ( $\mu \mathrm{V}$ ) | Area ( $\left.\mu \mathrm{V} *_{\text {min }}\right)$ | Area\% |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 |  | 7.665 | 1.120 | 1607.200 | 0.97 |
| 2 |  | 12.557 | 130.566 | 2671387.250 | 99.03 |
| Total |  |  | 131.686 | 2672994.450 | 100.00 |

Fig. S13 HPLC spectrum of DSA-pn8


Fig. S14 ${ }^{1} \mathrm{H}$ NMR spectrum of DSA-pn16


Fig. S15 ${ }^{13} \mathrm{C}$ NMR spectrum of DSA-pn16
Acquisition Parameter

| Source Type | ESI |
| :--- | :--- |
| Focus | Active |
| Scan Begin | $300 \mathrm{~m} / \mathrm{z}$ |
| Scan End | 1000 m |

Ion Polarity
Set Capillary
Set End Plate Offset
Set Collision Cell RF
Positive
4500 V
-500 V
150.0 Vpp

| Set Nebulizer | 0.4 Bar |
| :--- | :--- |
| Set Dry Heater | $190^{\circ} \mathrm{C}$ |
| Set Dry Gas | $4.0 \mathrm{I} / \mathrm{min}$ |
| Set Divert Valve | Waste |



Fig. S16 HRMS spectrum of DSA-pn16


| Peak | RT (min) | Height ( $\mu \mathrm{V}$ ) | Area ( $\mu \mathrm{V}$ *min) | Area\% |
| :---: | :---: | :---: | :---: | :---: |
| 1 | 8. 358 | 21.818 | 381709.938 | 3.89 |
| 2 | 11.623 | 468.182 | 9440492.000 | 96.11 |
| Total |  | 490.000 | 9822201. 938 | 100.00 |

Fig. S17 HPLC spectrum of DSA-pn16

## 5 Crystallographic data

Table S2 Crystallographic data of DSA-pn4

$$
\begin{aligned}
& \text { Empirical formula } \\
& \text { Formula weight } \\
& \text { Temperature } \\
& \text { Wavelength } \\
& \text { Crystal system, space group } \\
& \text { Unit cell dimensions } \\
& \mathrm{C}_{38} \mathrm{H}_{38} \mathrm{O}_{2} \\
& 526.68 \\
& \text { 292.1(4) K } \\
& 0.7107 \mathrm{~A} \\
& \text { Triclinic, P-1 } \\
& \mathrm{a}=9.6115(5) \AA \text { alpha }=110.013(6) \mathrm{deg} . \\
& \mathrm{b}=12.5777(8) \text { A beta }=107.490(5) \text { deg. } \\
& \mathrm{c}=14.3804(10) \mathrm{A} \text { gamma }=99.474(5) \mathrm{deg} . \\
& \text { Volume } \\
& \text { Z, Calculated density } \\
& \text { Absorption coefficient } \\
& \text { F (000) } \\
& \text { Crystal size } \\
& \text { Theta range for data collection } \\
& \text { Limiting indices } \\
& \text { 1486.96(15) A }{ }^{3} \\
& 2,1.150 \mathrm{Mg} / \mathrm{m}^{3} \\
& 0.069 \mathrm{~mm}^{-1} \\
& 552 \\
& 0.3900 \times 0.2100 \times 0.0700 \mathrm{~mm} \\
& 3.00 \text { to } 26.37 \mathrm{deg} \text {. } \\
& -11<=\mathrm{h}<=12,-15<=\mathrm{k}<=15,- \\
& \text { Reflections collected / unique } \quad 12303 / 6064[\mathrm{R}(\mathrm{int})=0.0251] \\
& \text { Completeness to theta }=26.37 \\
& \text { Absorption correction } \\
& \text { Max. and min. transmission } \\
& \text { Refinement method } \\
& \text { Data / restraints / parameters } \\
& \text { Goodness-of-fit on } \mathrm{F}^{2} \\
& \text { Final } \mathrm{R} \text { indices [ } \mathrm{I}>2 \text { sigma ( } \mathrm{I} \text { ] } \\
& \text { R indices (all data) } \\
& \text { Extinction coefficient } \\
& \text { Largest diff. peak and hole } \\
& 99.7 \text { \% } \\
& \text { Semi-empirical from equivalents } \\
& 1.00000 \text { and } 0.64979 \\
& \text { Full-matrix least-squares on } \mathrm{F}^{\wedge} 2 \\
& 6064 \text { / } 5 \text { / } 364 \\
& 1.040 \\
& \mathrm{R}_{1}=0.0674, \mathrm{wR}^{2}=0.1743 \\
& \mathrm{R}_{1}=0.1051, \mathrm{wR}^{2}=0.2033 \\
& \text { 0.014(2) } \\
& 0.429 \text { and }-0.311 \text { e. } A^{-3}
\end{aligned}
$$

