Electronic Supplementary Material (ESI) for RSC Advances. This journal is © The Royal Society of Chemistry 2015

Supplementary Material (ESI) for RSC Advances

## Remove of Chlorpyrifos from Waste Water by Wheat Straw-Derived Biochar Synthesized through Oxygen-Limited Method

Peifang Wang\*, Yayun Yin, Yong Guo\*, Chao Wang

Key Laboratory of Integrated Regulation and Resource Development on Shallow Lakes, Ministry of Education, College of Environment, Hohai University, P.R. China. E-mail: <u>pfwang2005@hhu.edu.cn</u>; E-mail: <u>guoyong@hhu.edu.cn</u>;

## Regents

Wheat straw was acquired from a farm of LiuHe district, Nanjing City, China. Chlorpyrifos (HPLC grade) and methanol (chromatographic grade) were purchased from the TEDIA Company, America. Hydrochloric acid (analytical grade), sodium hydrate (analytical grade) and calcium chloride (analytical grade) were purchased from the Sinopharm Chemical Reagent Limited Corporation, P.R. China.



Fig.S1 FTIR spectra of WS250-WS750 samples.



Fig.S2 The standard curve of chlorpyrifos in water.



Fig. S3 TEM image of WS750 sample.



Fig. S4. EDS analysis for the ash acquired through heating WS750 sample at 800 °C under air condition for two hours.



Fig. S5 Kinetic adsorption datas that were fitted by (a) pseudo-first-order model, and (b) pseudo-second-order model, respectively.

| F F F F F F F F F F F F F F F F F F F |                                      |                    |
|---------------------------------------|--------------------------------------|--------------------|
| Model                                 | Parameters                           | Values             |
| Pseudo-first-order                    | $k_1, h^{-1}$                        | $0.024 \pm 0.004$  |
|                                       | q <sub>e</sub> , mg/g                | $11.080 \pm 0.560$ |
|                                       | $\mathbb{R}^2$                       | 0.841              |
|                                       |                                      |                    |
| Pseudo-second-order                   | $k_2$ , $kg \cdot (mg \cdot h)^{-1}$ | $0.016 \pm 0.002$  |
|                                       | q <sub>e</sub> , mg/g                | 12.195±0.593       |
|                                       | R <sup>2</sup>                       | 0.991              |
|                                       |                                      |                    |

Table S1 Kinetic parameters for chlorpyrifos adsorption by WS750.



**InC**<sub>e</sub> Fig S6 Adsorption isotherm datas that were fitted by Freundlich model.

Table S2 Adsorption isotherm parameters for chlorpyrifos adsorption by WS750.

| Model      | Parameters     | Values             |
|------------|----------------|--------------------|
| Freundlich | k              | 30.265±4.852       |
|            | 1/n            | $0.413 \pm 0.0187$ |
|            | $\mathbb{R}^2$ | 0.968              |



Fig. S7 Recycle experiments of chlorpyrifos adsorption by WS750. The concentration of chlorpyrifos is 0.80 mg/L.