Supporting Information

New a-Glucosidase Inhibitors from a Marine Sponge-derived Fungus

Aspergillus sp. OUCMDZ-1583

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Anti-influenza A Virus (H1N1) Assay

The antiviral activity against H1N1 was evaluated by the CPE inhibition assay.⁶ Confluent MDCK cell monolayers were firstly incubated with influenza virus (A/Puerto Rico/8/34 (H1N1), PR/8) at 37 °C for 1 h. After removing the virus dilution, cells were maintained in infecting media (RPMI 1640, 4 μ g/mL of trypsin) containing different concentrations of test compounds at 37 °C. After 48 h incubation at 37 °C, the cells were fixed with 100 μ L of 4% formaldehyde for 20 min at room temperature. After removal of the formaldehyde, the cells were stained with 0.1% crystal violet for 30 min. The plates were washed and dried, and the intensity of crystal violet staining for each well was measured in a microplate reader (Bio-Rad, USA) at 570 nm. The IC₅₀ value was calculated as the compound concentration required inhibiting influenza virus yield at 48 h post-infection by 50%. Ribavirin was used as the positive control with an IC₅₀ value of 137.3±0.4 μ M.

a-Glucosidase Inhibitory Effect Assay

The inhibitory effects were assayed as described preciously.²² The sample was dissolved in sodium phosphate buffer (PBS, pH 6.8) at three concentrations. A volume of 10 μ L of the sample solution, 20 μ L of PBS and 20 μ L of 2.5 *m*M *p*-nitrophenyl- α -D-glucopyranoside (PNPG) solution (in phosphate buffer) were mixed in a 96-well microplate and incubated at 37 °C for 5 min. A volume of 10 μ L of α -glucosidase diluted to 0.2 U/mL by 0.01 M PBS was then added to each well. After incubating at 37 °C for 15 min, the absorbance at 405 nm was recorded by a Spectra max 190 micro plate reader (Molecular Devices Inc.). The blank was prepared by adding phosphate buffer instead of the α -glucosidase and acarbose was used as a positive control. Blank readings (no enzyme) were subtracted from each well and results were compared to the control. The inhibition (%) was calculated as $[1-(OD_{sample}/OD_{control})] \times 100 \%$. The IC₅₀ value was calculated as the compound concentration that is required for 50% inhibition and the IC₅₀ value of the acarbose was 0.95 *m*M.

Antimicrobial Assays

The antimicrobial activities against *Escherichia coli*, *Enterobacter aerogenes*, *Pseudomonas aeruginosa*, *Bacillus subtilis*, *Staphylococcus aureus*, and *Candida albicans* were evaluated by an agar dilution method.²³ The tested strains were cultivated in LB agar plates for bacteria and in YPD agar plates for *C. albicans* at 37 °C. Compounds **1–23** and positive controls (ciprofloxacin lactate for bacteria and ketoconazole for *C. albicans*) were dissolved in MeOH at different concentrations from 100 to 0.05 μ g/mL by the continuous 2-fold dilution methods. A 10 μ L quantity of test solution was

absorbed by a paper disk (5 mm diameter) and placed on the assay plates. After 24 h incubation, zones of inhibition (mm in diameter) were recorded. The minimum inhibitory concentrations (MICs) were defined as the lowest concentration at which no microbial growth could be observed.

Cytotoxic Assays

Cytotoxicity was assayed by the MTT²⁴ and CCK-8 methods.²⁵ In the MTT assay, A549 and MCF-7 cell lines was grown in RPMI-1640 supplemented with 10% FBS under a humidified atmosphere of 5% CO₂ and 95% air at 37 °C, respectively. Cell suspension, 100 μ L, at a density of 3 × 10⁴ cell mL⁻¹ was plated in 96-well microtiter plates, allowed to attach overnight, and then exposed to varying concentrations (10⁻⁵-10⁻¹² M) of compounds for 72 h. The MTT solution (20 μ L, 5 mg/mL in IPMI-1640 medium) was then added to each well and incubated for 4 h. Old medium containing MTT was then gently replaced by DMSO and pipetted to dissolve any formazan crystals formed. Absorbance was then determined on a Spectra Max Plus plate reader at 540 nm. In the CCK-8 assay, K562 cell line was grown in RPMI-1640 supplemented with 10% FBS under a humidified atmosphere of 5% CO₂ and 95% air at 37 °C. Cell suspension, 100 μ L, at a density of 5 × 10⁵ cell mL⁻¹ was plated in 96-well microtiter plates and then exposed to varying concentrations (10⁻⁵-10⁻¹² M) of compounds after cultivation for 24 h. Three days later, 10 μ L of CCK-8 solution was added 4 h before detection. Then the absorbency (A450 value) was measured, and the growth rates of cells were computed.

