

Supporting Information

A reversible rhodamine 6G-based fluorescence turn-on probe for Fe³⁺ in water and its application in living cells imaging

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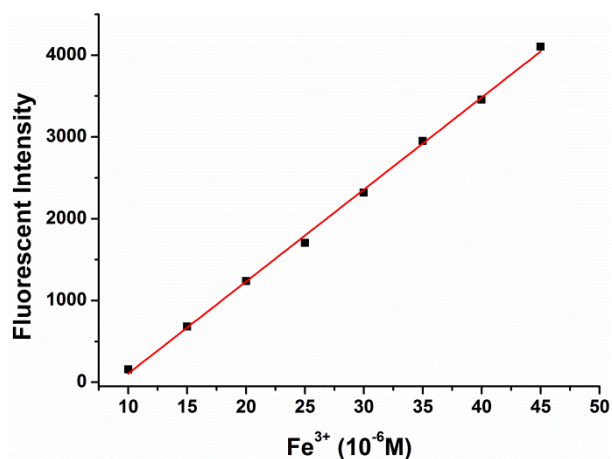


Fig. S1 Linear response of fluorescence intensity at 560 nm of L (10 μM) to the Fe³⁺ concentrations changes in distilled water.

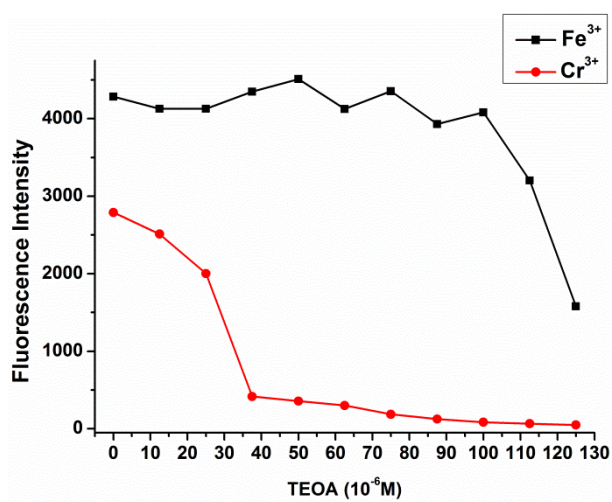


Fig. S2 Fluorescence intensity at 560 nm of Fe³⁺, Cr³⁺ (50 μM) add in L (10 μM) with different TEOA concentrations.

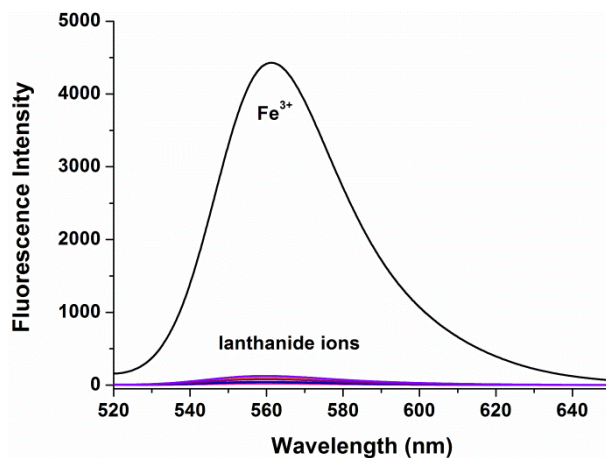


Fig. S3 Fluorescence spectra of L (10 μM) in the presence of all lanthanide ions (50 μM) except for Pm (radioactive elements) in distilled water.

