## Facile Synthesis of Structure-controllable, N-doped Graphene

## Aerogels and Their Application in Supercapacitors

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**Preparation of graphene oxide:** Graphene oxide was synthesized from natural graphite by a modified Hummers method. 3 g graphite and 1.5 g sodium nitrate were mixed with 90 mL of concentrated sulfuric acid (98%) in a 500 mL flask. The mixture was stirred for 1 h in an ice bath. 9 g potassium permanganate was added to the suspension under vigorous stirring. The rate of addition was carefully controlled to keep the reaction temperature below 20 °C. After removal of ice bath, the mixture was stirred at 35 °C for 5 h. Additional 6 g potassium permanganate was added in the mixture, and the reaction was stirred for 2 h at 35 °C. Then, the reaction mixture was poured into 400 mL deionized water with vigorous agitation. The diluted suspension was stirred for 30 min. At the end, 10 mL of  $H_2O_2$  (30%) were added to the mixture. The mixture was washed with 5% HCl and deionized water, and dried at 60°C to obtain GO powder.



Figure S1. XRD patterns of graphite, GO and NGA-3 (a); Raman spectra of NGOA and NGAs (b).



Figure S2.  $N_2$  adsorption/desorption isotherms of all the NGAs.



Figure S3. Nyquist impedance plots of NGA-1 and NGA-3.



Figure S4. Cyclic performance of NGA-3.



Figure S5. The two-electrode electrochemical performance of NGA-3: CV curves (a), galvanostatic charge/discharge curves (b), cyclic performance (c) and Nyquist impedance plots (d).