ITS sequences of Aspergillus sp. OUCMDZ-1583

CGTAGGTGAACCTGCGGAAGGATCATTACTGAGTGCGGGCTGCCTCCGGGCGCCCAACCTCCCACCCG TGAATACCTAACACTGTTGCTTCGGCGGGGAACCCCCTCGGGGGGCGAGCCGCGGGGACTACTGAACT TCATGCCTGAGAGTGATGCAGTCTGAGTCTGAATATAAAATCAGTCAAAACTTTCAACAATGGATCTCT TGGTTCCGGCATCGATGAAGAACGCAGCGAACTGCGATAAGTAATGTGAATTGCAGAATTCAGTGAAT CATCGAGTCTTTGAACGCACATTGCGCCCCCTGGCATTCCGGGGGGGCATGCCTGTCCGAGCGTCATTGC TGCCCATCAAGCCCGGCTTGTGTGTGGGTCGTCGTCCCCCCCGGGGGGACGGGCCCGAAAGGCAGC GGCGGCACCGTGTCCGGGTCCTCGAGCGTATGGGGCTTTGTCACCCGCTCGACTAGGGCCGGGCG CCAGCCGACGTCTCCAACCATTTTCTTCAGGTTGACCTCGGATCAGGTAGGGATACCCGCTGAACTTA AGCATATCAATAAGCGGAG



Neighbor-joining phylogenetic tree of strain OUCMDZ-1583 based on ITS gene sequences (ca. 565 bp). The values at each node represent the bootstrap values from 1000 replicates, and the scale bar represents 0.0005 substitutions per nucleotide. Phylogenetic analyses were conducted in MEGA4.

Theory and Calculation Details. The calculations were performed by using the density functional theory (DFT) as carried out in the Gaussian 03.^{S1} The preliminary conformational distributions search was performed by HyperChem 7.5 software. All ground-state geometries were optimized at the B3LYP/6-31G(d) level. Solvent effects of methanol solution were evaluated at the same DFT level by using the SCRF/PCM method. ^{S2} TDDFT ^{S3} at B3LYP/6-31G(d) was employed to calculate the electronic excitation energies and rotational strengths in methanol.

(S1) Gaussian 03, Revision E.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, J. A. Montgomery, Jr., T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez and J. A. Pople, Gaussian, Inc., Wallingford CT, 2004.

(S2) (a) S. Miertus and J. Tomasi, Chem. Phys., 1982, **65**, 239–245. (b) J. Tomasi and M. Persico, Chem. Rev., 1994, **94**, 2027–2094. (c) R. Cammi and J. Tomasi, J.Comp.Chem., 1995, **16**, 1449–1458.

(S3) (a) M. E. Casida, In Recent Advances in Density Functional Methods, part I; D. P. Chong, World Scientific: Singapore, 1995; pp 155–192. (b) E. K. U. Gross, J. F. Dobson and M. Petersilka, Top. Curr. Chem., 1996, 181, 81–172. (c) E. K. U. Gross and W. Kohn, Adv. Quantum Chem., 1990, 21, 255–291. (d) E. Runge and E. K. U. Gross, Phys. Rev. Lett., 1984, 52, 997–1000.