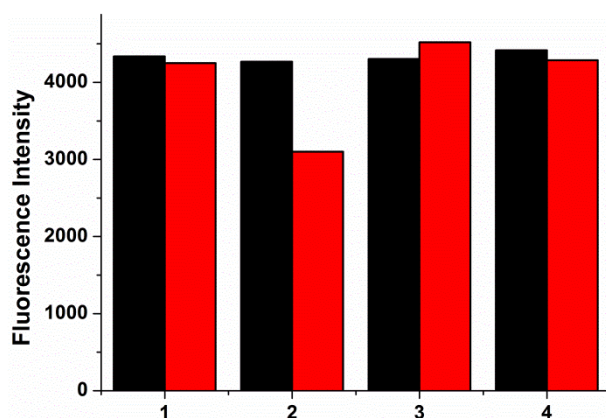


Fig. S4 Black bars: fluorescence intensity of L- Fe^{3+} (10 μM) at 560 nm; red bars: fluorescence intensity of L- Fe^{3+} (10 μM) at 560 nm in the presence of 100eq. ethanedioic acid (1), EDTA (2), tartaric acid (3) and propanedioic acid (4).

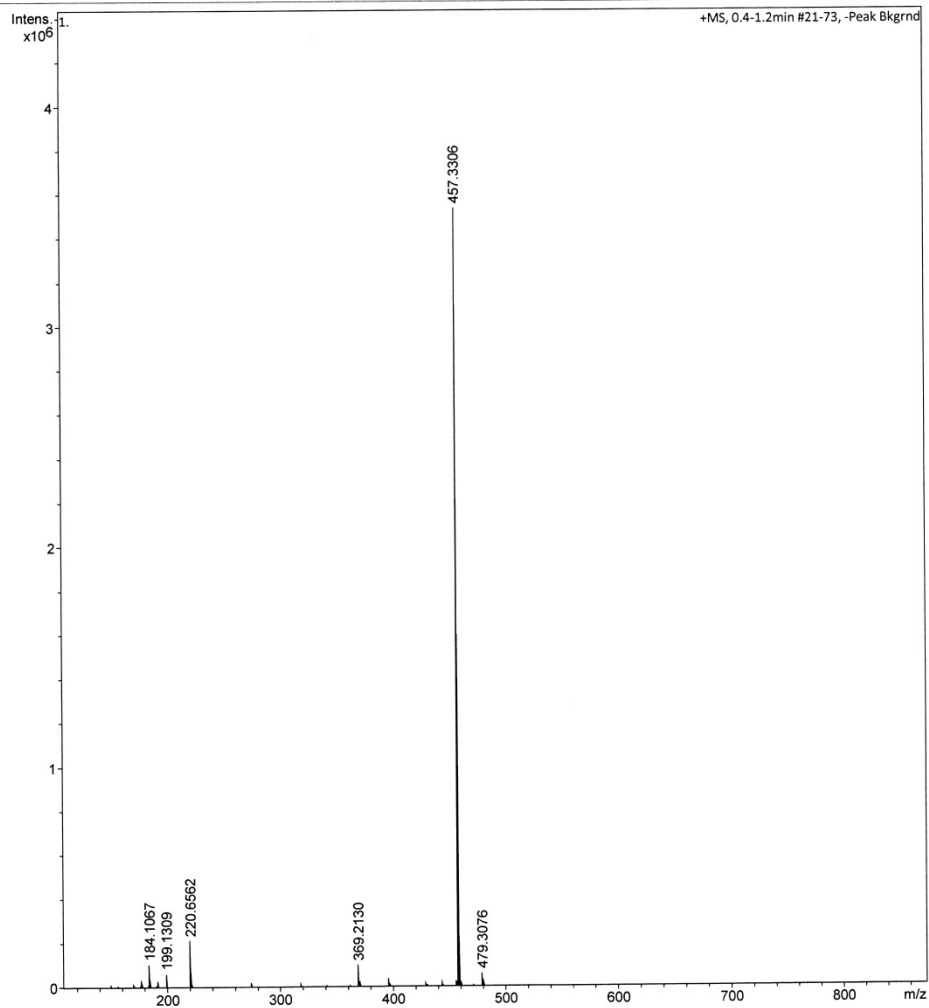
