

Supplementary Information

Influence of the alkenyl structures on the epoxidation of unsaturated fatty acid methyl esters and vegetable oils

Supplementary figures

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质量 : 99
ID : 9-Octadecenoic acid, methyl ester, (E)-

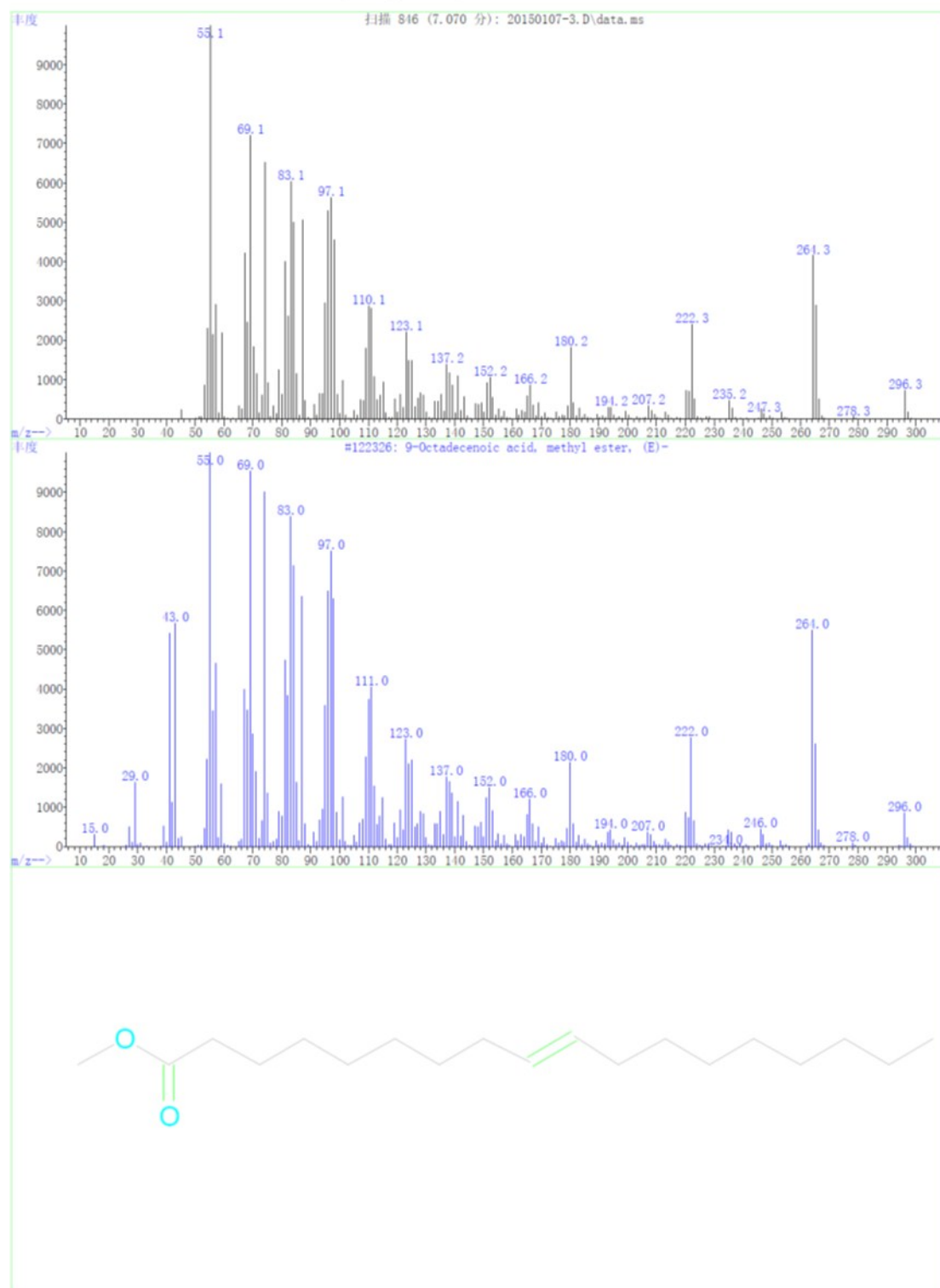


Fig. S1 GC-MS results of 1 in the trace of the epoxidation of methyl oleate

已检索库: C:\Database\NIST05a.L

质量 : 90

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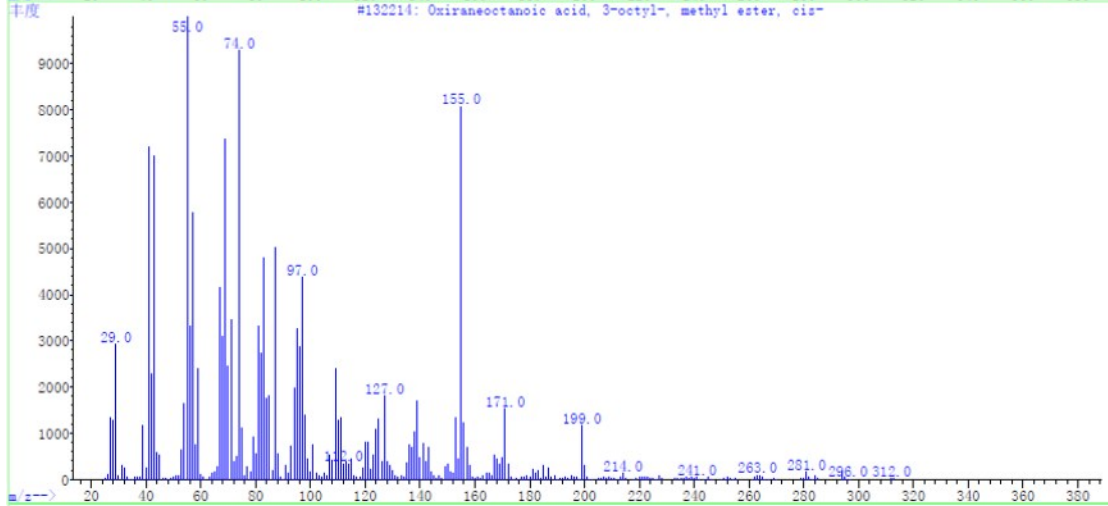
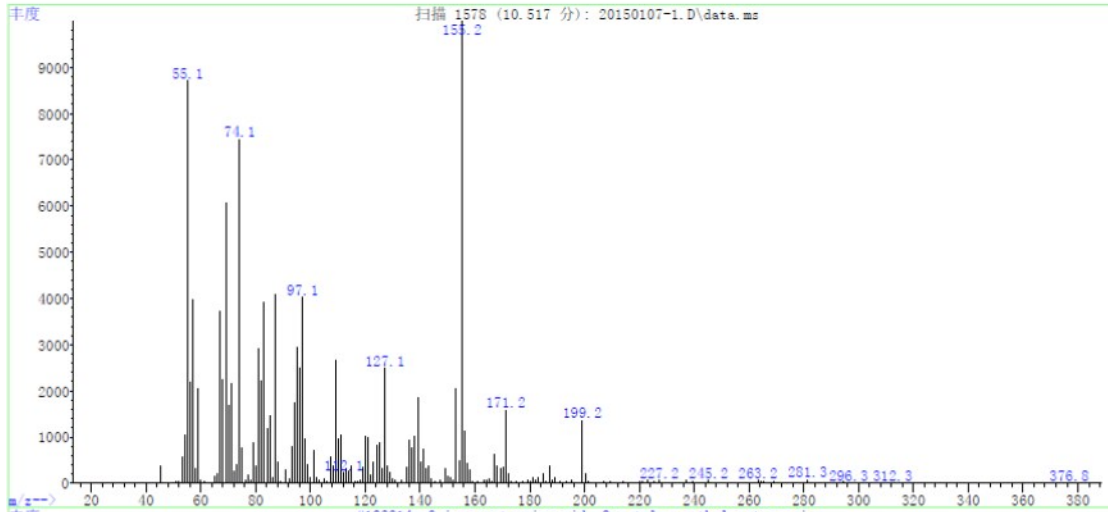