position	19 ^a		20 ^b		21 ^b		22 ^a		23 ^a	
	$\delta_{ m C}$	$\delta_{\rm H}(J \text{ in Hz})$	$\delta_{ m C}$	$\delta_{\rm H}(J \text{ in Hz})$	$\delta_{ m C}$	$\delta_{\rm H}(J \text{ in Hz})$	$\delta_{ m C}$	$\delta_{ m H}(J ext{ in Hz})$	$\delta_{ m C}$	$\delta_{\rm H}(J \text{ in Hz})$
1	128.1, C		168.0, C		167.8, C		170.2, C		169.6, C	
2	148.9, C									
3	195.1, C		81.7, CH	5.05, br d	81.2, CH	5.03, m	77.5, CH	4.73, m	79.8, CH	4.67, m
4	52.5, CH	3.47, d (4.0)	74.5, CH	4.52, br d	74.4, CH	4.52, d (3.0)	42.8, CH ₂	2.90, dd (16.0, 4.0); 2.83, dd (16.0, 12.0)	65.7, CH	4.46, br d (2.5)
4a			127.0, C		131.2, C		131.7, C		136.1, C	
5	55.8, CH	3.74, d (1.6, 4.0)	104.2, CH	6.59, s	104.4, CH	6.57, s	103.0, CH	6.51, s	103.9, CH	6.72, s
6	61.9, CH	4.86, d (1.6)	152.3, C		158.6, C		154.0, C		158.8, C	
7	128.1, CH	6.60, d (16.2)	134.3, C		137.2, C		132.8, C		139.0, C	
8	136.2, CH	6.47, dq (16.2, 6.8)	149.5, C		156.2, C		150.6, C		155.2, C	
8a			101.8, C		102.0, C		102.6, C		102.3, C	
9	19.7, CH	1.87, d (6.8)	19.2, CH ₂	2.57, m; 2.13, m	19.2, CH ₂	2.57, m; 2.13, m	33.0, CH ₂	1.81, ddd (14.7, 10.4, 2.8); 1.56, ddd (14.7, 10.8, 3.6)	38.6, CH ₂	1.94, m; 1.58,m
10	53.3, CH ₂	4.07, d (11.3); 4.27, d (11.3)	78.7, CH	4.09, m	78.7, CH	4.10, m	65.7, CH	3.70, m	66.1, CH	3.70, m
11			39.1, CH ₂	1.67, m; 1.54, m	39.8, CH ₂	1.67, m; 1.52, m	40.4, CH ₂	1.38, m; 1.35, m	$40.5, CH_2$	1.40, m; 1.38, m
12			38.1, CH ₂	1.39, m; 1.34, m	38.0, CH ₂	1.40, m; 1.31, m	18.9, CH ₂	1.33, m; 1.30, m	18.9, CH ₂	1.32, m; 1.37m
13			14.1, CH ₃	0.88, t (7.5)	19.2, CH ₃	0.89, t (7.4)	14.4, CH ₃	0.86, t (7.2)	14.6, CH ₃	0.89, t (7.1)
6-OCH ₃			56.4, CH ₃	3.93, s	56.2, CH ₃	3.87, s	56.5, CH ₃	3.83, s	56.7, CH	3.88, s
7-OCH ₃					60.7, CH ₃	3.93, s			60.5, CH ₃	3.70, s
8-OH				11.11, s		11.3, br s		10.87, br s		11.03, br s

Table S1.	^I H (500 MHz)) and ^{13}C ((125 MHz)	NMR Data	of Compounds	19–23
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^a recorded in DMSO- d_6 . ^b recorded in CDCl₃.







Figure S2. The DEPTQ spectrum of compound 1 in CDCl₃



Figure S3. The HSQC spectrum of compound 1 in CDCl₃



Figure S4. The HMBC spectrum of compound 1 in CDCl₃



Figure S5. The ¹H-¹H COSY spectrum of compound 1 in CDCl₃

Figure S6. The NOE difference spectrum of compound 1 in CDCl₃





Figure S7. The ¹H-NMR spectrum of compound **2** in CDCl₃





Figure S9. The DEPT spectrum of compound 2 in CDCl₃















Figure S13. The NOE difference spectrum of compound 2 in CDCl₃



Figure S14. The ¹H-NMR spectrum of compounds 3 and 4 in CDCl₃

S23



Figure S16. The DEPT spectrum of compounds 3 and 4 in CDCl₃



S25

Figure S17. The HMQC spectrum of compounds 3 and 4 in CDCl₃





Figure S18. The ¹H-¹H COSY spectrum of compounds 3 and 4 in CDCl₃











Figure S21. The ¹H-NMR spectrum of compound 5 in CDCl₃

Figure S22. The ¹³C-NMR spectrum of compound 5 in CDCl₃







Figure S24. The HMQC spectrum of compound 5 in CDCl₃



Figure S25. The ¹H-¹H COSY spectrum of compound **5** in CDCl₃



Figure S26. The HMBC spectrum of compound 5 in CDCl₃



Figure S27. The NOE difference spectrum of compound 5 in CDCl₃ (I)