Generic Display Report

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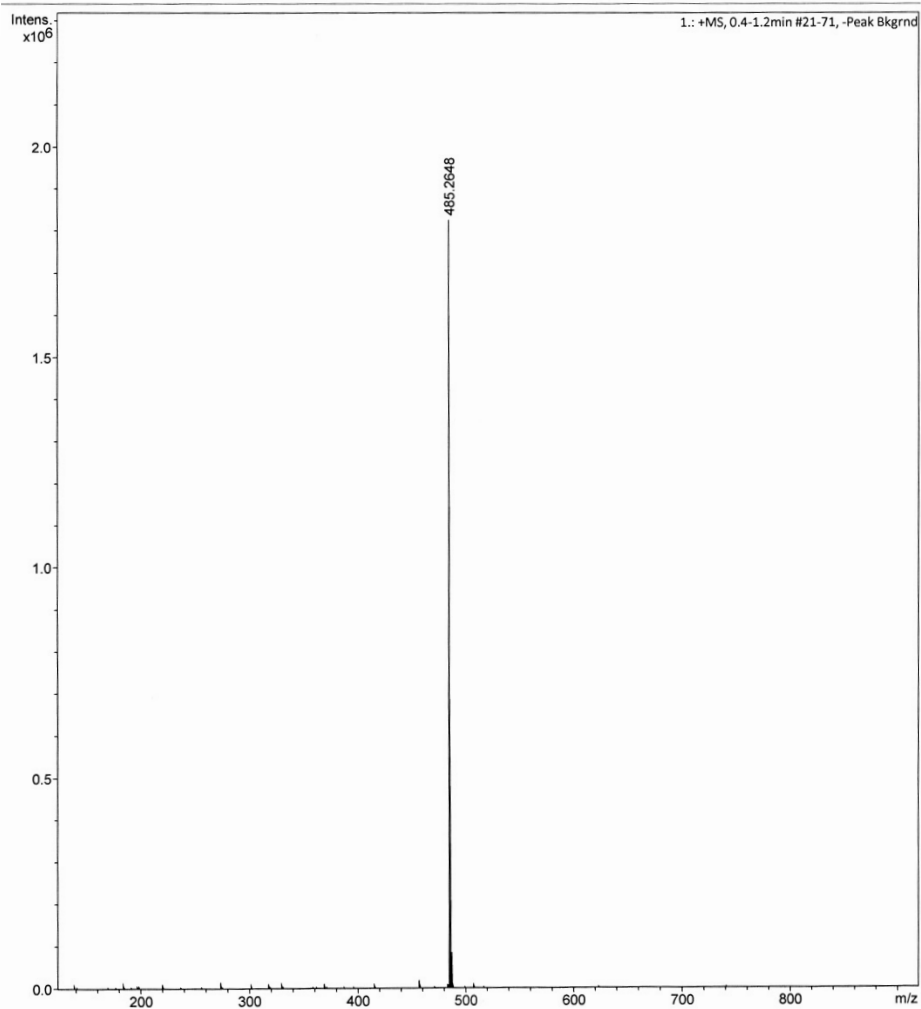
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Figure S5. ESI-MS of Compound 1.

Generic Display Report

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Figure S6. ESI-MS of Compound L.