Fig. S2 GC-MS results of 2 in the trace of the epoxidation of methyl oleate

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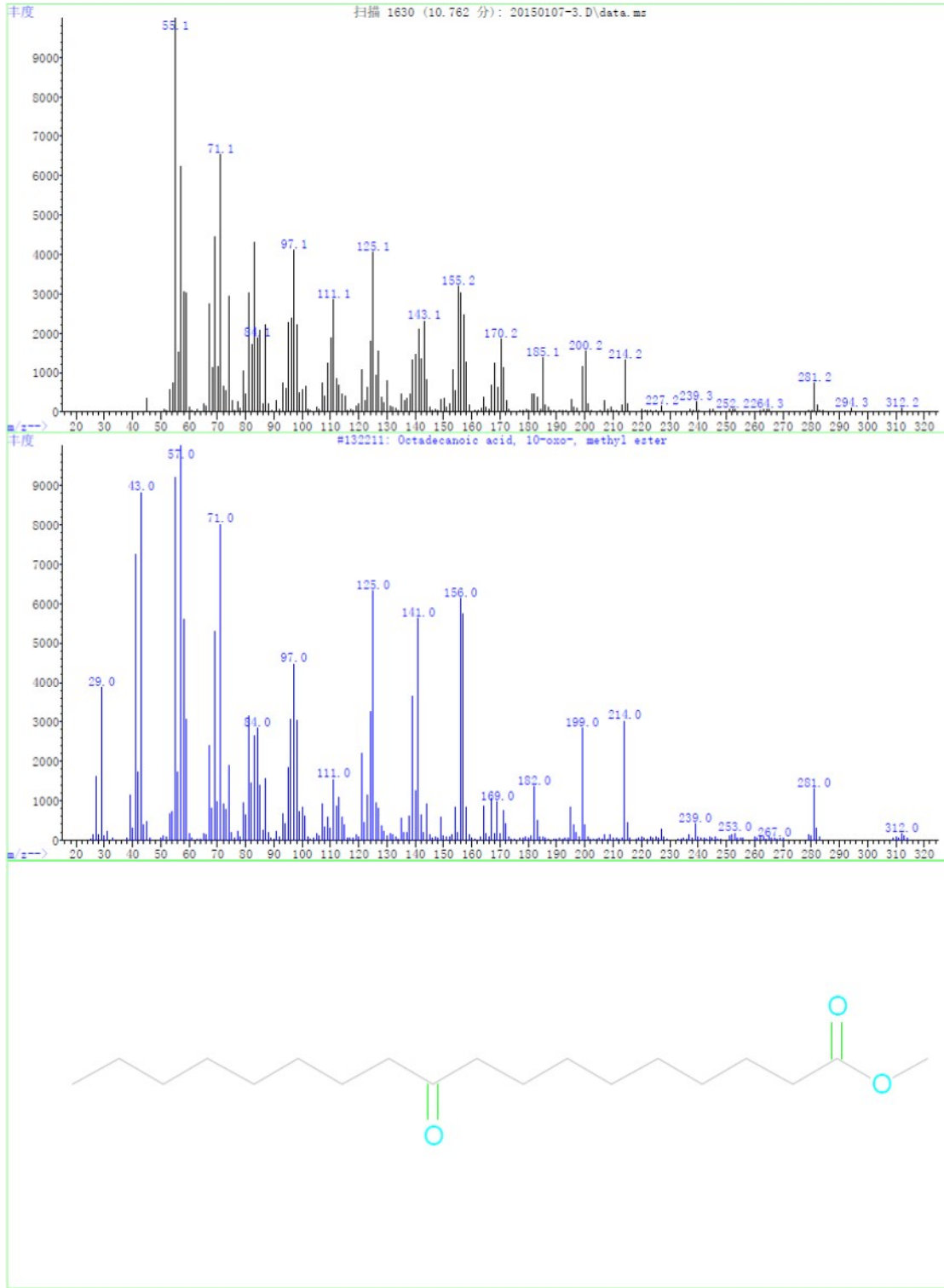


