## SUPPLEMENTARY INFORMATION

## Influence of electron acceptors on the kinetics of metoprolol photocatalytic degradation in TiO<sub>2</sub> suspension. A combined experimental and theoretical study

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**Fig. S1.** Base peak chromatograms of metoprolol (MET) in presence of TiO<sub>2</sub> Wackherr or Degussa P25, with different electron acceptors, in positive mode.







**Fig. S3.** Positive mode  $MS^2$  spectra of MET.



**Fig. S4.** Positive mode MS<sup>1</sup> spectra of degradation product P1.



**Fig. S5.** Positive mode  $MS^2$  spectra of degradation product P1.



**Fig. S6.** Positive mode MS<sup>1</sup> spectra of degradation product P2.



**Fig. S7.** Positive mode  $MS^2$  spectra of degradation product P2.



**Fig. S8.** Positive mode MS<sup>1</sup> spectra of degradation product P3.



**Fig. S9.** Positive mode  $MS^2$  spectra of degradation product P3.



**Fig. S10.** Positive mode  $MS^1$  spectra of degradation product P4a.



**Fig. S11.** Positive mode  $MS^1$  spectra of degradation product P4b.



**Fig. S12.** Positive mode  $MS^1$  spectra of degradation product P5.



**Fig. S13.** Positive mode  $MS^1$  spectra of degradation product P6.



**Fig. S14.** Positive mode composite MS<sup>2</sup> spectra of degradation products P4a, P4b, P5 and P6, respectively.



**Fig. S15.** Positive mode  $MS^1$  spectra of degradation product P7.



**Fig. S16.** Positive mode  $MS^2$  spectra of degradation product P7.



**Fig. S17.** Positive mode  $MS^1$  spectra of degradation product P8.



**Fig. S18.** Positive mode  $MS^2$  spectra of degradation product P8.



**Fig. S19.** Positive mode  $MS^1$  spectra of degradation product P9.



**Fig. S20.** Positive mode  $MS^2$  spectra of degradation product P9.



**Fig. S21.** Positive mode  $MS^1$  spectra of degradation product P10.



**Fig. S22.** Positive mode  $MS^2$  spectra of degradation product P10.

Compound	Precursor ion [M+H] <sup>+</sup>	Molecular formula	Name of compound	V <sub>col</sub> (V)	Product ions m/z (relative abundance)
MET	268	C <sub>15</sub> H <sub>25</sub> NO <sub>3</sub>	1-[4-(2 methoxyethyl)phenoxy] -3-(propan-2-ylamino)propan-2-ol	10 20 30 40	226(9), 191(76), 176(13), 159(62), 147(11), 133(56), 121(75), 116(100), 98(45), 74(54), 72(62), 56(20) 191(29), 159(59), 148(11), 133(78), 121(91), 116(68), 105(20), 98(43), 74(86), 72(100), 58(16), 56(87) 133(57), 121(42), 116(23), 105(44), 103(21), 74(59), 72(40), 56(100), 133(28), 121(18), 105(20), 103(47), 91(48), 79(26), 77(47), 74(14), 56(100)
P1	134	$C_6H_{15}NO_2$	3-(propan-2-ylamino)propane-1,2-diol	10 20 30 40	134(35), 116(28), 92(51), 74(100), 72(32), 56(46), 74(26), 72(9), 56(100), 56(100) 56(100)
P2*	254	$C_{13}H_{19}NO_4$	Hydroxy derivative 4-[2-hydroxy-3-(propan- 2- ylamino)propoxy]benzaldehyde	10 20 30 40	212(100), 177(24), 133(14), 105(4), 72(8), 56(27) 177(34), 133(38), 105(70), 74(17), 72(28), 56(100) 121(9), 105(19), 79(16), 74(10), 56(100) 121(30), 79(28), 77(24), 65(100), 56(42)