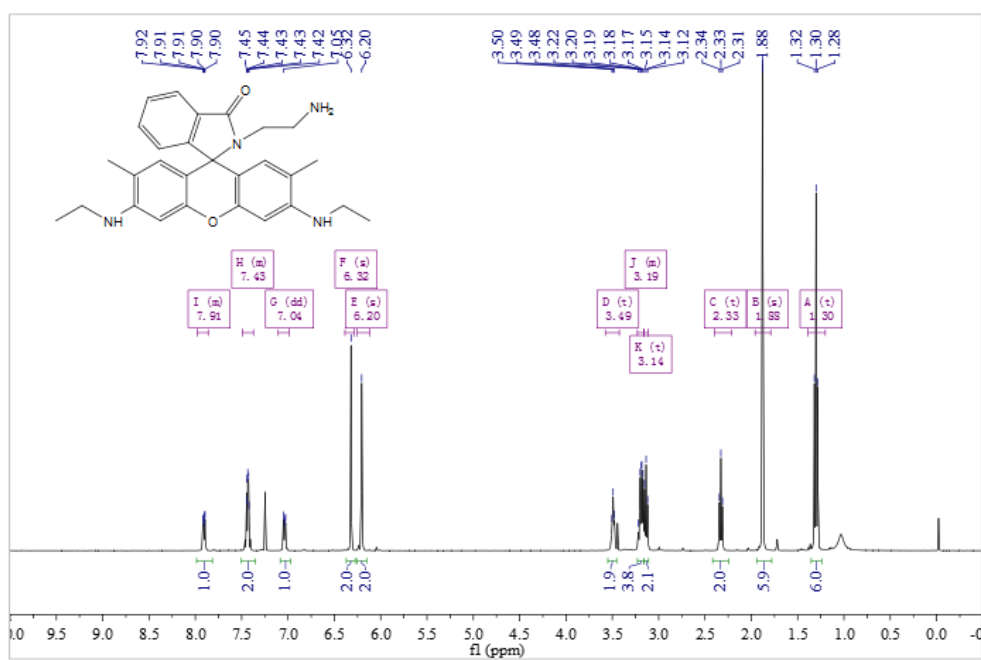


Fig. S7 ^1H NMR spectrum of 1 in CDCl_3 .

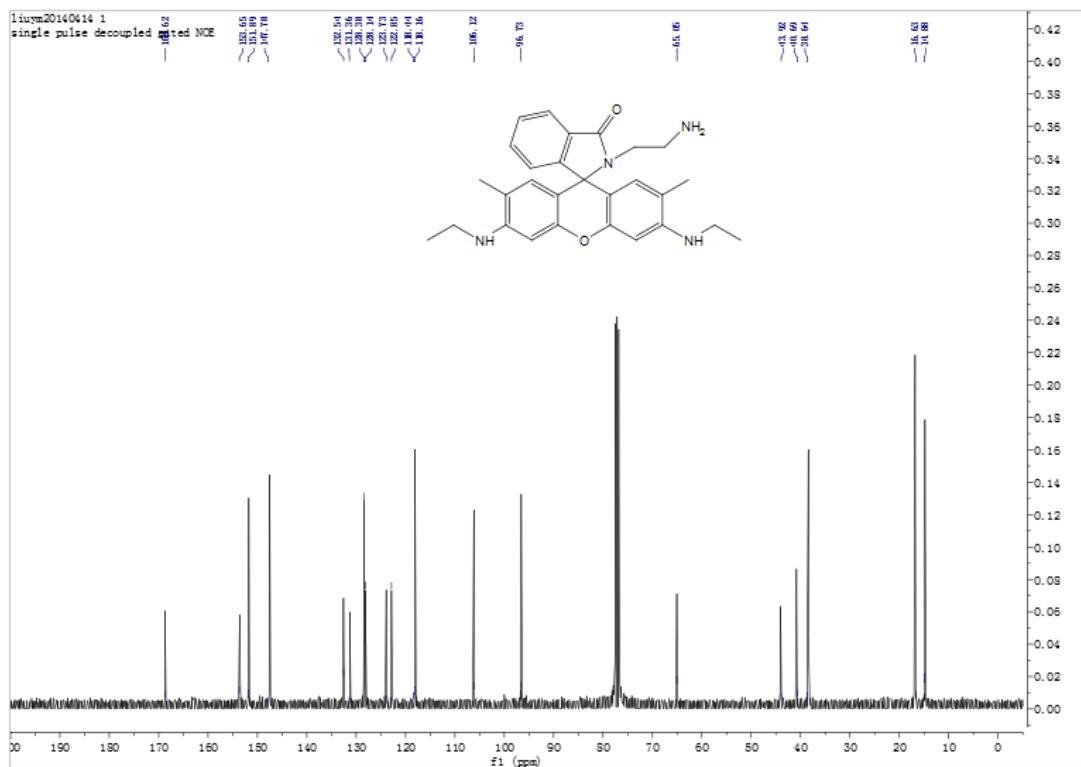


Fig. S8 ^{13}C NMR spectrum of 1 in CDCl_3 .

