Supporting Information

Tube-like α-Fe₂O₃@Ag/AgCl Heterostructure: Controllable Synthesis and Enhanced Plasmonic Photocatalytic Activity

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Figure S1. SEM images of α -Fe₂O₃@Ag SNTs with different Ag ions addition, (a) 0.1 mM, (b)

0.2 mM, (c) 0.4 mM and (d) 0.8 mM



Figure S2. SEM images of α-Fe₂O₃@Ag/AgCl SNTs with different FeCl₃ oxidant addition. (a) 0.5 mL (sample S1,), (b) 1.0 mL (sample S2), (c) 1.5 mL (sample S3), (d) 2.0 mL (sample S4)



Figure S3. SEM images of α -Fe₂O₃@Ag/AgCl SNTs which are produced by oxidation of the corresponding α -Fe₂O₃@Ag SNTs with different Ag ions addition in Figure S1. (a) 0.1 mM (sample S5), (b) 0.2 mM (sample S6), (c) 0.4 mM (sample S2) and (d) 0.8 mM (sample S7)



Figure S4. (a) UV-visible absorption spectra of α -Fe₂O₃ a), α -Fe₂O₃@Ag b) and α -

 $Fe_2O_3@Ag/AgCl$ (c, sample 2); (b) The $(\alpha hv)^2$ -hv curves of α - Fe_2O_3 a), α - $Fe_2O_3@Ag$ b) and α -

Fe₂O₃@Ag/AgCl c).