Table S3 Crystallographic data of DSA-pn8

| Empirical formula | $\mathrm{C}_{46} \mathrm{H}_{54} \mathrm{O}_{2}$ |
| :---: | :---: |
| Formula weight | 638.89 |
| Temperature | $293(2) \mathrm{K}$ |
| Wavelength | 1.54184 A |
| Crystal system, space group | Triclinic, $\mathrm{P}-1$ |
| Unit cell dimensions | $\mathrm{a}=9.0161(4) \mathrm{A} \mathrm{alpha}=78.313(4) \mathrm{deg}$. |
|  | $\mathrm{b}=16.5141(8) \mathrm{A} \mathrm{beta}=79.260(4) \mathrm{deg}$. |
|  | $\mathrm{c}=19.8096(9) \mathrm{A} \mathrm{gamma}=78.543(4) \mathrm{deg}$. |
| Volume | $2797.8(2) \mathrm{A}^{\wedge 3}$ |
| Z, Calculated density | $2,1.138 \mathrm{Mg} / \mathrm{m}^{\wedge 3}$ |
| Absorption coefficient | $0.512 \mathrm{~mm} \mathrm{~m}^{\wedge-1}$ |

$$
\begin{array}{cc}
\mathrm{F}(000) & 1038 \\
\text { Crystal size } & 0.4800 \times 0.1000 \times 0.0800 \mathrm{~mm} \\
\text { Theta range for data collection } & 3.29 \text { to } 67.08 \mathrm{deg} . \\
\text { Limiting indices } & -9<=\mathrm{h}<=10,-19<=\mathrm{k}<=19,-21<=\mathrm{l}<=23 \\
\text { Reflections collected / unique } & 18454 / 9928[\mathrm{R}(\mathrm{int})=0.0316] \\
\text { Completeness to theta }=67.08 & 99.3 \% \\
\text { Absorption correction } & \text { Semi-empirical from equivalents } \\
\text { Max. and min. transmission } & 1.00000 \text { and } 0.58251 \\
\text { Refinement method } & \text { Full-matrix least-squares on } \mathrm{F}^{\wedge 2} \\
\text { Data / restraints / parameters } & 9928 / 2 / 736 \\
\text { Goodness-of-fit on } \mathrm{F}^{\wedge} 2 & 1.077 \\
\text { Final R indices [I>2sigma(I)] } & \mathrm{R} 1=0.0714, \mathrm{wR} 2=0.2035 \\
\text { R indices (all data) } & \mathrm{R} 1=0.0944, \mathrm{wR} 2=0.2364 \\
\text { Extinction coefficient } & 0.0041(5) \\
\text { Largest diff. peak and hole } & 0.684 \text { and }-0.338 \mathrm{e} . \mathrm{A}^{\wedge} \wedge
\end{array}
$$

Table S4 Crystallographic data of DSA-pn16

| Empirical formula | $\mathrm{C}_{62} \mathrm{H}_{86} \mathrm{O}_{2}$ |
| :---: | :---: |
| Formula weight | 863.30 |
| Temperature | 100.00(10) K |
| Wavelength | 0.7107 A |
| Crystal system, space group | p Triclinic, P-1 |
| $\begin{array}{ll} \text { Unit cell dimensions } & \mathrm{a}=5.4009(2) \mathrm{A} \text { alpha }=89.974(3) \mathrm{deg} . \\ & \mathrm{b}=7.3944(3) \mathrm{A} \text { beta }=89.926(3) \mathrm{deg} . \\ & \mathrm{c}=31.9700(11) \text { A gamma }=88.837(3) \mathrm{deg} . \end{array}$ |  |
| Volume | 1276.50 (8) A^3 |
| Z, Calculated density | $2,1.123 \mathrm{Mg} / \mathrm{m}^{\wedge} 3$ |
| Absorption coefficient | $0.065 \mathrm{~mm}^{\wedge}$-1 |
| F(000) | 474 |
| Crystal size 0 | $0.48 \times 0.12 \times 0.01 \mathrm{~mm}$ |
| Theta range for data collection | 3.19 to 26.37 deg . |
| Limiting indices -6< | $-6<=\mathrm{h}<=6,-9<=\mathrm{k}<=9,-39<=1<=39$ |
| $1 / 5234[\mathrm{R}(\mathrm{int})=0.0325]$ |  |
| Completeness to theta $=26.37$ | 99.8 \% |
| Absorption correction | Semi-empirical from equivalents |
| Max. and min. transmission | 1.00000 and 0.55454 |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{\wedge} 2$ |
| Data / restraints / parameters | 5234 / 0 / 290 |
| Goodness-of-fit on $\mathrm{F}^{\wedge} 2$ | 1.031 |
| Final R indices [ $\mathrm{I}>2 \operatorname{sigma}(\mathrm{I})$ ] | $\mathrm{R} 1=0.0509, \mathrm{wR} 2=0.1275$ |
| R indices (all data) | $\mathrm{R} 1=0.0714, \mathrm{wR} 2=0.1406$ |
| Largest diff. peak and hole | 0.221 and -0.238 e. $\mathrm{A}^{\wedge}-3$ |