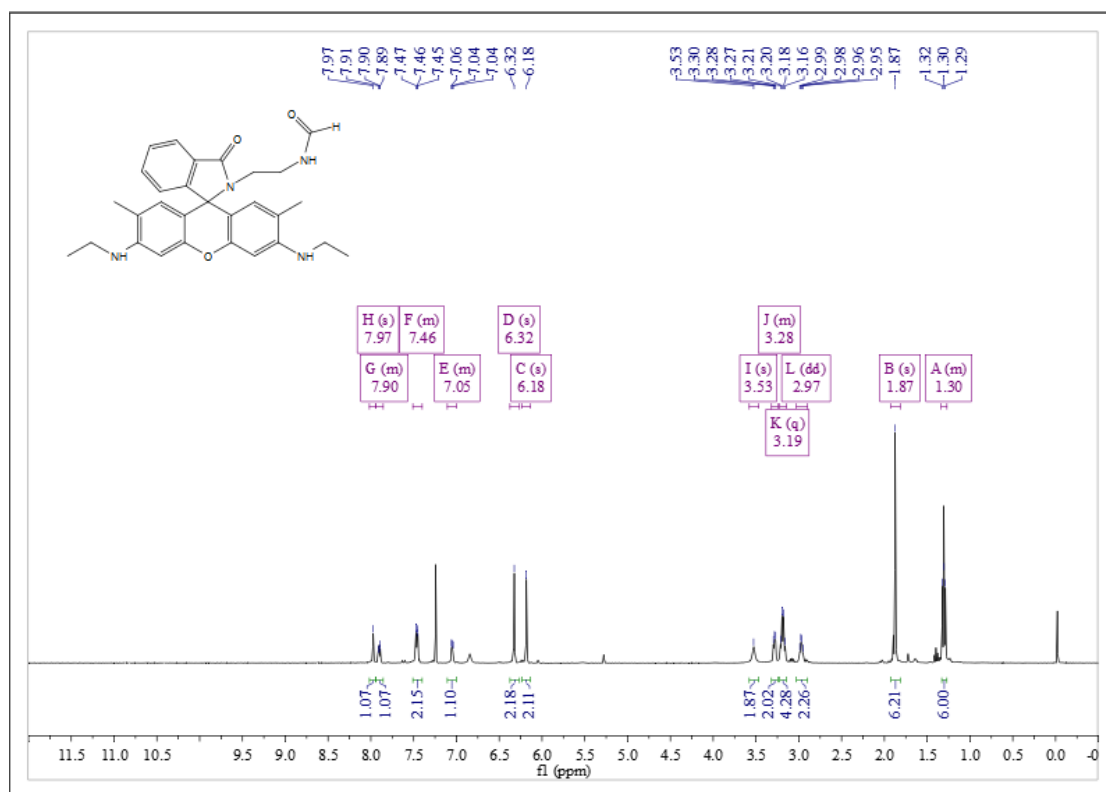


Fig. S9 ¹H NMR spectrum of L in CDCl₃.