Fig. S3 GC-MS results of 3 for the epoxidation of methyl oleate

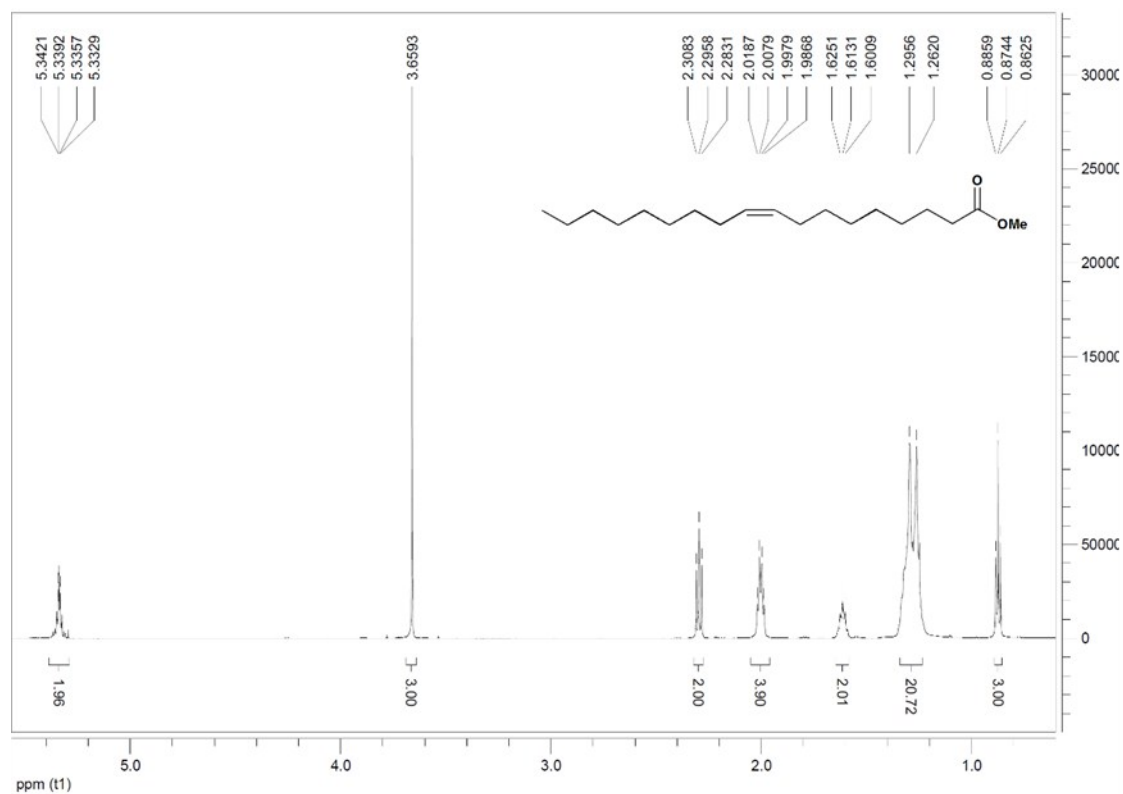


Fig. S4 $^1\text{H-NMR}$ spectra of 1 for the epoxidation of methyl oleate

$^1\text{H-NMR}$ (600MHz, CDCl_3): δ (ppm) = 5.34(m, $-\text{CH}=\text{CH}-$), 3.66(s, $-\text{OCH}_3$), 2.30(t, $J=7.56\text{Hz}$, $-\text{CH}_2-\text{CO}-$), 2.02(m, $-\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2-$), 1.62(t, $J=7.26\text{Hz}$, $-\text{CH}_2-\text{CH}_3$), 1.29(m, $-(\text{CH}_2)_n-$) and 0.87(t, $J=7.02\text{Hz}$, $-\text{CH}_3$).

