

Fig. S1: ^{31}P MAS NMR spectra of representative samples

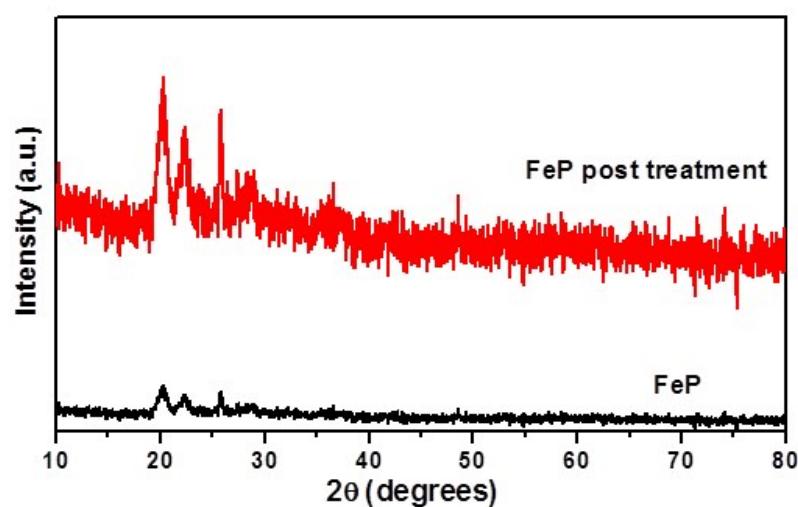


Fig. S2: XRD patterns of FeP and FeP-post treatment samples

Table S1: Chemical composition of samples

Catalyst	Elemental (ICP-MS) analysis (molar %)		
	Fe	P	Mo
FeP	36.9	20.4	0.0
1Mo-FeP	36.7	20.3	0.5
2Mo-FeP	36.4	20.4	1.5
3Mo-FeP	36.2	20.2	2.3
4Mo-FeP	36.1	20.1	3.5
5Mo-FeP	35.6	20.0	4.4

Table S2: Activation energies (E_a) and TOF for the catalytic reaction conducted over different catalysts

Catalyst	Ea (kJ mol ⁻¹)	TOF X 10 ² (h ⁻¹)
FeP	78.6	9.5
1Mo-FeP	69.3	15.3
2Mo-FeP	66.6	38.6
3Mo-FeP	62.3	43.5
4Mo-FeP	59.8	63.0
5Mo-FeP	50.5	93.3
FeMoP	64.4	31.2
Fe ₂ O ₃	71.2	30.6
MoO ₃	74.6	3.2

Benzyl chloride: benzne stoichiometric ratio = 1:15 and 50 mg of catalyst

Table S3: Comparison of Mo-FePO₄ catalyst with other reported iron based catalysts

Authors	Catalyst	Reaction conditions	Con. of benzyl chloride (%)	Remarks
Arafat et al.	Fe ₂ O ₃ /MCM-41	T=80°C, Time=10 min.	95	Leaching of active phase
Sun et al.	Fe ₂ O ₃ /SBA-15	T=60°C, Time=20 min.	100	Role support is not clear
Leng et al.	Fe ₂ O ₃ /H-modernite	T=70°C, Time=30 min.	100	Poor stability and weak acidity
Shinde & Sawant	NiFe ₂ O ₄	T=70°C, Time=70 min.	100	Low surface area and pore size
Koyande et al.	Fe ₂ O ₃ /sulfated ZrO ₂	T=80°C, Time=30 min.	80	Usage of strong mineral acid
Choudary & Jana	Fe ₂ O ₃ /H-ZSM-5	T=80°C, Time=8.5 min.	90	A significant reaction induction period was observed.
Bachari et al	Fe-HMS	T=75°C, Time=245 min.	50	Low activity, high selectivity
Ali et al	Fe ₂ O ₃ /ZrO ₂ (nano)	T=75°C, Time=480 min.	91	Stable activity and reusability
Present work	MoO _x -FePO ₄	T=80°C, Time=45 min.	100	Stable activity and reusability