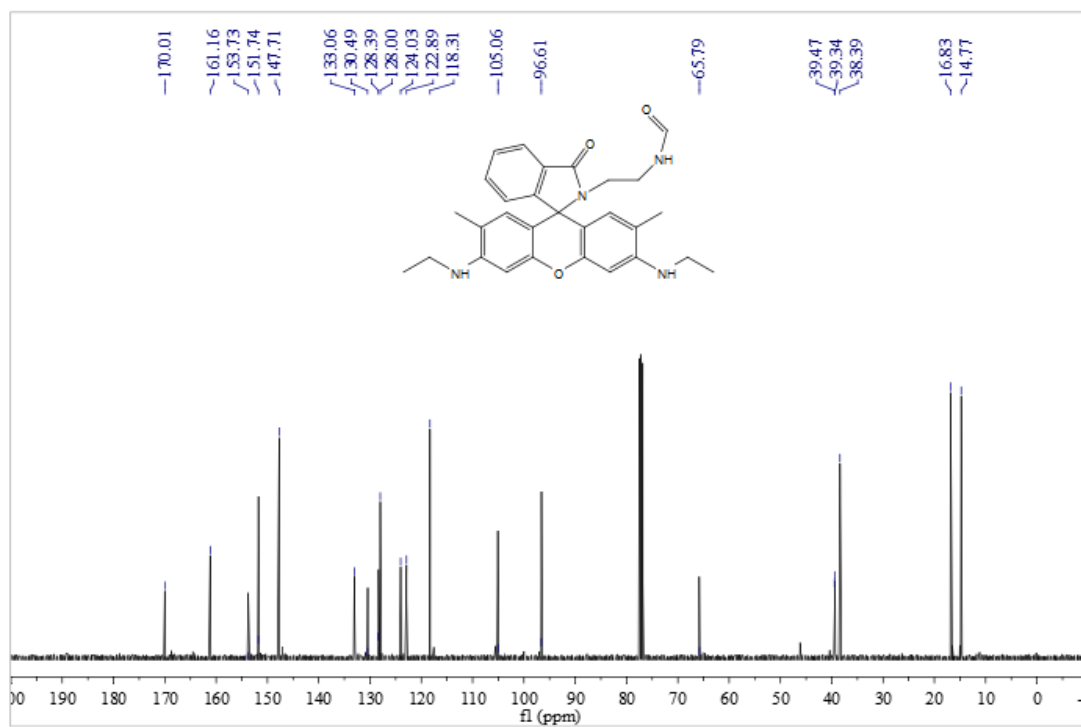


Fig. S10 ¹³C NMR spectrum of L in CDCl₃.