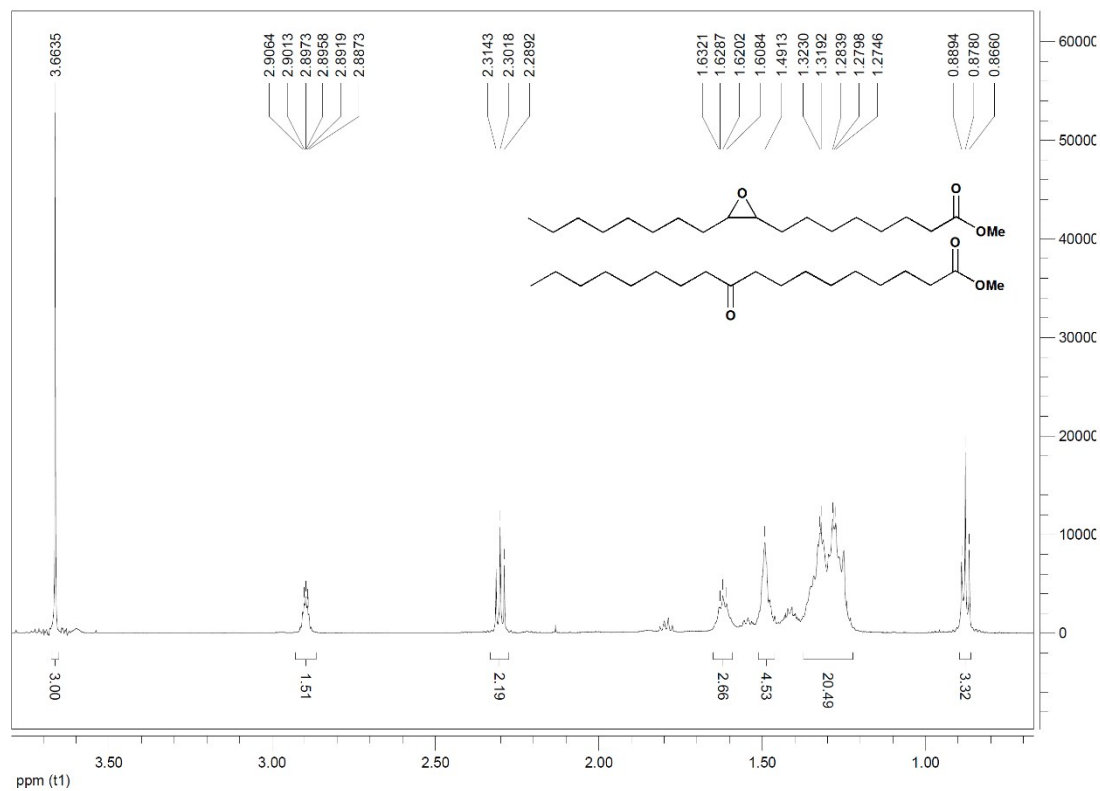


Fig. S5 $^1\text{H-NMR}$ spectra of 2 from epoxidation of methyl oleate

$^1\text{H-NMR}$ (600MHz, CDCl_3): δ (ppm) = 3.66(s, $-\text{OCH}_3$), 2.89(m, $-\text{CHOCH-/-CH}_2\text{-CO-CH}_2-$), 2.30(t, $J=7.53\text{Hz}$, $-\text{CH}_2\text{-CO-}$), 1.49(m, $-\text{CH}_2\text{-CH=CH-CH}_2-$), 1.62(m, $-\text{CH}_2\text{-CH}_3$), 1.29(m, $-(\text{CH}_2)_n-$) and 0.87(t, $J=7.02\text{Hz}$, $-\text{CH}_3$).

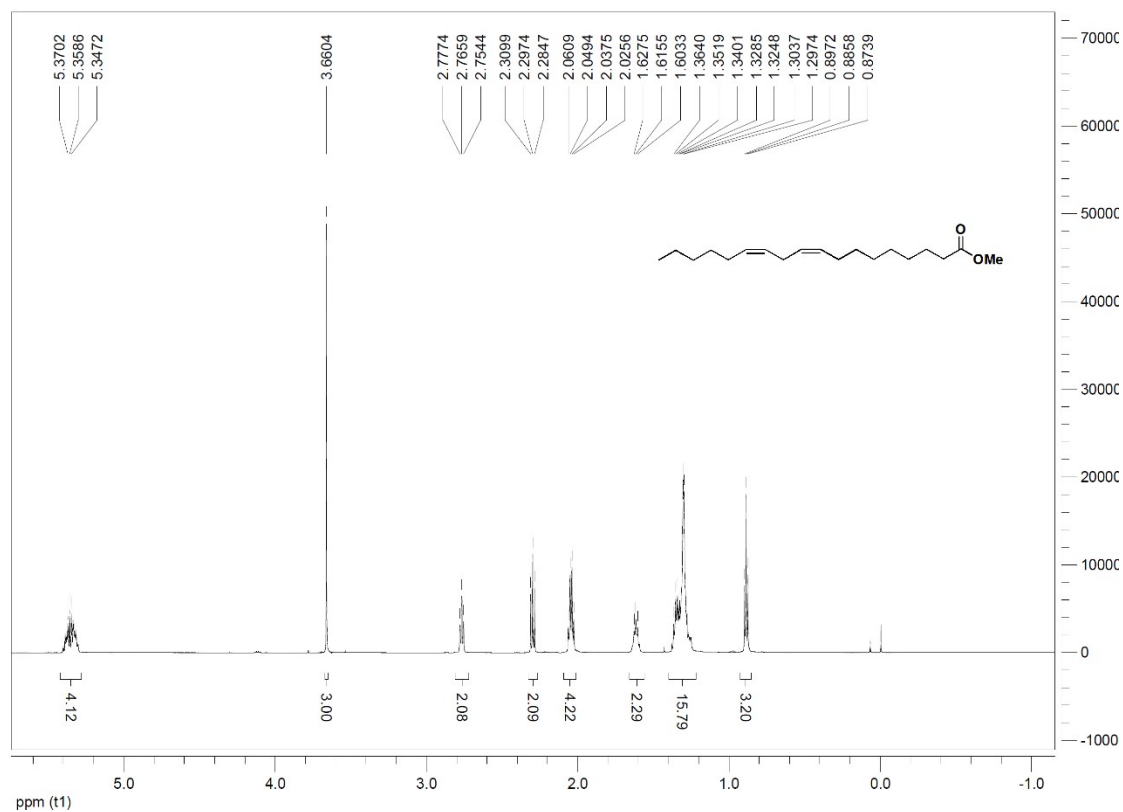


Fig. S6 $^1\text{H-NMR}$ spectra of 4 for the epoxidation of methyl linoleate

$^1\text{H-NMR}$ (600MHz, CDCl_3): δ (ppm) = 5.35(m, $-\text{CH=CH-}$), 3.66(s, $-\text{OCH}_3$), 2.76(t, $J=6.90\text{Hz}$, $-\text{CH=CH-CH}_2\text{-CH=CH-}$), 2.30(t, $J=7.56\text{Hz}$, $-\text{CH}_2\text{-CO-}$), 2.07(m, $-\text{CH}_2\text{-CH=CH-CH}_2\text{-CH=CH-CH}_2-$) 1.62(t, $J=7.26\text{Hz}$, $-\text{CH}_2\text{-CH}_3$), 1.30(m, $-(\text{CH}_2)_n-$) and 0.87(t, $J=6.99\text{Hz}$, $-\text{CH}_3$).