Figure S29. The ¹H-NMR spectrum of compound 6 in DMSO- d_6







Figure S31. The DEPT spectrum of compound 6 in DMSO- d_6



-10 -20 -20 -30 -40 -60 -60 -60 -100 -110 -1100 -1100 -1100 -1110 -1120 -1100 -1120 0ò-~ ŌН ß 0 8 MIL Ţ gHSQCAD_01 KFD-3-18

Figure S32. The HMQC spectrum of compound 6 in DMSO- d_6

-0.5

0.0

0.5

1.0

1.5

2.0

2.5

3.0

3.5 f2 (ppm)

4.0

4.5

5.0

5.5

6.0

6.5

7.0

7.5

Figure S33. The ¹H-¹H COSY spectrum of compound **6** in DMSO- d_6



S42

Figure S34. The HMBC spectrum of compound 6 in DMSO- d_6









Figure S36. The NOESY difference spectrum of compound 6 in DMSO- d_6 (II)

S45

Figure S37. The ¹H-NMR spectrum of compounds 7 and 8 in CDCl₃



Figure S38. The ¹³C-NMR spectrum of compounds 7 and 8 in CDCl₃



Figure S39. The DEPT spectrum of compounds 7 and 8 in CDCl₃



Figure S40. The HMQC spectrum of compounds 7 and 8 in CDCl₃



Figure S41. The ¹H-¹H COSY spectrum of compounds 7 and 8 in CDCl₃



Figure S42. The HMBC spectrum of compounds 7 and 8 in CDCl₃









Figure S44. The ¹H-NMR spectrum of compound 9 in CDCl₃

Figure S45. The ¹³C-NMR spectrum of compound 9 in CDCl₃





Figure S46. The DEPT spectrum of compound 9 in CDCl₃



Figure S47. The HMQC spectrum of compound 9 in CDCl₃

Figure S48. The ¹H-¹H COSY spectrum of compound 9 in CDCl₃



Figure S49. The HMBC spectrum of compound 9 in CDCl₃





Figure S50. The ¹H-NMR spectrum of compound **10** in DMSO- d_6



Figure S51. The ¹³C-NMR spectrum of compound 10 in DMSO- d_6

Figure S52. The DEPT spectrum of compound 10 in DMSO- d_6



Figure S53. The HMQC spectrum of compound 10 in DMSO- d_6



Figure S54. The ¹H-¹H COSY spectrum of compound **10** in DMSO- d_6



Figure S55. The HMBC spectrum of compound 10 in DMSO- d_6





Figure S56. The ¹H-NMR spectrum of compound **11** in DMSO- d_6



Figure S57. The ¹³C-NMR spectrum of compound 11 in DMSO- d_6

Figure S58. The DEPT spectrum of compound 11 in DMSO- d_6



Figure S59. The HMQC spectrum of compound 11 in DMSO- d_6





Figure S60. The ¹H-¹H COSY spectrum of compound **11** in DMSO- d_6







Figure S62. The ¹H-NMR spectrum of compound **12** in DMSO- d_6



Figure S63. The ¹³C-NMR spectrum of compound **12** in DMSO- d_6
Figure S64. The DEPT spectrum of compound 12 in DMSO- d_6



-110 -140-100-120-130-90 -10 -30 -40 -50 -60 -70 -80 -20 HΟ 0 0 HO 3.0 0 4.0 (ppm) 20130710 gradient enhanced HNQC with X-decoupling 0 -8