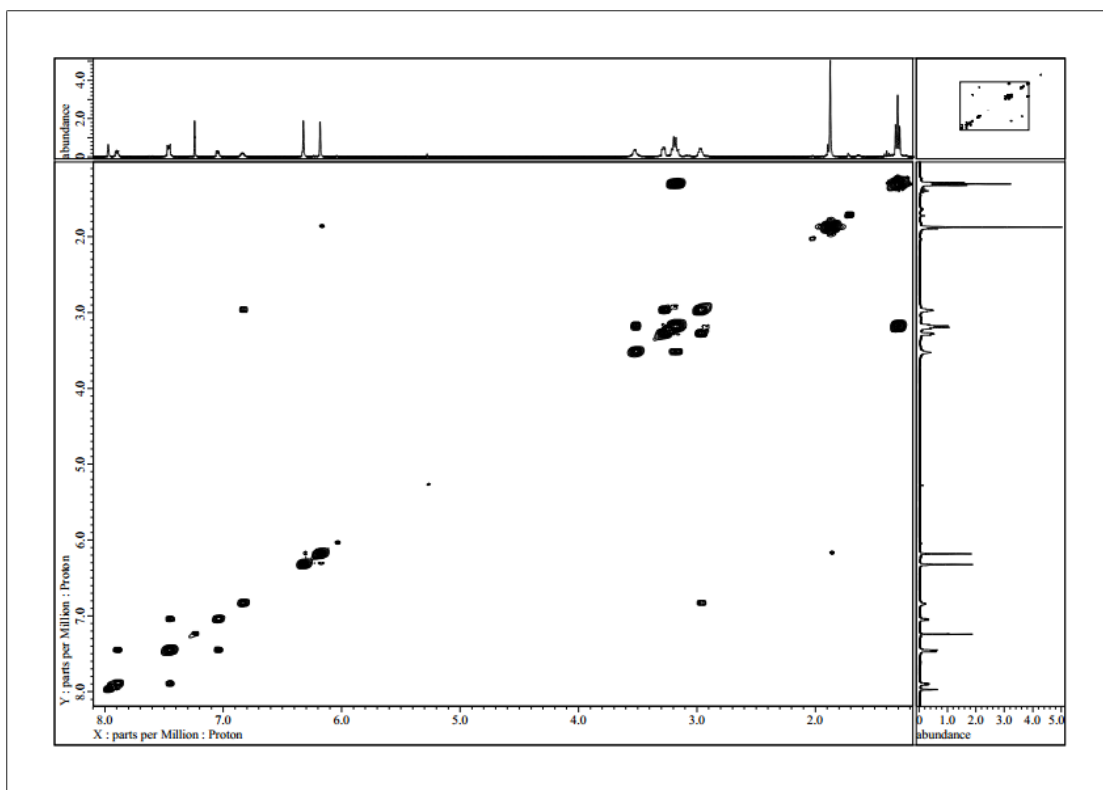


Fig. S11 ^1H - ^1H cosy NMR spectrum of L in CDCl_3 .

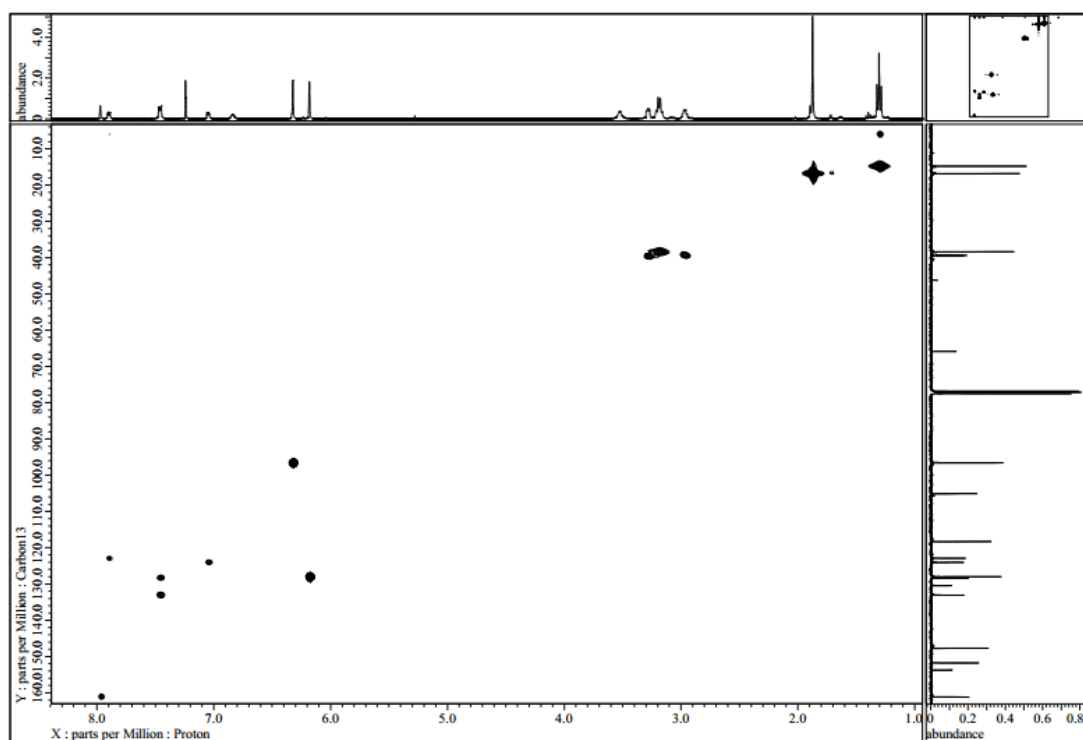


Fig. S12 ^{13}C - ^1H cosy NMR spectrum of L in CDCl_3 .