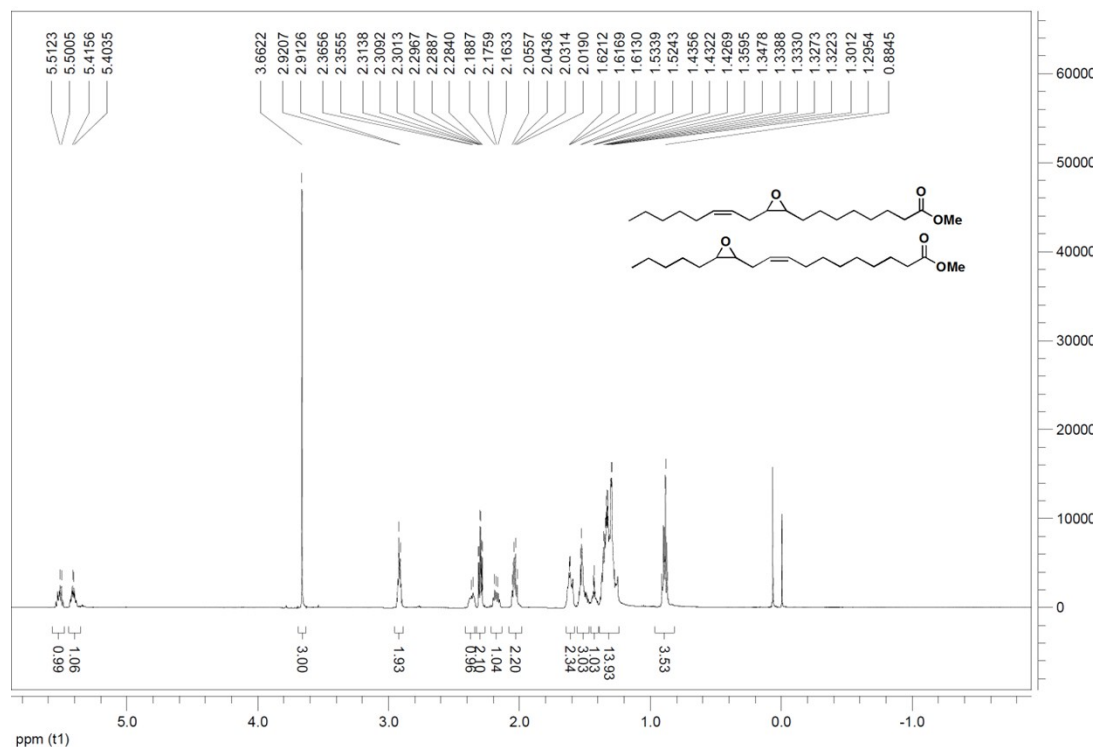


Fig. S7 $^1\text{H-NMR}$ spectra of **5** for the epoxidation of methyl linoleate

$^1\text{H-NMR}$ (600MHz, CDCl_3): δ (ppm) = 5.40(m, $-\text{CH}=\text{CH}-$), 3.66(s, $-\text{OCH}_3$), 2.91(m, $-\text{CH-O-CH-}$), 2.35(m, $-\text{CH}_2-\text{CH-O-CH-}$), 2.29(m, $-\text{CH}_2-\text{CO-}$), 2.05(m, $-\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2-\text{CH-O-CH-CH}_2-$), 1.61(m, $-\text{CH}_2-\text{CH}_3$), 1.30(m, $-(\text{CH}_2)_n-$) and 0.87(m, $-\text{CH}_3$).

