

Supporting Information (SI)

A novel biosensor for silver(I) ion detection based on nanoporous gold and duplex-like DNA scaffolds with anionic intercalator

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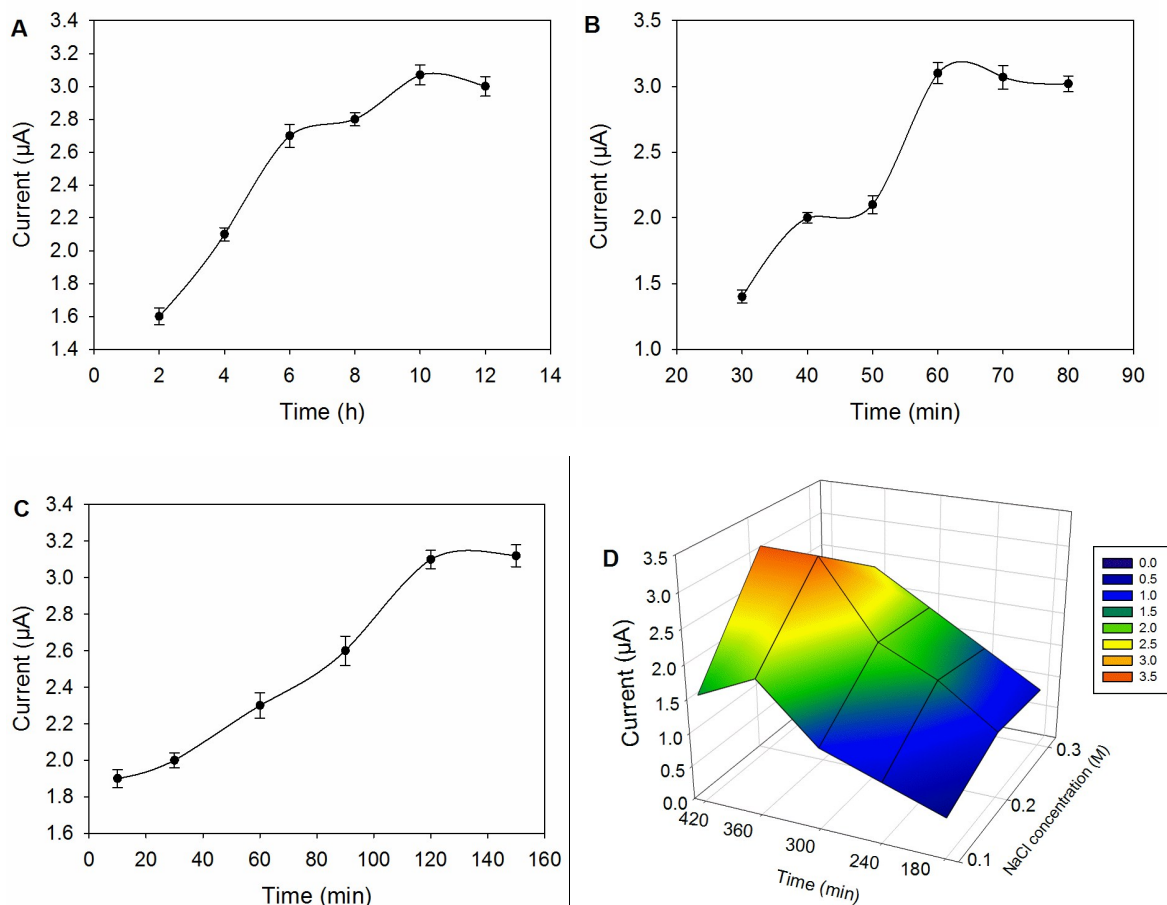


Fig. S-1. Optimization of experimental conditions: (A) Effect of the self-assembly time of capture probe; (B) Effect of the hybridization time of (S2+S3); (C) Effect of the time-course of the Ag⁺ hybridized with C bases; (D) Effect of the salt concentration and immersing time to AQDS intercalating; All tested electrodes were fabricated by immobilizing 2 µL capture probe on electrodes surfaces at 4 °C. Error bars indicate standard deviations from three replicative tests.

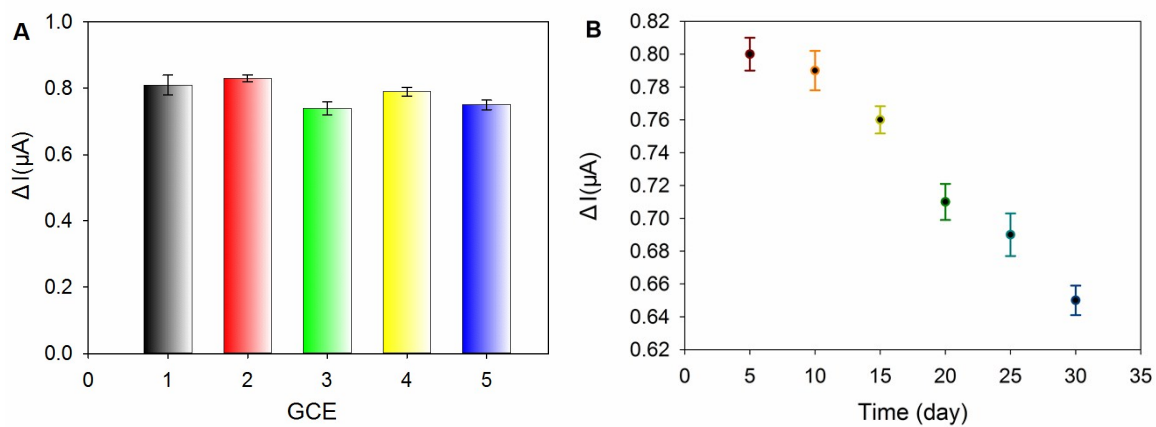


Fig. S-2. (A) Five different GCEs constructed by the same procedure on response of biosensor for Ag^+ (10^{-8} M), (B) Stability of the biosensor in detecting 10^{-8} M Ag^+ solution. Error bars indicate standard deviations from three replicative tests.

Table S–1 Equivalent circuit element values for the films in Fig. 3C^a

Equivalent circuit elements				
electrode	$R_{CT}(\Omega)$	$R_S (\Omega)$	$C_{dl}(\mu F)$	$W(m\Omega/s^{1/2})$
GCE	760.0 (± 3)	154.7(± 4.5)	4.024(± 0.5)	1.504(± 0.052)
GCE/NPG	19.9(± 2)	140.5(± 1.5)	3.368(± 0.38)	1.580(± 0.064)

^aThe values in parentheses represent the standard deviations from at least 3 electrode measurements.

Table S–2 Ag⁺ ions in environmental samples determined by this sensor and atomic absorption spectroscopy (AAS).

Sample number of Taozi Lake	Addition concentration of Ag ⁺ (nM)	Biosensor (mean ^a ± SD ^b) (nM)	atomic absorption spectroscopy (AAS) (mean ^a ± SD ^b) (nM)	Relative Standard Deviation (%)
1	65	64.12±3.5	66.89±4.3	2.99
2	5	5.15±0.3	4.91±0.21	3.37
3	20	21.35±1.1	21.92±1.5	1.86
4	150	153.26±4.2	157.03±3.9	1.72

^a An average of three replicate measurement. ^bSD =standard deviation