

Supporting Information

Ammonia Borane in External Electric Field: Structure, Charge transfer, and Chemical Bonding

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Table S1. The optimized structural parameters and Dipole moment (μ) of BH_3NH_3 in different external electric fields^a.

| $E_{ext} \times 10^{-4}$ au | r (B-N) | r(B-H) | r(N-H) | H-B-H | H-N-H | H ₁ -B-H ₂ -H ₃ | H ₁ '-N-H ₂ '-H ₃ ' | μ_1 | μ_2 |
|-----------------------------|---------|--------|--------|-------|-------|--|--|---------|---------|
| -767 | 1.653 | 1.263 | 1.045 | 106.7 | 101.0 | 113.9 | 103.7 | 13.33 | 6.30 |
| -750 | 1.634 | 1.263 | 1.044 | 106.2 | 100.8 | 112.7 | 103.4 | 13.01 | 6.36 |
| -700 | 1.617 | 1.258 | 1.040 | 106.1 | 101.1 | 112.7 | 103.7 | 12.26 | 6.35 |
| -650 | 1.611 | 1.253 | 1.037 | 106.5 | 101.5 | 113.5 | 104.4 | 11.63 | 6.31 |
| -600 | 1.608 | 1.247 | 1.034 | 107.1 | 102.0 | 114.6 | 105.2 | 11.04 | 6.25 |
| -550 | 1.606 | 1.242 | 1.032 | 107.7 | 102.5 | 115.9 | 106.0 | 10.49 | 6.18 |
| -519 | 1.606 | 1.239 | 1.031 | 108.1 | 102.8 | 116.8 | 106.5 | 10.15 | 6.14 |
| -500 | 1.606 | 1.237 | 1.030 | 108.3 | 102.9 | 117.3 | 106.8 | 9.96 | 6.12 |
| -450 | 1.607 | 1.233 | 1.028 | 108.9 | 103.4 | 118.7 | 107.6 | 9.46 | 6.05 |
| -400 | 1.609 | 1.229 | 1.026 | 109.5 | 103.9 | 120.1 | 108.5 | 8.98 | 5.99 |
| -350 | 1.611 | 1.226 | 1.024 | 110.1 | 104.4 | 121.5 | 109.3 | 8.52 | 5.92 |
| -300 | 1.614 | 1.222 | 1.023 | 110.6 | 104.9 | 122.9 | 110.2 | 8.06 | 5.86 |
| -250 | 1.617 | 1.219 | 1.022 | 111.1 | 105.4 | 124.3 | 111.1 | 7.61 | 5.79 |
| -200 | 1.621 | 1.217 | 1.021 | 111.6 | 105.8 | 125.7 | 112.1 | 7.18 | 5.73 |
| -150 | 1.626 | 1.214 | 1.020 | 112.1 | 106.3 | 127.2 | 113.0 | 6.74 | 5.66 |
| -100 | 1.633 | 1.212 | 1.019 | 112.6 | 106.8 | 128.7 | 114.1 | 6.31 | 5.59 |
| -50 | 1.641 | 1.210 | 1.018 | 113.1 | 107.4 | 130.3 | 115.3 | 5.87 | 5.51 |
| 0 | 1.649 | 1.208 | 1.017 | 113.6 | 107.9 | 131.9 | 116.3 | 5.44 | 5.44 |
| 50 | 1.660 | 1.207 | 1.016 | 114.1 | 108.5 | 133.6 | 117.8 | 5.44 | 5.35 |
| 100 | 1.674 | 1.205 | 1.015 | 114.6 | 109.1 | 135.6 | 119.2 | 4.99 | 5.25 |
| 150 | 1.692 | 1.204 | 1.015 | 115.2 | 109.8 | 137.7 | 120.9 | 4.52 | 5.14 |
| 200 | 1.716 | 1.202 | 1.014 | 115.8 | 110.6 | 140.4 | 122.9 | 4.04 | 4.99 |
| 250 | 1.751 | 1.201 | 1.013 | 116.5 | 111.6 | 143.8 | 125.5 | 3.51 | 4.80 |
| 300 | 1.820 | 1.199 | 1.011 | 117.5 | 113.2 | 149.2 | 130.7 | 2.90 | 4.42 |
| 320 | 1.902 | 1.197 | 1.009 | 118.3 | 114.9 | 154.7 | 136.8 | 1.80 | 3.97 |
| 321 | 3.207 | 1.192 | 1.021 | 119.7 | 102.3 | 168.3 | 105.7 | 4.20 | 2.01 |

^aThe bond distance are in angstroms; angles, in degrees; dipole moment, in Debye.