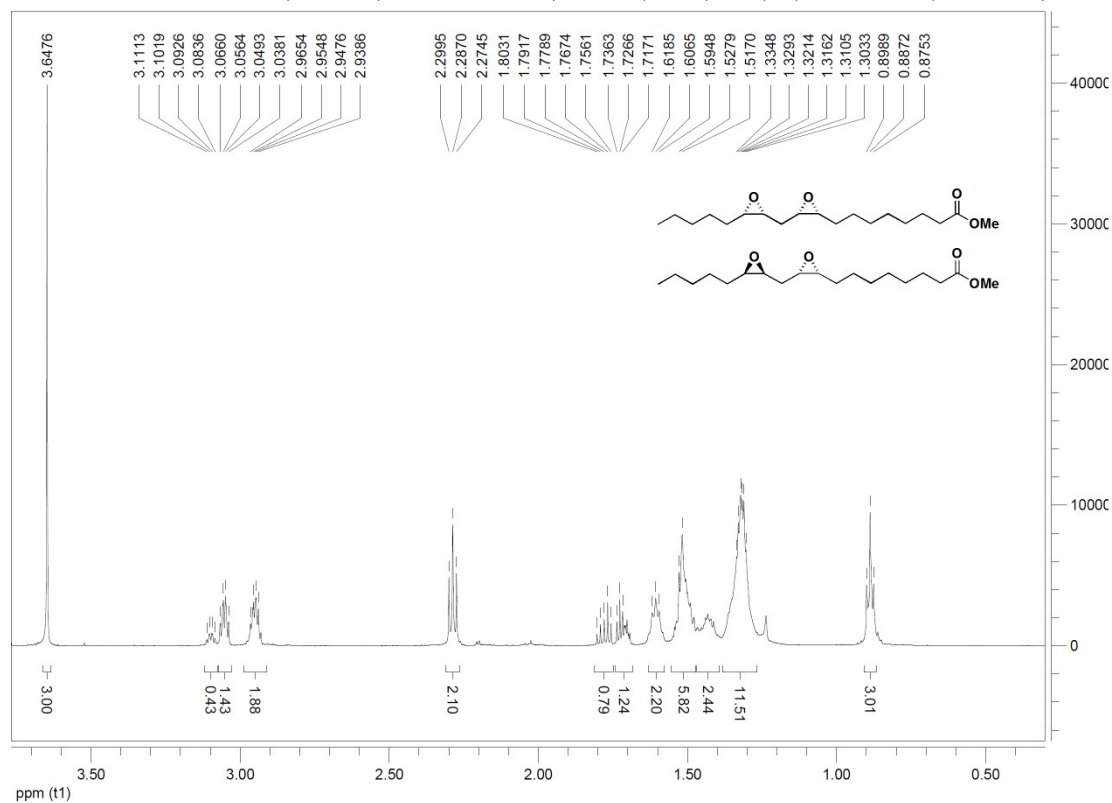


Fig. S8 $^1\text{H-NMR}$ spectra of **5** for the epoxidation of methyl linoleate

$^1\text{H-NMR}$ (600MHz, CDCl_3): δ (ppm) = 3.65(s, $-\text{OCH}_3$), 2.96-3.09(m, $-\text{CH-O-CH-}$), 2.29(t, $J=7.50\text{Hz}$, $-\text{CH}_2\text{-CO-}$), 1.60(m, $-\text{CH}_2\text{-CH-O-CH-CH}_2\text{-CH-O-CH-CH}_2\text{-}$), 1.31(m, $-(\text{CH}_2)_n\text{-}$) and 0.88(t, $J=7.08\text{Hz}$, $-\text{CH}_3$).

