## **Supporting Information**

## Novel Organic-Inorganic Hybrids Based on T<sub>8</sub> and T<sub>10</sub> Silsesquioxanes: Synthesis, Cage-Rearrangement and Properties

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a) H<sub>2</sub>O, HCl; b) H<sub>2</sub>O, CF<sub>3</sub>SO<sub>3</sub>H c) acetone, precipitate d) acetone solution e) CF<sub>3</sub>SO<sub>3</sub>H, DMSO f) decanoyl chloride, DMF, Et<sub>3</sub>N

Scheme S1. Synthesis of 1–5.



Figure S1. Powder XRD patterns for 1–5.



Figure S2. Powder XRD patterns for 1–5 after heating to the decomposition temperature.



Figure S3. <sup>1</sup>H NMR (500 MHz, DMSO- $d_6$ , 300 K) spectrum of 1, s = solvent.



Figure S4. <sup>13</sup>C NMR (126 MHz, DMSO-d<sub>6</sub>, 300 K) spectrum of 1.









Figure S7. Simulated (A) calcd for  $C_{24}H_{65}N_8O_{12}Si_8$ , [M -8HCl + H]<sup>+</sup> and measured (B) HR-MS (ESI+, TOF, MeOH) spectra of 1.



**Figure S8.** EDS spectrum of **1** (copper content is derived from the high-purity conducting Cu grid).



Figure S9. Powder XRD pattern of 1.



**Figure S10.** <sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>, 300 K) spectrum of **2**, s = solvent. Chemical shifts were referenced to tetramethylsilane (TMS) (δ 0.0).





**Figure S12.** <sup>29</sup>Si NMR (59.6 MHz, DMSO-d<sub>6</sub>, 300 K) spectrum of **2**. Chemical shifts were referenced to 4,4-dimethyl-4-silapentane-1-sulfonic acid (DSS) (δ 1.316).





Figure S14. Simulated (A) calcd for  $C_{24}H_{65}N_8O_{12}Si_8$ , [M -8CF<sub>3</sub>SO<sub>3</sub>H + H]<sup>+</sup> and measured (B) HR-MS (ESI+, TOF, MeOH) spectra of **2**.



Figure S15. EDS spectrum of 2 (copper content is derived from the high-purity conducting Cu grid).



Figure S16. Powder XRD pattern of 2.



Figure S17. <sup>1</sup>H NMR (500 MHz, DMSO- $d_6$ , 300 K) spectrum of 3, s = solvent.





Figure S20. <sup>29</sup>Si NMR (59.6 MHz, DMSO-d<sub>6</sub>, 300 K) spectrum of **3** obtained in method B. Chemical shifts were referenced to 4,4-dimethyl-4-silapentane-1-sulfonic acid (DSS) ( $\delta$  1.316).



Figure S21. <sup>29</sup>Si NMR (59.6 MHz, DMSO-d<sub>6</sub>, 300 K) spectrum of **3** obtained in method C (by cage-rearrangement  $1 \rightarrow 3$ ). Chemical shifts were referenced to tetramethylsilane (TMS) ( $\delta$  0.00).



Figure S22. <sup>29</sup>Si NMR (59.6 MHz, DMSO-d<sub>6</sub>, 300 K) spectrum of **3** obtained in method D (by cage-rearrangement  $2 \rightarrow 3$ ). Chemical shifts were referenced to tetramethylsilane (TMS) ( $\delta$  0.00).





**Figure S24.** Simulated (A) calcd for  $C_{30}H_{82}N_{10}O_{15}Si_{10}$ , [M -10CF<sub>3</sub>SO<sub>3</sub>H + 2H]<sup>2+</sup> and measured (B) HR-MS (ESI+, TOF, MeOH) spectra of **3** obtained in method A.



Figure S25. Simulated (A) calcd for  $C_{30}H_{83}N_{10}O_{15}Si_{10}$ , [M -10CF<sub>3</sub>SO<sub>3</sub>H + 3H]<sup>3+</sup> and measured (B) HR-MS (ESI+, TOF, MeOH) spectra of **3** obtained in method B.



Figure S26. HR-MS (ESI+, TOF, MeOH) spectra of **3** obtained in method C (by cage-rearrangement  $1 \rightarrow 3$ ). Simulated (A) calcd for  $C_{30}H_{81}N_{10}O_{15}Si_{10}$ , [M -10CF<sub>3</sub>SO<sub>3</sub>H + 1H]<sup>+</sup> and measured (B).



Figure S27. HR-MS (ESI+, TOF, MeOH) spectra of 3 obtained in method D (by cage-rearrangement  $2 \rightarrow 3$ ).



Figure S28. EDS spectrum of 3 (copper content is derived from the high-purity conducting Cu grid).



Figure S29. Powder XRD pattern of 3.



Figure S30. <sup>1</sup>H DOSY (600 MHz, DMSO-d<sub>6</sub>, 300 K) spectrum of 2 (top) and 3 (bottom).











Figure S36. FT-IR (KBr pellets) spectrum of 4.



Figure S37. Simulated (A) calcd for  $C_{104}H_{210}N_8O_{20}Si_8$ ,  $[M + 2H]^{2+}$  and measured (B) HR-MS (ESI+, TOF, CHCl<sub>3</sub>) spectra of 4.