Р3	254	$C_{14}H_{23}NO_3$	1-[4-(1-hydroxyethyl)phenoxy]-3-(propan-2- ylamino)propan-2-ol	10 20 30 40	177(100), 116(55), 74(63) 177(8), 159(100), 131(74), 116(8), 72(42), 56(5) 131(100), 105(39), 57(39) 91(59), 56(100)
P4a	282	C <sub>15</sub> H <sub>23</sub> NO <sub>4</sub>	methyl {4-[2-hydroxy-3-(propan-2- ylamino)propoxy]phenyl}acetate; or 1-{4-[2-hydroxy-3-(propan-2- ylamino)propoxy]phenyl}-2-methoxyethanone	composite	240(9), 205(40), 159(20), 149(11), 145(9), 133(100), 116(29), 91(7), 74(8), 56(44)
P4b	282	C <sub>15</sub> H <sub>23</sub> NO <sub>4</sub>	methyl {4-[2-hydroxy-3-(propan-2- ylamino)propoxy]phenyl}acetate; or 1-{4-[2-hydroxy-3-(propan-2- ylamino)propoxy]phenyl}-2-methoxyethanone	composite	240(56), 205(9), 159(74), 145(73), 133(84), 116(45), 103(47), 74(65), 56(100)
Р5	282	C <sub>15</sub> H <sub>23</sub> NO <sub>4</sub>	2-hydroxy-3-[4-(2-methoxyethyl)phenoxy]-N-(propan-2- yl)propanamide	composite	205 (5), 177(19), 159(64), 145(25), 121(20), 103(20), 98(16), 91(40), 74(64), 72(25), 58(26), 56(100)
Р6	282	C <sub>15</sub> H <sub>23</sub> NO <sub>4</sub>	4-(2-methoxyethyl)phenyl 2-hydroxy-3-(propan-2- ylamino)propanoate	composite	264(18), 205(6), 159(53), 145 (16), 133(15) 121(17), 103(13), 116(33), 98(20), 91(33), 74(33), 72(22), 56(100)
Р7	300	C <sub>15</sub> H <sub>25</sub> NO <sub>5</sub>	1-{4-[2-hydroxy-3-(propan-2-ylamino)propoxy]phenyl}-2- methoxyethane-1,2-diol	10 20 30 40	238(100) 238(100), 196(12), 74(7) 196(30), 105(20), 74(70), 72(100) 105(65), 56(100)
P8**	300	C <sub>15</sub> H <sub>25</sub> NO <sub>5</sub>	Dihydroxy derivate 1-[4-(2 methoxyethyl)phenoxy] -3-(propan-2-ylamino)propan-2-ol	10 20 30 40	282(100) 282(100), 116(14) 159(32), 74(100) 159(21), 133(100), 74(19), 72(30), 56(17)
P9***	274	C <sub>12</sub> H <sub>19</sub> NO <sub>6</sub>	Trihydroxy derivate 1-amino-3-[4-(2- methoxyethyl)phenoxy]propan-2-ol	10 20 30 40	116(69), 98(100), 56(13) 116(73), 98(47), 74(23), 56(100) 116(18), 56(100) 56(100)
P10****	316	C <sub>15</sub> H <sub>25</sub> NO <sub>6</sub>	Trihydroxy derivate 1-[4-(2-methoxyethyl)phenoxy] -3-(propan-2-ylamino)propan-2-ol	10 20 30 40	116(32), 98(100) 151(17), 116(80), 98(50), 74(43), 56(100) 74(15), 72(32), 56(100) 67(21), 56(100)

Table S1. Proposed structures of intermediates for the photocatalytic degradation of MET in the presence of TiO<sub>2</sub> Wackherr or Degussa P25, with different electron acceptors

P1, P3–P7 were identified in all cases. P2<sup>\*</sup> was identified, in all cases, only with TiO<sub>2</sub> Wackherr P8<sup>\*\*</sup> was identified in all cases except TiO<sub>2</sub> Wackherr/O<sub>2</sub>/H<sub>2</sub>O<sub>2</sub> P9<sup>\*\*\*</sup> was identified in all cases except TiO<sub>2</sub> Wackherr/O<sub>2</sub> and Degussa P25/O<sub>2</sub>/KBrO<sub>3</sub>