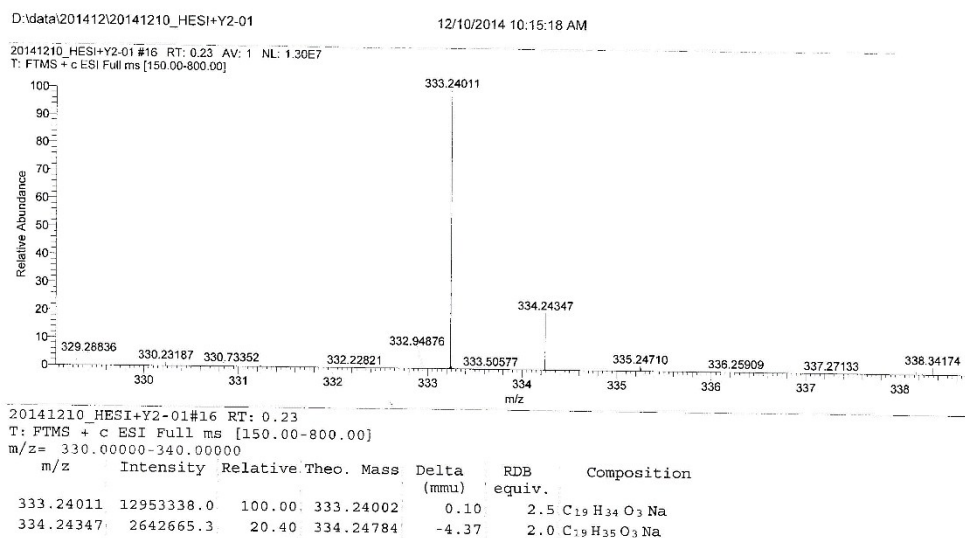


Fig. S9 HRMS spectrum of 5 for the epoxidation of methyl linoleate

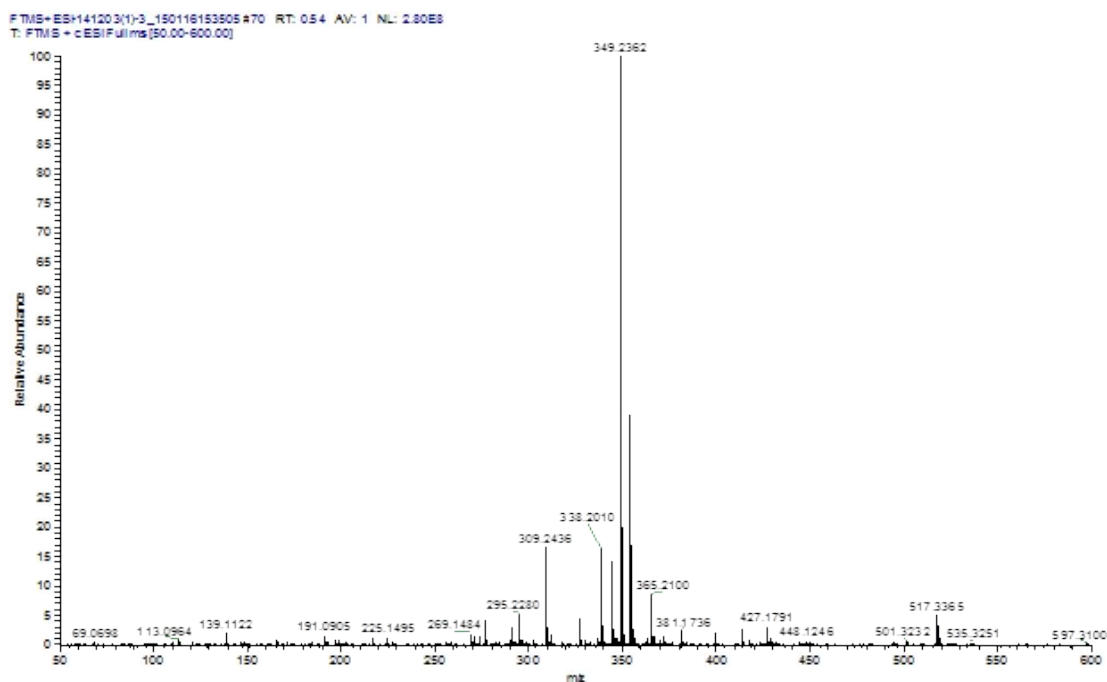


Fig. S10 HRMS spectrum of 6 for the epoxidation of methyl linoleate

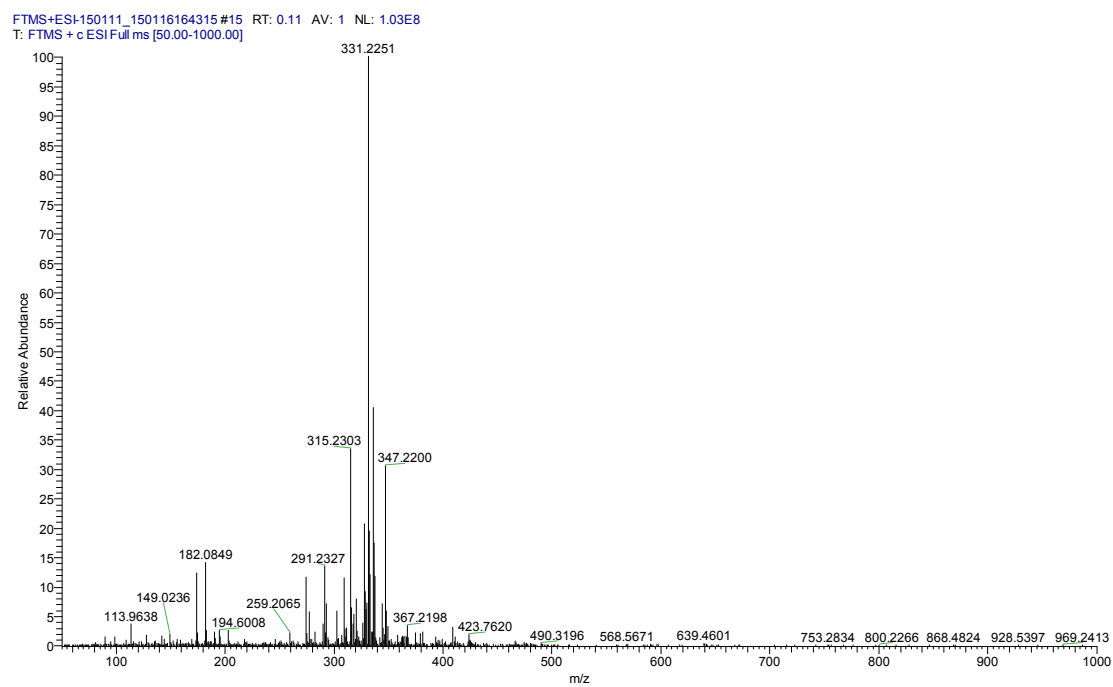


Fig. S11 HRMS spectrum for the epoxidation of methyl linolenate