Controlling the cooperative self-assembly of graphene oxide quantum dots in aqueous solutions

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Table of Contents

S1
S3
S4
S4
S5
S6
S6
S7

Preparation and characterization of GOQDs



Figure S1. The synthetic scheme for preparation of the GOQDs



Figure S2. a) DSC analysis of the GOQDs in N_2 atmosphere, b) TGA (solid line) and DTG (dots line)

curves in N₂ atmosphere.



Figure S3. Raman spectra of Synthesized GOQD

Particle sizes for GOQD solutions

Table S1. Average sizes of the GOQD and GOQD assemblies at different concentrations in deionized H_2O as determined by DLS at 25 °C.

GOQD [mg/ml]	Average size in diameter [nm]
0.01	0.7 +/- 0.2
0.02	0.9 +/- 0.1
0.05	1.2 +/- 0.7
0.067	66.7 +/- 4.1
0.1	66.7 +/- 9.4
1.0	78.3 +/- 19

Table S2. Average sizes of the GOQDs and GOQD assemblies at a concentration of 0.05 mg/ml after addition of HNO_3 (pH 2.15), only deionized H_2O (pH 6.93) and after addition of NaOH (pH 11.9) as determined by DLS at 25 °C.

рН	Average size in diameter [nm]
2.1	73.1 +/- 9.42

6.9	1.2 +/- 0.7
11.9	93.3 +/- 9.42

Zeta potentials (ζ) of GOQD solutions



Figure S4. Zeta potential (ζ) measurements of the GOQD solution in the 0.05 mg/ml (a) and 0.1 mg/ml (b) concentrations at deionized water.

TEM images



Figure S5. HR-TEM images of the GOQDs at concentration a) under CAC (0.05 mg/ml) and b) above CAC (1 mg/ml).

UV-Vis transmittance





AFM images and size analysis in the dried state



Figure S7. Approximate size and thickness of the single layer and self-assembled GOQDs through AFM images as a function of pH determined by DLS at 25 °C. With HNO₃ addition (pH=2.1), no addition (pH=6.9) and with NaOH (pH=11.9) at a concentration of 0.05 mg/ml in deionized H_2O .

Fluorescence spectroscopy



Figure S8. Fluorescence of 0.05 mg/ml (blue) and 1 mg/ml (red) solutions of GOQDs in deionized H_2O at different excitations and emissions.