Table S2. NBO charge q_1 of BH_3NH_3 in external electric fields and NBO charge q_2 of BH_3NH_3 after external electric fields. The Laplacian of the electron density ($\nabla^2\rho(r)$) and electron density (ρ) of the bond critical points (BCPs).

| $E_{ext} \times 10^{-4}$ au | $q_1(\text{NH}_3)$ | $q_1(\text{BH}_3)$ | $q_2(\text{NH}_3)$ | $q_2(\text{BH}_3)$ | $\nabla^2\rho(\text{B-N})$ | $\nabla^2\rho(\text{B-H})$ | $\nabla^2\rho(\text{N-H})$ | $\rho(\text{B-N})$ | $\rho(\text{B-H})$ | $\rho(\text{N-H})$ |
|-----------------------------|--------------------|--------------------|--------------------|--------------------|----------------------------|----------------------------|----------------------------|--------------------|--------------------|--------------------|
| -767 | 0.706 | -0.706 | 0.318 | -0.318 | 0.124 | -0.053 | -1.809 | 0.134 | 0.140 | 0.310 |
| -750 | 0.618 | -0.618 | 0.322 | -0.322 | 0.152 | -0.050 | -1.811 | 0.138 | 0.140 | 0.312 |
| -700 | 0.529 | -0.529 | 0.325 | -0.325 | 0.198 | -0.050 | -1.818 | 0.141 | 0.142 | 0.316 |
| -650 | 0.490 | -0.490 | 0.327 | -0.327 | 0.235 | -0.053 | -1.823 | 0.141 | 0.145 | 0.319 |
| -600 | 0.472 | -0.472 | 0.330 | -0.330 | 0.267 | -0.056 | -1.826 | 0.139 | 0.147 | 0.323 |
| -550 | 0.457 | -0.457 | 0.330 | -0.330 | 0.297 | -0.060 | -1.828 | 0.137 | 0.150 | 0.325 |
| -519 | 0.449 | -0.449 | 0.329 | -0.329 | 0.314 | -0.063 | -1.828 | 0.136 | 0.151 | 0.327 |
| -500 | 0.444 | -0.444 | 0.329 | -0.329 | 0.324 | -0.065 | -1.828 | 0.135 | 0.152 | 0.328 |
| -450 | 0.430 | -0.430 | 0.328 | -0.328 | 0.350 | -0.070 | -1.826 | 0.133 | 0.154 | 0.330 |
| -400 | 0.419 | -0.419 | 0.327 | -0.327 | 0.374 | -0.075 | -1.824 | 0.130 | 0.156 | 0.332 |
| -350 | 0.406 | -0.406 | 0.326 | -0.326 | 0.128 | -0.081 | -1.820 | 0.396 | 0.158 | 0.334 |
| -300 | 0.393 | -0.393 | 0.325 | -0.325 | 0.308 | -0.087 | -1.816 | 0.125 | 0.160 | 0.336 |
| -250 | 0.380 | -0.380 | 0.324 | -0.324 | 0.435 | -0.093 | -1.811 | 0.122 | 0.161 | 0.337 |
| -200 | 0.370 | -0.370 | 0.323 | -0.323 | 0.452 | -0.099 | -1.805 | 0.119 | 0.163 | 0.339 |
| -150 | 0.356 | -0.356 | 0.322 | -0.322 | 0.466 | -0.106 | -1.798 | 0.115 | 0.165 | 0.340 |
| -100 | 0.344 | -0.344 | 0.321 | -0.321 | 0.479 | -0.113 | -1.791 | 0.112 | 0.166 | 0.341 |
| -50 | 0.331 | -0.331 | 0.319 | -0.319 | 0.487 | -0.121 | -1.785 | 0.108 | 0.168 | 0.342 |
| 0 | 0.316 | -0.316 | 0.316 | -0.316 | 0.494 | -0.128 | -1.775 | 0.103 | 0.169 | 0.342 |
| 50 | 0.300 | -0.300 | 0.313 | -0.313 | 0.494 | -0.137 | -1.770 | 0.099 | 0.170 | 0.343 |
| 100 | 0.284 | -0.284 | 0.309 | -0.309 | 0.495 | -0.146 | -1.763 | 0.094 | 0.172 | 0.343 |
| 150 | 0.267 | -0.267 | 0.303 | -0.303 | 0.487 | -0.155 | -1.757 | 0.088 | 0.173 | 0.344 |
| 200 | 0.248 | -0.248 | 0.295 | -0.295 | 0.469 | -0.167 | -1.751 | 0.081 | 0.174 | 0.344 |
| 250 | 0.225 | -0.225 | 0.284 | -0.284 | 0.436 | -0.180 | -1.748 | 0.073 | 0.176 | 0.344 |
| 300 | 0.191 | -0.191 | 0.262 | -0.262 | | -0.200 | -1.757 | | 0.178 | 0.344 |
| 320 | 0.161 | -0.161 | 0.237 | -0.237 | | -0.217 | -1.775 | | 0.180 | 0.344 |
| 321 | 0.005 | -0.005 | 0.018 | -0.018 | | -0.288 | -1.621 | | 0.186 | 0.335 |