Figure S38. EDS spectrum of 4 (copper content is derived from the high-purity conducting Cu grid).



Figure S39. Powder XRD pattern of 4.







**Figure S43.** <sup>29</sup>Si NMR (59.6 MHz, DMSO-d<sub>6</sub>, 300 K) spectrum of **5**. Chemical shifts were referenced to 4,4-dimethyl-4-silapentane-1-sulfonic acid (DSS) (δ 1.316).



Figure S44. FT-IR (KBr pellets) spectrum of 5.



Figure S45. Simulated (A) calcd for  $C_{130}H_{262}N_{10}O_{25}Si_{10}$ ,  $[M + 2H]^{2+}$  and measured (B) HR-MS (ESI+, TOF, CHCl<sub>3</sub>) spectra of 5 (method A).



Figure S46. HR-MS (ESI+, TOF, CHCl<sub>3</sub>) spectra of 4 and 5.



Figure S47. EDS spectrum of 5 (copper content is derived from the high-purity conducting Cu grid).



Figure S48. Powder XRD pattern of 5.





Figure S51. TG (black line), and DTA (blue line) thermogram of 1 10 °C/min (in the air atmosphere: 60% N<sub>2</sub>, 40% O<sub>2</sub>).



Figure S52. TG (black line), and DTA (blue line) thermogram of 2 10 °C/min (in the air atmosphere: 60% N<sub>2</sub>, 40% O<sub>2</sub>).



**Figure S53.** TG (black line), and DTA (blue line) thermogram of **3** 10 °C/min (in the air atmosphere: 60% N<sub>2</sub>, 40% O<sub>2</sub>).



**Figure S54.** TG (black line), and DTA (blue line) thermogram of **4** 10 °C/min. Left graph in the air atmosphere (60% N<sub>2</sub>, 40% O<sub>2</sub>), right in nitrogen.



Figure S55. TG (black line), and DTA (blue line) thermogram of 5 10 °C/min. Left graph in the air atmosphere (60% N<sub>2</sub>, 40% O<sub>2</sub>), right in nitrogen.



**Figure S56.** TG (black line), and DTA (blue line) thermogram of **6** 10 °C/min. Left graph in the air atmosphere (60% N<sub>2</sub>, 40% O<sub>2</sub>), right in nitrogen.



**Figure S57.** First derivative of TG (DTG) thermograms of **1–6** 10 °C/min (in the air atmosphere: 60% N<sub>2</sub>, 40% O<sub>2</sub>).



Figure S58. DSC of 5, 2<sup>nd</sup> heat & cooling cycle (5 °C/min in the nitrogen atmosphere).



Figure S59. DSC of 6, 2<sup>nd</sup> heat & cooling cycle (5 °C/min in the nitrogen atmosphere).



Figure S60. Selected HR-TEM images of 4 (a) and 5 (b).



**Figure S61.** <sup>29</sup>Si NMR (59.6 MHz, DMSO-d<sub>6</sub>, 300 K) spectrum of **1** after reaction with 0, 1, 4, 8, 12 and 16 equivalents of CH<sub>3</sub>SO<sub>3</sub>H. Chemical shifts were referenced to tetramethylsilane (TMS).



**Figure S62.** Simulated (up) calcd for  $C_{18}H_{52}N_6O_{11}Si_6Na$ ,  $[M + Na]^+$  and measured (down) HR-MS (ESI+, TOF, MeOH) spectra of **C**.



**Figure S63.** Simulated (up) calcd for  $C_{10}H_{16}F_{12}N_2O_{13}S_4Si_2Na$ ,  $[M + Na]^+$  and measured (down) HR-MS (ESI+, TOF, MeOH) spectra of **D**.



Figure S64. Simulated (up) calcd for  $C_{24}H_{68}N_8O_{14}Si_8$ ,  $[M + H]^+$  and measured (down) HR-MS (ESI+, TOF, MeOH) spectra of E.



Figure S65. Molecular models of compound 4 (a), compound 5 (b).

Table S1. MM2 calculations of the minimization energies of Compound 2, 3, 4, 5.

	Amine		Amide	
Energy parameters*	Compound 2	Compound 3	Compound 4	Compound 5
Stretch:	1.7969	7.6789	7.0276	11.1709
Bend:	7.0679	58.1066	27.8027	155.4612
Stretch-Bend:	-0.6755	-4.5902	0.9053	0.4549
Torsion:	0.9477	3.1937	29.6476	39.5012
Non-1,4 VDW:	-35.4194	-53.9810	-52.8458	-123.3177
1,4 VDW:	2.4715	15.4807	44.0530	66.8356
Dipole/Dipole:	35.5346	35.0298	-25.4860	-47.7560
Total Energy:	11.7237	60.9185	31.1044	102.3501

\*units of energy are kcal/mol



**Figure S66**. <sup>29</sup>Si NMR (59.63 MHz, DMSO-d<sub>6</sub>, 300 K) spectrum of DSS (4,4-dimethyl-4-silapentane-1-sulfonic acid). Chemical shifts were referenced to TMS (Tetramethylsilane) ( $\delta$  0.000).