Figure S65. The HMQC spectrum of compound 12 in DMSO- d_6

1.0

1.5

2.0

2.5

3.5

f24.5

5.0

5.5

6.0

6.5

7.0



Figure S66. The ¹H-¹H COSY spectrum of compound **12** in DMSO- d_6



Figure S68. The NOESY spectrum of compound 12 in DMSO- d_6





Figure S69. The ¹H-NMR spectrum of compound **13** in DMSO- d_6



Figure S70. The ¹³C-NMR spectrum of compound 13 in DMSO- d_6

Figure S71. The DEPT spectrum of compound 13 in DMSO- d_6



Figure S72. The HMQC spectrum of compound 13 in DMSO- d_6





Figure S73. The ¹H-¹H COSY spectrum of compound **13** in DMSO- d_6

Figure S74. The HMBC spectrum of compound 13 in DMSO- d_6



Figure S75. The NOESY spectrum of compound 13 in DMSO- d_6









Figure S77. The ¹³C-NMR spectrum of compound 14 in DMSO- d_6

Figure S78. The DEPT spectrum of compound 14 in DMSO- d_6



Figure S79. The HMQC spectrum of compound 14 in DMSO- d_6



Figure S80. The ¹H-¹H COSY spectrum of compound **14** in DMSO- d_6





Figure S81. The HMBC spectrum of compound 14 in DMSO- d_6



Figure S82. The ¹H-NMR spectrum of compound **15** in DMSO- d_6



Figure S83. The ¹³C-NMR spectrum of compound **15** in DMSO- d_6



Figure S84. The DEPT spectrum of compound 15 in DMSO- d_6

Figure S85. The HMQC spectrum of compound 15 in DMSO- d_6





Figure S86. The ¹H-¹H COSY spectrum of compound **15** in DMSO- d_6



Figure S87. The HMBC spectrum of compound 15 in DMSO- d_6



Figure S88. The ¹H-NMR spectrum of compound **16** in DMSO- d_6



Figure S89. The ¹³C-NMR spectrum of compound **16** in DMSO- d_6



Figure S90. The DEPT spectrum of compound 16 in DMSO- d_6

Figure S91. The HMQC spectrum of compound 16 in DMSO-*d*₆



Figure S92. The ¹H-¹H COSY spectrum of compound **16** in DMSO- d_6



Figure S93. The HMBC spectrum of compound 16 in DMSO- d_6





Figure S94. The ¹H-NMR spectrum of compound **17** in DMSO- d_6



Figure S95. The ¹³C-NMR spectrum of compound 17 in DMSO- d_6





Figure S97. The HMQC spectrum of compound 17 in DMSO- d_6












Figure S100. The ¹H-NMR spectrum of compound 18 in CDCl₃



Figure S101. The ¹³C-NMR spectrum of compound 18 in CDCl₃

Figure S102. The DEPT spectrum of compound 18 in CDCl₃













Figure S105. The HMBC spectrum of compound 18 in CDCl₃



Figure S106. The NOESY spectrum of compound 18 in CDCl₃

Figure S107. X-ray data and structure of compound 20

7-O-Demethylmonocerin (20): Colorless Monoclinic crystals from MeOH-H₂O (1:1, v/v) with molecular formula of C₁₅H₁₈O₆; space group *P*2(1) with *a* = 7.7687(7) Å, *b* = 7.4049(5) Å, *c* = 12.6454(9) Å, *V* = 703.08(9) Å³, *Z* = 2, $D_{calcd} = 1.390 \text{ Mg/m}^3$, $\mu = 0.906 \text{ mm}^{-1}$, and *F*(000) = 312; crystal size: $0.32 \times 0.27 \times 0.07 \text{ mm}^3$. *T* = 293(2) K. Absolute structure parameter: 0.0(3). These data were obtained on a Bruker APEX DUO area detector diffractometer with graphite monochromatic Cu-K\alpha radiation ($\lambda = 1.54178$ Å) and have been deposited in the Cambridge Crystallographic Data Centre with supplementary publication No 995362. The structure was solved by direct methods (SHELXS-97) and expanded using Fourier techniques (SHELXL-97). The final cycle of full-matrix least-squares refinement was based on 1549 unique reflections ($2\theta < 50^{\circ}$) and 193 variable parameters and converged with unweighted and weighted agreement factors of R1 = 0.0409, wR2 = 0.1015, and *R* = 0.0499 for I > $2\sigma(I)$ data. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via <u>www.ccdc.cam.ac.uk/data_request/cif</u>.



ORTEP drawing of 20



Figure S108. The ¹H-NMR spectrum of compound 12a in CDCl₃



Figure S109. The DEPTQ spectrum of compound 12a in CDCl₃

Figure S110. The HMQC spectrum of compound 12a in CDCl₃



Figure S111. The ¹H-¹H COSY spectrum of compound 12a in CDCl₃



Figure S112. The HMBC spectrum of compound 12a in CDCl₃

















0.

Figure S116. The ¹H-NMR spectrums of compounds 6a and 6b in CDCl₃

0

Figure S117. The ¹H-NMR spectrums of compounds 7a and 7b in CDCl₃











490

460 470 480

500 510 520

530 540 560

Figure S120. The HPLC analysis of compound 16 and synthetic 15 (ODS, 10% MeOH/H₂O, v/v).



Figure S121. The HPLC analysis of 14 and the synthetic 14 (ODS, 20% MeOH/H₂O, v/v).



Figure S122. The HPLC analysis of 15 and the synthetic 15 (ODS, 15% MeOH/H₂O, v/v).