Table S3. The fragment contribution to HOMO-2 of BH_3NH_3 molecule in a series of $E_{\text{ext}} \times 10^{-4}$ au.

| $E_{\text{ext}} \times 10^{-4}$ au | -767 | -750 | -700 | -650 | -600 | -550 | -519 | -500 | -450 | -400 | -350 | -300 | -250 |
|------------------------------------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| $\text{BH}_3\%$ [HOMO-2] | 41.27 | 42.28 | 44.05 | 45.69 | 47.34 | 49.07 | 50.18 | 50.88 | 52.72 | 54.58 | 56.42 | 58.2 | 59.91 |
| $\text{NH}_3\%$ [HOMO-2] | 58.73 | 57.72 | 55.95 | 54.31 | 52.65 | 50.92 | 49.82 | 49.13 | 47.28 | 45.42 | 43.58 | 41.8 | 40.09 |
| $E_{\text{ext}} \times 10^{-4}$ au | -200 | -150 | -100 | -50 | 0 | 50 | 100 | 150 | 200 | 250 | 300 | 320 | 321 |
| $\text{BH}_3\%$ [HOMO-2] | 61.5 | 62.97 | 64.3 | 65.33 | 66.37 | 67.09 | 67.52 | 67.61 | 67.11 | 65.66 | 60.97 | 53.57 | 3.03 |
| $\text{NH}_3\%$ [HOMO-2] | 38.5 | 37.03 | 35.7 | 34.66 | 33.63 | 32.91 | 32.48 | 32.39 | 32.89 | 34.34 | 39.03 | 46.43 | 96.97 |

Table S4. The optimized structural parameters of BH₃NH₃ dimers in different external electric fields^a.

| $E_{ext} \times 10^{-4}$ au | r(B-N) | H ₁ -B-H ₂ -H ₃ | H ₁ '-N-H ₂ '-H ₃ ' |
|-----------------------------|--------|--|--|
| -200 | 1.606 | 112.4 | 123.1 |
| -150 | 1.610 | 124.2 | 124.3 |
| -100 | 1.614 | 114.2 | 114.2 |
| -50 | 1.620 | 115.1 | 115.1 |
| 0 | 1.626 | 128.1 | 116.1 |
| 50 | 1.634 | 129.5 | 117.2 |
| 100 | 1.643 | 130.9 | 118.4 |
| 150 | 1.654 | 132.5 | 119.5 |
| 200 | 1.668 | 134.2 | 121.0 |

^aThe bond distance are in angstroms; angles, in degrees.

Optimized Cartesian Coordinates

BH₃NH₃ without external electric field

| | | | |
|---|-------------|-------------|-------------|
| B | 0.00000000 | 0.00000000 | -0.10573900 |
| N | 0.00000000 | 0.00000000 | 1.54374200 |
| H | 1.16752400 | 0.00000000 | -0.41674100 |
| H | -0.58376200 | -1.01110600 | -0.41674100 |
| H | -0.58376200 | 1.01110600 | -0.41674100 |
| H | -0.94926200 | 0.00000000 | 1.90888300 |
| H | 0.47463100 | 0.82208500 | 1.90888300 |
| H | 0.47463100 | -0.82208500 | 1.90888300 |

Table S5. Potential energy E of eclipsed forms of BH₃NH₃ in different external electric fields.

| | | | | | |
|-----------------------------|----------|----------|----------|----------|----------|
| $E_{ext} \times 10^{-4}$ au | -200 | -150 | -100 | -50 | 0 |
| E (au) | -82.6896 | -82.6754 | -82.6619 | -82.6490 | -82.6368 |
| $E_{ext} \times 10^{-4}$ au | 50 | 100 | 150 | 200 | |
| E (au) | -82.6253 | -82.6145 | -82.6043 | -82.5949 | |